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1 Petrology and isotope geochemistry of the Mesoproterozoic
2 anorthosite and related rocks of the Mesoproterozoic Kunene
3 Intrusive Complex, NW Namibia

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Abstract

The intrusion of the large (c. 18.000 km²) anorthosite body of the Kunene Intrusive Complex, Angola/Namibia, which is associated with minor marginal granite bodies and syenite/syenodiorite dykes, marked the beginning of Mesoproterozoic igneous activity in NW Namibia. Anorthosites display an igneous mineral assemblage of dominating plagioclase, together with minor olivine, orthopyroxene or clinopyroxene, Fe-Ti oxides, whereas biotite and pargasite are late-magmatic phases. Apatite is a common accessory phase. In regionally associated felsic rocks K-feldspar, plagioclase, clinopyroxene and/or hastingsite crystallized first, followed by interstitial quartz, hastingsite, Fe-Ti oxides, and titanite. The general geochemical characteristics for the anorthositic rocks of the Kunene Intrusive Complex (i.e. high Al₂O₃, CaO, Sr and Eu, intermediate to high Mg numbers of 0.37-0.74, positive εNd (T) values of +3.0 to +1.0, and initial Sr ratios of 0.7028-0.7041) are in excellent agreement with their derivation from fractionated basaltic liquids. The δ¹⁸O values obtained for plagioclase (5.88 ± 0.19 ‰ δ¹⁸O) support their development from mantle-derived magmas. Relic crustal xenoliths of calc-silicate rocks and Sr and Nd isotope data, however, suggest that the anorthosite magma underwent minor crustal contamination, which was most prominent during the early intrusion stage. A U-Pb-zircon age of 1376 ± 2 Ma was obtained for a felsic rock associated with the 1385 ± 25 Ma anorthosites, which suggests that the two suites were emplaced during the same igneous event. The geochemical and isotope signatures of the granites, syenites, and syenodiorites provide convincing evidence against consanguinity of the anorthositic and the felsic rock suites, i.e.: (1) a compositional gap exists between the major and trace element contents of the two suites, (2) trace element and Sr, and Nd isotopic data of the felsic rocks point to a crustal source (εNd: +2.0 to -0.4, initial Sr ratio: 0.7024-0.7063), (3) the REE patterns of the felsic rocks do not exhibit ubiquitous negative Eu-anomalies, which

1 would be expected from fractionation products of melts that previously formed extensive
2 plagioclase cumulates, and (4) $\delta^{18}\text{O}$ values of magmatic feldspar from the felsic rock suite fall
3 in a restricted range (7.20-7.92 ‰) that is about 1.6 ‰ higher than the average igneous
4 plagioclase $\delta^{18}\text{O}$ of the anorthosites and suggests a formation of the felsic melts by anatexis of
5 the lower crust.

6

7 *Keywords:* Congo Craton, felsic rocks, isotope geochemistry, leuconorite, leucotroctolite,
8 oxygen isotopes.

1 **1. Introduction**

2

3 Massif-type anorthosite complexes, commonly associated with minor volumes of mafic
4 and granitic rocks, are characteristic features of the Proterozoic crust. Their apparent temporal
5 restriction to the Proterozoic suggests that specific tectono-thermal regimes and crust-forming
6 processes have been operative during this period. Many anorthosite massifs underwent a
7 prograde amphibolite to granulite facies metamorphism after their emplacement, which
8 complicates the reconstruction of their magmatic evolution. Thus, the petrogenesis of
9 Proterozoic anorthosite complexes is still a matter of debate. No general agreement has been
10 reached on issues such as the source of parental magmas of the anorthosite and granite suites,
11 monobaric vs. polybaric anorthosite crystallization, mode and depth of emplacement and
12 cooling history (*e.g.* Ashwal, 1993, for a review).

13 In general, the petrogenesis of the huge anorthosite massifs is interpreted in terms of
14 partial melting of the upper mantle, followed by plagioclase flotation in a magma chamber at
15 the mantle-crust boundary and buoyant ascent of the plagioclase-rich crystal mushes to their
16 final depth of emplacement (Duchesne, 1984; Emslie, 1985, Emslie et al., 1994; Longhi and
17 Ashwal, 1985; Ashwal, 1993; Wiebe, 1994; Mitchell et al., 1995). Recently, however, many
18 authors favour the hypothesis of anorthosite generation by lower crustal anatexis (Taylor et
19 al., 1984; Duchesne et al., 1999; Longhi et al., 1999; Schiellerup et al., 2000; Selbekk et al.,
20 2000, Wiszniewska et al., 2002), which is supported by Re-Os isotope data and experimental
21 investigations. With the results presented here we contribute new data to this debate.

22 Most anorthosite massifs contain a considerable volume of temporally and spatially
23 associated granitic plutonic rocks, characterized by a composition ranging from diorite to
24 granite and a mostly anhydrous mineralogy. Commonly, intrusive contact relationships
25 suggest these granitoids to be younger than the associated anorthosite (*e.g.* de Waard, 1970;

1 Seifert, 1978). The fact, that the mineral chemistry of the Fe-Mg silicates and plagioclase in
2 the silicic suites appears to extend the compositional ranges of the anorthositic minerals to
3 higher X_{Fe} and lower An-contents, respectively, has been interpreted in terms of
4 consanguinity between the two rock suites (e.g. Fuhrman et al., 1988). An alternative model
5 considers the granitoids to represent a chemically independent, crustal derived intrusive suite
6 (e.g. Anderson, 1980; Anderson & Bender, 1989; Emslie, 1991; Emslie et al., 1994).

7 Among the anorthosite bodies of the world, the Kunene Intrusive Complex (KIC), NW
8 Namibia, and the Lac-Saint-Jean Complex, Canada, are presumably the largest exposed
9 massifs, both covering a total surface area of *c.* 18,000 km². Nonetheless, the Kunene
10 Intrusive Complex is one of the least known anorthosite bodies of the world. In clear contrast
11 to most anorthosite bodies world-wide, the Mesoproterozoic KIC experienced no prograde
12 metamorphic overprint after its emplacement (e.g. Ashwal and Twist, 1994; Morais et al.,
13 1998; Drüppel, 1999, 2003; Drüppel et al., 2001; Slejko et al., 2002; Mayer et al., 2004) and
14 hence allows direct study of the igneous processes that led to its formation.

15 We describe and interpret the mineralogical and geochemical signatures and Nd, Sr, and O
16 isotope data from the anorthositic rocks of the Kunene Intrusive Complex and the associated
17 granitoid suite and provide a U-Pb zircon age for the emplacement of the associated syenites.
18 Our results place constraints on the tectono-metamorphic evolution of the Congo Craton
19 during the Proterozoic and on some aspects of the petrogenesis of massif-type anorthosites in
20 general.

21

22 **2. Geological framework**

23

24 Anorthositic rocks of the Kunene Intrusive Complex (KIC) cover an estimated surface
25 area of 18,000 km² in NW Namibia and SW Angola and are regarded to form one of the

1 largest massif-type anorthosite bodies in the world (e.g. Ashwal and Twist, 1994; Morais et
2 al., 1998; Drüppel et al., 2001; Mayer et al., 2004). The anorthosite massif was emplaced at
3 the southern margin of the Congo Craton in the Mesoproterozoic. In Namibia, the KIC
4 intruded the high-grade metamorphic rocks of the Epupa Complex, which mainly consists of
5 upper amphibolite facies rocks of the Orue Unit, comprising migmatitic granitic orthogneisses
6 and subordinate migmatitic metasedimentary and metavolcanic rocks (Brandt, 2003; Brandt et
7 al., 2003). South of the KIC, two limited areas of ultrahigh-temperature granulite-facies rocks,
8 the Epembe Unit, have been recognized by Brandt et al. (2003, 2007), that mainly consist of a
9 volcano-sedimentary sequence of interlayered felsic and mafic granulites, intercalated with
10 subordinate migmatitic metasediments that contain rare lenses of sapphirine-bearing
11 granulites (Brandt et al., 2003, 2007). The Epembe Unit is separated from the Orue Unit by
12 subvertical ductile shear zones. Both units preserve Palaeoproterozoic protolith ages (Seth et
13 al., 2003, 2005), that were also recognized in other parts of the Epupa Complex and the
14 Congo craton (see reviews in Tegtmeier and Kröner, 1985; Carvalho et al., 2000). SHRIMP
15 U-Pb zircon dating of granulite facies rocks from the Epembe Unit reveals Mesoproterozoic
16 ages between 1520 Ma and 1510 Ma for prograde zircon growth under granulite-facies
17 conditions, whereas slightly younger Pb-Pb garnet ages of 1450 ± 50 Ma date the peak of
18 UHT metamorphism (Seth et al., 2003). Distinctly younger Pb-Pb garnet and U-Pb SHRIMP
19 ages of zircon rims of 1340-1320 Ma have been recorded for part of the amphibolite facies
20 rocks of the Orue Unit by Seth et al. (2005).

21 Rarely exposed contacts between the KIC and the metamorphic basement are largely
22 obliterated by broad ductile shear zones and late brittle faults, which were subsequently
23 intruded by syenite and/or dolerite dykes. Direct intrusion contacts are only observed between
24 the white anorthosite and the amphibolite facies Orue Unit at the southeastern margin of the
25 KIC. Along this contact zone, a narrow contact-thermal aureole of several metres in width is

1 developed in the Orue Unit, which is recorded by Grt-Crd-Sill hornfelses (Brandt, 2003;
2 Brandt et al., 2003). For this rock a Pb-Pb garnet age of 1341 ± 47 Ma Ma has been recorded
3 (Seth et al., 2005).

4

5 At least two major anorthosite units can be distinguished, separated by a NE-SW trending
6 belt of granitoid intrusions (Morais et al., 1998; Drüppel et al., 2001; Fig. 1):

7 (1) The majority of the KIC exposed in Angola displays a N-S elongation and is mainly
8 composed of dark, weakly altered anorthositic rocks (mainly anorthosites and leucotroctolites;
9 *e.g.* Carvalho and Alves, 1990, Silva, 1990, 1992, Ashwal and Twist, 1994, Morais et al.,
10 1998, Slejko et al., 2002, Mayer et al., 2004). Based on satellite images Ashwal & Twist
11 (1994) concluded that even this northern part of the KIC represents a composite massif,
12 composed of at least eight separate anorthosite plutons. A similar interpretation has also been
13 reached by Morais et al. (1998) and Slejko et al. (2002), who distinguished at least two
14 individual anorthosite bodies in the Angolan part of the KIC, a northern and a southern zone,
15 that display remarkable differences in their geochemical characteristics and mineralogy.

16 (3) The southernmost KIC, on the other hand, including the study area in Namibia (Fig. 1,
17 2), has a pronounced E-W elongation. In this area the KIC mainly consists of heavily
18 tectonised and pervasively altered pale anorthosite, the 'white anorthosite' (*cf.* Köstlin, 1967,
19 1974, Menge, 1998, Drüppel et al., 2001). The coarse white anorthosite is characterised by a
20 white or pale violet and greenish colour and comprises anorthosites *sensu stricto* as well as
21 leuconorites with abundant orthopyroxene of up to 8 cm in diameter. This white anorthosite
22 massif is intruded by subordinate, sheet-like bodies of dark, weakly altered anorthosites,
23 leucogabbronorites and leucotroctolites ('dark anorthosite') that typically contain xenoliths of
24 the white anorthosite. Anorthosites and leucogabbronorites of the dark anorthosite suite
25 mainly occur in the centre of the white anorthosite massif, forming the Zebra Mountains, and

1 grade into small bodies of leucotroctolite and olivine-bearing anorthosite towards the margin
2 of the KIC. Cumulate textures are relatively widespread on a local scale; yet no evidence for
3 extensive layering or igneous lamination was recognized, in contrast to the interpretation of
4 aerial photographs by Menge (1996, 1998). If transected by major faults dark anorthosite is
5 always strongly sericitised and can then be easily confused with the white anorthosite.
6 Towards the north, the dark anorthosite massif is terminated by a major NE trending, ca. 6 km
7 wide dextral shear zone, the ‘Serpa Pinto Lineament’ of Menge, 1998, which is partly
8 intruded by small dark anorthosite bodies. Along this major fault bundle the white anorthosite
9 is strongly tectonized and partly intermingled with structurally isolated and incorporated
10 blocks of the Orue Unit (Fig. 2).

11 For the emplacement of the KIC at the southern margin of the Congo craton, an internal
12 biotite-plagioclase whole-rock Rb-Sr isochron date of 1347 ± 13 Ma has been determined for
13 an anorthosite sample of the northern Angolan part of the complex (Mayer et al., 2000). This
14 age has been superseded by an almost concordant U-Pb age of 1371 ± 2.5 Ma using zircons
15 from a cogenetic mangerite vein (Mayer et al., 2004). The latter age is in accordance with a
16 preliminary, subconcordant U/Pb single-zircon age of 1385 ± 25 Ma, obtained for a
17 leucogabbronorite sample of the dark anorthosite suite from Namibia (Drüppel et al., 2000). A
18 Sm-Nd mineral (amphibole-biotite-plagioclase-pyroxene) bulk rock age of 1319 ± 13 Ma,
19 determined on an Angolan anorthosite, marks the end of the anorthosite activity (Mayer et al.,
20 2004). Radiometric dates for the white anorthosite are lacking, so far. The age of the
21 emplacement and cooling of the KIC between *c.* 1385 and 1319 Ma covers the ages of both
22 contact metamorphism at ca. 1341 Ma and the regional metamorphic overprint of the Orue
23 Unit at 1340-1320 Ma, suggesting that the emplacement of the anorthositic magma had a
24 regional thermal effect on the bordering Orue Unit.

1 Massive Fe-Ti ore bodies have been reported from all parts of the KIC (e.g. Morais et al.,
2 1998, von Seckendorff et al., 2000) but are, in our study area, largely restricted to the white
3 anorthosite. Several small mafic (gabbro) and ultramafic bodies (pyroxenite, hyperite,
4 peridotite) which are inferred to be genetically related to the KIC, crop out within the
5 basement near the contact to the KIC (Fig. 2). Close to its southernmost margin, the Namibian
6 KIC as well as the neighbouring basement are transected by numerous SE and NE trending
7 felsic dykes (mainly alkali granites, syenites, syenodiorites). These felsic intrusives may be
8 subdivided into two units (Fig. 2): (1) an older generation of mainly ENE trending bodies of
9 up to 40 km long and 2-3 km wide is exposed in the Epupa Complex and (2) a second
10 generation of mainly NW trending dykes exposed over 100 m in length and 3-10 m in width
11 are observed exclusively within the marginal zones of the KIC. The emplacement of the felsic
12 rocks in both areas may be related to regional granitic magmatism in NW Namibia in the
13 Mesoproterozoic, i.e. the granite belt that separates the northern and southern part of the KIC
14 (Rb-Sr whole-rock isochrons of 1411-1302 Ma: Torquato et al., 1979) and mangerite veins in
15 anorthosite in Angola (U/Pb single-zircon age of 1371 ± 2.5 Ma: Mayer et al., 2004). Near the
16 southern margin of the KIC, the basement rocks are intruded by several nepheline syenite
17 stocks with U-Pb zircon ages of ca. 1216-1213 Ma (Littmann et al., 2000; Fig. 2) In the
18 southeastern part of the KIC, in the Swartbooisdrif area, the subsequent emplacement of ca.
19 1140 Ma ferrocarnatites as dykes was locally responsible for a local but pervasive
20 fenitisation of the older felsic rocks (Drüppel et al., 2005).

21

22 **3. Analytical methods**

23

24 All electron microprobe analyses reported here were performed on a CAMECA SX50
25 instrument, with three WDS spectrometers, housed in the Institute of Mineralogy, University

1 of Würzburg or on a CAMECA SX50 and SX100 at the GeoForschungsZentrum Potsdam
2 (GFZ). Operating conditions were 15 kV accelerating voltage and 15 nA beam current with 1-
3 5 μm spot size of Würzburg and 20 nA beam current with 2-5 μm for amphibole, pyroxene,
4 epidote and magnetite/ilmenite to 10 μm for feldspar at the GFZ Potsdam. The accessory
5 minerals apatite and zircon were measured with an accelerating voltage of 20 kV, using a
6 beam current of 10 nA for apatite and 40 nA for zircon at the GFZ Potsdam.

7 The X-ray fluorescence (XRF) analyses of the major and trace element content
8 anorthositic and granitic rocks were determined from fused glass discs with a Philips PW
9 1480 XRF spectrometer at the Institute of Mineralogy, Würzburg. Matrix effects were
10 corrected automatically by the Philips X40 software. The relative analytical error for major
11 and trace elements is 1 % and 1-8 %, respectively. FeO contents were analysed using a ZEISS
12 PMD 2 spectral photometer.

13 Additional trace elements of selected samples were analysed by Inductively Coupled
14 Plasma Mass Spectrometry (ICP-MS) using a VG Plasma Quad PQ²⁺ at the GFZ Potsdam,
15 Germany. Sample powders were digested in HF-HClO₄, dried and then redissolved in dilute
16 HNO₃. Calibration was carried out using multi-element solutions and routinely checked for
17 accuracy with international geological reference materials. Precision is generally better than \pm
18 10% (rel.) for each element (Plessen, 1997).

19 Rare earth element analyses were carried out at the GFZ Potsdam. Sample preparation
20 followed the method of Zuleger and Erzinger (1988). Measurements of REE and Y
21 concentrations were conducted by ICP-AES. Details on ICP equipment, operating conditions,
22 background wavelength corrections and background equivalent concentrations as well as
23 precision of the method are given in Zuleger and Erzinger (1988).

24 Stable isotope ratios are reported for hand-picked mineral separates of feldspar. The
25 silicates were treated with ClF₃ according to the method described by Borthwick and Harmon

1 (1982) at the Geochemical Institute, University of Göttingen. All $\delta^{18}\text{O}$ values are given
2 relative to SMOW. The reproducibility of $\delta^{18}\text{O}$ values is better than $\pm 0.2\text{‰}$.

3 The Sr and Nd isotope composition of the whole rock samples was determined at the GFZ
4 Potsdam. The whole rock powders were dissolved in HF on a hot plate (160°C). After drying,
5 residues were redissolved in 6N HCl and split for the determination of the isotopic
6 composition (IC) and for concentration determination by isotope dilution (ID). The ID splits
7 were spiked with ^{87}Rb - ^{84}Sr tracer and a ^{150}Nd - ^{149}Sm tracer. Strontium and rubidium were
8 analysed on Ta filaments using a VG Sector 54-30 mass spectrometer and neodymium and
9 samarium were analysed with a Finnigan MAT 262 mass spectrometer on Re filaments. All
10 Sr ratios were normalized to $^{86}\text{Sr}/^{88}\text{Sr} = 0.1194$. The average value for $^{87}\text{Sr}/^{86}\text{Sr}$ obtained from
11 NBS SRM 987 standard during the period of this study was 0.710246 ± 10 (2σ , $n = 20$
12 independent measurements). Mass fractionation for Nd and Sm was corrected by normalizing
13 to $^{146}\text{Nd}/^{144}\text{Nd} = 0.7219$ and $^{147}\text{Sm}/^{152}\text{Sm} = 0.5608$, respectively. Repeated analysis of the La
14 Jolla standard yielded a value of $^{143}\text{Nd}/^{144}\text{Nd} = 0.511855 \pm 8$ (2σ , $n = 15$ independent
15 measurements). For details see Romer et al. (2005). Initial Sr and Nd isotopic ratios were
16 calculated using Rb/Sr and Sm/Nd ratios determined by isotope dilution.

17 Zircon was used for the U-Pb age determination of the syenite suite at the GFZ Potsdam.
18 The zircon separates were produced using Wilfley table, heavy liquids, and Frantz isodynamic
19 magnetic separator. Concentrates were purified by hand under the binocular microscope. The
20 individual analyses were performed on euhedral, unbroken crystals. A mixed ^{205}Pb - ^{235}U tracer
21 was added before sample dissolution in HF. The samples were dissolved in teflon capsules in
22 a Parr® autoclave at 220°C for four days. After drying the samples on the hot plate, samples
23 were taken up in 6N HCl and redissolved overnight in the Parr® autoclave at 220°C. U and
24 Pb were separated using the ion-exchange chromatography of Krogh (1973). Pb and U were
25 loaded together with colloidal silica and H_3PO_4 on a single Re-filament. The isotope ratios for

1 Pb and U were determined at 1200-1260°C and 1300-1560°C, respectively, on a Finnigan
2 MAT 262 multi-collector mass spectrometer using an ion-counter and Faraday collectors.
3 Analytical precisions on the 2 σ level, including uncertainties for the mass fractionation, are
4 generally better than 0.1%, except for ratios involving ^{204}Pb . The total procedure blanks were
5 5 to 10 pg for Pb and below 1 pg for U. Ages were calculated using the constants
6 recommended by IUGS (Steiger and Jäger, 1977).

