



Originally published as:

Jacob, D. E., Wirth, R., Soldati, A. L., Wehrmeister, U., Schreiber, A. (2011): Amorphous calcium carbonate in the shells of adult Unionoida. - *Journal of Structural Biology*, 173, 2, 241-249

DOI: [10.1016/j.jsb.2010.09.011](https://doi.org/10.1016/j.jsb.2010.09.011)

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Amorphous calcium carbonate in the shells of adult *Unionoida*

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29 **Abstract**

30 Shells of adult individuals from two different bivalve families, *Hyriopsis cumingii* and *Diplodon*
31 *chilensis patagonicus*, were studied by Micro Raman spectroscopy and Focussed Ion Beam-
32 assisted TEM. The shells contain amorphous calcium carbonate in a zone at the interface
33 between the periostracum and the prismatic layer. In this area, the initial prism structures
34 protrude from the inner periostracum layer and it is demonstrated that these structures
35 systematically consist of highly disordered and amorphous calcium carbonate. Within this
36 zone, ordered and disordered areas are intermingled discounting the existence of a
37 crystallization front and favouring models of domainal crystallization processes via so-called
38 mesocrystals. These observations are the first documentation of the use of amorphous
39 calcium carbonate as a precursor phase by adult mollusc species and lend further support to
40 hypotheses postulating widespread use of amorphous phases as building material of skeletal
41 tissue in biology.

42

43 **Keywords:** Amorphous calcium carbonate, *Hyriopsis cumingii*, *Diplodon chilensis*
44 *patagonicus*, Raman spectroscopy, FIB-TEM, vaterite

45

46 **1. Introduction**

47 *1.1 The nanostructure of bivalve shells*

48 The mechanisms of bivalve shell growth have fascinated and inspired generations of
49 researchers. Intensive research across disciplines unveiled a complex picture of a highly
50 organised structure involving both organic and inorganic components. However, the majority
51 of the exact metabolic pathways as well as the molecular mechanisms which lead to the
52 development of this bio-engineered material are still enigmatic.

53 Major advances are tightly coupled to the development of high resolution instrumental
54 analytical methods like, for example, Transmission Electron or Atomic Force Microscopy that
55 allow observation at the nanometre scale. Studies using such nano-scale methods showed
56 that nacre platelets and prisms are not single crystals (Mutvei, 1977), but consist of highly
57 aligned crystals of aragonite or calcite at the nanometre scale (Metzler et al., 2007) which
58 have been termed mesocrystals (Niederberger and Cölfen, 2006). The basic building units of
59 these and all higher structures in bivalve shells are minute CaCO₃ vesicles in the order of ca.
60 20-50 nanometres (e.g. Addadi et al, 2006, Dauphin, 2008, Jacob et al., 2008). Element
61 mapping using electron energy-loss spectrometry by TEM provided direct evidence for an
62 organic membrane coating each individual CaCO₃ vesicle (Jacob et al., 2008). These organic
63 membranes permanently coat the CaCO₃ and are incorporated in the crystal lattice of
64 aragonite and calcite upon assemblage of the much larger tablet and prism structures. This
65 leads to characteristic distortions of the aragonite or calcite crystal lattice that can be traced
66 by X-ray diffraction (Pokroy et al., 2006).

67 Descriptions of such CaCO₃-bearing vesicles in cells close to the mineralization site can be
68 found throughout the literature (e.g. Cartwright and Checa, 2007; Neff, 1972; Watabe et al.,
69 1976) and imply active cellular transport of CaCO₃ to the site of mineralization, for example
70 by haemocytes (Mount et al., 2004), rather than precipitation from oversaturated solutions.

71

72 *1.2 Amorphous Calcium Carbonate in mollusc shells*

73 Jacob et al. (2008) showed for freshwater cultured pearls from *Hyriopsis* sp. that CaCO₃
74 contained in these vesicles is initially an amorphous phase which subsequently crystallizes to
75 form aragonite. Amorphous calcium carbonate (ACC) is frequently discussed to be the
76 precursor phase in biology in general (Weiner, 2008) and in mollusc shell formation
77 processes in particular: larval shells of the marine bivalves *Mercenaria mercenaria*
78 (Linnaeus, 1758) and *Crassostrea gigas* (Thunberg, 1793) for example, were found to form
79 entirely from amorphous calcium carbonate that subsequently transformed into the crystalline
80 mineral phase (Weiss et al., 2002). Similarly, larval shells of the freshwater gastropod
81 *Biomphalaria glabrata* (Preston, 1910) initially consist completely of ACC (Hasse et al.,
82 2000). The finding of ACC in larval shells of such diverse species of molluscs led to the

83 hypothesis that the use of amorphous CaCO₃ as the precursor of the crystalline phase may
84 be a basic strategy (Addadi et al., 2003; Weiss et al., 2002). However, up to now findings of
85 ACC as a constituent phase in shells were restricted to the larval stage, while there is only
86 one report of ACC in an adult mollusc shell (Nassif et al., 2005). These authors found an
87 ACC coating around a nacre tablet in the shell of *Haliotis laevigata* and interpreted this
88 amorphous coating to be stabilized by impurities which were expelled from the amorphous
89 precursor phase upon crystallization into aragonite.

90 ACC in freshwater cultured pearls from *Hyriopsis cumingii* x *Hyriopsis schlegelii* hybrids are
91 found close to the innermost organic layer in the pearls (Jacob et al. 2008). Pearl culturing
92 makes use of the identical natural processes to those that form the mussel shell, and pearls
93 can therefore be used as structural analogues to shells, although showing a distinctly
94 inverted geometry. The zone containing ACC in the pearls is the area in which pearl
95 mineralization starts and corresponds structurally to the interface between the inner
96 periostracum and the prismatic layer in bivalve shells. Consequently, we report here on the
97 systematic occurrence of amorphous calcium carbonate in this interface zone in shells in two
98 different *Unionoid* species and argue that ACC could be the regular building material,
99 supporting the hypothesis that it is possibly widespread in the phylum.

100

101 **2. Materials and Methods**

102 *2.1 Bivalve samples*

103 Valves of two different mussel species, *Hyriopsis cumingii* (Lea, 1852) and *Diplodon*
104 *chilensis patagonicus* (d'Orbigny, 1835) were studied. The bivalves are of the order of
105 *Unionoida* (Stoliczka, 1871) and belong to the *Unionidae* and *Hyriidae* families, respectively.

106 Valves of *Hyriopsis schlegelii* and *Diplodon chilensis patagonicus* are equivalent, covered
107 with a thick periostracum and consist exclusively of aragonite. The genus *Hyriopsis* is native
108 to East Asia. *Hyriopsis cumingii* (Lea, 1852) and *Hyriopsis schlegelii* (Martens, 1861) are the
109 two main pearl-producing species native to Japan and China, respectively (Graf and
110 Cummings, 2007). A non-specified hybrid of these two species is increasingly used for pearl
111 culturing in Japan as this yields better harvests (Fiske and Shepherd, 2007). Valves can be
112 as large as 25 cm and individuals can live as long as 40 years (Strack, 2006); the sample of
113 *Hyriopsis cumingii* used here was collected alive from a pearl farm in Wuhan, China in 2007
114 and was six years old at the time of collection.

115 Bivalves of the genus *Diplodon* are geographically widely distributed and occur in most areas
116 of the southern hemisphere. The subspecies *Diplodon chilensis patagonicus* is native to
117 Patagonia and populations are reported in Argentina between Mendoza (32° 52' S; 68° 51'
118 W) and Chubut (45° 51' S; 67° 28' W; Castellanos, 1959). Samples used in this study were
119 collected alive in 2007 from the lakes of the Nahuel Huapi National Park in Patagonia,

120 Argentina (41°S; 71°W). Individuals are long-lived (90 years and more; Soldati et al., 2009)
121 and valves reach sizes of ca. 12 cm. The age of the shell used in this study was determined
122 as at least 26 years by counting the internal growth rings after cutting the shell dorsal-ventral
123 along the axis of maximum growth and staining with *Alcian Blue* to highlight the annual
124 growth lines (for a detailed description see Soldati et al., 2009).

125 The shells were cut and mounted in epoxy resin (Struers, Willich, Germany), then polished
126 using 800 and 1200 Al₂O₃ powder followed by a last polishing step with 1µm Al₂O₃ powder
127 on a Buehler G-cloth. Light microscopy investigations were carried out with a VHX-600 digital
128 microscope (KEYENCE, Neu-Isenburg; Germany), equipped with a VH-Z25 zoom lens
129 (magnification from 25x to 175x) or a VH-Z50 long-distance, high-performance zoom lens
130 with a magnification of up to 5000x.