7

8 **4. Petrography and Mineral Chemistry**

9

10 Data on the mineral chemistry of selected minerals were obtained for 20 rock samples out
11 of a total of 67 samples of the anorthosite suite and 6 rock samples out of 57 samples of the
12 granite-syenite suite. Representative analyses of feldspar, pyroxene and amphibole are
13 compiled in Tables 1 to 3. The full set of EMP analyses is given in Drüppel (2003) and may
14 be obtained from the first author, on request.

15

16 *4.1. Anorthositic rocks*

17

18 The petrography and mineral chemistry of the anorthositic rocks of the KIC were
19 discussed in some detail by Drüppel et al. (2001). The authors have shown that both the white
20 and the dark anorthosite suite display the primary magmatic assemblage of cumulus
21 plagioclase (An_{43-53} and An_{43-75} , respectively) \pm olivine \pm orthopyroxene \pm clinopyroxene \pm
22 biotite \pm pargasite/magnesio-hastingsite + ilmenite + magnetite \pm apatite \pm zircon \pm sulfides.

23 Crustal xenoliths are represented by relic xenocrysts of andradite-rich garnet (average
24 composition: $\text{Adr}_{72}\text{Grs}_{27}$), which are exclusively observed in samples of white anorthosite and

1 in leucogabbronorite of the dark anorthosite suite. The fractured garnet grains are partially
2 replaced by chlorite and surrounded by late-magmatic pargasite. The mineralogical
3 compositions of the xenoliths point to a contamination of the parental melt with calc-silicate
4 gneisses of the country rock.

5

6 **4.1.1 White anorthosite**

7

8 White anorthosite represents the first recognisable generation of anorthosite magmatism.
9 A characteristic feature of the white anorthosite is the dominance of coarse (commonly 1-5
10 cm in diameter), anhedral cumulus plagioclase (87-99 vol.%). Plagioclase crystals commonly
11 display embayed grain boundaries and suffered an almost complete hydrothermal alteration
12 (i.e., saussuritisation, albitisation and/or sericitisation), responsible for the white to pale violet
13 or greenish colour of these rocks. Unaltered plagioclase relicts (An_{43-53}) mostly exhibit a weak
14 normal or oscillatory zoning with an average An-decreases from core to rim of about 2
15 mol.%.

16 Cumulus to interstitial orthopyroxene (X_{Mg} : 0.71), in cases forming orthopyroxene-
17 megacrysts of up to 8 cm in diameter, is the main Fe-Mg silicate present in the leuconorites
18 and is commonly completely replaced by chlorite that encloses preserved ilmenite exsolution
19 lamellae. Accessory intercumulus clinopyroxene is always completely replaced by late
20 actinolite and enveloped by a narrow rim of late-magmatic pargasite (Al p.f.u.: 2.42-2.97;
21 X_{Mg} : 0.65-0.70; Table 3). Magnetite and ilmenite are minor constituents of the white
22 anorthosite and occur as discrete anhedral grains in the interstices of cumulus plagioclase.
23 Olivine or its alteration products have not been observed in the investigated samples.

24 Small-scaled ductile shear zones transect part of the samples. Locally, plagioclase is
25 strongly deformed or even recrystallised to granular mosaics of inversely zoned albite (An_{3-4}

1 to An₅₋₁₀), that presumably formed during the hydrothermal alteration of the rocks. Late
2 cracks are filled by carbonate and/or epidote.

3

4 **4.1.2 Dark anorthosite**

5

6 In the subsequently emplaced dark anorthosite suite subhedral tabular coarse-grained
7 (commonly 0.5-3 cm in diameter) cumulus plagioclase is the dominant mineral (65-99 vol.%).
8 The vast majority of the analysed plagioclase crystals display minor chemical zoning, with the
9 anorthite contents falling in the compositional range of andesine to labradorite (An₄₃₋₇₅; Table
10 1). The An-contents of plagioclase increase in the sequence (1) leucogabbro/pyroxene
11 anorthosite (An₄₃₋₅₄), (2) leucotroctolite/olivine anorthosite (An₄₇₋₆₀), and (3) garnet-bearing
12 leucotroctolite (An₅₅₋₇₅). Minor orthopyroxene and clinopyroxene are the dominant mafic
13 minerals (3-12 vol.%) in the leucogabbronorite samples (Fig. 3a). They occur as large,
14 subhedral to anhedral cumulus to intercumulus phases or form rims around olivine. Following
15 the IMA-classification (Morimoto, 1988) clinopyroxene of the leucogabbronorites
16 (Wo_{44.4}En_{32.6}Fs_{18.3} – Wo_{45.7}En_{27.8}Fs_{20.7}) and leucotroctolites (Wo_{40.03}En_{34.4}Fs_{21.5} –
17 Wo_{45.7}En_{39.1}Fs_{12.3}) is diopside and subordinately augite (Table 2). The Al₂O₃ content of
18 clinopyroxene increases from leucogabbronorite (1.5-1.9 wt.%) to leucotroctolite (1.7-3.3
19 wt.%). Orthopyroxene of the anorthositic suite is enstatite with high X_{Mg} (0.54-0.63) but with
20 lower Al₂O₃ contents (0.4-1.8 wt.%; Table 2). In the olivine anorthosite and leucotroctolite
21 subordinate olivine (5-8 vol.%; X_{Mg}:0.54-0.65) occurs as small subhedral inclusions in
22 plagioclase or as large, subhedral to anhedral, interstitial grains, partially rimmed by pyroxene
23 (Fig. 3a, 3b). Magnetite is observed as minute inclusions in plagioclase and olivine, or,
24 together with ilmenite, as large, anhedral grains in the interstices between plagioclase crystals
25 (up to 3 vol.%). Especially in the leucogabbronorite, both, ilmenite and pyroxene are
26 surrounded by narrow rims of late-magmatic biotite (X_{Mg}: 0.39-0.67) and amphibole. The vast

1 majority of late-magmatic amphibole of the leucogabbronorite and pyroxene anorthosite of
2 the dark anorthosite suite is magnesio-hastingsite/pargasite to ferro-pargasite/hastingsite (Al
3 p.f.u.: 1.82-2.51; X_{Mg} : 0.46-0.73; Table 3). Anorthosite *senso stricto* contains only minor
4 amounts of Mg-Fe silicates and Fe-Ti oxides (0.5-3 vol.%). Zircon is extremely rare and has
5 only been observed in two of the 67 investigated anorthosite samples. Apatite is commonly
6 present as an accessory phase in the leucogabbronorites, where it occurs as inclusions in the
7 late-magmatic amphibole and biotite. In some leucotroctolite samples garnet-bearing corona
8 textures are developed around olivine and Fe-Ti oxides (Drüppel et al., 2001).

9

10 4.2. Felsic rock suite

11

12 The felsic rocks are characterised by a coarse equigranular, massive appearance. A weak
13 foliation is exclusively developed in the marginal zones of the large granite bodies in the
14 basement and is attributed to syn-emplacement deformation. Small felsic dykes transecting
15 the KIC, on the other hand, locally show igneous lamination, defined by a preferred
16 orientation of feldspar. The samples investigated are mainly composed of subhedral to
17 anhedral, tabular alkali feldspar (55-75 vol.%; microcline and occasional orthoclase: Or_{91-98}),
18 displaying strong micro- to mesoperthitic albite exsolution (An_{1-4} ; Fig. 3c). Reintegrated
19 compositions of alkali feldspar cores fall in the range of $Ab_{43.3}An_{1.9}Or_{54.8}$ – $Ab_{85.8}An_{4.1}Or_{10.1}$
20 whereas the feldspar rims have a more albite-rich composition ($Ab_{86.8}An_{2.6}Or_{10.6}$ –
21 $Ab_{88.3}An_{3.1}Or_{8.6}$) or are the almost pure albite end member (Ab_{95-99}). String perthite,
22 transgressing into patch and block perthite is most common, whereas other exsolution textures
23 like braid, vein, plate or flame perthite are rare. A second generation of later microcline may
24 occur, forming separate, anhedral grains that lack exsolution textures. Sericitization is
25 concentrated in the crystal cores. Platy plagioclase with weak reverse zonation (An_{9-18} ; Table

1 1), showing mostly subhedral to euhedral shape occurs in variable amounts (5-35 vol.%), but
2 is always less abundant than alkali feldspar. Secondary albite, grown at the expense of alkali
3 feldspar, exhibits a weak normal zonation (An_{4-5} to An_{0-2}) or is the unzoned, almost pure
4 albite end member. In some of the samples that underwent hydrothermal alteration secondary
5 albite may be present in significant amounts of > 50 vol.%.

6 The dominant mineral among the Fe-Mg silicates (5-23 vol.%) in most of the felsic rock
7 samples is hastingsite, which is either enclosed by plagioclase or fills interstices between
8 individual feldspar crystals. Amphibole of both textures displays low Al and X_{Mg} (Al p.f.u.:
9 1.91-2.10; X_{Mg} : 0.01-0.06) when compared to amphibole of the anorthosite suite (Table 3). In
10 quartz-poor or -free samples greenish clinopyroxene (3-8 vol.%; X_{Mg} : 0.45-0.51) may be
11 present, which fills interstices between feldspar crystals (Fig. 3d), but may also occur as
12 inclusions in feldspar. The clinopyroxene is generally more hedenbergitic in composition
13 ($Wo_{43.8}En_{23.1}Fs_{28.0}$ – $Wo_{45.1}En_{25.4}Fs_{24.3}$) when compared to that of the anorthositic rocks
14 (Table 2). Its Al_2O_3 contents are low (0.4-0.8 wt.%), reflecting the Al-poor whole-rock
15 composition of the rocks. In most samples clinopyroxene is partially to completely replaced
16 by fibrous ferro-hornblende to ferro-actinolite (X_{Mg} : 0.23-0.31). Anhedral magnetite, either
17 spatially associated with ilmenite or displaying oxidative exsolution of ilmenite after lamellae
18 as well as anhedral quartz (granite 13-25 vol.%, quartz-bearing syenodiorite: 5-8 vol.%) fill
19 interstices between feldspar crystals. Titanite (aTtn: 0.87-0.90), forming large anhedral grains
20 or as rims around Fe-Ti oxides, is the most common accessory phase, associated with
21 euhedral zircon. Apatite occurs as stubby prisms, commonly enclosed by plagioclase.

22 Secondary enrichment of epidote (Ps_{21-34}), locally associated by carbonate and/or chlorite
23 (X_{Mg} : 0.09-0.11), is common in samples, where hydrothermal alteration is more pronounced.
24 A conspicuous feature are symplectitic intergrowths of green hastingsite (X_{Mg} : 0.01-0.06),
25 plagioclase (An_{9-18}), ilmenite and magnetite replacing interstitial hastingsite.

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5. Geochemical characteristics

Major and trace element geochemical data were obtained for 56 rock samples of the anorthosite suite and 33 rock samples of the granitoid suite. Representative analyses of the major and trace element data and of the REE data are compiled in Tables 4 and 5, respectively. The full set of analyses is given in Drüppel (2003) and may be obtained from the first author, on request.

5.1. Anorthositic rocks

Both the white and the dark anorthosite suite are rich in Al_2O_3 (21-29 and 16-30 wt.%), SiO_2 (45-51 and 38-54 wt.%), CaO (4-13 and 7-13 wt.%), and Na_2O (4-9 and 3-6 wt.%), reflecting high amounts of plagioclase, whereas they are deficient in all other oxides (Fig. 4). The $\text{Fe}^{\text{tot}}/(\text{Fe}^{\text{tot}}+\text{Mg})$ ratio is > 0.28 , pointing to anorthosite crystallization from a fractionated melt (Fig. 5). The highest average $\text{Fe}^{\text{tot}}/(\text{Fe}^{\text{tot}}+\text{Mg})$ values occur in the white anorthosite (0.51-0.74) and leucogabbronorite of the dark anorthosite suite (0.31-0.66). The geochemical trends of the major elements of the least altered samples of the dark anorthosite suite, as demonstrated in various Harker diagrams, correlate well with varying modal proportions of plagioclase and Fe-Mg silicates and Fe-Ti oxides and are hence unlikely to reflect different degrees of fractionation. Decreases of Si, Al, Ca, and Na are accompanied by increases of Fe, Mn and Mg from anorthosite and troctolite towards leucogabbronorite of the dark anorthosite suite (Fig. 4, 5). Leucogabbronorites generally display the highest values of K and P, reflecting generally high amounts of late-magmatic biotite and apatite, which crystallized from pockets of intercumulus melt. Trends of the white anorthosite are mostly

1 indistinguishable from those of the dark anorthosite suite, with the exception of Na and K that
2 were apparently strongly enriched during the hydrothermal alteration.

3 The highest values for Ni are obtained for olivine-bearing rock samples, whereas the high
4 V and Zn values of the leucogabbronorites are reflected by higher modes of ilmenite and
5 pyroxene. The highest Sr concentrations have been obtained for white anorthosite and
6 pyroxene anorthosite/leucogabbronorite of the dark anorthosite suite, which also exhibit the
7 highest overall trace element abundances. Regarding their rare earth element patterns, all
8 anorthositic rocks are characterized by a variable enrichment of the light REE (La_N/Yb_N : 3.3-
9 26.9), a strong positive Eu (Eu/Eu^* : 1.3-11.2) anomaly and flattening in the heavy REE (Fig.
10 6), typical of plagioclase cumulate rocks. In the dark anorthosite, the overall abundance of
11 REE increases in the sequence leucotroctolite/olivine-bearing anorthosite towards pyroxene-
12 bearing anorthosite/leucogabbronorite, resulting in a relative decrease of the Eu anomaly. The
13 overall increase of the REE can be partly explained by a higher modal abundance of apatite in
14 the leucogabbronorite. The white anorthosite shows strongly varying REE patterns, one (Ku-
15 97-45; Eu/Eu^* : 7.1, La_N/Yb_N : 26.9) resembling that of the leucotroctolites of the dark
16 anorthosite suite whereas two other patterns (Ku-98-44, Ku-97-33b; Eu/Eu^* : 1.6-2.3;
17 La_N/Yb_N : 5.9-12.6) agree well with those of the leucogabbronorites and pyroxene
18 anorthosites. It may thus be concluded that the white anorthosite, like the dark anorthosite, is
19 not only composed of anorthosites *sensu stricto* as was proposed by Menge (1998) but also
20 comprises leucotroctolites and leucogabbronorites. When compared to the corresponding
21 lithologies of the dark anorthosite suite, all three white anorthosite samples display a slight
22 enrichment in the light REE and a relative decrease of the heavy REE, indicating either an
23 elevated amount of crustal contamination or a later alteration of the rocks.

24 The major and trace element data of the anorthosite suite permit comparison with large
25 Proterozoic anorthosite massifs of N America, like Harp Lake, Marcy, Nain, and Lac Saint-

1 Jean (see Ashwal, 1993, for a review; Fig. 5). When compared to the anorthosites investigated
2 in the Angolan part of the KIC (Morais et al., 1998), the Namibian rock suite is characterised
3 by a similar variability of geochemical compositions but higher average amounts of Na₂O.

4 5 *5.2. Felsic rock suite*

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7 The geochemical characteristics of the felsic rock suite have been investigated for dykes
8 transecting both the basement and the KIC. The SiO₂ contents of the felsic rocks exposed in
9 the basement range between 71-76 wt.%, whereas those of the dykes transecting the
10 anorthositic KIC are distinctly lower (51-66 wt.%). According to the classification of Cox et
11 al. (1979; Fig. 7), the granitic bodies intruding the basement can be classified as alkali
12 granites to granites, whereas the felsic dykes transecting the anorthositic KIC are syenites and
13 syenodiorites. The syenites/syenodiorites are either slightly quartz-normative or slightly
14 nepheline-normative. Quartz-normative syenite and syenodiorite samples can be classified as
15 jotunites to mangerites, whereas the granites resemble charnockites following the
16 nomenclature of Streckeisen (1974) for charnockitic rocks that is often adopted for felsic rock
17 suites associated with anorthosite massifs.

18 Samples of the felsic rock suite are generally enriched in K₂O (1.3-5.8 wt.%) and/or Na₂O
19 (3.6-8.3 wt.%) relative to CaO (<5.9 wt.%, except for Ku-98-204: 10.0 wt.% CaO). The data
20 for samples of the felsic rock suite form coherent arrays on major and trace element variation
21 diagrams (Fig. 4), involving a negative correlation of Si with Ti, Fe, Mn, Mg, Ca, and P. Al
22 and Na of the syenites and syenodiorites are positively correlated with Si, whereas Al and Na
23 of the alkali granites decrease with increasing Si. Syenites and syenodiorites affected by a
24 later alteration display higher Na₂O (7.2-10.5 wt.%) and lower K₂O (0.1-2.3 wt.%), reflected
25 by the high modal amounts of secondary albite in these samples. Moreover these samples

1 partly record higher amounts of the trace elements V, 17-181 ppm (unaffected: 11-76 ppm),
2 Zn, 32-532 ppm (unaffected: 25-201 ppm), Zr, 117-3326 ppm (unaffected: 49-2598 ppm),
3 Nb, 52-497 ppm (unaffected: 9-178 ppm), Sr, 173-1309 ppm (unaffected: 40-674 ppm), and
4 sums of the REE, 638-1424 ppm (unaffected: 325-399 ppm). An extremely high Sr value of
5 4514 ppm is recorded for the weakly altered but strongly deformed syenodiorite sample Ku-
6 98-204, that also contains high amounts of Al₂O₃ (23.4 wt.%) and CaO (10.0 wt.%).

7 The trends observed for the felsic rock suite generally differ from those obtained for the
8 anorthositic rocks. The ratio FeO^{tot}/(MgO+FeO^{tot}) is constantly high, ranging between 0.91-
9 0.96 in the granite and 0.80-0.99 in the syenite/syenodiorite (except sample Sy-18 with a
10 value of 0.71; Fig. 8a), thus plotting in the ferroan field of Frost et al. (2001). The syenites
11 and syenodiorites are alkalic in composition, whereas the alkali granites are alkali-calcic (Fig.
12 8b). Most of the felsic rock samples analysed are metaluminous to slightly peraluminous Fig.
13 8c). They generally contain abundant LIL (Rb, Ba, REE) and HFS elements (Zr, Y, Nb),
14 typical of A-type granites (Fig. 8d). In the discrimination diagrams Y vs. Nb and (Y+Nb) vs.
15 Rb of Pearce et al. (1984) most samples plot in the field of within-plate granites (Fig. 8e, f).
16 Chondrite-normalized REE plots of the felsic rock samples are almost identical (Fig. 6e).
17 La_N/Yb_N ratios of 5.9-60.1 indicate a variable but moderate LREE enrichment. With Eu/Eu*
18 ratios varying between 0.72 and 1.06, the rocks lack accentuated Eu anomalies. An extensive
19 plagioclase fractionation can thus be ruled out.

20 The major and trace element data of the granitoid suite of NW Namibia is similar to that
21 recorded by felsic rock suites associated with large Proterozoic anorthosite massifs like for
22 example the adamellite suite of Harp Lake (Emslie, 1980; see Ashwal, 1993, for a review;
23 Fig. 5).

24

1 **6. U-Pb Geochronology**

2

3 One weakly altered syenodiorite sample (Ku-98-201) was chosen for the U-Pb age
4 determination on zircon. The 80 to 150 μm long zircon crystals of this syenite are slightly
5 rounded at tips and edges and show irregular cores. The cores have weak or no
6 cathodoluminescence (CL), whereas the margins show strong CL emission. The contrasting
7 CL intensity suggests higher amounts of uranium and thorium in the cores (Poller et al.,
8 2001). The zircon samples have uranium contents between 135 and 172 ppm and total Pb
9 contents between 41 and 48 ppm. Their $\text{Th}/\text{U}_{\text{atomic}}$, as calculated from $(^{208}\text{Pb}/^{206}\text{Pb})_{\text{radiogenic}}$ and
10 the age, ranges from 0.87 to 0.93, which is typical for mafic and mantle-derived magmatic
11 rocks (e.g., Heaman et al., 1990). Four multi-grain short-prismatic, pink to pale-brown zircon
12 samples of igneous origin (Table 6) yield slightly discordant data that define an upper
13 concordia intercept at 1376 ± 2 Ma (2σ , MSWD = 0.4; forced through 0 Ma; Fig. 9), which is
14 identical with the weighted $^{207}\text{Pb}/^{206}\text{Pb}$ average age of 1376 ± 2 Ma. The low-U margins
15 apparently had no effect on the age determination irrespective of the timing of their formation,
16 i.e. late-magmatic or during the late hydrothermal alteration. The age of 1376 ± 2 Ma (2σ) is
17 interpreted to date the emplacement of the felsic rock suite. It coincides with the concordant
18 U/Pb single-zircon age of 1371 ± 2.5 Ma determined for a mangerite vein of the Angolan part
19 of the KIC (Mayer et al., 2004) and the less precise U/Pb single-zircon age of 1385 ± 25 Ma,
20 obtained for a leucogabbro sample from Namibia (Drüppel et al., 2000).

21

22 **7. Isotope characteristics**

23

24 *7.1. Oxygen isotopic data*

25

1 The utility of stable isotopes as monitors for the source rocks and processes like crustal
2 assimilation and hydrothermal alteration of anorthosites and related rocks has been
3 demonstrated in various studies (Taylor, 1968; Valley and O'Neill, 1982, Morrison and
4 Valley, 1988; O'Connor and Morrison, 1999; Peck and Valley, 2000). Since both the
5 anorthosite and the felsic rock suite are dominantly composed of feldspar, the obtained
6 plagioclase and alkali feldspar $\delta^{18}\text{O}$ values are a good proxy for the bulk rock $\delta^{18}\text{O}$. For
7 analyses of the oxygen isotopic composition of the anorthositic suite, plagioclase was
8 separated from 8 samples of the dark anorthosite and 6 samples of the white anorthosite.
9 Oxygen isotopic compositions of the felsic suite were determined for hand-picked K-feldspar
10 and plagioclase from one quartz-syenite and 4 syenite samples. Analytical results are listed in
11 Table 7 and shown in Fig. 10.

12

13 *7.1.1. Anorthosite suite*

14

15 Most plagioclase separates from rock samples of the dark anorthosite suite exhibit a
16 narrow range in $\delta^{18}\text{O}$ (5.61-6.13 ‰), in accordance with typical values of 5.8-7.6 ‰ for
17 magmatic anorthosite from occurrences world-wide (e.g. Ashwal, 1993). The obtained $\delta^{18}\text{O}$
18 values are similar to those from typical basalts and gabbros and are thus consistent with
19 uncontaminated mantle-derived parental melts. Just one leucotroctolite sample (Ku-97-105),
20 taken from a fault zone at the contact between the Epupa Complex and the Kunene Intrusive
21 Complex has a considerably lower plagioclase $\delta^{18}\text{O}$ of 3.19 ‰. Extremely low $\delta^{18}\text{O}$ values of
22 down to -6 ‰ have been reported for the Boehls Butte (Idaho) and Bitterroot (Montana)
23 anorthosite complexes (Mora and Valley, 1988) and have been interpreted to be due to
24 subsolidus alteration of these rocks. For the KIC, these low values could also be related to an
25 exchange with meteoric or hydrothermal fluids. In addition, the low $\delta^{18}\text{O}$ values of the dark

1 anorthosite samples might be attributed to the abundance of very fine grained Fe-Ti oxide
2 inclusions in plagioclase of these samples that might have caused a shift of the O isotope
3 ratios to lower values.

4 Three of five samples of the white anorthosite display plagioclase $\delta^{18}\text{O}$ values of 5.35-
5 6.14, i.e. in the range of unaltered plagioclase from the dark anorthosite. Thus, a pervasive
6 alteration by circulating crustal fluids and/or meteoric water seems rather unlikely. A more
7 convincing explanation is that the alteration of the white anorthosite results from an
8 interaction with fluids released by younger anorthositic melts, since fluids like these would
9 have no effects on the isotopic composition of plagioclase. This interpretation is in agreement
10 with the preservation of the magmatic REE patterns in white anorthosite. A conspicuously
11 high $\delta^{18}\text{O}$ value of 7.30 has been determined for plagioclase from a white anorthosite sample,
12 which is $\sim 1.4\text{‰}$ higher than the values found in the 'normal', magmatic plagioclase of the
13 white and dark anorthosite. This high $\delta^{18}\text{O}$ value presumably reflects contamination with
14 crustal calc-silicate material of this rock sample. Another white anorthosite sample, taken
15 from a major fault zone, exhibits a lower $\delta^{18}\text{O}$ value of 2.36, which agrees well with the one
16 obtained for the fault-related leucotroctolite sample. In conclusion, the intermediate
17 plagioclase $\delta^{18}\text{O}$ values (average: $5.88 \pm 0.19\text{‰}$ $\delta^{18}\text{O}$) of rocks from both the dark and the
18 white anorthosite suite are interpreted to be magmatic in origin. Lower plagioclase $\delta^{18}\text{O}$
19 values (2.36-3.19 ‰ $\delta^{18}\text{O}$) seem to have resulted from the influx of meteoric waters restricted
20 to faults and the higher $\delta^{18}\text{O}$ value (7.30 ‰ $\delta^{18}\text{O}$) seem to have been caused by crustal
21 contamination.

22

23 *7.1.2. Felsic rock suite*

24

1 The $\delta^{18}\text{O}$ values of four alkali feldspar separates from both the syenites and
2 syenodiorites fall in a restricted range of 7.22-7.92 ‰ (mean: 7.50 ± 0.26 ‰). Almost
3 identical values have been obtained for plagioclase separates of the respective samples (7.20-
4 7.52 ‰; mean: 7.36 ± 0.16 ‰), suggesting equilibrium crystallization of the two feldspars.
5 The obtained values are interpreted to be magmatic, since the small shift of $\delta^{18}\text{O}$ values and
6 the coincidence of the $\delta^{18}\text{O}$ values of K-feldspar and plagioclase in the felsic rocks are
7 inconsistent with low-temperature alteration. It has to be mentioned, however, that the
8 average $\delta^{18}\text{O}$ value of feldspar from the felsic rock suite is about 1.6 ‰ higher than the
9 average plagioclase $\delta^{18}\text{O}$ of the anorthosites (Fig. 10). Although there is a general increase in
10 $\delta^{18}\text{O}$ with increasing SiO_2 during closed-system crystal fractionation, since silica tends to
11 concentrate the heavy oxygen isotope (Taylor and Sheppard, 1986), this effect is significantly
12 smaller than the observed contrast in $\delta^{18}\text{O}$ between the felsic and anorthositic suites of the
13 Swartbooisdrif area.

14

15 7.2. Sr and Nd isotopic data

16

17 Anorthosites and the associated felsic rocks, which locally occur as dykes within the
18 anorthosites, show Sr(*T*) and Nd(*T*) isotopic compositions that fall into two groups, with
19 overlapping $\epsilon\text{Nd}_{(T)}$ values of +3.0 to -0.4 (Table 8). The $^{87}\text{Sr}/^{86}\text{Sr}_{(T)}$ values of the anorthosites
20 define a narrow range of 0.70284 to 0.70413. This seems to be real as all samples have low
21 $^{87}\text{Rb}/^{86}\text{Sr}$ values and high Sr contents, which resulted in only very small corrections for in situ
22 ^{87}Sr growth (< 0.0015) and made the Sr isotopic composition little sensitive to later Sr
23 additions. The highest $^{87}\text{Sr}/^{86}\text{Sr}_{(T)}$ values of 0.70385 and 0.70413 are recorded by the white
24 anorthosites, which underwent both crustal contamination and later alteration. In contrast, the
25 syenites/syenodiorites define a broader range in $^{87}\text{Sr}/^{86}\text{Sr}_{(T)}$ than the anorthosites, i.e., 0.70245

1 to 0.70761, whereby the relatively radiogenic $^{87}\text{Sr}/^{86}\text{Sr}_{(\text{T})}$ values derive from samples with
2 high Sr contents and low $^{87}\text{Rb}/^{86}\text{Sr}$ values. The highest $^{87}\text{Sr}/^{86}\text{Sr}_{(\text{T})}$ ratios of 0.70718, 0.70761
3 and 0.70627 are recorded by the felsic rock samples Ku-98-201 (197 ppm Sr), Ku-98-204
4 (4514 ppm Sr), and Ku-98-200 (1301 ppm Sr), that underwent a variably severe secondary
5 alteration and/or deformation. Samples with less radiogenic $^{87}\text{Sr}/^{86}\text{Sr}_{(\text{T})}$ ratios than those of the
6 anorthosites are characterized by both low Sr contents and relatively high $^{87}\text{Rb}/^{86}\text{Sr}$ ratios.
7 Thus, the low $^{87}\text{Sr}/^{86}\text{Sr}_{(\text{T})}$ values are bound to samples that are sensitive to Rb and Sr addition,
8 which would result in a direct shift of the isotope ratios and indirectly – for $\text{Rb}_{\text{addition}} > \text{Sr}_{\text{addition}}$
9 – in an overcorrection of ^{87}Sr growth for the time before Rb and Sr addition. Hence, it is
10 possible that the felsic rocks are derived from a source that had more radiogenic $^{87}\text{Sr}/^{86}\text{Sr}_{(\text{T})}$
11 values than the anorthosites.

12 In comparison with other geochemical reservoirs present at 1.38 Ga, the Nd and Sr
13 isotopic composition of the anorthosites and syenites have distinctly more radiogenic Nd
14 isotopic compositions and typically less radiogenic Sr isotopic composition than the northern
15 Namibian crust (Seth et al., 2005) and fall at the low- $^{143}\text{Nd}/^{144}\text{Nd}$ end of the field defined by
16 sub-continental lithospheric mantle and depleted mantle (Fig. 11). Regarding their $\epsilon\text{Nd}_{(\text{T})}$
17 values of +3.0 to + 1.1, the anorthosites from the Namibian part of KIC strongly differ from
18 two anorthosite samples of the Angolan KIC, for which distinctly lower $\epsilon\text{Nd}_{(\text{T})}$ values of -5.72
19 and -6.36 have been calculated by Slejko et al. (2002). On the other hand, the $^{87}\text{Sr}/^{86}\text{Sr}_{(\text{T})}$
20 ratios of 0.7038 to 0.7045 constrained for the Angolan anorthosites by Morais et al. (1998) are
21 in the range of the values of the Namibian samples presented in this study (i.e. 0.7028 to
22 0.7041).

23

24 **8. Summary and conclusions**

25

1 8.1. Mineralogical and geochemical characteristics of the anorthosites and related rocks

2

3 The part of the KIC exposed in NW Namibia mainly comprises pale coloured, massive,
4 white anorthosites (i.e., anorthosites *sensu stricto* and leuconorites) that are characterised by a
5 strong deformation and alteration, i.e., sericitisation, saussuritisation, and albitisation of
6 plagioclase and chloritization of Fe-Mg silicates. The white anorthosite massif is intruded by
7 massive, moderately altered dark anorthositic rocks, comprising anorthosites,
8 leucogabbronorites and leucotroctolites.

9 There is a small volume of associated felsic rocks, whereas gabbroic rocks are rare. Felsic
10 rocks transecting the metapelitic rocks and orthogneisses of the Orue Unit are alkali granites
11 and always more SiO₂-rich than the felsic rocks emplaced within the KIC, i.e., syenites and
12 syenodiorites, suggesting that the composition of the melt parental to the felsic rock suite is
13 modified by a veritable contamination by the respective bordering country rock. Moreover
14 part of the felsic rock samples, especially those exposed as dykes within the KIC, show a
15 severe late hydrothermal alteration that caused their transformation into albitites. The U-Pb
16 zircon age of 1376 ± 2 Ma, determined for an altered syenodiorite intruding the KIC, is
17 interpreted to date the emplacement of the felsic rock suite. It coincides with the concordant
18 U-Pb single-zircon age of 1371 ± 2.5 Ma determined for a mangerite vein of the Angolan part
19 of the KIC (Mayer et al., 2004) and the less precise U-Pb single-zircon age of 1385 ± 25 Ma,
20 obtained for a leucogabbronorite sample from Namibia (Drüppel et al., 2000), suggesting that
21 the anorthosite and the felsic rock suites are formed during the same igneous event. The age
22 of the KIC and the respective compositions of the anorthositic and the felsic rock suites
23 resemble that of other large Proterozoic anorthosite massifs worldwide like, for example,
24 Harp Lake, Marcy, and Nain (see Ashwal, 1993, for a review). However, in clear contrast to

1 many other Proterozoic massif-type anorthosites, the KIC experienced no prograde
2 metamorphic overprint after its emplacement at 1.38 Ga.

3 The early emplaced white anorthosite as well as leucogabbro and pyroxene-bearing
4 anorthosite of the dark anorthosite suite generally display higher Fe/Mg ratios of the Fe-Mg
5 silicates and higher absolute amounts of late-magmatic biotite and amphibole and of
6 accessory apatite when compared to the olivine-bearing anorthosites and leucotroctolites of
7 the dark anorthosite suite. Moreover, both units frequently display crustal xenoliths of calc-
8 silicate material. The highest Sr concentrations have been obtained for the white anorthosite
9 and the pyroxene-bearing anorthosite/leucogabbro of the dark anorthosite suite, which,
10 at the same time, exhibit the highest overall trace element abundances. Lowest average Sr
11 concentrations (470-640 ppm) are recorded for the olivine-bearing anorthosites and
12 leucotroctolites, corresponding with the highest average An-contents of plagioclase and
13 lowest total sums of incompatible trace elements. An inverse correlation between Sr and the
14 An-content is a common feature of layered intrusions and is generally attributed to plagioclase
15 fractionation, causing a decrease of the Sr content in the melt (see Ashwal, 1993, for a
16 review). On the other hand it has been demonstrated by Duchesne (1978) for the layered
17 Bjerkrem-Sogndal bodies of Norway that plagioclase fractionation may cause an increase of
18 Sr in the melt. In the case of the KIC, however, this trend is more likely caused by contrasting
19 contamination with calc-silicate crustal material. Younger anorthosite melts may have used
20 already existing pathways through the crust provided by the previously intruded anorthositic
21 magma, i.e., ascending mainly through anorthositic country rock. In case of the white
22 anorthosite, however, the trace element patterns may additionally be obscured by later
23 secondary processes like the severe hydrothermal alteration and deformation recorded by
24 most samples.

25

1 8.2. Origin of the anorthositic and felsic rock suites

2

3 The respective mineral compositions of the granite-syenite suite appear to extend the
4 compositional range of the anorthositic rocks towards higher Fe/Mg ratios and lower Al of the
5 Fe-Mg silicates (pyroxene, amphibole, epidote and chlorite) and lower An contents of
6 plagioclase. The relative changes in the mineral compositions could provide evidence for
7 consanguinity between the two rock suites. However, intermediate mineral compositions,
8 bridging the gap between those of the anorthositic and the felsic rock suite, are missing. Like
9 the Harp Lake massif, Labrador, the Namibian felsic and anorthositic rock suites are
10 characterized by chemical discontinuities, reflecting the lack of rock types of intermediate
11 chemical composition (Fig. 5), which is a common feature of unmetamorphosed anorthosite
12 massifs (Ashwal, 1993). The difference in the magmatic $\delta^{18}\text{O}$ values of feldspar from
13 granitoids and anorthositic rocks also supports a non-comagmatic relationship of the two
14 suites. The rather homogeneous but high $\delta^{18}\text{O}$ values of feldspar from the felsic rock suite
15 ($7.36 \pm 0.16 \text{‰}$) when compared to those of the anorthositic rock suite ($5.88 \pm 0.19 \text{‰}$ $\delta^{18}\text{O}$)
16 point to a crustal source for the felsic suite. On the other hand, the values obtained for
17 plagioclase of the anorthosite suites are consistent with uncontaminated mantle-derived
18 parental melts. The contrast in $^{87}\text{Sr}/^{86}\text{Sr}_{\text{(T)}}$ ratios between anorthositic (0.7028-0.7041) and the
19 felsic rock suites (0.7025-0.7076) also suggest a different source of the two units, in line with
20 $\delta^{18}\text{O}$ values. If the variable Sr, Nd, and O isotopic signatures of both the anorthositic and
21 felsic rocks are due to the assimilation of crustal material, the isotopic composition of Sr and
22 O were more affected than the isotopic composition of Nd. Such a Sr-Nd-O decoupling
23 indicates that the assimilated rocks should have had relatively high Sr contents, but low Nd
24 contents, which is typically found for marine carbonates, but not in silicic crustal material.
25 Assimilation of carbonates is a particularly attractive explanation as there locally occur

1 abundant xenoliths of crustal calc-silicate material and carbonate assimilation not only
2 accounts for the elevated $^{87}\text{Sr}/^{86}\text{Sr}$, but also for the heavy $\delta^{18}\text{O}$ values. In case of the felsic
3 rocks this trend is rather caused by the late hydrothermal alteration of these rocks. In
4 conclusion, our dataset demonstrates that the anorthositic lithologies and the felsic rock suite
5 represent two chemically independent suites. The source of the anorthositic rocks is a mantle-
6 derived liquid that, during their uprise, was modified by minor crustal contamination. Hereby
7 crustal contamination, as evident from the abundance of crustal xenoliths, low $\text{Ca}/(\text{Ca}+\text{Na})$
8 ratios, and high amounts of late-magmatic biotite, amphibole and apatite, is predominantly
9 displayed by the early emplaced white anorthosite and the leucogabbro of the dark
10 anorthosite suite. In contrast, leucotroctolites and olivine anorthosites are characterized by
11 high An-contents of plagioclase and a mostly unhydrous mineralogy; late-magmatic biotite
12 and amphibole are rare. Melts parental to the felsic rocks presumably originate by anatexis of
13 the lower crust or at least experienced a stronger crustal contamination than the anorthosites,
14 either in the magma chamber or during their ascent. Granite bodies emplaced in the Epupa
15 Complex were additionally affected by contamination of the granitic magma with
16 metasedimentary and metavolcanic crustal material during their uprise.

17 The ϵNd values of the anorthositic rocks, calculated back to their emplacement age of
18 1385 Ma given by U-Pb zircon data (Drüppel et al., 2000), range between 1.1 and 3.0,
19 indicating a time-integrated weakly depleted character of the parental magma with respect to
20 CHUR. Corresponding Nd model ages of 2.01-1.61 Ga, calculated with respect to a depleted
21 mantle curve, are significantly older than the crystallization age of the anorthosites. Model
22 ages older than the intrusion ages have also been reported for several anorthosite massifs and
23 are either ascribed to crustal contamination effects (e.g. Ashwal, 1993; Ashwal et al., 1998),
24 in which case the model ages lack geological significance (cf. Arndt & Goldstein, 1987), or to
25 a lower crustal origin of the anorthosites (e.g. Schiellerup et al., 2000; Wiszniewska et al.,

1 2002). The hypothesis that anorthosites originate in the lower crust has found increasing
2 support in recent years (Taylor et al., 1984; Duchesne et al., 1999; Longhi et al., 1999;
3 Schiellerup et al., 2000; Selbekk et al., 2000, Wiszniewska et al., 2002), based on Re-Os
4 isotope data and experimental investigations. In case of the KIC, the geochemical
5 characteristics as well as O, Sr, and Nd isotope data indicate a mantle origin of the
6 anorthosites and hence suggest that the magma parental to the anorthosites formed by partial
7 melting of the upper mantle. The same signatures, however, may have been preserved also
8 during lower crustal melting of a previously emplaced mantle-derived mafic rock under dry
9 conditions. Indeed, igneous protolith ages of 2.0-1.7 Ga, calculated for granulite and
10 amphibolite facies mafic rocks of the Epupa Complex (Seth et al., 2005), would fit in this
11 model, suggesting the following tectono-metamorphic scenario: Emplacement of a large
12 mafic bodies at ca. 2.0-1.7 Ga, subsequent granulite facies metamorphism of the mafic rocks
13 metamorphism at 1.52 Ga and finally partial melting with formation of the anorthositic
14 magma at 1.38 Ga. A scenario like this has been invoked by Schiellerup et al. (2000) for the
15 Rogaland anorthosites, Norway, and by Wiszniewska et al. (2002) for the Mazury complex,
16 Poland. In case of the KIC, however, this alternative model seems rather unlikely. With an
17 estimated surface area of ca. 18.000 km² the KIC has an enormous size when compared to the
18 anorthosites from Norway and Poland. Moreover the KIC is characterized by an extraordinary
19 compositional and textural homogeneity, being almost entirely composed of anorthosite,
20 leucogabbro, and leucotroctolite with high-Ca plagioclase, whereas related gabbroic or
21 felsic lithologies are rare. If the anorthositic magmas of the KIC originated by partial melting
22 of a lower continental mafic crust or foundered mafic plutons, these mafic source regions
23 must be of unrealistically large and homogeneous in composition. Moreover, the production
24 of such large volumes of anorthositic magma by partial melting would presumably lead to a
25 gravitational collapse of the lower crustal source regions. Therefore, we interpret the parental

1 magma of the anorthosites of the Kunene Intrusive Complex to be derived from partial
2 melting of the upper mantle with anatexis of crustal material leading to the formation of the
3 associated felsic rocks.