131

132 *2.2 Raman spectroscopy and XRD*

133 Raman spectra were recorded at room temperature using a Horiba Jobin Yvon LabRAM HR
134 spectrometer equipped with a Si-based CCD-detector (Peltier-cooled) and an integrated
135 Olympus BX41 optical microscope (100x magnification) with automated stage. The 532.2 nm
136 line of a Nd-YAG laser line was used for excitation with a laser power of 90 mW. The
137 Rayleigh radiation of the laser was blocked using edge filters and the scattered light was
138 dispersed by a grating with 1800 grooves/mm. Data acquisition and spectra treatment was
139 carried out with the commercially available program LabSpec v4.02 (Horiba Jobin Yvon).
140 Due to the sensitive nature of amorphous calcium carbonate, Raman spectra were recorded
141 once and with short acquisition times (2s per window). Laser spot size was 2µm. Peak
142 analysis was performed with Origin-lab® 7.5 Pro by fitting the peaks as either single or
143 overlapping Lorentzian and Gaussian curves.

144

145 *2.3 Focused Ion Beam-assisted Transmission Electron Microscopy*

146 For Transmission Electron Microscope analyses (TEM) at the GeoForschungsZentrum
147 Potsdam, ultra-thin foils of ca. 10 by 15 µm and 0.150 – 0.200 µm thicknesses were prepared
148 using the focussed ion beam device (FIB) of a FEI FIB200 instrument following procedures
149 given in Wirth (2004). After milling, the foil was cut free, lifted out, and placed on a carbon
150 coated Cu grid. No further carbon coating was necessary. The FIB-milling method is carried
151 out by sputtering the material surrounding the target area with gallium ions. The target area
152 itself is protected by a platinum coating. Heating of the sample during sputtering has been
153 shown to be less than 10 K, due to the very low angle between Ga beam and sample
154 (Ishitani and Yaguchi, 1996). However, it is well known that FIB-milling generates a thin
155 amorphous layer of 10- 20 nm thickness on both sides of the foil and that its thickness shows
156 a strict linear relationship to the accelerating voltage of the Ga beam (e.g. Kato 2004). The

157 thickness of this layer is much less than the total thickness of the foil (150-200 nm) and
158 causes nothing more than a slight blurring effect in high resolution TEM images, if not
159 removed. Amorphisation introduced by FIB milling across the complete thickness (or
160 considerable parts) of the foil has never been observed, an observation which is supported
161 based on the experience with producing and examining more than 2000 FIB foils in this
162 laboratory. We consider it therefore highly unlikely that the amorphous phases studied here
163 are introduced by the FIB-milling procedure.

164 TEM imaging and analysis were undertaken with a FEI Tecnai™G2 F20 X-Twin transmission
165 electron microscope with a field emission gun electron source. The TEM was operated at
166 200 kV acceleration voltage. A Gatan Tridiem™ filter allowed energy-filtered imaging
167 applying a 20 eV window to the zero loss peak. The Gatan system was also used for electron
168 energy-loss spectroscopy (EELS). Analytical electron microscopy (AEM) was performed with
169 an energy dispersive X-Ray analyzer (EDX). Analyses were usually carried out in scanning
170 transmission mode (STEM) scanning the beam in a pre-selected window thus avoiding mass
171 loss during the spectrum acquisition. Counting time was 60 -120 seconds. Beam size was
172 approximately 1 nm in diameter.

173 Great care was taken to minimize radiation damage to the material during analysis. At the
174 start of the analyses, an overview picture was taken and the sample was examined critically
175 at low magnification under a defocused beam. In this way, crystalline and amorphous areas
176 were recorded and episodically repeated careful visual examination during the analytical
177 session confirmed the unchanged nature of the sample. Furthermore, all analyses requiring a
178 focused electron beam and high resolution TEM (HRTEM) were carried out at the very end of
179 the analytical session. Nevertheless, the last of these HRTEM analyses is suspected to have
180 created radiation damage to the sample which is described in detail in the discussion section.

181

182 **3. Results**

183 *3.1 Light microscopy: Shell structure*

184 All shells studied here consist entirely of aragonite and display a so-called “simple prismatic
185 structure” (Carter, 1980), consisting of a dorsal prismatic layer followed by a nacreous layer
186 towards the inside (Fig. 1a). Prism and nacre layer have different thicknesses in the different
187 specimens with the largest (*Hyriopsis cumingii*) showing the thickest nacre layer (5-8 mm)
188 and a relatively thin prismatic layer of ca. 200 μm , while the much smaller *Diplodon chilensis*
189 *patagonicus* has a ca. 1000 μm thick prismatic layer and a nacre layer of 2-4 mm thickness.
190 Single prisms are typically 40 by 200 μm in *Hyriopsis* and ca. 10 by 300 μm in *Diplodon*
191 *chilensis patagonicus*.

192 Although all shell structures in molluscs are a combination of inorganic and organic material
193 at the nanometre-scale (e.g. Lowenstam and Weiner, 1989), the prism structures contain

194 considerably less chitin than the nacre layer (Nudelman et al., 2007). The periostracum,
195 which has been shown to play a major role in the growth of the prism layer (Checa, 2000;
196 Petit et al., 1980a, b), is a two-layered structure in unionid bivalves, formed by the mantle
197 epithelium and covering the mineral part of the shell towards the outside (Bevelander and
198 Nakahara, 1967; Checa, 2000). Initial CaCO₃ prisms protrude from within the inner
199 periostracum as spherulitic, rounded structures (e.g. Checa, 2000; Ubukata, 1994), before
200 the prismatic morphology develops. Not all “prism seeds”, however, develop into even prisms
201 (Fig. 1b, c). Depending on the growth rate of the shell and geometric factors, such as the
202 angle between individual prism and periostracum and curvature of the shell, competition for
203 space and geometrical selection leads to growth of some prisms at the expense of others
204 (Ubukata, 1994). Examined under reflected light, initial prisms protruding from the inner
205 periostracum in all shells appear darker (i.e. have lower reflectance) than those further
206 towards the ventral side of the shell (Fig. 1c).

207 A section of the shell in a dorsal-ventral direction along the axis of maximum growth reveals
208 that periostracum and prism layers are duplicated in some areas of the shell (Fig. 1a,
209 arrows). Overlapping prism layers are divided by growth lines that extend from the prismatic
210 into the nacreous layer and can be accompanied by extension of the periostracum into the
211 nacre layer (Checa, 2000). These dark organic layers have been described as “annual
212 bands” or “pseudoannuli” (Coker et al., 1921) or as “intra-shell periostracum” (Checa, 2000)
213 and are caused by prolonged interruption and subsequent resumption of shell growth. In
214 adult *Diplodon chilensis patagonicus* as well as in many other bivalve species (e.g.
215 *Margaritifera margaritifera*, *Arctica islandica*, etc.), major growth lines, with or without intra-
216 shell periostracum, have been found to occur regularly once a year and are caused by
217 reduction of growth during the reproduction period of the individuals (e.g. Schöne et al.,
218 2004; Soldati et al., 2009). Additional finer growth lines (Fig. 1c, arrow) delineate subannual
219 shell growth increments (e.g. Barker, 1964; Clark II, 1974) and cause the characteristic
220 horizontal “striping” of the prisms frequently observed during SEM imaging (e.g. Checa,
221 2000; Ubukata, 1994). The fact that these growth lines traverse prisms and nacre platelets
222 alike attests to the simultaneous formation of prism and nacre layer during shell growth.

223

224 3.2 Micro-Raman Spectroscopy

225 Figure 2 compares the typical Raman spectrum of crystalline biogenic aragonite, represented
226 by the nacre layer of *Diplodon chilensis patagonicus* (spectrum d), with spectra from the
227 initial prisms at the interface with the periostracum of *Hyriopsis cumingii* (a) and *Diplodon*
228 *chilensis patagonicus* (b). Plotted in Fig. 2a and b for further comparison is a spectrum of
229 ACC from the sternal CaCO₃ deposits of a woodlouse (*Porcellio scaber*, crustacea)
230 measured at identical conditions (spectrum c). Woodlice have been shown to form cuticular

231 deposits of stable amorphous calcium carbonate during moulting (Ziegler 1994, 1997). The
232 amorphous character of this phase has been confirmed and is well documented in several
233 studies (e.g. Becker et al., 2003, Hild et al., 2008, Tao et al., 2009).