4

5 *8.3. Comparison to other parts of the Kunene Intrusive Complex*

6

7 Due to difficult access during the civil war, most studies of the northern part of the KIC
8 are based on old field data and sampling (Carvalho & Alves, 1990, 1993; Silva, 1990, 1992).
9 Recent detailed field work in combination with petrographical investigations is lacking for
10 most of the Angolan part of the KIC, except restricted areas around Lufinda, Dongue and
11 Chiange in Angola (Ashwal & Twist, 1994; Morais et al., 1998; Slejko et al., 2002; Mayer et
12 al., 2004; Fig. 1). Based on satellite imagery Ashwal & Twist (1994) suggested that the KIC
13 represents a composite massif-type anorthosite, composed of at least eight distinct anorthosite
14 bodies with probably individual petrogenetic evolution.

15 Regarding their geochemical composition the Namibian rock suite is characterised by a
16 similar variability of geochemical compositions and higher average amounts of Na₂O when
17 compared to the anorthosites from the Angolan part of the KIC (Silva, 1990, 1992; Morais et
18 al., 1998; Fig. 12). Based on the mineral chemistry data and isotope characteristics of the
19 anorthosites from Angola a northern (average An-content of plagioclase: 70 ± 5 mol.%;
20 ⁸⁷Sr/⁸⁶Sr_(T): 0.7038; εNd: -5.72) and a southern zone (average An-content of plagioclase: 53 ±
21 5 mol.%; ⁸⁷Sr/⁸⁶Sr_(T): 0.7045; εNd: -6.36) have been distinguished, that are separated by a belt
22 of granitoid intrusions (Morais et al., 1998; Slejko et al., 2002). Likewise the Namibian KIC
23 can be subdivided into two units, the white anorthosite (An-content of plagioclase: 43-53
24 mol.%; ⁸⁷Sr/⁸⁶Sr_(T): 0.7039-0.70413; εNd: 1.1-2.0) and the dark anorthosite massif (An-
25 content of plagioclase: 43-75 mol.%; ⁸⁷Sr/⁸⁶Sr_(T): 0.7028-0.7036; εNd: 1.7-3.0). Both rock

1 units presumably represent individual anorthosite intrusions that underwent a distinct
2 petrogenetic evolution. These are marked by a contrasting contamination with crustal material
3 and also strongly differ from the two anorthosite bodies identified in Angola.

4 The petrographical features of the Namibian leucotroctolite, leucogabbronorite, and
5 anorthosite samples of the dark anorthosite suite investigated in this study resemble those
6 reported for the northern, Angolan, part of the KIC (Silva, 1990, 1992; Ashwal & Twist,
7 1994; Morais et al., 1998; Slejko et al., 2002). However, subsolidus reaction textures like
8 garnet-bearing corona structures of the dark anorthosites from NW Namibia (Drüppel et al.,
9 2001) have not yet been observed. In the Lufinda-Dongue area, the presence of cogenetic
10 dolerites and mangerites is interpreted in favour of a shallow emplacement level of the
11 anorthosite (Morais et al., 1998; Slejko et al., 2002). Pressures of 3-5 kbar have been
12 estimated by Slejko et al. (2002) for the subsolidus re-equilibration of orthopyroxene-
13 clinopyroxene pairs. In contrast, distinctly higher pressures of 7.3 ± 0.9 kbar and of 6.5 ± 0.6
14 kbar were constrained for the subsolidus formation of garnet-bearing corona structures in the
15 Namibian leucotroctolites (Drüppel et al., 2001; Drüppel, 2003) and the magmatic
16 crystallization of hastingsite in the associated felsic rocks (Drüppel, 2003), respectively.
17 Similar pressures of 6.5 ± 0.6 kbar were constrained for a contact-metamorphic hornfels of
18 the Orue Unit by Brandt (2003). As is evident from Fig. 1, however, the Angolan and
19 Namibian study areas are at least 150 km apart from each other and detailed field work as
20 well as mineralogical investigations of the anorthosites in the intermediate southern Angolan
21 part of the KIC are lacking so far. Future investigations will help to decide whether the
22 apparent differences in the erosion level of ca. 9 km is due to a continuous gradient in
23 exposed crustal depth or due to offsets by deep fracture zones as for example the Serpa-Pinto
24 lineament or the Kunene River fault.

1 The emplacement of the KIC and the associated granitic rocks at the southern margin of
2 the Congo Craton marks the beginning of widespread Mesoproterozoic igneous activity in
3 southern Africa. The Mesoproterozoic KIC is possibly related to the Mesoproterozoic Kibaran
4 Belt of the Democratic Republic of Congo and Tanzania, that is characterised by voluminous
5 Mesoproterozoic mafic and A-type granite magmatism at 1383-1371 Ma (Tack *et al.*, 2002;
6 see Johnson *et al.*, 2005, for a review). This age interval resembles that constrained for the
7 anorthosites of the KIC and related rocks, however, intermediate zones between the KIC and
8 the Kibaran belt are not exposed. If the KIC and the Kibaran belt were related in the
9 Mesoproterozoic, they might have been part of a major northeast striking extensional belt that
10 formed during the early stages of separation of the Congo and the southern Kapvaal-
11 Zimbabwe cratons. On the other hand, the KIC may also represent a postorogenic pluton, with
12 the mantle and crustal melts being formed by lithospheric delamination (Black and Liégeois,
13 1993) during a post-orogenic collapse that followed widespread Palaeoproterozoic orogenic
14 activity in central and southern Africa at 2.2-1.6 Ga, the so-called Eburnian Orogeny (e.g.
15 Cahen *et al.*, 1984; Carvalho *et al.*, 2000).

16

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4

5 **References**

6

7 Anderson, J.L., 1980. Mineral equilibria and crystallization conditions in the late Precambrian
8 Wolf River rapakivi massif, Wisconsin. *Am. J. Sci.* 280, 289-332.

9 Anderson, J.L., Bander, E.E., 1989. Nature and origin of Proterozoic A-type granitic
10 magmatism in the southwestern United States of America. *Lithos* 23, 19-52.

11 Arndt, N.T., Goldstein, S.L., 1987. Use and abuse of crust-formation ages. *Geology* 15, 893-
12 895.

13 Ashwal, L.D., 1993. Anorthosites. *Minerals and Rocks Series* 21. Springer-Verlag, Berlin.

14 Ashwal, L.D., Hamilton, M.A., Morel, V.P.I., Rambeloson, R.A., 1998. Geology, petrology
15 and isotope geochemistry of massif-type anorthosites from southwest Madagascar.
16 *Contrib. Miner. Petrol.* 133, 389-401.

17 Ashwal, L.D., Twist, D., 1994. The Kunene complex, Angola/Namibia: a composite massif-
18 type anorthosite complex. *Geol. Mag.* 131, 579-591.

19 Black, R., Liégeois, J., 1993. Cratons, mobile belts, alkaline rocks and continental lithospheric
20 mantle: the Pan-African testimony. *Journal of the Geological Society* 150, 89-98.

21 Borthwick, J., Harmon, R.S., 1982. A note regarding ClF₃ as an alternative to BrF₅ for
22 oxygen isotope analysis. *Geochim. Cosmoch. Acta* 46, 1665-1668.

23 Brandt, S., 2003. Metamorphic evolution of ultrahigh-temperature granulite facies and upper
24 amphibolite facies rocks of the Epupa Complex, NW Namibia. Dr. rer. nat. thesis,
25 University of Würzburg

- 1 Brandt, S., Klemm, R., Okrusch, M., 2003. Ultrahigh-temperature metamorphism and
2 multistage evolution of garnet-orthopyroxene granulites from the Proterozoic Epupa
3 Complex, NW Namibia. *J. Petrol.* 44, 1121-1144.
- 4 Brandt, S., Will, T.M., Klemm, R., 2007. Ultrahigh-temperature metamorphism and
5 anticlockwise PT paths of sapphirine-bearing orthopyroxene-sillimanite gneisses from the
6 Proterozoic Epupa Complex, NW Namibia. *Precamb. Res.* (in press).
- 7 Cahen, L., Snelling, N.J., Delhal, J., Vail, J.R., Bonhomme, M., Ledent, D., 1984. The
8 Geochronology and evolution of Africa. Oxford University Press, Oxford.
- 9 Carvalho, H. de, Alves, P., 1990. Gabbro-Anorthosite Complex of SW Angola/ NW Namibia.
10 *Inst. Inv. Científica Tropical, Série de Ciências da Terra, Comunicações* 2, 1-66.
- 11 Carvalho, H. de, Alves, P., 1993. The Precambrian of SW Angola and NW Namibia. Garcia
12 de Orta, *Série de Ciências da Terra, Comunicações* 4, 1-38.
- 13 Carvalho, H. de, Tassinari, C., Alves, P.H., Guimarães, F., Simões, M.C., 2000.
14 Geochronological review of the Precambrian in western Angola: links with Brazil. *J. Afr.*
15 *Earth Sci.* 31, 383-402.
- 16 de Waard, D., 1970. The anorthosite-charnockite suite of rocks of Roaring Brook Valley in
17 the eastern Adirondacks (Marcy Massif). *Am. Miner.* 55, 2063-2075.
- 18 Cox, K.G., Bell, J.D., Pankhurst, R.J., 1979. The interpretation of igneous rocks. George
19 Allen & Unwin Ltd, London.
- 20 Drüppel, K., 1999. Petrologie und Geochemie von Anorthositen des Kunene-Intrusiv-
21 Komplexes, NW Namibia. Diploma thesis, University of Würzburg.
- 22 Drüppel, K., 2003. Petrogenesis of the Mesoproterozoic anorthosite, syenite and carbonatite
23 suites of NW Namibia and their contribution to the metasomatic formation of the
24 Swartbooisdrif sodalite deposits. Dr. rer. nat. thesis, University of Würzburg.

- 1 Drüppel, K., Littmann, S., Okrusch, M., 2000. Geo- und isotopechemische Untersuchungen
2 der Anorthosite des Kunene-Intrusiv-Komplexes (KIC) in NW-Namibia. Berichte der
3 Deutschen Mineralogischen Gesellschaft, Beiheft zum Eur. J. Miner. 12, 37.
- 4 Drüppel, K., Hoefs, J., Okrusch, M., 2005. Fenitizing processes induced by ferrocarnatite
5 magmatism at Swartbooisdrif, NW Namibia. J. Petrol. 46, 377-406.
- 6 Drüppel, K., von Seckendorff, V., Okrusch, M., 2001. Subsolidus reaction textures in
7 anorthosites of the Kunene Intrusive Complex, NW Namibia. Eur. J. Miner. 13, 289-309.
- 8 Duchesne, J.-C., 1978. Quantitative modelling of Sr, Ca, Rb and K in the Bjerkrem-Sogndal
9 layered lopolith (S.W. Norway). Contrib. Miner. Petrol. 66, 175-184.
- 10 Duchesne, J.-C., 1984. Massif anorthosites: another partisan review. In: Brown W.L. (Ed.),
11 Feldspars and feldspathoids. Reidel, Dordrecht, 411-433.
- 12 Duchesne, J.C., Liégeois, J.P., Vander Auwera, J., Longhi, J., 1999. The crustal tongue
13 melting model and the origin of massive anorthosites. Terra Nova 11, 100-105.
- 14 Dulski, P., 1994. Interferences of oxide, hydroxide and chloride analyte species in the
15 determination of rare earth elements in geological samples by inductively coupled plasma-
16 mass spectrometry. Fresenius J. Analytical Chemistry 350,194-203.
- 17 Emslie, R.F., 1980. Geology and petrology of the Harp Lake Complex, Central Labrador: an
18 example of Elsonian magmatism. Geological Survey of Canada Bulletin 293, 1-136.
- 19 Emslie, R.F., 1985. Proterozoic anorthosite massifs. In: Tobi, A., Touret, J.L.R. (Eds.), The
20 Deep Proterozoic Crust in the North Atlantic provinces. NATO ASI Series C 158, Reidel,
21 Dordrecht, 39-60.
- 22 Emslie, R.F., 1991. Granitoids of rapakivi granite-anorthosite and related associations.
23 Precamb. Res. 51, 173-192.

- 1 Emslie, R.F., Hamilton, M.A., Therault, R.J., 1994. Petrogenesis of mid-Proterozoic
2 anorthosite-mangerite-charnockite-granite, AMCG complex: isotopic and chemical
3 evidence from the Nain Plutonic Suite. *J. Geol.* 102, 539-558.
- 4 Frost, B.R., Barnes, C.G., Collins, W.J., Arculus, R.J., Ellis, D.J., Frost, C.D., 2001. A
5 geochemical classification for granitic rocks. *J. Petrol.* 42, 2033-2048.
- 6 Fuhrmann, M.L., Frost, B.R., Lindsley, D.H., 1988. Crystallization conditions of the Sybille
7 Monzosyenite, Laramie Anorthosite Complex, Wyoming. *J. Petrol.* 29, 699-729.
- 8 Gerstenberger, H., Haase, G., 1997. A highly effective emitter substance for mass
9 spectrometric Pb isotope ratio determinations. *Chem. Geol.* 136, 309–312.
- 10 Heaman, L.M., Bowins, R., Crocket, J., 1990. The chemical composition of igneous zircon
11 suites: Implications for geochemical tracer studies. *Geochim. Cosmoch. Acta* 55, 1597-
12 1607.
- 13 Jacobsen, S.B., Wasserburg, G.J., 1979. Nd and Sr isotopic study of the Bay of Islands
14 ophiolite complex and the evolution of the source of midocean ridge basalts. *J. Geophys.*
15 *Res.* 84, 7429-7445.
- 16 Jaffey, A.H., Flynn, K.F., Glendenin, L.E., Bentley, W.C., Essling, A.M., 1971. Precision
17 measurement of half-lives and specific activities of ^{235}U and ^{238}U . *Physics Reviews C*4,
18 1889–1906.
- 19 Johnson, S.P., Rivers, T., De Waele, B., 2005. A review of the Mesoproterozoic to early
20 Palaeozoic magmatic and tectonothermal history of south-central Africa: implications for
21 Rodinia and Gondwana. *J. Geol. Soc. London* 162, 433-450.
- 22 Köstlin, E.C., 1967. The geology of part of the Kunene basic complex, Kaokoveld, South
23 West Africa. M.Sc. Thesis, University of Cape Town.

- 1 Köstlin, E.C., 1974. The Kunene basic complex, northern South West Africa. In: Kröner, A.
2 (Ed.), Contributions to the Precambrian geology of southern Africa. University of Cape
3 Town Bulletin 15, Capetown, 123-135.
- 4 Krogh, T.E., 1973. A low contamination method for hydrothermal decomposition of zircon
5 and extraction of U and Pb for isotopic age determinations. *Geochim. Cosmoch. Acta* 37,
6 485-494.
- 7 Littmann, S., Romer, R.L., Okrusch, M., 2000. Nephelinsyenite der Epembe-Swartbooisdrif-
8 Alkali-Provinz (ESAP)/ NW Namibia. *Berichte der deutschen Mineralogischen*
9 *Gesellschaft, Beihefte zum Eur. J. Miner.* 12: 115.
- 10 Longhi, J., Ashwal, L.D., 1985. Two-stage models for lunar and terrestrial anorthosites:
11 Petrogenesis without a magma ocean. *J. Geophys. Res.* 90, C571-C584.
- 12 Longhi, J., Vander Auwera, J., Fram, M.S., Duchesse, J.-C., 1999. Some Phase Equilibrium
13 Constraints on the Origin of Proterozoic (Massif) Anorthosites and Related Rocks. *J.*
14 *Petrol.* 40, 339-362.
- 15 Mayer, A., Sinigoi, S., Miguel, L.G., Morais, E., Petrini, R., 2000. Kibaran ages in the
16 Kunene anorthositic complex. *GEOLUANDA 2000*, Abstract volume, 106.
- 17 Mayer, A., Hofmann, A.W., Sinigoi, S., Morais, E., 2004. Mesoproterozoic Sm–Nd and U–Pb
18 ages for the Kunene Anorthosite Complex of SW Angola. *Precambr. Res.* 133, 187-206.
- 19 Menge, G.F.W., 1996. The eastern portion of the Kunene complex, its satellite intrusions and
20 the alkaline suite between Epembe and Swartbooisdrif. Unpublished Report, University of
21 Cape Town.
- 22 Menge, G.F.W., 1998. The antiformal structure and general aspects of the Kunene Complex.
23 *Z. Dt. Geol. Ges.* 149, 431-448.

- 1 Mitchell J.N., Scoates J.S., Frost C.D., 1995. High-Al gabbros in the Laramie anorthosite
2 complex, Wyoming: implications for the composition of melts parental to Proterozoic
3 anorthosites. *Contrib. Miner. Petrol.* 119, 166-180.
- 4 Mora, C.I., Valley, J.W., 1988. Channelized meteoric-hydrothermal alteration of the Boehls
5 Butte and Bitterroot anorthosites, northern ID and MT. *Eos* 69, 1489.
- 6 Morais, E., Sinigoi, S., Mayer, A., Mucana, A., Miguel, L.G., Rufino Neto, J., 1998. The
7 Kunene gabbro-anorthosite Complex: preliminary results based on new field and chemical
8 data. *Afr. Geosci. Rev.* 5, 485-498.
- 9 Morimoto, N., 1988. Nomenclature of pyroxenes. *Miner. Mag.* 52, 535-550.
- 10 Morrison, J., Valley, J.W., 1988. Contamination of the Marcy anorthosite massif, Adirondack
11 Mountains, NY: petrologic and isotopic evidence. *Contrib. Miner. Petrol.* 98, 97-108.
- 12 O'Connor, Y.-L., Morrison, J., 1999. Oxygen isotope constraints on the petrogenesis of the
13 Sybille intrusion of the Proterozoic Laramie anorthosite complex. *Contrib. Miner. Petrol.*
14 136, 81-91.
- 15 Pearce, J.A., Harris, N.B.W., Tindle, A.G., 1984. Trace element discrimination diagrams for
16 the tectonic interpretation of granitic rocks. *J. Petrol.* 25, 956-983.
- 17 Pearce, N.J.G., Perkins, W.T., Westgate, J.A., Gorton, M.P., Jackson, S.E., Neal, C.R.,
18 Chenery, S.P., 1997. A compilation of new and published major and trace element data
19 for NIST SRM 610 and NIST SRM 612 glass reference materials. *Geostandards*
20 *Newsletter* 21, 115-144.
- 21 Peck, W.H., Valley, J.W., 2000. Large crustal input into high $\delta^{18}\text{O}$ anorthosite massifs of the
22 southern Grenville Province: new evidence from the Morin Complex, Quebec. *Contrib.*
23 *Miner. Petrol.* 139, 402-417.
- 24 Plessen, H.-G., 1997. Analytik und Geochemie der Platingruppenelemente in magmatischen
25 Gesteinen. Dr. rer. nat. thesis, University of Potsdam.

- 1 Poller, U., Uher, P., Janák, M., Plasienka, D., 2001. Late Cretaceous age of the Rochovce
2 granite, Western Carpathians, constrained by U/Pb single-zircon dating in combination
3 with cathodoluminescence imaging. *Geologica Carpathica* 52, 41-47.
- 4 Romer, R.L., Heinrich, W., Schröder-Smeibidl, B., Meixner, A., Fischer, C.-O., Schulz, C.,
5 2005. Elemental dispersion and stable isotope fractionation during reactive fluid-flow and
6 fluid immiscibility in the Bufa del Diente aureole, NE-Mexico: Evidence from
7 radiographies and Li, B, Sr, Nd, and Pb isotope systematics. *Contrib. Miner. Petrol.* 149,
8 400-429.
- 9 Schiellerup, H., Lambert, D., Presvik, T., Robins, B., McBride, J.S., Lars, R.B., 2000. Re–Os
10 isotopic evidence for a lower crustal origin of massif-type anorthosites. *Nature* 405, 781-
11 784.
- 12 Seifert, K.E., 1978. Anorthosite-mangerite relations on Barker Mountain, New York. *Geol.*
13 *Soc. Am. Bull.* 89, 245-250.
- 14 Selbekk, R.S., Skjerlie, K.P., Pedersen, R.B., 2000. Generation of anorthositic magma by
15 H₂O-fluxed anatexis of silica-undersaturated gabbro: An example from the north
16 Norwegian Caledonides. *Geol. Mag.* 137, 609–621.
- 17 Seth, B., Armstrong, R.A., Brandt, S., Villa, I.M., Kramers, J.D., 2003. Mesoproterozoic U-
18 Pb and Pb-Pb ages of granulites in NW Namibia: reconstructing a complete orogenic
19 cycle. *Precamb. Res.* 126, 147-168.
- 20 Seth, B., Armstrong, R.A., Büttner, A., Villa, I.M., 2005. Time constraints for
21 Mesoproterozoic upper amphibolite facies metamorphism in NW Namibia: a multi-
22 isotopic approach. *Earth Plan. Sci. Lett.* 230, 355-378.
- 23 Silva, Z.C.G., 1990. Geochemistry of the gabbro-anorthosite Complex of Angola. *J. Afr.*
24 *Earth Sci.* 10, 683-692.

- 1 Silva, Z.C.G., 1992. Mineralogy and cryptic layering of the Kunene anorthosite complex of
2 SW Angola and Namibia. *Miner. Mag.* 56, 319-327.
- 3 Slejko, F.F., Demarchi, G., Morais, E., 2002. Mineral chemistry and Nd isotopic composition
4 of two anorthositic rocks from the Kunene complex (South western Angola. *J. Afr. Earth*
5 *Sci.* 35, 77-88.
- 6 Steiger, R.H., Jäger, E., 1977. Subcommittee on geochronology: Convention on the use of
7 decay constants in geo- and cosmochronology. *Earth Plan. Sci. Lett.* 36, 359-362.
- 8 Streckeisen, A., 1974. How should charnockitic rocks be named? In: Bellière, J., Duchesne,
9 J.C. (Eds.), *Géologie des Domaines Cristallins*. Société Géologique de Belgique,
10 Centenary volume, Liège, 349-360.
- 11 Tack, L., Fernandez-Alonso, M., Tahon, M., Wingate, M.T.D., Barritt, S., 2002. The
12 'northeastern Kibaran belt' (NKB) and its mineralisations reconsidered: new constraints
13 from a revised lithostratigraphy, a GIS-compilation of existing geological maps and a
14 review of recently published as well as unpublished igneous emplacement ages in
15 Burundi. Abstracts, 11th IAGOD Quadrennial Symposium and Geocongress, Windhoek, 6.
- 16 Taylor, H.P., 1968. Oxygen isotope studies of anorthosites, with particular reference to the
17 origin of bodies in the Adirondack Mountains, New York. In: Isachsen, Y.W. (ed.) *Origin*
18 *of anorthosite and related rocks*. N.Y. State Museum and Science Service Memoir 18,
19 111-134.
- 20 Taylor, S.R., Campbell, I.H., McCulloch, M.T., McLennan S.M., 1984. A lower crustal origin
21 for massif-type anorthosites. *Nature* 311, 372-374.
- 22 Taylor, H.P., Sheppard, S.M.F., 1986. Igneous rocks I. Processes of isotopic fractionation and
23 isotope systematics. In: Valley, J.W., Taylor, H.P., O'Neil, J. (Eds.), *Stable isotopes in*
24 *high temperature geological processes*. Mineralogical Society of America, Chelsea,
25 Michigan, 227-269.

- 1 Tegtmeier, A., Kröner, A., 1985. U-Pb zircon ages for granitoid gneisses in northern Namibia
2 and their significance for Proterozoic crustal evolution of Southwestern Africa. *Precamb.*
3 *Res.* 28, 311-326.
- 4 Torquato, J.R., Silva, A.T.S.F. da, Cordani, U.G., Kawashita, K., 1979. A evolução geológica
5 do Cinturão Móvel do Quipungo no Ocidente de Angola. *Anais Academia Brasileira*
6 *Ciências* 51, 133-143.
- 7 Valley, J.W., O'Neil, J.R., 1982. Oxygen isotope evidence for shallow emplacement of
8 Adirondack anorthosite. *Nature* 300, 497-500.
- 9 Von Seckendorff, V., Drüppel, K., Okrusch, M., Littmann, S., Cook, N., 2000. Oxide-
10 sulphide relationships in sodalite-bearing metasomatites of the Epembe-Swartbooisdrif
11 Alkaline Province, North-West Namibia. *Mineralium Deposita* 35, 430-450.
- 12 Whalen, J.B., Currie, K.L., Chappell, B.W., 1987. A-type granites: geochemical
13 characteristics, discrimination and petrogenesis. *Contrib. Miner. Petrol.* 95, 407-419.
- 14 Wiebe, R.A., 1994. Proterozoic anorthosite complexes. In: Condie, K.C. (Ed.), *Proterozoic*
15 *crustal Evolution*. Elsevier, Amsterdam, 215-261.
- 16 Wilson, M., 1989. *Igneous petrogenesis – a global tectonic approach*. Harper Collins
17 Academic, London.
- 18 Wiszniewska, J., Claesson, S., Stein, H.J., Vander Auwera, J., Duchesne, J.-C., 2002. The
19 north-eastern Polish anorthosite massifs: petrological geochemical and isotopic evidence
20 for a crustal derivation. *Terra Nova* 14, 451-460.
- 21 Zuleger, E., Erzinger, J., 1988. Determination of the REE and Y in silicate materials with
22 ICP-AES. *Fresenius Z. Analytische Chemie* 332, 140-143.
- 23

1 **FIGURE CAPTIONS**

2 Fig. 1. Geologic overview map of the Kunene Intrusive Complex in SW Angola and NW
3 Namibia. Map simplified and modified after Carvalho & Alves (1990). Outlined areas
4 mark regions of detailed mapping in the Angolan part (Ashwal & Twist, 1994; Morais et
5 al., 1998; Slejko et al., 2002; Mayer et al., 2004) and in the Namibian part of the KIC
6 (Menge, 1998; Drüppel et al., 2001; Drüppel, 2003).

7

8 Fig. 2. Geologic overview map of the southern Kunene Intrusive Complex in NW Namibia.
9 Map modified after Menge (1998) based on new field data and satellite images. Locations
10 for geochemical analyses and age determinations are marked.

11

12 Fig. 3. Polished-section photomicrographs of the anorthositic rocks of the KIC (a, b) and the
13 related felsic rocks (c, d). Plane-polarized light (a). Cross-polarized light (b-d). (a)
14 Subhedral, interstitial olivine is surrounded by irregular rims of orthopyroxene and late
15 amphibole and biotite that fill the open spaces between plagioclase crystals. (b) Interstices
16 between coarse-grained, anhedral cumulus plagioclase are filled by interstitial olivine that
17 is surrounded by broad clinopyroxene rims. (c) Subhedral, tabular plagioclase displays
18 straight grain boundaries against micropertthitic K-feldspar. (d) Euhedral clinopyroxene
19 inclusion in plagioclase. (*Am* amphibole, *Bt* biotite, *Kfs* K-feldspar, *Ol* olivine, *Opx*
20 orthopyroxene, *Pl* plagioclase).

21

22 Fig. 4. Variations of selected major and trace element contents of the anorthositic rocks of the
23 KIC and the related felsic rocks.

24

1 Fig. 5. Compositional variability of both the anorthositic and the felsic rock suite compared to
2 similar lithologies of the Harp Lake massif, Labrador (stippled boundaries; Emslie, 1980).
3 Same symbols as Fig. 4. (a) Plot of molar $\text{Fe}^{\text{tot}}/(\text{Fe}^{\text{tot}}+\text{Mg})$ vs. SiO_2 and (b) $\text{Na}_2\text{O}+\text{K}_2\text{O}$ vs.
4 SiO_2 . (*S* syenite, *SD* syenodiorite).

5
6 Fig. 6. Diagrams showing the variations in chondrite-normalized rare earth element
7 abundances for the anorthositic rocks of the KIC and the associated felsic rock suite
8 (normalisation factors after Pearce et al., 1997).

9
10 Fig. 7. Composition of the felsic rock suite expressed in the $\text{Na}_2\text{O} + \text{K}_2\text{O}$ vs. SiO_2
11 classification diagram for plutonic rocks after Cox et al. (1979).

12
13 Fig. 8. Compositional characteristics of the felsic rock suite expressed in various binary
14 discrimination diagrams. (a) Plot of $\text{FeO}^{\text{tot}}/(\text{FeO}^{\text{tot}}+\text{MgO})$ vs. SiO_2 with the boundary
15 between ferroan and magnesian fields given by Frost et al. (2001). (b) Plot of the modified
16 alkali-lime index $\text{Na}_2\text{O}+\text{K}_2\text{O}-\text{CaO}$ vs. SiO_2 (Frost et al., 2001). (c) Shand's alkalinity
17 index (plot of $\text{Al}_2\text{O}_3/(\text{Na}_2\text{O}+\text{K}_2\text{O})_{\text{molar}}$ vs. $\text{Al}_2\text{O}_3/(\text{Na}_2\text{O}+\text{K}_2\text{O}+\text{CaO})_{\text{molar}}$). (d) A-type
18 granitoid discrimination diagram after Whalen et al. (1987). (e) and (f) Tectonic
19 environment discrimination diagrams after Pearce et al. (1984). (*ORG* orogenic granite,
20 *syn-COLG* syn-collisional granite; *VAG* volcanic arc granite, *WPG* within-plate granite).

21
22 Fig. 9. Isochron $^{206}\text{Pb}/^{204}\text{Pb}$ vs $^{208}\text{Pb}/^{204}\text{Pb}$ diagram for zircon samples of syenodiorite Ku-98-
23 201, Swartbooisdrif, Namibia (data from Table 6)

24

1 Fig. 10. Plot of $\delta^{18}\text{O}$ vs. SiO_2 for the anorthositic rocks of the KIC and the felsic rock suite
2 (see text for discussion).

3

4 Fig. 11. Initial Sr-Nd isotope systematics of c. 1.38 Ga anorthosites and related felsic rocks in
5 comparison with depleted mantle (light grey shaded area) and Palaeoproterozoic crust of
6 northern Namibia (dark grey shaded area). Note the sub-horizontal arrangement of the
7 anorthosite and felsic rock samples. The Sr and Nd isotopic composition of the crust was
8 estimated from 1.9 Ga orthogneisses (Seth *et al.*, 2005) recalculated to 1.38 Ga. The
9 composition of the mantle reservoirs was estimated, assuming Rb/Sr and Sm/Nd typical
10 for these reservoirs (e.g., Wilson, 1989).

11

12 Fig. 12. Diagram of MgO vs. $\text{CaO}/(\text{CaO}+\text{Na}_2\text{O})$ for available geochemical data on the
13 Kunene Complex. (a) Geochemical composition of the white and dark anorthosite suites
14 from NW Namibia. (b) Geochemical composition of anorthositic rocks from Angola
15 (Silva, 1992 and Morais *et al.*, 1998).

1 **TABLE CAPTIONS**

2 Table 1a Representative electron microprobe analyses of plagioclase of the anorthosite suite.

3 Table 1b Representative electron microprobe analyses of feldspar of the felsic rock suite.

4

5 Table 2 Representative electron microprobe analyses of pyroxene of the anorthosite and the
6 felsic rock suites.

7

8 Table 3 Representative electron microprobe analyses of amphibole of the anorthosite and the
9 felsic rock suites.

10

11 Table 4 Selected major and trace element data of the anorthosite and the felsic rock suites.

12

13 Table 5 Rare earth element data of the anorthosite and the felsic rock suites.

14

15 Table 6 U-Pb analytical results for zircon from the syenodiorite sample Ku-98-201,
16 Swartbooisdrif, Namibia.

17

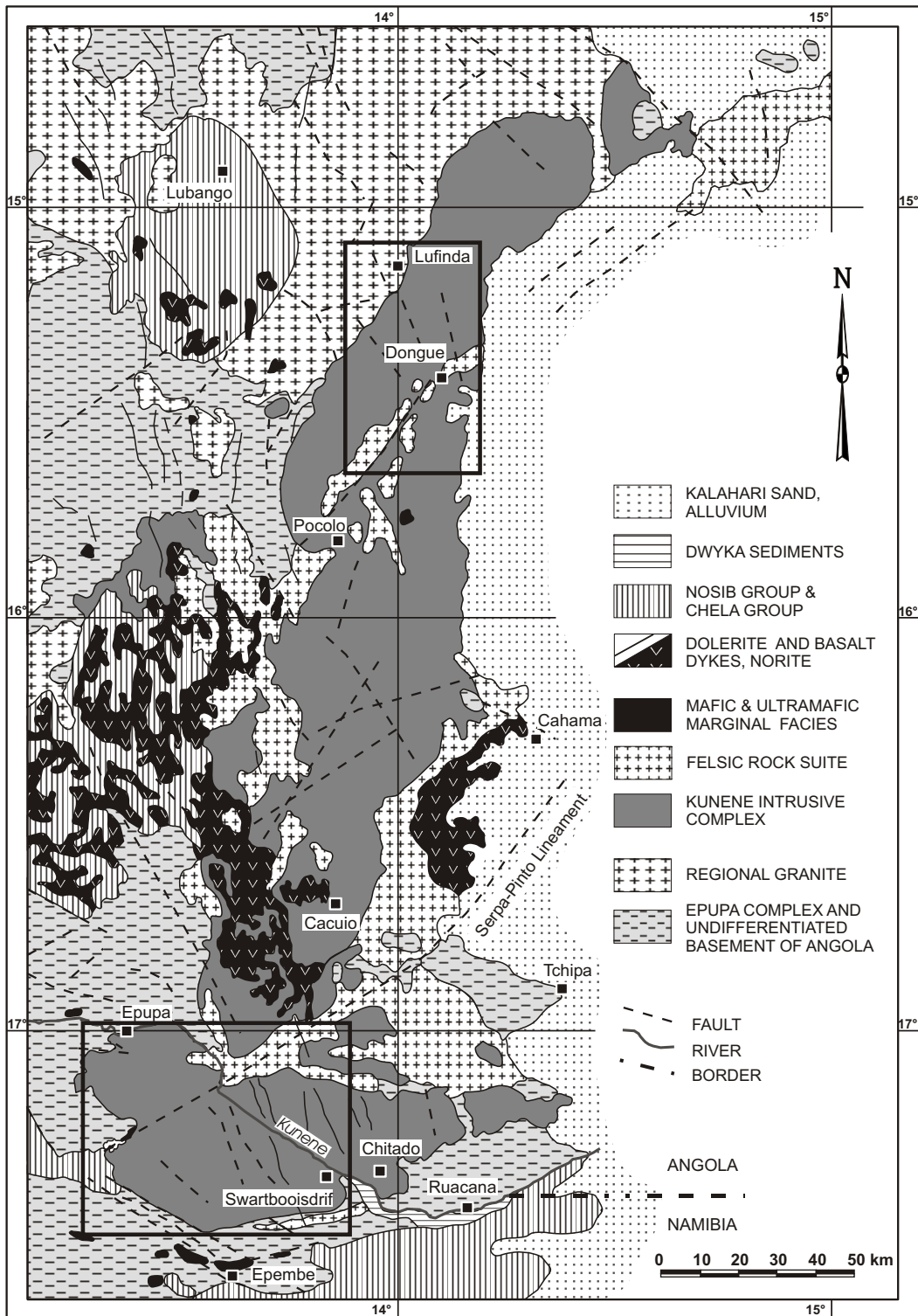
18 Table 7 Oxygen isotopic data of feldspar of the anorthosite and the felsic rock suites.

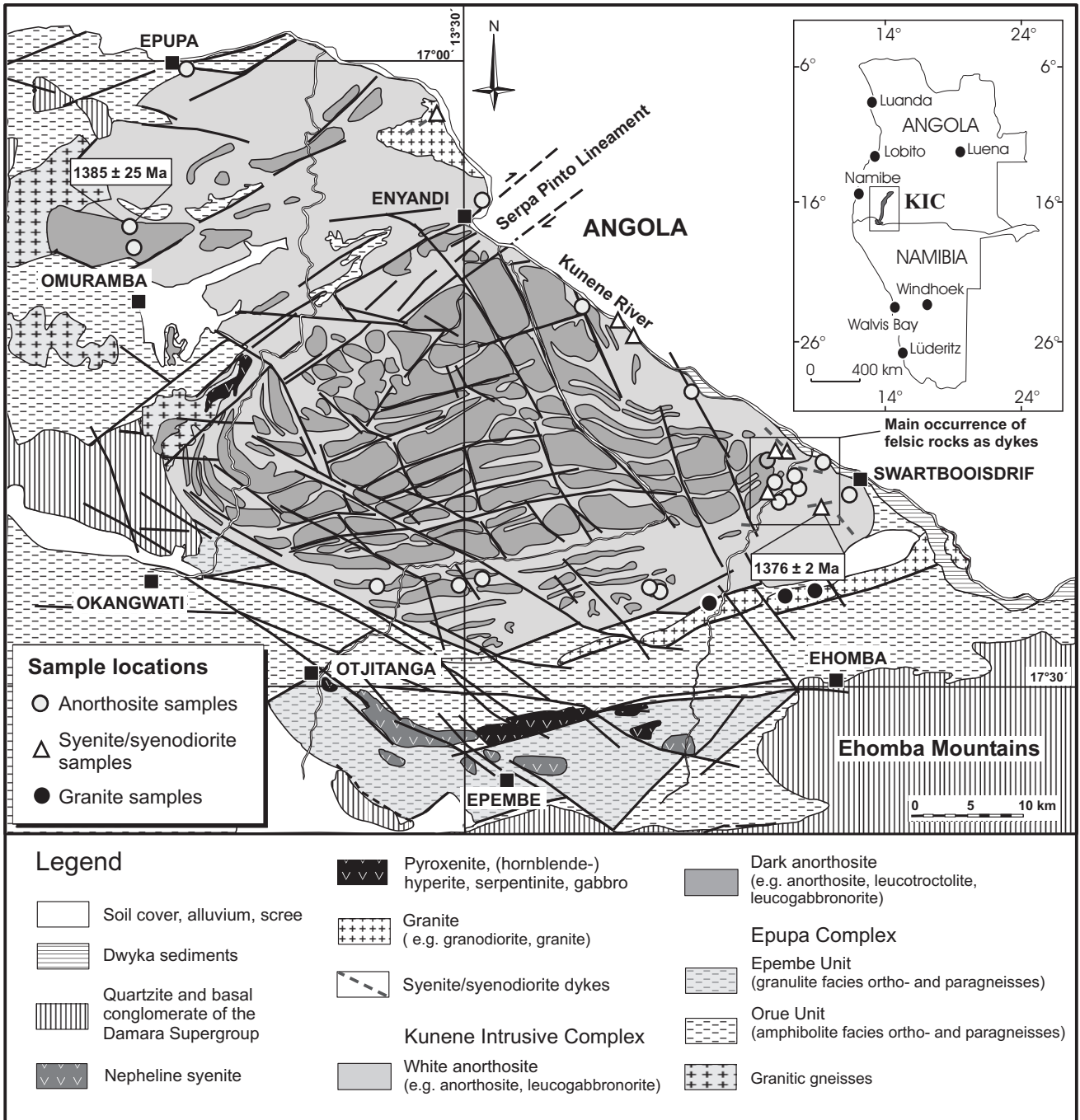
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20 Table 8 Sr and Nd isotope data for anorthosites and felsic rocks from NW Namibia.