234 Compared to crystalline aragonite from the shell nacre layer, the Raman spectra of the initial
235 prisms show a much lower intensity for all aragonite Raman bands. Furthermore, the main
236 Raman band of the carbonate ion (ν_1 at 1085cm^{-1} , White, 1974) is shifted towards lower
237 wavenumbers and is broader (i.e. shows a higher value for the full width at half maximum;
238 FWHM) than for well crystallized aragonite (Fig. 2b). Lastly, the bands in the $100\text{-}300\text{cm}^{-1}$
239 region of the Raman spectra of crystalline aragonite, which are caused by vibrations of the
240 aragonite crystal lattice (White, 1974), are not detectable in the three prism spectra. These
241 features are typical for highly disordered (i.e. amorphous) materials and agree closely with
242 the features of the ACC spectrum of *Porcellio scaber* (Fig. 2a, spectrum c).

243

244 3.3 TEM analyses

245 To study the ACC areas at higher resolution, a section of a *Hyriopsis cumingii* shell was
246 chosen exemplarily for FIB-assisted TEM analysis. Figure 1d shows the pit in the shell
247 section left by Focused Ion Beam (FIB) milling. The FIB-foil (Fig. 3a) straddles the
248 periostracum-prismatic layer interface and was cut free from the positions marked with "X".
249 These marks are used during FIB-milling for re-alignment following instrumental drift. TEM
250 analyses concentrated on an area in one of the initial prisms adjacent to the periostracum,
251 marked with a white square in Fig. 3a. The calcium carbonate in the prism shows the typical
252 vesicular nanostructure (Fig. 3b-f) common to many mollusc species (Addadi et al., 2006;
253 Dauphin, 2008, Jacob et al., 2008) as well as to sea urchins (Beniash et al., 1997). Several
254 authors have shown that the vesicles in bivalves are contained in individual organic sheaths
255 (Dauphin et al., 2008, Jacob et al., 2008) and are held together by an organic mesh (Jacob
256 et al., 2008), some of whose fibres can be seen in Fig. 3c (arrows).

257 TEM dark field imaging reveals that the vesicles represent a mixture of amorphous areas
258 without diffraction contrast (Fig. 3d, arrows) and crystallized domains in which crystallites
259 show up as white spots as small as 5-10 nm in Fig. 3d. It is important to note that the
260 material showing these differences in crystallinity consists mainly of calcium carbonate as
261 shown by EELS mapping, with high calcium (Fig. 3e) and low carbon concentrations (Fig. 3f)
262 and that organic material is only present in minor amounts.

263 One vesicle (Fig. 3c, d, black frame) with very few crystallites was used for HRTEM analysis.
264 The vesicle was analysed along a profile from rim to core in three areas of ca. 35 nm in size
265 (Fig. 4a, white line). The HRTEM image of the vesicle rim (Fig. 4b) shows that this area is
266 almost entirely free of crystalline areas, marked by a nearly complete lack of lattice fringes.
267 Only a very small area of ca. 5 nm was found to show lattice fringes (circled area). These

268 findings, i.e. dominant amorphous areas with minor crystallized domains, apply also to other
269 areas towards the centre of the vesicle (Fig. 4c, d). The area in the centre of the vesicle
270 which was analysed last, however, shows the highest amount of crystalline nanodomains.
271 These are arranged in an intimate mixture of crystalline and amorphous areas in which the
272 crystalline domains appear to be randomly oriented as shown by their lattice fringes (Fig. 4d).
273 Lattice-spacing (d-spacing) values for all three areas, calculated from the respective FFT
274 (Fast Fourier Transform) analyses (Figs. 4f, g) are tabulated in Table 1. Most values are
275 consistent with literature values for aragonite as well as with those for vaterite. However, one
276 characteristic value of 3.16 Å for the d-spacing in the centre of the vesicle is clearly indicative
277 of calcite.

278

279 **4. Discussion**

280 Micro-Raman spectroscopy identified ACC at the periostracum-prismatic layer interface in
281 the shells of both studied species. This structural context is in accordance with what was to
282 be expected from our studies on pearls, where ACC was identified at the interface between
283 the innermost organic lamella and the prismatic layer (Jacob et al., 2008). These structures
284 are the pearl analogues to periostracum and prismatic layer in the shells, and support the
285 view that pearls can be studied as pathological, but nevertheless structural analogues to
286 shells.

287 A closer look with TEM shows that the crystallinity of CaCO₃ in this interface area is variable:
288 amorphous areas are intermingled with nano- and micro-domains of crystalline material (Fig.
289 3d). The vesicle chosen for HRTEM profiling (Fig. 4a) consists mainly of ACC with
290 subordinate crystalline nano-domains. This strongly suggests that ACC is the original
291 carbonate phase and precursor to the crystallized polymorph. It should be noted, however,
292 that compared to the dark field TEM overview image (Fig. 3d), the high resolution images
293 show more crystalline areas in the centre of the vesicle (Fig. 4d). As the high resolution
294 analysis at this spot was carried out last, we think it likely that crystallization here was
295 induced by the electron beam and that the centre of the vesicle originally contained more
296 disordered CaCO₃, similar to the rim. Calcite detected in the centre of the vesicle (Table 1) is
297 taken as further evidence for radiation damage, as the occurrence of this phase is restricted
298 to the centre spot which was longest exposed to the electron beam and otherwise has never
299 been reported in *Hyriopsis cumingii* shells.

300

301 A number of studies reported vaterite in biominerals (e.g. Hasse et al., 2000, Soldati et al.,
302 2008, Lowenstam and Weiner 1989), and some have argued that this mineral is a regular
303 intermediate phase in the biomineralization pathway to aragonite or calcite (e.g. Gago-Duport
304 et al., 2008; Hasse et al., 2000). However, in contrast to this study, most of these

305 occurrences are apparent repair structures which do not represent regular shell growth
306 pathways. Therefore, growth structures here were carefully analysed for the possible
307 occurrence of vaterite.

308 At the resolution of Micro-Raman spectroscopy of ca. 2 μ m, vaterite was never encountered
309 along the ca. 20 μ m wide ACC-bearing zone at the periostracum interface of either of the two
310 bivalve species studied. This precludes a major role of vaterite during aragonite
311 crystallization from ACC. At nanometre resolution, the occurrence of vaterite can neither be
312 confirmed nor ruled out from the HRTEM analyses. All d-spacing values (except for one
313 indicative of calcite) can be assigned to lattice planes of aragonite as well as to vaterite, if
314 measurement uncertainties for FFT analyses of ca. 0.01 nm are taken into account (Table 1).
315 It should be noted that generally only few reflections differentiate vaterite clearly from
316 aragonite and not all reflections are observed in every measurement. In our case, it is mainly
317 the vaterite (100) = 0.357 nm reflection, as opposed to the aragonite (111) = 0.397 nm
318 reflection (Table 1). An observed d-spacing value of 0.346 nm for the centre of the vesicle
319 (Fig. 4g and Table 1) could possibly indicate vaterite in a nano-domain. However, this
320 evidence is weak, especially since this area is believed to have experienced some radiation
321 damage.

322 Taking all evidence together, it appears that ACC is the precursor phase of aragonite in the
323 shells and that biomineralization proceeds directly without evidence of vaterite as an
324 intermediate product.

325
326 The TEM analyses (Fig. 3d) show that no coherent mineralisation front is developed, a
327 finding that compares favourably with studies on sea urchin larval spicules. Using X-ray
328 photoelectron emission spectromicroscopy, Politi et al. (2008) showed that amorphous
329 calcium carbonate and the corresponding crystalline phase, in this case calcite, were
330 juxtaposed on a scale of tens of nanometres. These authors speculated that crystallization
331 emanated from nano-domains, 40-100 nm in size, which stimulated further crystallization in
332 adjacent domains. This observation is supported for the nanostructure of nacre platelets in
333 *Haliotis refuscens* (Li and Huang, 2009) that show nanocrystals in similar sizes connected by
334 screw dislocations and unidentified amorphous material resulting in a pseudo-single crystal
335 or mesocrystal (Cölfen and Antonietti, 2005). The alignment mechanisms of the
336 nanocrystals, however, are not fully understood and possibilities comprise epitaxial
337 crystallization via mineral bridges (Schäffer et al., 1997), Ostwald ripening processes (e.g.
338 Gago-Duport et al., 2008) to alignment of freely crystallized nano-particles via short range
339 physical forces, such as interface energies, capillary forces, etc. Although it is currently
340 difficult to substantiate whether mineral bridges are the cause or the effect of mesocrystal
341 formation, they are omnipresent at all scales, i.e. between nano-particles, between

342 membrane coated vesicles (Jacob et al., 2008) and between individual tablets in mollusc
343 nacre (e.g. Li and Huang, 2009).