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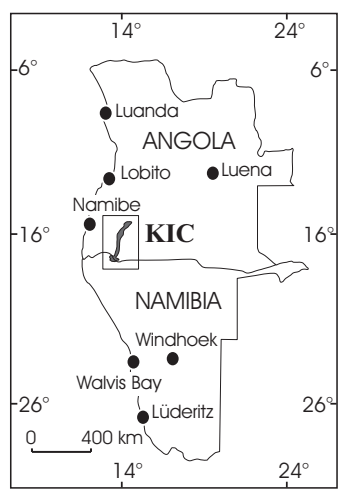
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1385 ± 25 Ma

1376 ± 2 Ma

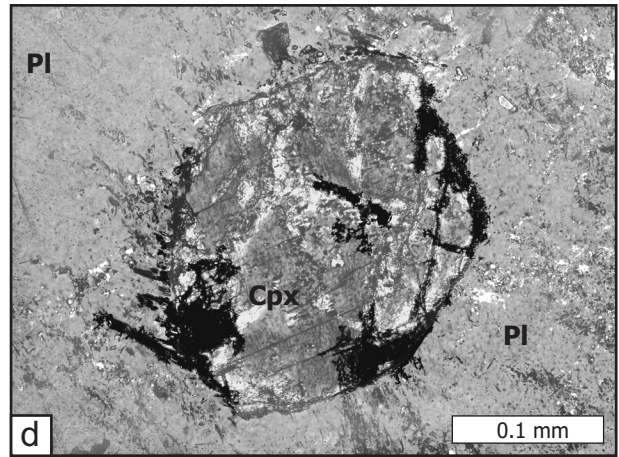
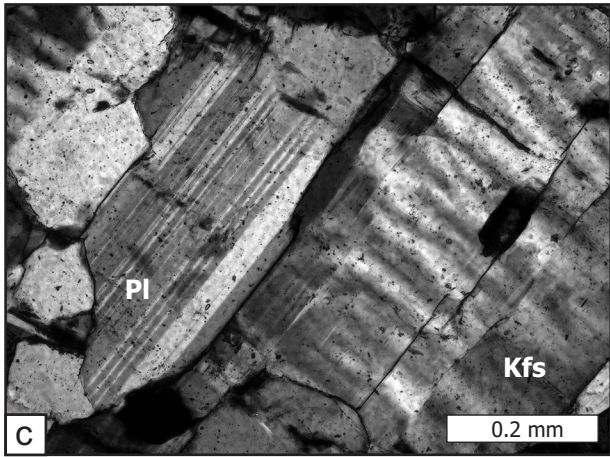
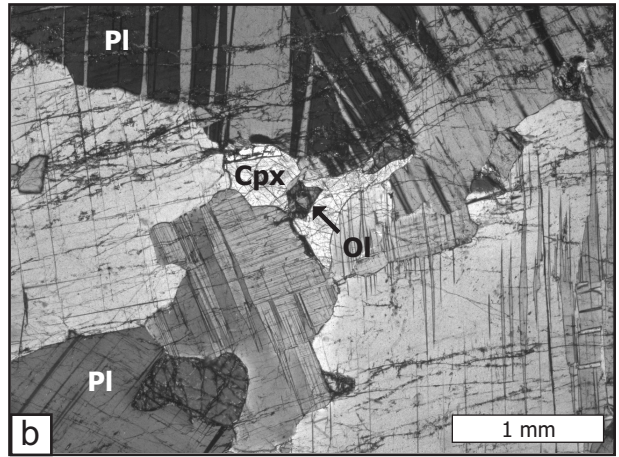
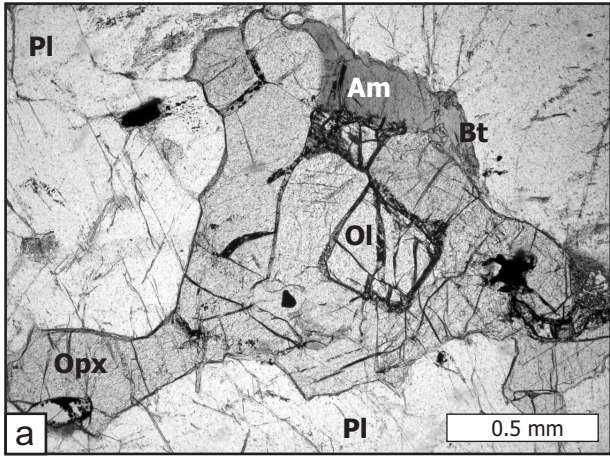


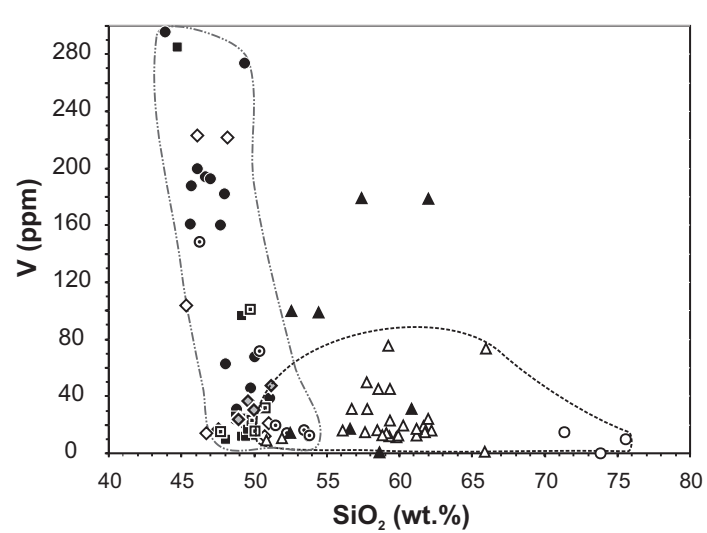
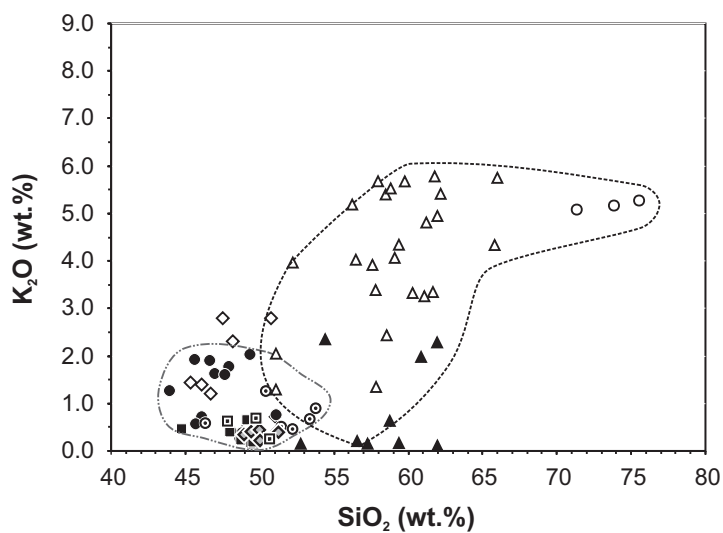
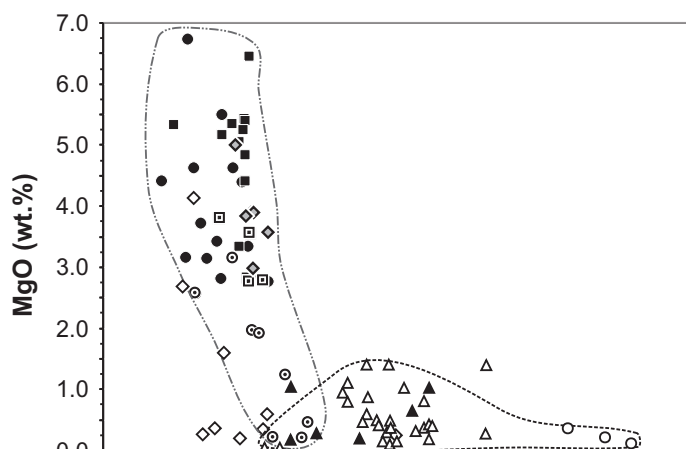
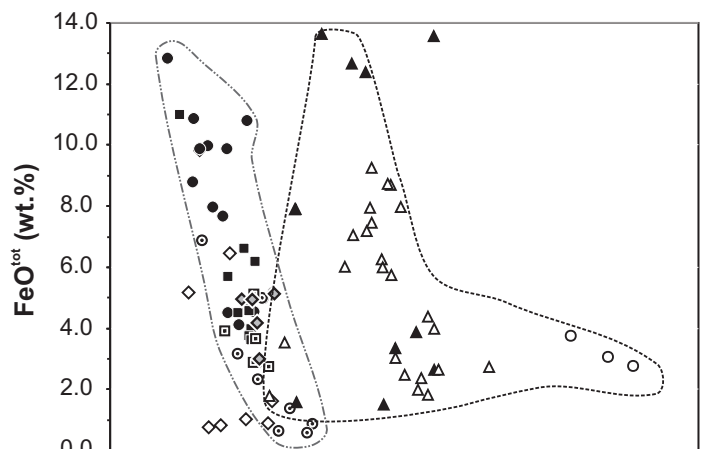
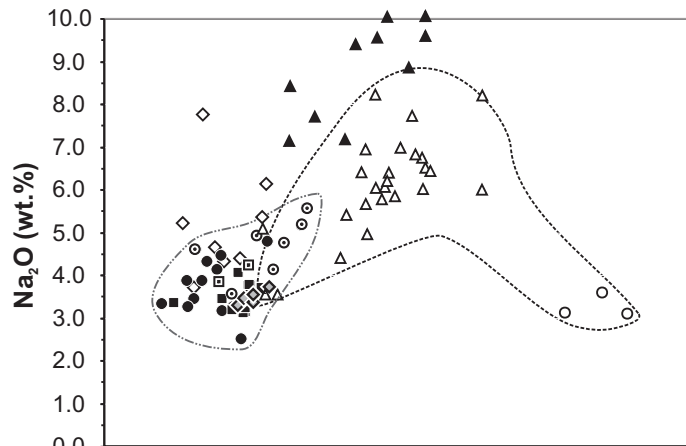
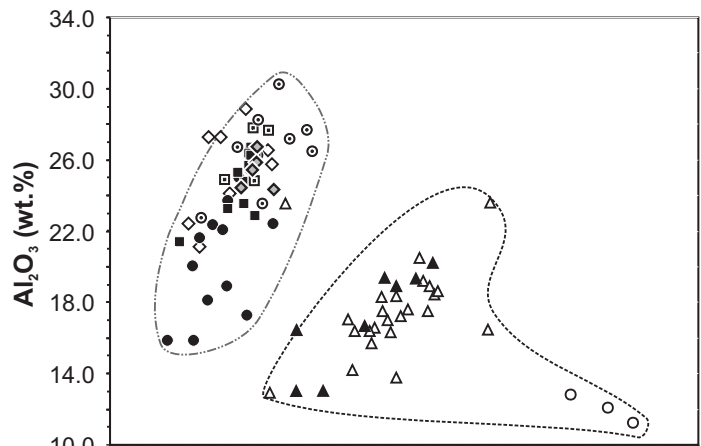
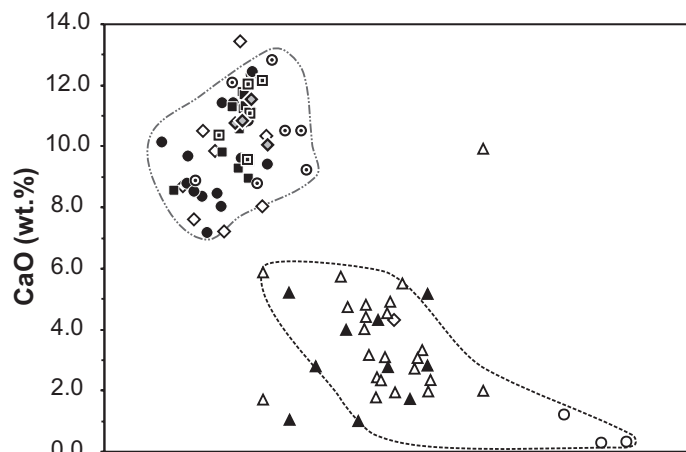
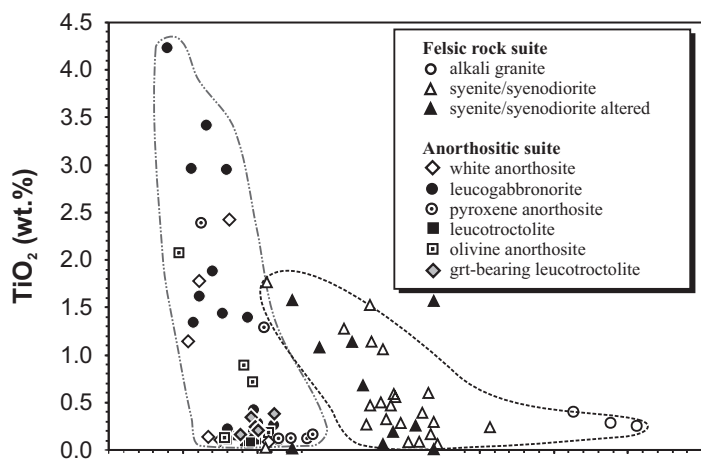
Main occurrence of felsic rocks as dykes

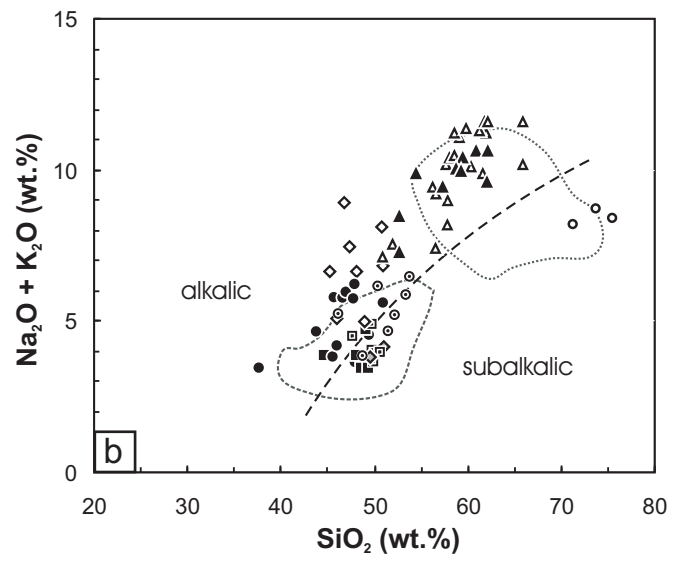
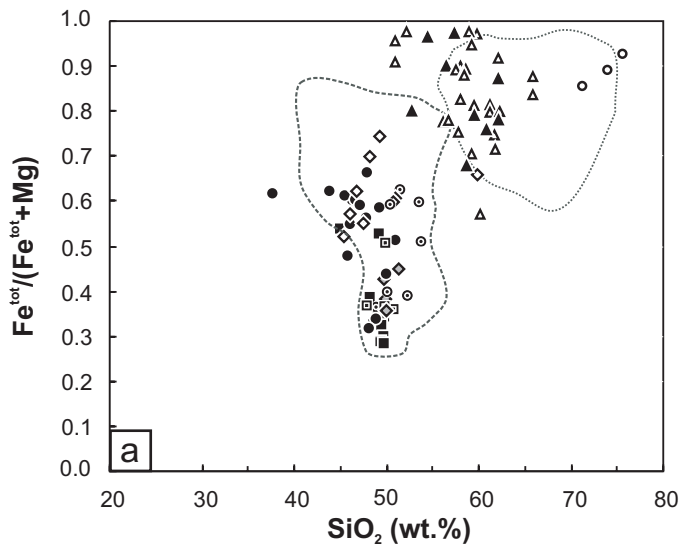
SWARTBOOISDRIF

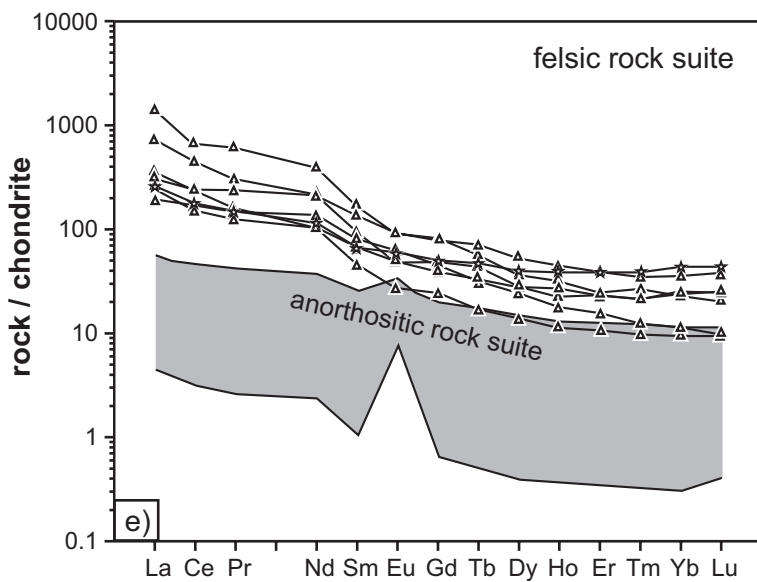
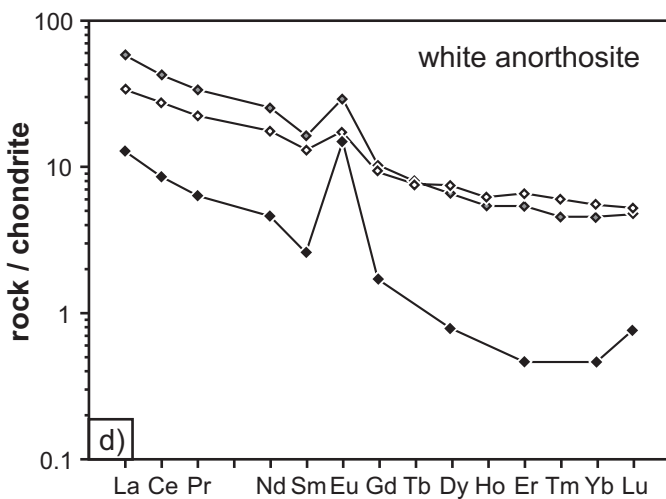
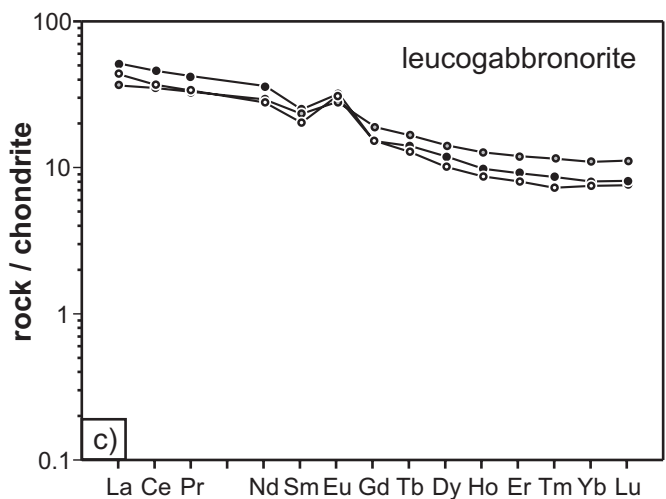
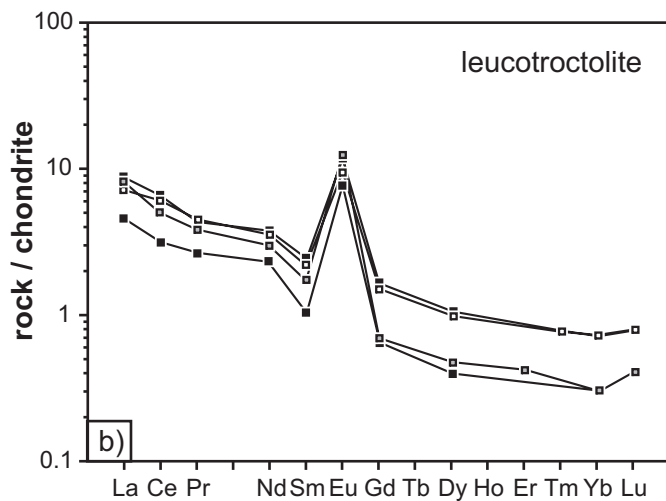
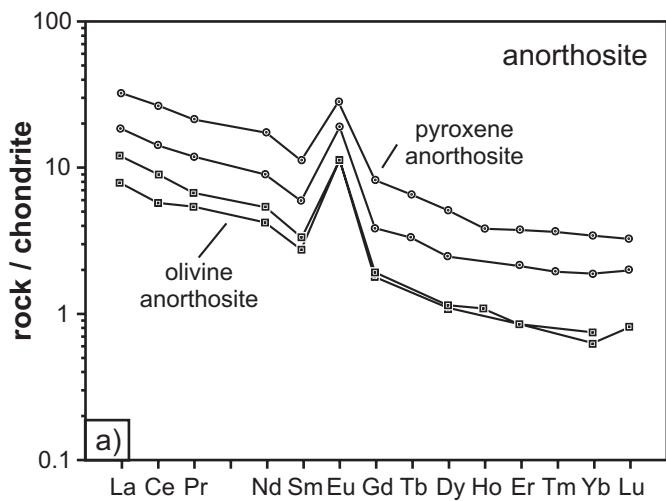
Ehomba Mountains

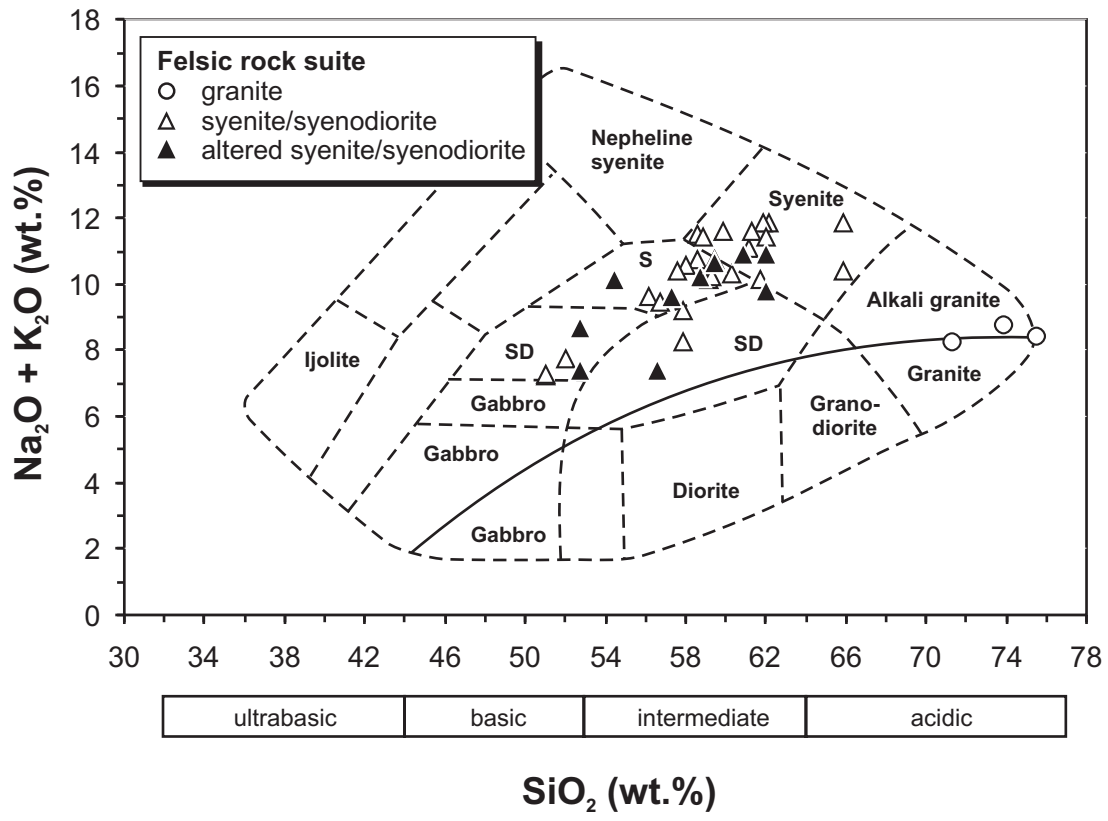


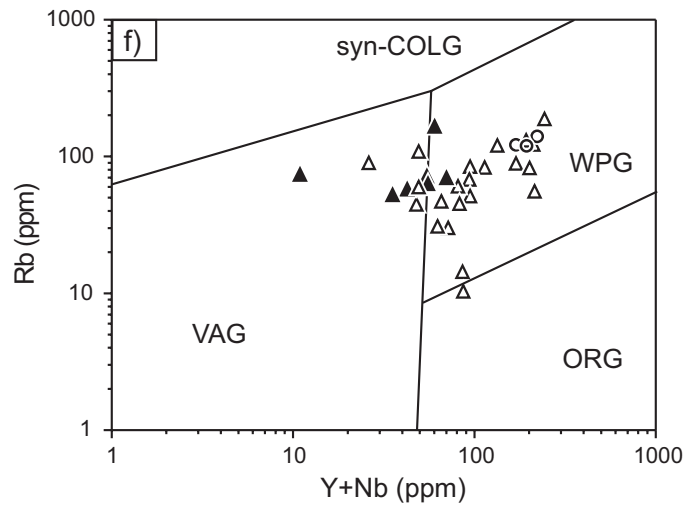
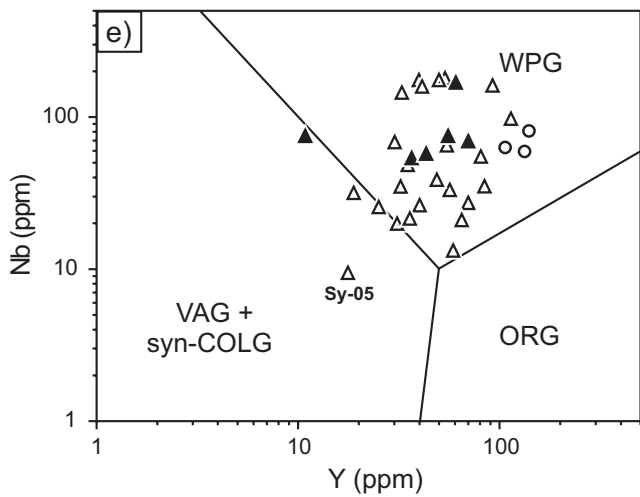
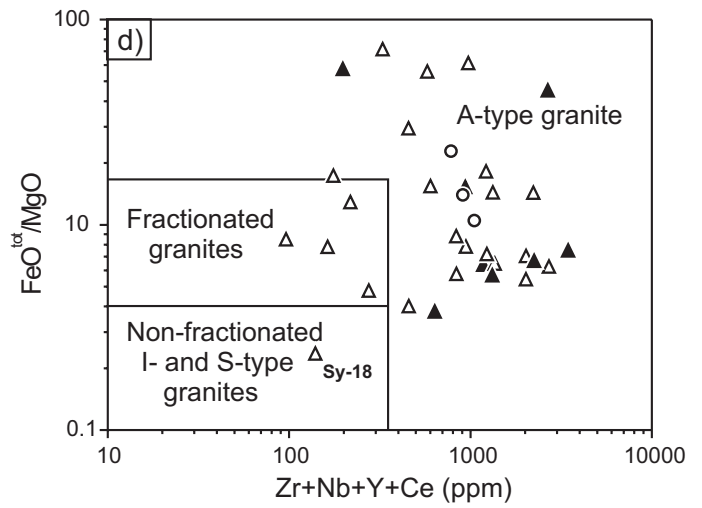
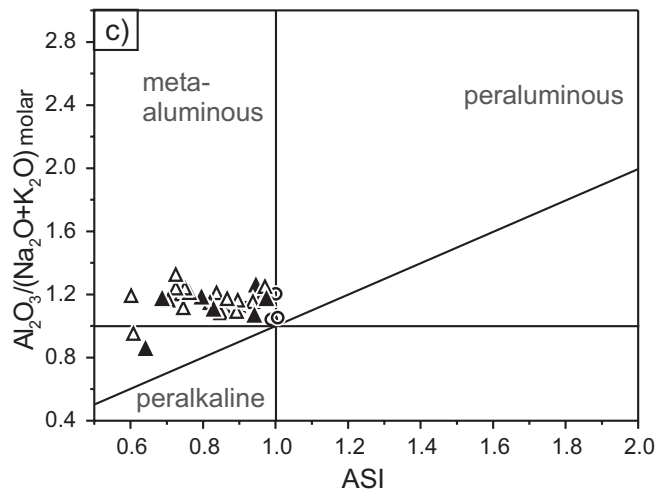
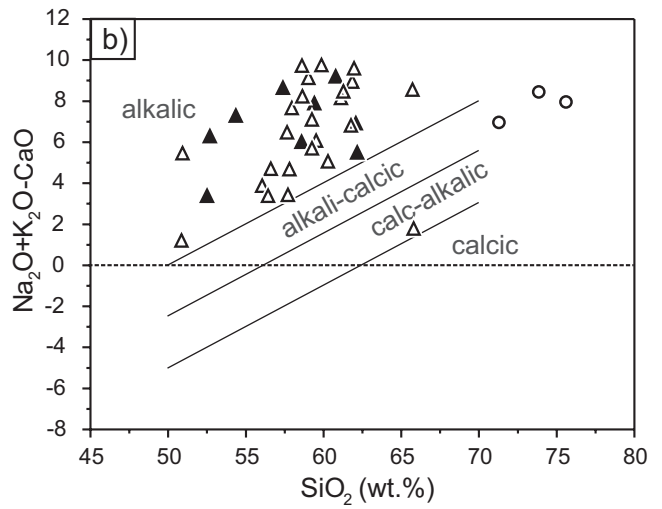
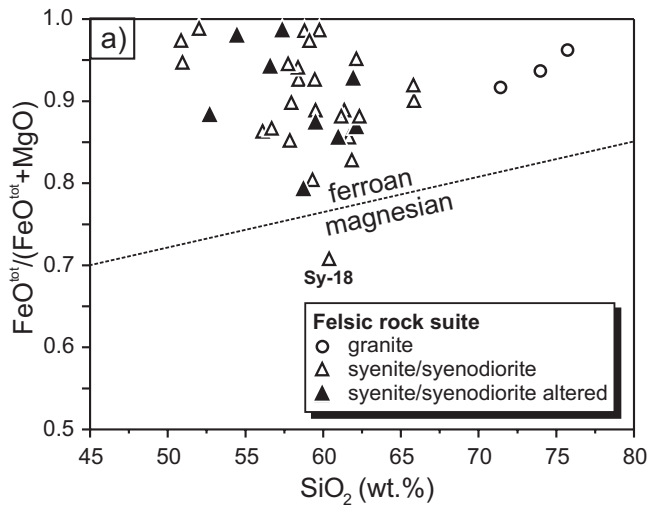


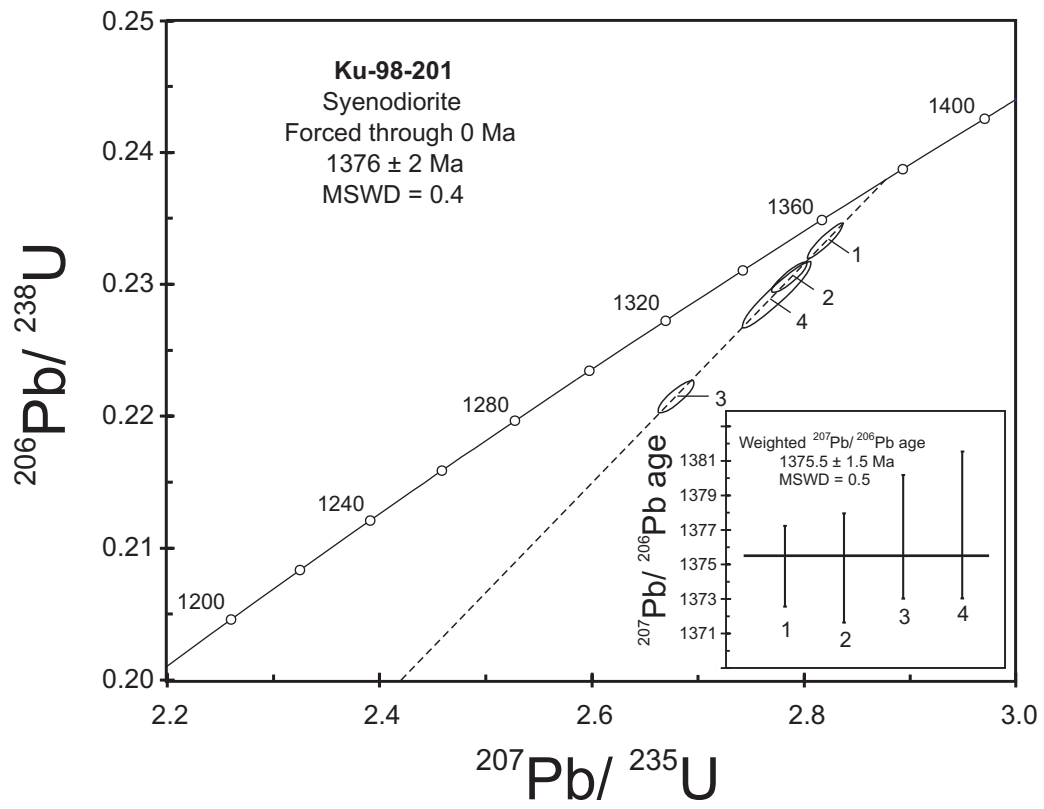


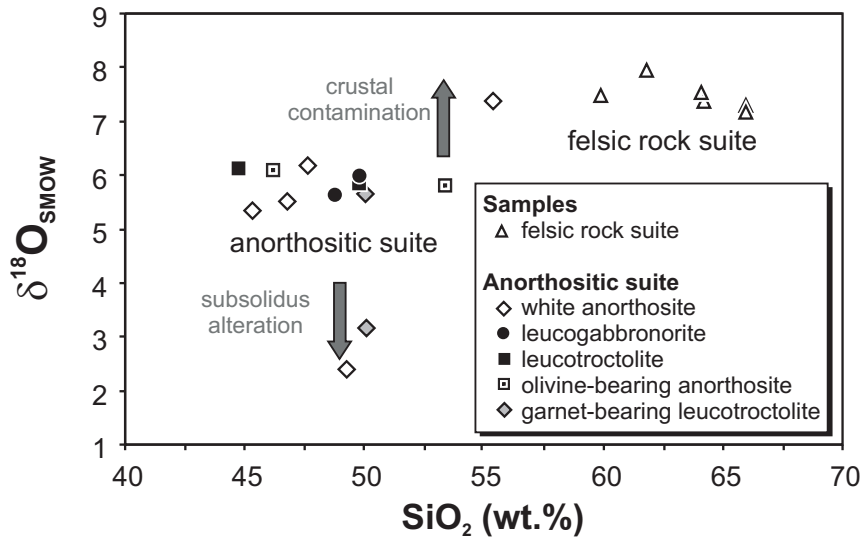


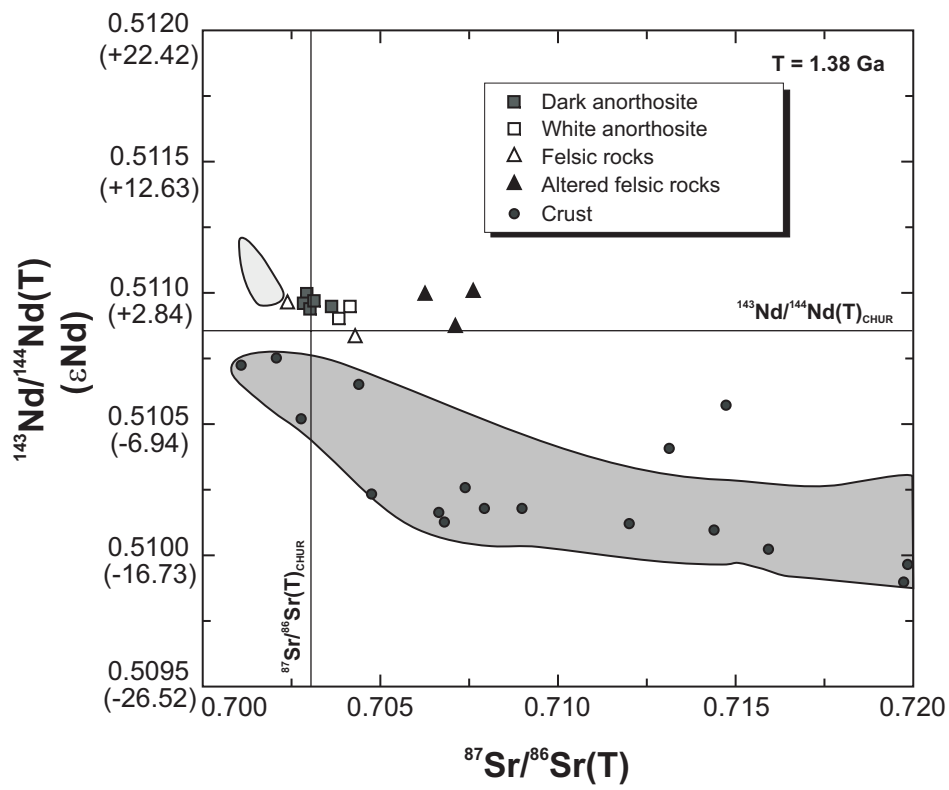












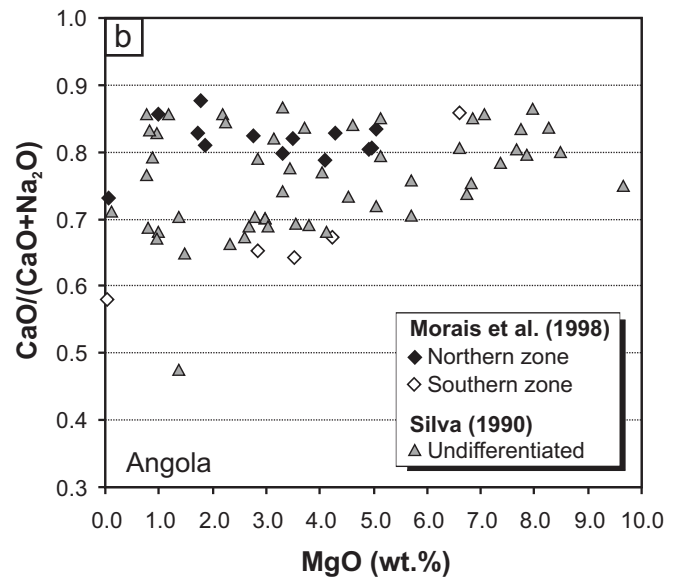
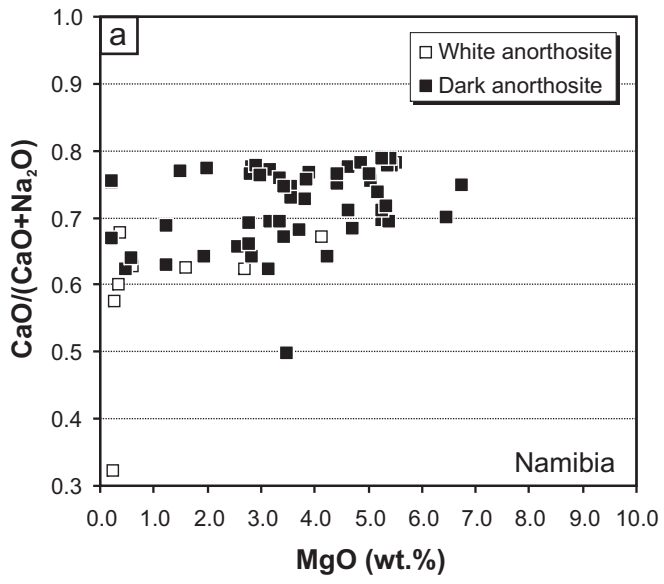


Table 1a. Representative electron microprobe analyses of plagioclase of the anorthosite suite.