344

345 Sizes of the nano-domains are well below the 2 μm resolution of the Micro-Raman
346 spectrometer. Raman spectra of ACC in this study, and probably in many other cases of
347 biogenic ACC, are therefore mixtures of amorphous and crystalline fractions in variable
348 amounts. The observed broadening of the ν_1 band at ca. 1080 cm^{-1} is therefore not only
349 caused by the disordered state of the material, but also results from addition of the ν_1 band of
350 the crystalline phase situated at slightly higher wavenumbers, resulting in an asymmetric
351 shape of the ν_1 band (Wehrmeister et al., in press).

352

353 The amino acid composition of the molluscan periostracum contains considerable amounts
354 of aspartic and glutamic acid (Meenakshi et al., 1969) which are thought to be the main
355 active bio-molecules in the stabilization of ACC (Aizenberg et al., 1996). Furthermore, Checa
356 (2000) demonstrated that the inner periostracum in bivalves is produced by the same outer
357 mantle epithelial cells that simultaneously secrete the prismatic and nacreous shell layers. It
358 is therefore proposed here that ACC, as the originally building material of the shell, is
359 preserved at the edge of the inner periostracum due to the high organic to mineral ratio in
360 this zone, whereas it is rapidly transformed into the crystalline phase further away. In this
361 light, it would be interesting to investigate possible chemical similarities of the inner
362 periostracum, interprismatic layers and nacreous interlamellar sheets further to clarify its role
363 in the biomineralization of mollusc shells.

364

365 **6. Conclusions**

366 Up to now, the occurrence of an amorphous phase in the shells of adult mollusc species was
367 often suspected, but with little success in providing direct proof for its existence. While ACC
368 has been observed in larval mollusc (Hasse et al., 2000; Weiss et al., 2002) and juvenile
369 bivalve shells (Baronnet et al., 2008), only one study exists that reports the occurrence of
370 amorphous calcium carbonate in the shell of an adult individual (Nassif et al., 2005). These
371 authors discovered continuous amorphous coatings of ca. 3 to 5 nm size around each nacre
372 platelet within shells of the gastropod *Haliotis leaevigata*. The structural context of these
373 coatings, however, remained unclear and it was speculated that they resulted from a process
374 similar to zone-refining that expelled impurities from the aragonite crystals which, in turn,
375 kinetically stabilized the amorphous phase.

376 Here we demonstrate the occurrence of ACC in shells of two adult bivalves belonging to
377 different families, while evidence for the involvement of vaterite is lacking. We show that the
378 amorphous phase is not randomly distributed, but is systematically found in a narrow zone at

379 the interface between periostracum and prism layer. This zone is the area where spherulitic
380 CaCO₃-structures protrude from the inner periostracum to form the initial prisms (Checa,
381 2000). These observations are in accordance with our earlier results on equivalent structures
382 in freshwater cultured pearls (Jacob et al., 2008) and show that the original building material
383 for the prisms is amorphous calcium carbonate, secreted in vesicles at the inner
384 periostracum layer. As the nacre layer of many other mollusc species consists of membrane-
385 coated crystalline CaCO₃ vesicle, too (Addadi et al., 2003; Dauphin 2008; Jacob et al., 2008),
386 it is viable to suspect a much broader role for ACC as building material of biological hard
387 tissues.

388 The growing evidence for a common use of ACC as a transient precursor phase of crystalline
389 CaCO₃ by adult individuals from different bivalve families, even by different evolutionary
390 unrelated phyla such as molluscs and echinoderms is highly intriguing. It should be noted
391 that even within the mollusc phylum, the use of ACC as a precursor to crystalline CaCO₃
392 appears to have developed independently during evolution (Jackson et al., 2010).

393 These findings clearly underline the general importance of amorphous phases as plastic
394 building material in hard tissues (e.g. Addadi et al., 2003). However, the exact mechanisms
395 of biomineralization are still largely unknown and it will have to be the aim of studies of both
396 biomimetic and natural systems to shed light on this area.

397

398

399 **5. Acknowledgments**

400 We thank the Johannes Gutenberg University (Mainz) and the Earth System Science
401 Research Centre (Geocycles) for financial support. Crystallographic discussions with R.
402 Meffert and H.-D. Werner are greatly appreciated. Comments from two anonymous
403 reviewers are gratefully acknowledged.

404

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555

556 **Figure Captions**

557 **Fig. 1:** Light microscope images of sectioned shells, showing the typical features of bivalve
558 shells in a *Diplodon chilensis patagonicus* shell (a): The periostracum covers the prism layer
559 which is followed by the nacre layer. Note the prominent growth lines extending into the
560 nacre layers, some of which are covered by intra-shell periostracum (arrow). (b) prisms
561 develop evenly or are terminated depending on geometric factors (*Hyriopsis cumingii* shell).
562 Initial prisms, protruding from the inner periostracum (c, here: *Hyriopsis cumingii*) show
563 prominent fine growth lines and have lower reflectance than prisms towards the ventral side
564 of the shell. These initial prisms at the periostracum interface (*Hyriopsis cumingii*) were
565 sampled by Focussed Ion Beam for TEM analysis (d). This method leaves a characteristic pit
566 in the material while the TEM-foil was extracted from the positions marked with “x” on both
567 sides of the pit. For a detailed description of the method see Wirth (2004).

568

569 **Fig. 2:** (A) Raman spectra of initial prisms in shells of the three bivalve genera (a: *Hyriopsis*
570 *cumingii*, b: *Diplodon chilensis patagonicus*) compared to ACC from *Porcellio scaber* (c) and
571 to crystalline biogenic aragonite from the nacre layer of *Diplodon chilensis patagonicus* (d).
572 The relative intensities are normalized to the highest peak in the spectrum, no baseline
573 correction was applied. The dashed line denotes the position of this band in ACC. 2(B)
574 shows the positions of the ν_1 band in the different spectra, highlighting the shifted position in
575 ACC compared to crystalline aragonite (spectrum d). Note the asymmetric ν_1 band in
576 spectrum c (*Porcellio scaber*) which is indicative of a mixture of crystalline and amorphous
577 CaCO_3 .

578

579 **Fig. 3:** (a) FIB-foil, extracted from the periostracum-prism interface (Fig. 1d). Slight bending
580 of the foil occurred during milling which resulted in small gaps along the sides of the initial
581 prism in the middle of the foil (see also b) and lead to the development of slightly thicker
582 areas (darker grey in the middle of the foil). White frames shows the area enlarged in (b) and
583 (c), respectively. The black frame in the bright field TEM image (c) is enlarged in Fig. 4a.
584 Black arrows in (c) point to organic fibers between the CaCO_3 vesicles. The dark field TEM
585 image of the area (d) shows crystalline (white patches) and amorphous (without diffraction
586 contrast: white arrows) nanodomains. Calcium (e) and carbon (f) EELS maps demonstrate
587 that the depicted area consists predominantly of CaCO_3 .

588

589 **Fig. 4:** A single vesicle (a, enlarged from Fig. 3c) was profiled along the white line by
590 HRTEM. While the rim (b) is nearly completely amorphous (see FFT analysis, e), except for a
591 ca. 5nm area circled in white, more nano-crystalline domains are observed toward the centre
592 of the vesicle (c, d). (e, f, g) are Fast Fourier Transform analyses of (b-d).

593 **Table 1:** Observed d-spacing values calculated from the HRTEM power spectra in Fig. 4 and
 594 comparison to literature data for aragonite (JCPDS-ICDD 41-1475), vaterite (JCPDS-ICDD
 595 33-268) and calcite (JCPDS-ICDD 5-586).

596

Observed			Literature					
Fig. 4e: Rim	Fig. 4f: Transition	Fig. 4g: Centre	Aragonite (6 ₂)Pnma		Vaterite (194)P6 ₃ /mnc		Calcite (167)R-3c	
d _{hkl} [nm]	d _{hkl} [nm]	d _{hkl} [nm]	d _{hkl} [nm]	hkl	d _{hkl} [nm]	hkl	d _{hkl} [nm]	hkl
-	-	0.434	0.4212	110	0.4226	004	-	-
-	0.340	0.346	0.3397	111	0.3573	110	-	-
0.322	0.325	-	0.3274	021	0.3294	112	-	-
-	-	0.316	-	-	-	-	0.3040	104
0.226	0.230	-	0.2330	022	0.2282	205	0.2285	113
0.222	0.222	0.223	0.2190	211	0.2212	116	-	-
0.189	-	-	0.1882	041	0.1854	304	0.1875	116
-	0.181	0.181	0.1815	132	0.1820	118	-	-
-	-	0.169	0.1698	222	0.1646	224	-	-

597

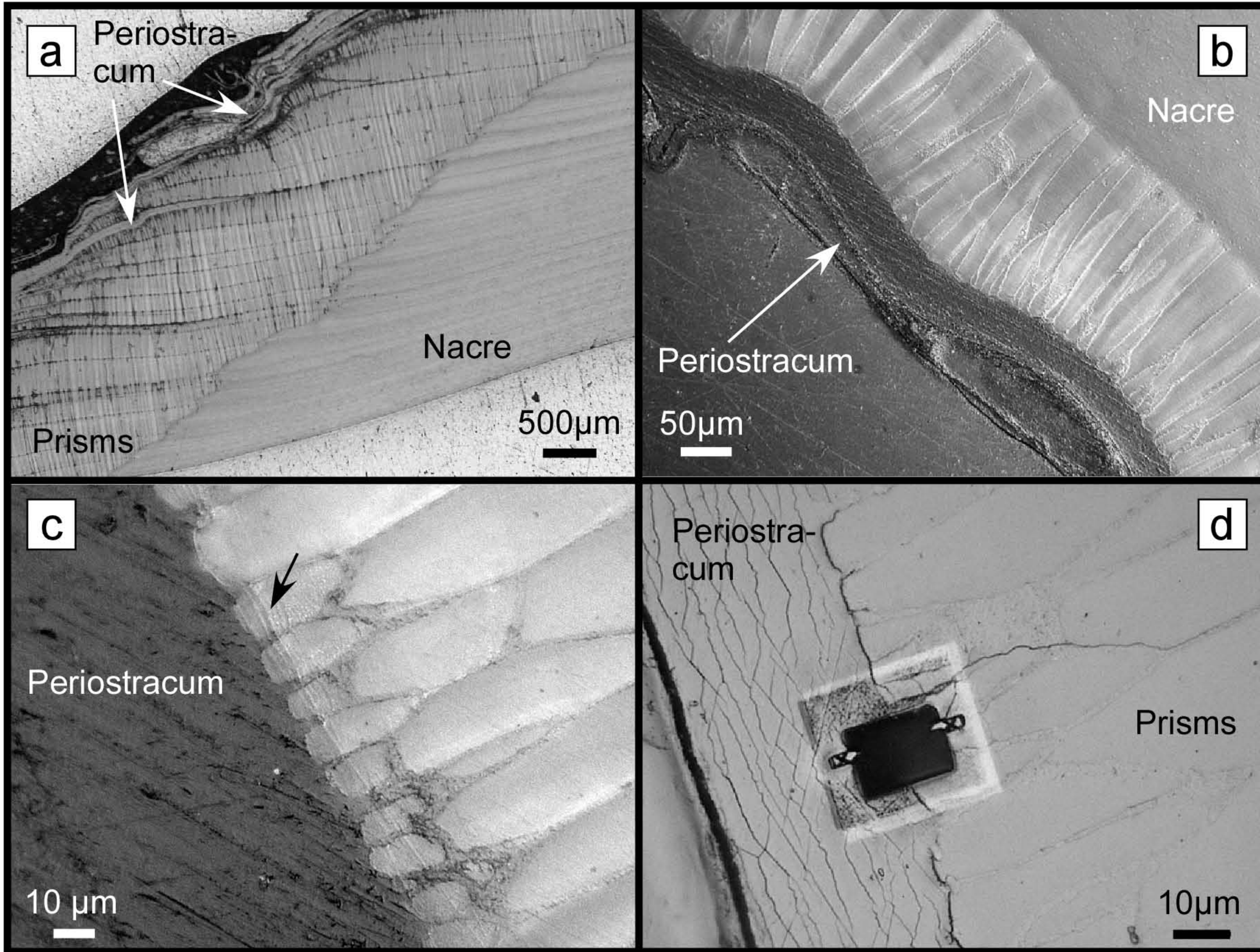


Fig. 1, Jacob, Wirth, Soldati, Wehrmeister, Schreiber

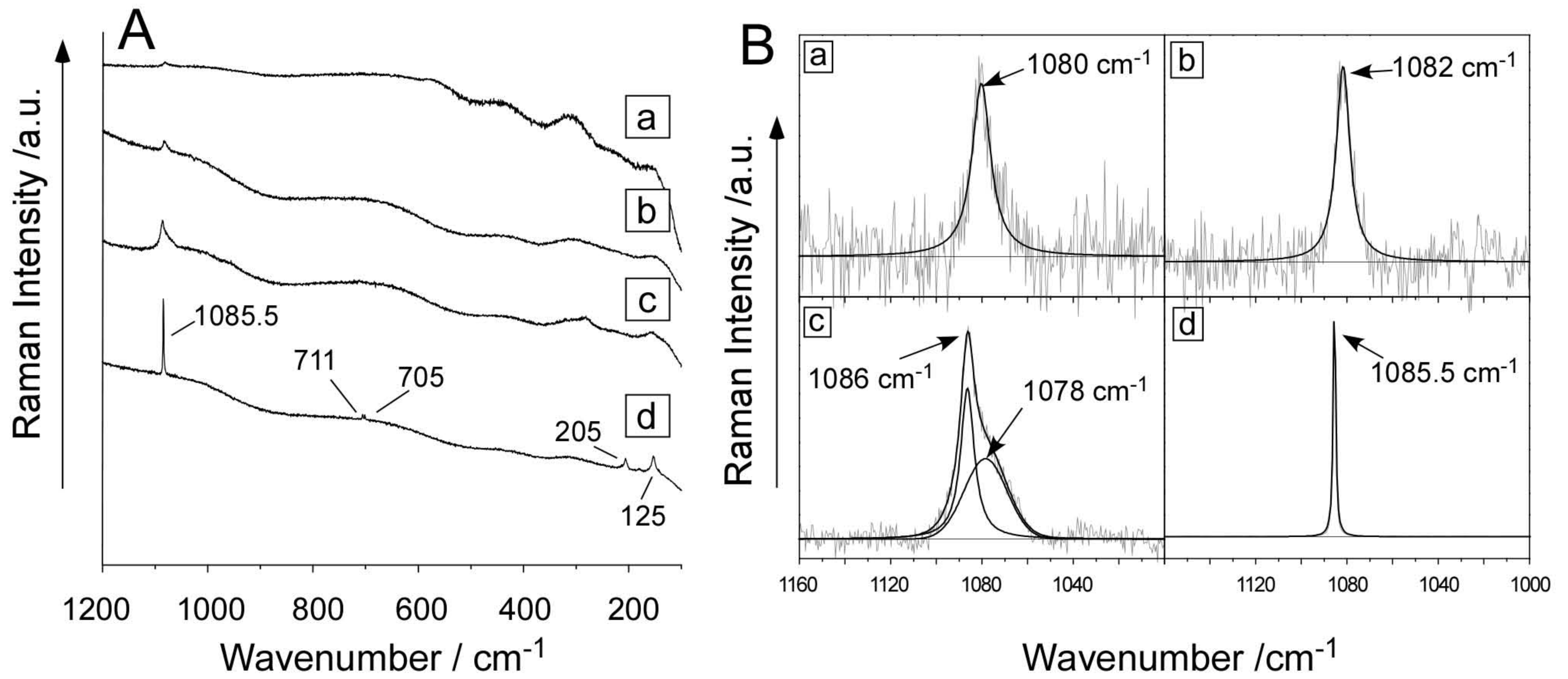


Fig. 2, Jacob, Wirth, Soldati, Wehrmeister, Schreiber

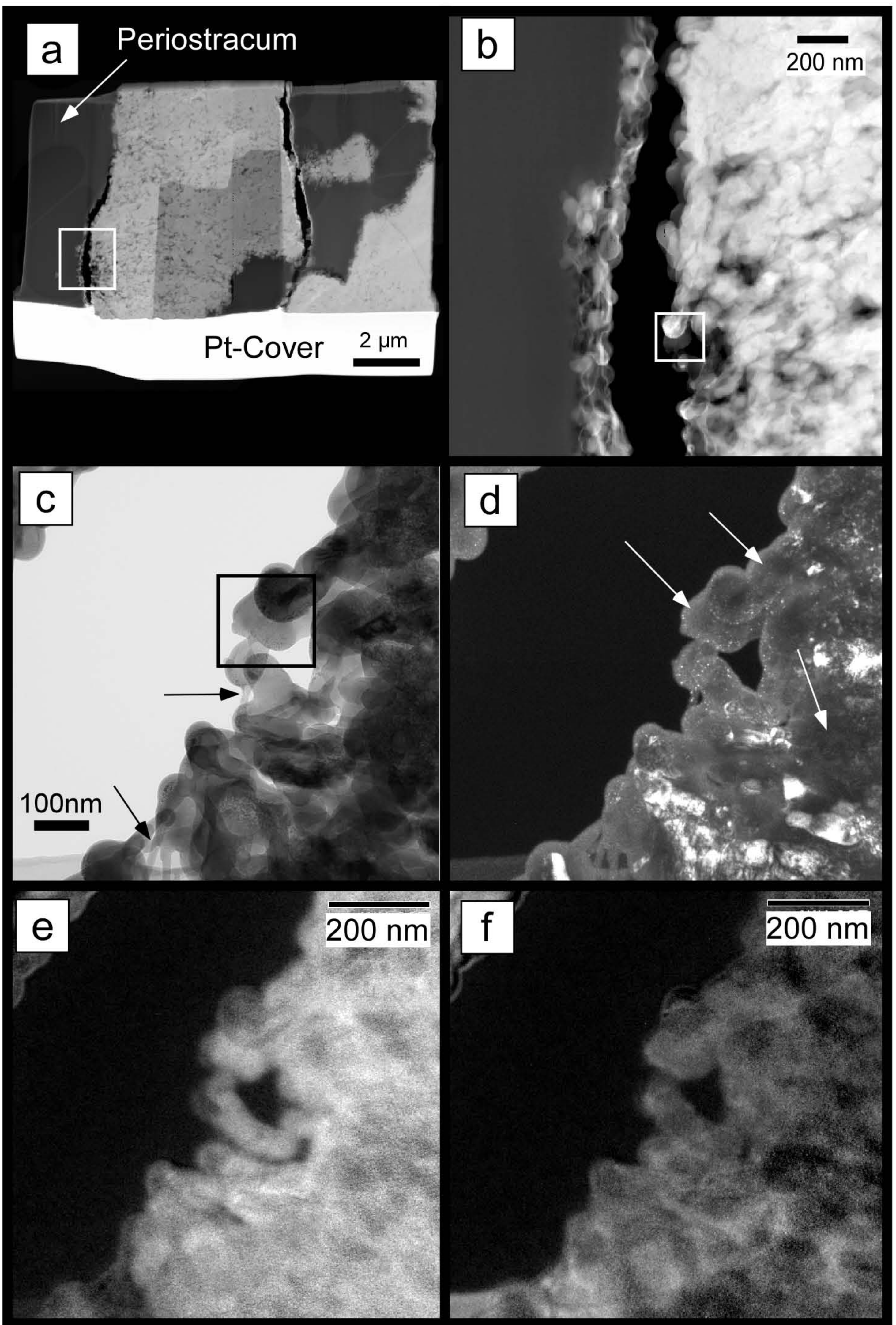


Fig. 3, Jacob, Wirth, Soldati, Wehrmeister, Schreiber

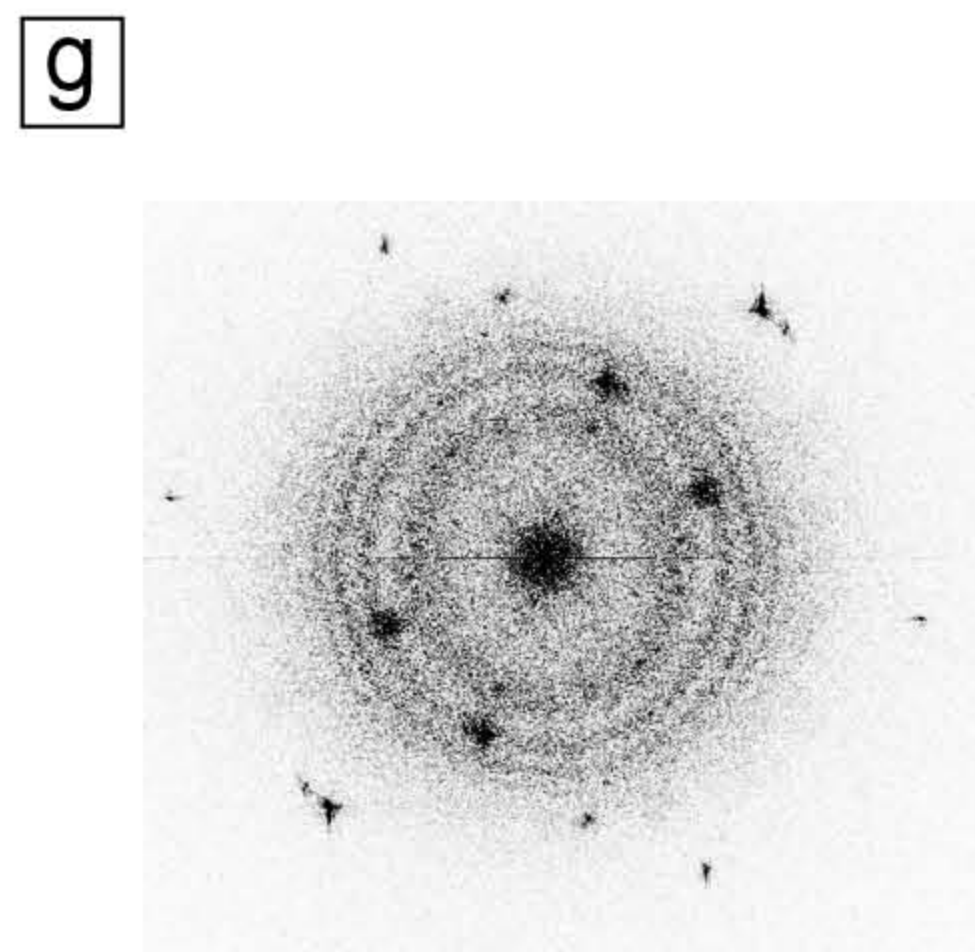
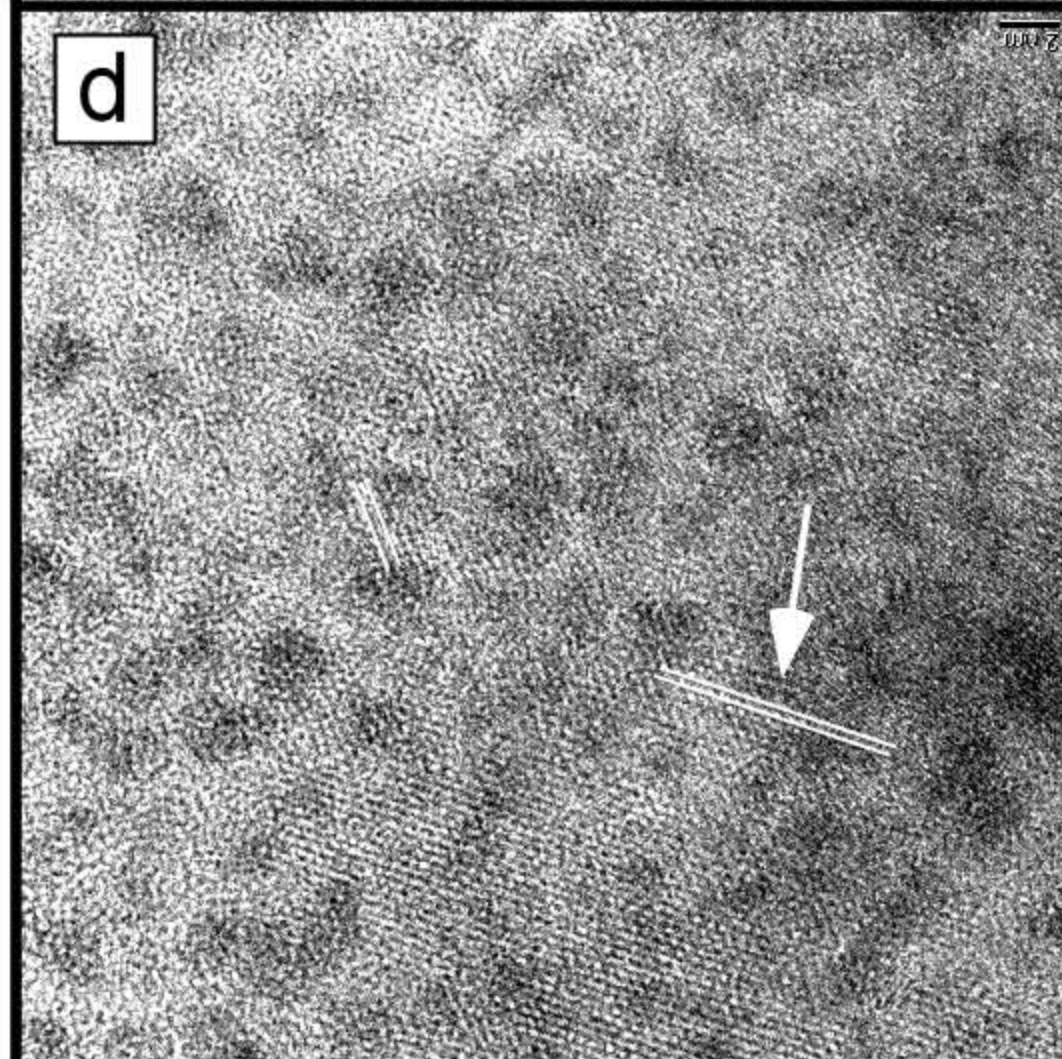
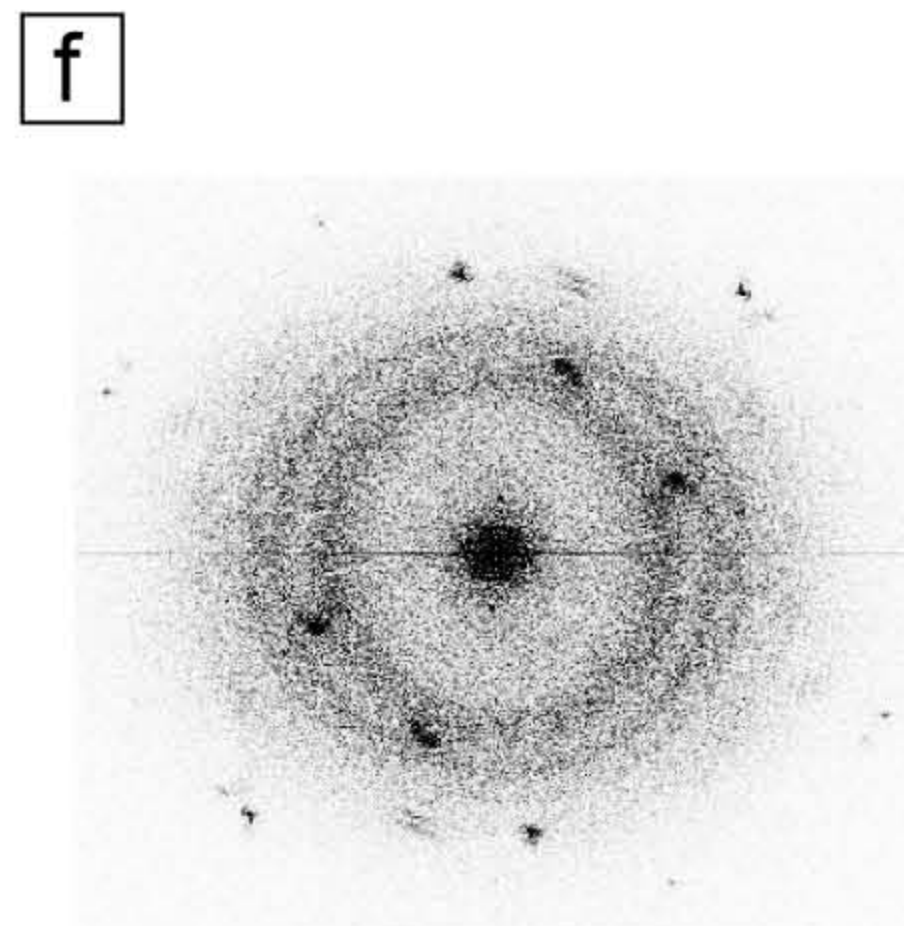
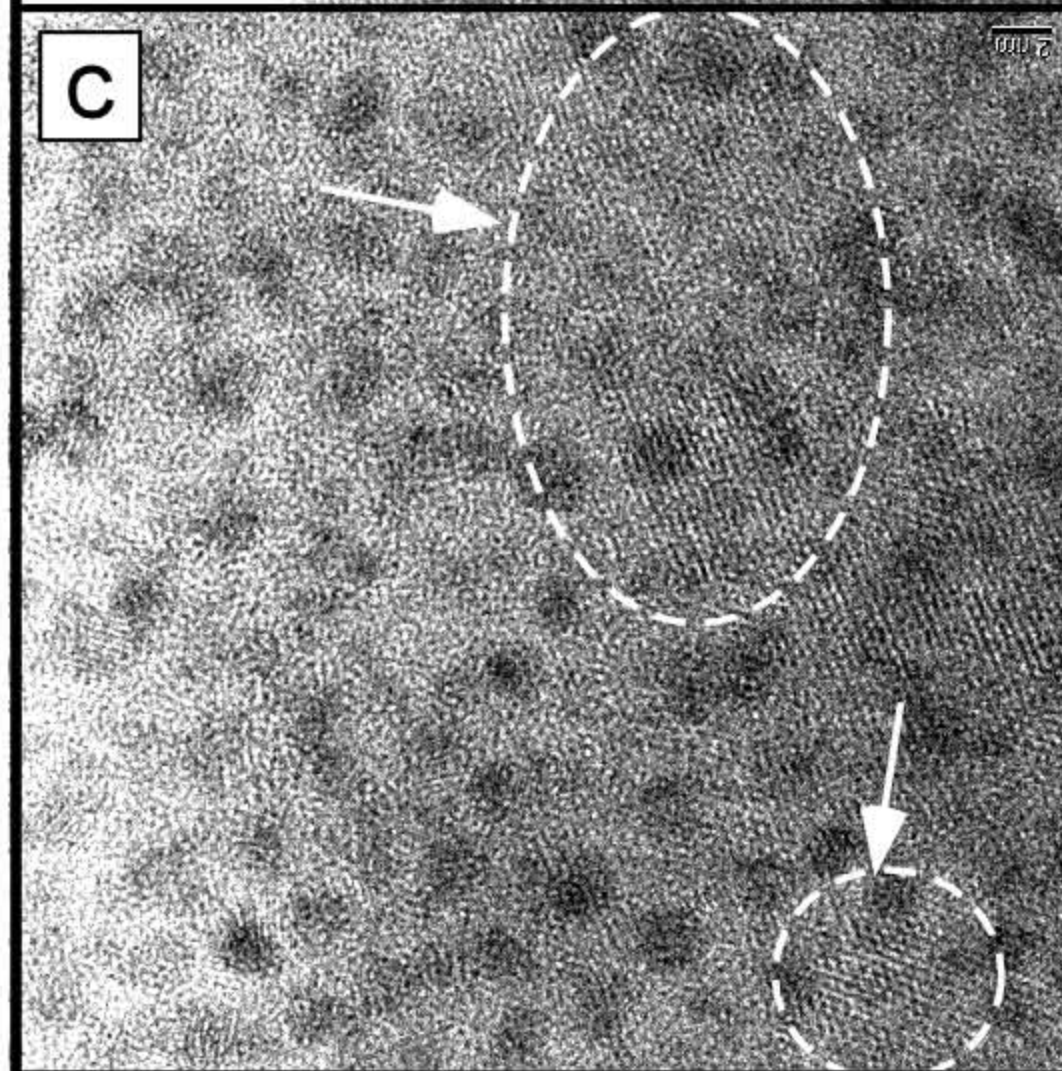
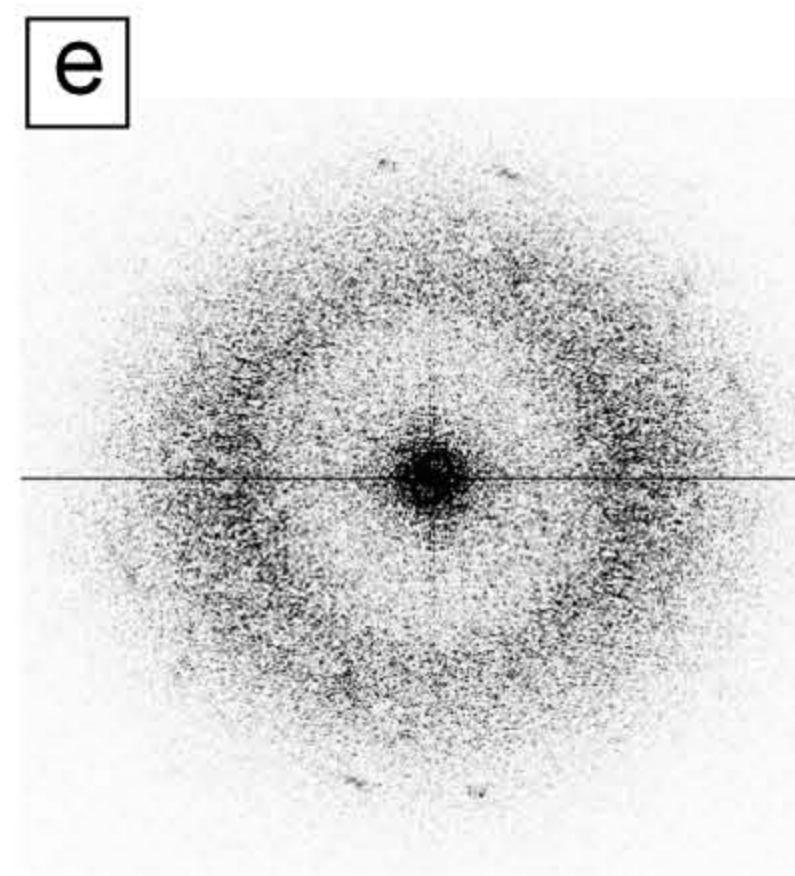
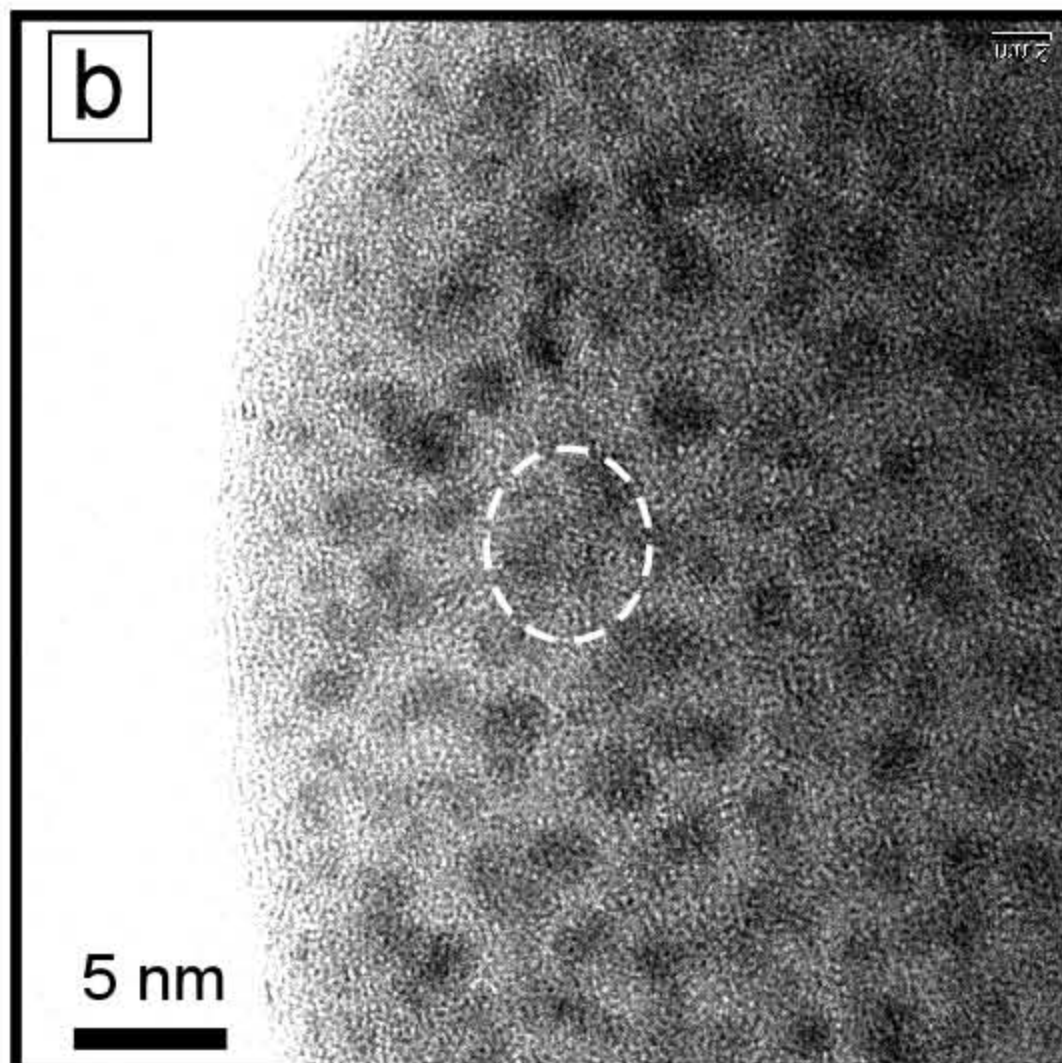
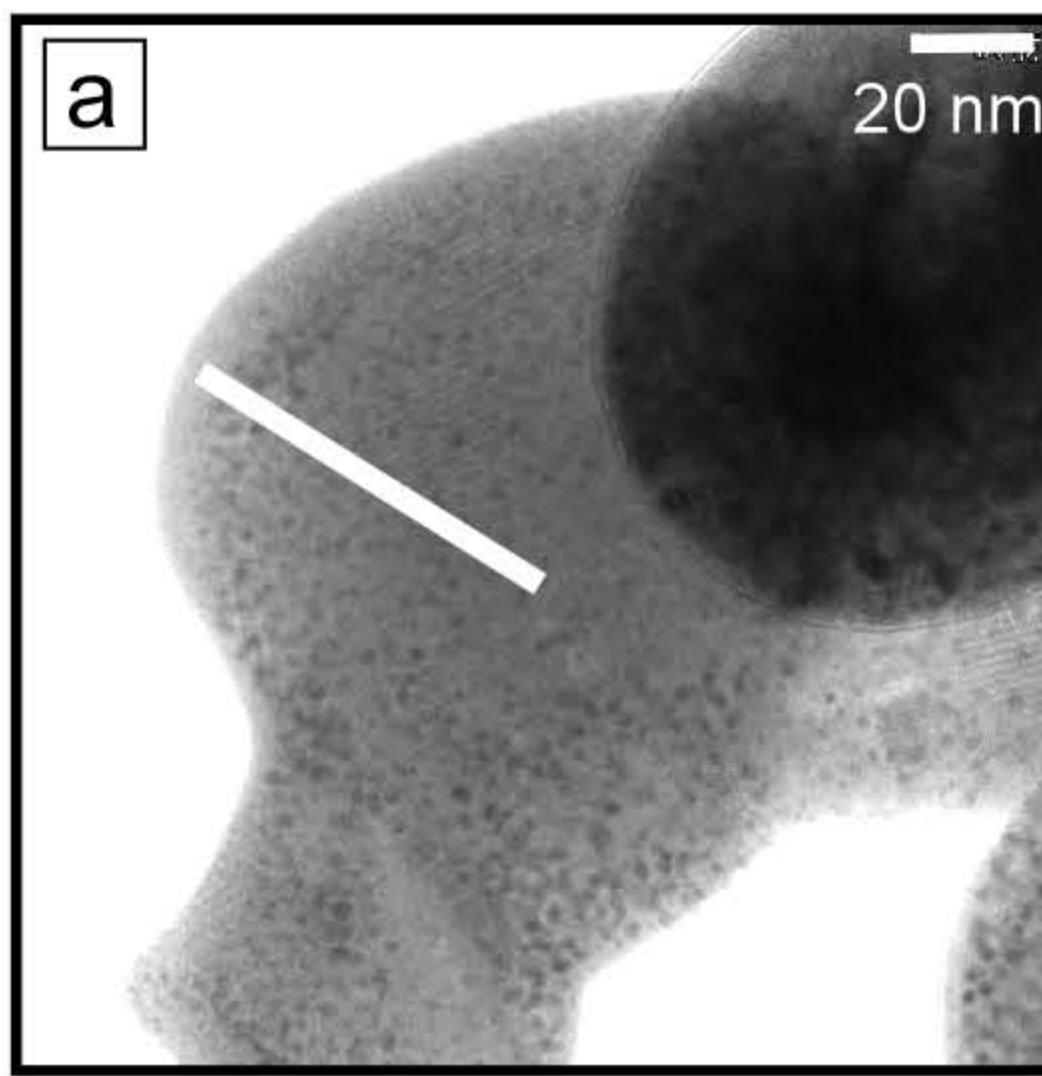


Fig. 4 Jacob, Wirth, Soldati, Wehrmeister, Schreiber