| Rock type | A,w | A,w | A,w | GN | GN | GN | GN | A,px | A,px | A,px | A,px | A,px | T | T | T | T | T | T | T | A,ol | A,ol |
|--------------------------------|--------|-------|-------|-------|-------|--------|-------|-------|-------|-------|-------|-------|--------|--------|-------|-------|---------|---------|---------|-------|--------|
| Sample | 97-08b | 97-33 | 97-44 | 97-03 | 98-68 | 98-71 | 98-84 | 97-13 | 97-93 | 97-93 | 98-78 | 98-78 | 97-92 | 97-105 | 98-52 | 98-52 | 98-221a | 98-221b | 98-221b | 97-95 | 97-95 |
| Mineral | fs1 | fs1 | fs3 | fs1 | fs1 | fs1 | fs2 | fs1 | fs1 | fs1 | fs1 | fs2 | fs1 | fs1 | fs1 | fs2 | fs2 | fs1 | fs3 | fs1 | fs2 |
| | Pl | Pl | Pl | Pl | Pl | Pl | Pl | Pl | Pl | Pl | Pl | Pl | Pl | Pl | Pl | Pl | Pl | Pl | Pl | Pl | Pl |
| | rim | core | rim | core | rim | core | rim | rim | core | rim | core | core | rim | rim | rim | core | core | core | core | core | rim |
| SiO ₂ | 54.52 | 67.23 | 57.11 | 55.88 | 54.90 | 54.42 | 50.55 | 55.46 | 53.62 | 54.87 | 54.61 | 53.91 | 54.05 | 52.28 | 55.09 | 55.53 | 52.42 | 57.55 | 53.19 | 53.65 | 54.51 |
| Al ₂ O ₃ | 28.39 | 20.17 | 26.92 | 27.53 | 28.25 | 29.00 | 31.48 | 27.93 | 29.29 | 28.08 | 28.45 | 29.34 | 29.42 | 30.08 | 27.89 | 27.92 | 29.98 | 27.05 | 29.56 | 28.83 | 29.00 |
| MgO | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 |
| CaO | 10.59 | 0.95 | 8.75 | 9.86 | 10.47 | 10.87 | 14.05 | 10.20 | 11.51 | 10.48 | 10.94 | 11.35 | 11.78 | 12.53 | 10.14 | 9.94 | 12.46 | 9.15 | 12.07 | 11.43 | 11.20 |
| MnO | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 |
| FeO | 0.33 | 0.30 | 0.03 | 0.03 | 0.14 | 0.48 | 0.06 | 0.06 | 0.04 | 0.04 | 0.05 | 0.06 | 0.13 | 0.13 | 0.09 | 0.06 | 0.04 | 0.18 | 0.13 | 0.13 | 0.01 |
| BaO | 0.04 | 0.00 | 0.05 | 0.09 | 0.01 | 0.10 | 0.00 | 0.04 | 0.03 | 0.02 | 0.00 | 0.00 | 0.00 | 0.00 | 0.06 | 0.04 | 0.07 | 0.00 | 0.02 | 0.00 | 0.04 |
| Na ₂ O | 5.39 | 11.13 | 6.43 | 5.91 | 5.52 | 5.27 | 3.49 | 5.80 | 4.88 | 5.50 | 5.35 | 5.04 | 4.81 | 4.25 | 5.72 | 5.82 | 4.32 | 6.30 | 4.67 | 4.99 | 5.24 |
| K ₂ O | 0.05 | 0.03 | 0.07 | 0.08 | 0.11 | 0.17 | 0.00 | 0.04 | 0.07 | 0.06 | 0.11 | 0.12 | 0.27 | 0.21 | 0.08 | 0.14 | 0.14 | 0.21 | 0.04 | 0.03 | 0.07 |
| Sum | 99.32 | 99.81 | 99.36 | 99.38 | 99.41 | 100.31 | 99.62 | 99.53 | 99.43 | 99.05 | 99.51 | 99.82 | 100.45 | 99.48 | 99.07 | 99.45 | 99.43 | 100.45 | 99.69 | 99.05 | 100.06 |
| Formula (O=8) | | | | | | | | | | | | | | | | | | | | | |
| Si | 2.48 | 2.95 | 2.57 | 2.53 | 2.49 | 2.45 | 2.31 | 2.51 | 2.44 | 2.49 | 2.47 | 2.44 | 2.43 | 2.38 | 2.50 | 2.51 | 2.39 | 2.57 | 2.41 | 2.45 | 2.46 |
| Al | 1.52 | 1.04 | 1.43 | 1.47 | 1.51 | 1.54 | 1.69 | 1.49 | 1.57 | 1.50 | 1.52 | 1.56 | 1.56 | 1.62 | 1.49 | 1.49 | 1.61 | 1.42 | 1.58 | 1.55 | 1.54 |
| Mg | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 |
| Ca | 0.52 | 0.04 | 0.42 | 0.48 | 0.51 | 0.53 | 0.69 | 0.49 | 0.56 | 0.51 | 0.53 | 0.55 | 0.57 | 0.61 | 0.49 | 0.48 | 0.61 | 0.44 | 0.59 | 0.56 | 0.54 |
| Mn | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 |
| Fe | 0.01 | 0.01 | 0.00 | 0.00 | 0.01 | 0.02 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.01 | 0.00 | 0.00 | 0.00 |
| Ba | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 |
| Na | 0.47 | 0.95 | 0.56 | 0.52 | 0.49 | 0.46 | 0.31 | 0.51 | 0.43 | 0.48 | 0.47 | 0.44 | 0.42 | 0.38 | 0.50 | 0.51 | 0.38 | 0.55 | 0.41 | 0.44 | 0.46 |
| K | 0.00 | 0.00 | 0.00 | 0.00 | 0.01 | 0.01 | 0.00 | 0.00 | 0.00 | 0.00 | 0.01 | 0.01 | 0.02 | 0.01 | 0.00 | 0.01 | 0.01 | 0.01 | 0.00 | 0.00 | 0.00 |
| Sum | 5.00 | 5.00 | 4.99 | 5.00 | 5.00 | 5.01 | 5.00 | 5.00 | 5.00 | 5.00 | 5.00 | 5.00 | 5.00 | 5.00 | 5.00 | 5.00 | 5.00 | 5.00 | 5.00 | 5.00 | 5.00 |
| Si+Al | 4.00 | 4.00 | 4.00 | 4.00 | 4.00 | 4.00 | 4.00 | 4.00 | 4.00 | 4.00 | 3.99 | 4.00 | 4.00 | 4.00 | 4.00 | 4.00 | 4.00 | 3.99 | 4.00 | 4.00 | 4.00 |
| Na+Ca+K | 0.99 | 0.99 | 0.99 | 1.00 | 1.00 | 1.00 | 1.00 | 1.01 | 0.99 | 1.00 | 1.01 | 1.00 | 1.00 | 1.00 | 1.00 | 1.00 | 1.00 | 1.00 | 1.00 | 1.00 | 1.00 |
| AN(Na/Ca) | 51.84 | 4.49 | 42.69 | 47.66 | 50.82 | 52.62 | 68.95 | 49.11 | 56.33 | 51.08 | 52.76 | 55.09 | 56.62 | 61.21 | 49.20 | 48.14 | 60.87 | 43.97 | 58.65 | 55.79 | 53.91 |
| AB(Na/Ca) | 47.78 | 95.35 | 56.79 | 51.74 | 48.51 | 46.21 | 31.04 | 50.58 | 43.23 | 48.54 | 46.63 | 44.23 | 41.83 | 37.56 | 50.24 | 50.99 | 38.21 | 54.81 | 41.06 | 44.05 | 45.62 |
| OR | 0.31 | 0.16 | 0.42 | 0.44 | 0.65 | 0.99 | 0.01 | 0.24 | 0.40 | 0.34 | 0.60 | 0.69 | 1.56 | 1.19 | 0.45 | 0.80 | 0.80 | 1.22 | 0.25 | 0.16 | 0.40 |
| CE | 0.07 | 0.00 | 0.09 | 0.16 | 0.02 | 0.18 | 0.00 | 0.07 | 0.05 | 0.04 | 0.00 | 0.00 | 0.00 | 0.04 | 0.11 | 0.07 | 0.13 | 0.00 | 0.04 | 0.00 | 0.07 |

Abbreviations: *A,w* white anorthosite; *A,ol* olivine anorthosite; *A,px* pyroxene anorthosite; *GN* leucogabbroanorthite; *T* leucotroctolite.

Table 1b. Representative electron microprobe analyses of plagioclase of the felsic rock suite.

| Sample | 98-07 | 98-40 | 98-40 | 99-11 | 99-11 | 99-11 | 99-11 | 99-12 | 99-13 | 99-13 | 99-13 | 99-14 |
|--------------------------------|-------|--------|--------|-------|--------|-------|-------|-------|-------|--------|-------|-------|
| Mineral | 3-fs4 | 2-fs1 | 2-fs1 | a-fs1 | a-fs1 | a-fs2 | a-fs2 | b-fs4 | 1-fs2 | 3-fs3 | 3-fs3 | a-fs2 |
| | Pl | Pl | Pl | Pl | Pl | Pl | Pl | Pl | Pl | Pl | Pl | Pl |
| | rim | rim | core | core | core | core | core | rim | rim | rim | -> | core |
| SiO ₂ | 67.32 | 69.02 | 64.11 | 65.27 | 66.07 | 64.88 | 65.58 | 66.99 | 66.55 | 67.97 | 66.92 | 67.12 |
| Al ₂ O ₃ | 19.85 | 19.71 | 22.87 | 21.12 | 21.18 | 21.32 | 21.15 | 20.24 | 19.80 | 19.95 | 20.02 | 20.57 |
| MgO | 0.00 | 0.01 | 0.00 | 0.00 | 0.02 | 0.00 | 0.00 | 0.02 | 0.03 | 0.00 | 0.00 | 0.00 |
| CaO | 0.27 | 0.17 | 3.73 | 1.96 | 1.86 | 2.36 | 2.11 | 0.85 | 0.58 | 0.59 | 0.88 | 0.94 |
| MnO | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 |
| FeO | 0.05 | 0.07 | 0.10 | 0.20 | 0.24 | 0.17 | 0.12 | 0.05 | 0.33 | 0.16 | 0.06 | 0.04 |
| BaO | 0.00 | 0.04 | 0.00 | 0.07 | 0.00 | 0.05 | 0.00 | 0.00 | 0.00 | 0.00 | 0.04 | 0.08 |
| Na ₂ O | 11.62 | 11.78 | 9.57 | 10.46 | 10.66 | 10.18 | 10.32 | 11.25 | 11.20 | 11.36 | 11.12 | 10.96 |
| K ₂ O | 0.03 | 0.15 | 0.32 | 0.14 | 0.12 | 0.20 | 0.13 | 0.08 | 0.06 | 0.10 | 0.11 | 0.17 |
| Sum | 99.14 | 100.95 | 100.69 | 99.22 | 100.16 | 99.15 | 99.40 | 99.48 | 98.55 | 100.12 | 99.16 | 99.88 |
| Formula (O=8) | | | | | | | | | | | | |
| Si | 2.97 | 2.99 | 2.81 | 2.89 | 2.90 | 2.88 | 2.90 | 2.95 | 2.96 | 2.97 | 2.96 | 2.94 |
| Al | 1.03 | 1.01 | 1.18 | 1.10 | 1.10 | 1.12 | 1.10 | 1.05 | 1.04 | 1.03 | 1.04 | 1.06 |
| Mg | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 |
| Ca | 0.01 | 0.01 | 0.18 | 0.09 | 0.09 | 0.11 | 0.10 | 0.04 | 0.03 | 0.03 | 0.04 | 0.04 |
| Mn | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 |
| Fe | 0.00 | 0.00 | 0.00 | 0.01 | 0.01 | 0.01 | 0.00 | 0.00 | 0.01 | 0.01 | 0.00 | 0.00 |
| Ba | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 |
| Na | 0.99 | 0.99 | 0.81 | 0.90 | 0.91 | 0.88 | 0.88 | 0.96 | 0.97 | 0.96 | 0.95 | 0.93 |
| K | 0.00 | 0.01 | 0.02 | 0.01 | 0.01 | 0.01 | 0.01 | 0.00 | 0.00 | 0.01 | 0.01 | 0.01 |
| Sum | 5.01 | 5.01 | 5.01 | 5.01 | 5.01 | 5.00 | 5.00 | 5.01 | 5.01 | 5.00 | 5.00 | 5.00 |
| Si+Al | 4.00 | 4.00 | 4.00 | 4.00 | 4.00 | 4.00 | 4.00 | 4.00 | 4.00 | 4.00 | 4.00 | 4.01 |
| Na+Ca+K | 1.01 | 1.01 | 1.01 | 1.00 | 1.00 | 1.00 | 0.99 | 1.00 | 1.00 | 1.00 | 1.00 | 0.99 |
| AN(Na/Ca) | 1.26 | 0.76 | 17.43 | 9.29 | 8.74 | 11.23 | 10.07 | 3.99 | 2.79 | 2.78 | 4.16 | 4.48 |
| AB(Na/Ca) | 98.55 | 98.36 | 80.81 | 89.81 | 90.57 | 87.57 | 89.18 | 95.57 | 96.89 | 96.64 | 95.14 | 94.42 |
| OR | 0.18 | 0.82 | 1.76 | 0.79 | 0.69 | 1.11 | 0.76 | 0.44 | 0.32 | 0.58 | 0.64 | 0.96 |
| CE | 0.00 | 0.06 | 0.00 | 0.12 | 0.00 | 0.09 | 0.00 | 0.00 | 0.00 | 0.00 | 0.07 | 0.13 |

Table 2. Representative electron microprobe analyses of pyroxene of the anorthosite and the felsic rock suites.

| Rock type | Leucogabbronorite | | Leucotroctolite | | | | Felsic rocks | | White anorthosite | Leucotroctolite | | | | | |
|--------------------------------|--------------------------|-------------------------|-------------------------|---------------------------|-------------------------|----------------------------|---------------------------|--------------------------|--------------------------|--------------------------|---------------------------|---------------------------|-------------------------|--------------------------|-------------------------------|
| Sample | Ku-97-03 cpx1 core | Ku-97-04 cpx1 rim | Ku-97-92 cpx1 rim | Ku-97-105 cpx1 core | Ku-98-52 cpx2 rim | Ku-98-221a cpx1 core | Ku-98-40 7-cpx1 rim | Ku-98-40 7-cpx2 -> | Ku-97-33 opx1 core | Ku-97-92 opx1 core | Ku-97-104 opx1 core | Ku-97-105 opx1 core | Ku-97-52 opx1 rim | Ku-98-221a opx1 -> | Ku-98-221b opx1 olivine |
| SiO ₂ | 51.04 | 51.37 | 48.95 | 50.30 | 51.11 | 51.29 | 50.22 | 51.08 | 53.70 | 52.19 | 54.52 | 53.13 | 52.02 | 52.74 | 53.36 |
| TiO ₂ | 0.42 | 0.23 | 0.99 | 0.81 | 0.36 | 0.43 | 0.08 | 0.05 | 0.06 | 0.12 | 0.00 | 0.24 | 0.20 | 0.21 | 0.00 |
| Al ₂ O ₃ | 1.71 | 1.55 | 5.45 | 3.16 | 1.67 | 2.27 | 0.62 | 0.41 | 0.22 | 3.24 | 0.68 | 1.58 | 1.55 | 1.59 | 0.48 |
| Cr ₂ O ₃ | 0.01 | 0.00 | 0.09 | 0.14 | 0.03 | 0.00 | 0.02 | 0.01 | 0.03 | 0.01 | 0.01 | 0.02 | 0.01 | 0.00 | 0.01 |
| MgO | 11.54 | 9.84 | 13.03 | 13.35 | 12.38 | 14.21 | 7.96 | 8.69 | 25.84 | 24.73 | 26.13 | 24.75 | 20.98 | 22.91 | 24.99 |
| CaO | 21.96 | 22.53 | 21.99 | 21.97 | 21.78 | 23.11 | 20.99 | 21.50 | 0.09 | 0.59 | 0.12 | 0.76 | 0.86 | 0.96 | 0.14 |
| MnO | 0.29 | 0.47 | 0.21 | 0.22 | 0.53 | 0.22 | 1.49 | 1.49 | 0.37 | 0.31 | 0.29 | 0.40 | 0.68 | 0.39 | 0.44 |
| FeO | 11.56 | 13.11 | 8.06 | 9.00 | 11.22 | 7.99 | 17.23 | 15.62 | 18.34 | 18.58 | 18.59 | 18.29 | 23.45 | 21.58 | 19.62 |
| Na ₂ O | 0.51 | 0.80 | 0.68 | 0.37 | 0.33 | 0.26 | 0.56 | 0.66 | 0.00 | 0.00 | 0.00 | 0.02 | 0.08 | 0.00 | 0.00 |
| K ₂ O | 0.02 | 0.00 | 0.00 | 0.00 | 0.02 | 0.00 | 0.02 | 0.00 | 0.01 | 0.01 | 0.01 | 0.01 | 0.00 | 0.00 | 0.02 |
| Total | 99.07 | 99.89 | 99.44 | 99.33 | 99.43 | 99.78 | 99.18 | 99.50 | 98.66 | 99.78 | 100.35 | 99.19 | 99.82 | 100.37 | 99.05 |
| Formula (O=6) | | | | | | | | | | | | | | | |
| Si | 1.95 | 1.97 | 1.84 | 1.90 | 1.94 | 1.92 | 1.98 | 1.99 | 1.99 | 1.91 | 1.98 | 1.96 | 1.96 | 1.95 | 1.98 |
| Al | 0.05 | 0.03 | 0.16 | 0.10 | 0.06 | 0.08 | 0.02 | 0.01 | 0.01 | 0.09 | 0.02 | 0.04 | 0.04 | 0.05 | 0.02 |
| Sum Z | 2.00 | 2.00 | 2.00 | 2.00 | 2.00 | 2.00 | 2.00 | 2.00 | 2.00 | 2.00 | 2.00 | 2.00 | 2.00 | 2.00 | 2.00 |
| Al | 0.03 | 0.04 | 0.08 | 0.04 | 0.02 | 0.02 | 0.01 | 0.01 | 0.00 | 0.05 | 0.01 | 0.03 | 0.02 | 0.02 | 0.00 |
| Cr | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 |
| Ti | 0.01 | 0.01 | 0.03 | 0.02 | 0.01 | 0.01 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.01 | 0.01 | 0.01 | 0.00 |
| Mg | 0.66 | 0.56 | 0.73 | 0.75 | 0.70 | 0.79 | 0.47 | 0.50 | 1.00 | 0.94 | 0.99 | 0.97 | 0.97 | 0.97 | 1.00 |
| Fe ²⁺ | 0.30 | 0.40 | 0.16 | 0.18 | 0.27 | 0.17 | 0.52 | 0.48 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 |
| Mn | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 |
| Sum Y | 1.00 | 1.00 | 1.00 | 1.00 | 1.00 | 1.00 | 1.00 | 1.00 | 1.00 | 1.00 | 1.00 | 1.00 | 1.00 | 1.00 | 1.00 |
| Mg | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.43 | 0.41 | 0.43 | 0.39 | 0.21 | 0.29 | 0.38 |
| Fe ²⁺ | 0.07 | 0.02 | 0.09 | 0.10 | 0.09 | 0.08 | 0.05 | 0.03 | 0.57 | 0.57 | 0.56 | 0.56 | 0.74 | 0.67 | 0.61 |
| Mn | 0.01 | 0.02 | 0.01 | 0.01 | 0.02 | 0.01 | 0.05 | 0.05 | 0.01 | 0.01 | 0.01 | 0.01 | 0.02 | 0.01 | 0.01 |
| Ca | 0.90 | 0.92 | 0.89 | 0.89 | 0.89 | 0.93 | 0.89 | 0.90 | 0.00 | 0.02 | 0.00 | 0.03 | 0.03 | 0.04 | 0.01 |
| Na | 0.04 | 0.06 | 0.05 | 0.03 | 0.02 | 0.02 | 0.04 | 0.05 | 0.00 | 0.00 | 0.00 | 0.00 | 0.01 | 0.00 | 0.00 |
| K | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 |
| Sum X | 1.02 | 1.02 | 1.04 | 1.02 | 1.02 | 1.03 | 1.03 | 1.02 | 1.01 | 1.01 | 1.00 | 1.00 | 1.01 | 1.01 | 1.01 |
| Total | 4.02 | 4.02 | 4.04 | 4.02 | 4.02 | 4.03 | 4.03 | 4.02 | 4.01 | 4.01 | 4.00 | 4.00 | 4.01 | 4.01 | 4.01 |

Table 3. Representative electron microprobe analyses of amphibole of the anorthosite and the felsic rock suites.

| Rock type | Leucogabbronorite | | | | White anorthosite | | Px anorthosite | | Felsic rock suite | | | | | |
|--------------------------------|-------------------|-----------------|-----------------|-----------------|-------------------|------------------|-----------------|-----------------|-------------------|-------------------|-------------------|-------------------|-------------------|-------------------|
| Sample | Ku-97-03 am1 | Ku-97-03 am8 | Ku-97-04 am1 | Ku-97-04 am4 | Ku-97-08b am1 | Ku-97-08b am3 | Ku-97-13 am1 | Ku-97-13 am2 | Ku-99-11 a-am2 | Ku-99-11 a-am2 | Ku-99-13 3-am1 | Ku-99-13 3-am1 | Ku-99-14 a-am1 | Ku-99-14 a-am1 |
| SiO ₂ | 40.73 | 41.41 | 39.07 | 40.63 | 41.69 | 39.82 | 40.65 | 43.24 | 36.79 | 36.72 | 37.51 | 36.67 | 36.89 | 36.89 |
| TiO ₂ | 3.24 | 2.92 | 3.43 | 2.45 | 0.73 | 1.89 | 0.80 | 0.16 | 1.87 | 1.94 | 0.80 | 0.75 | 0.70 | 0.76 |
| Al ₂ O ₃ | 11.90 | 11.88 | 12.84 | 11.16 | 14.29 | 14.83 | 12.21 | 10.22 | 10.33 | 10.23 | 9.86 | 10.26 | 10.84 | 10.78 |
| Cr ₂ O ₃ | 0.03 | 0.00 | 0.01 | 0.00 | 0.00 | 0.05 | 0.00 | 0.01 | 0.00 | 0.03 | 0.00 | 0.00 | 0.00 | 0.05 |
| Fe ₂ O ₃ | 1.19 | 0.84 | 0.23 | 1.01 | 4.07 | 3.43 | 4.55 | 5.54 | 9.27 | 10.53 | 11.52 | 13.70 | 11.79 | 11.92 |
| FeO | 14.52 | 13.58 | 19.17 | 18.43 | 10.12 | 10.53 | 16.17 | 14.34 | 22.68 | 22.24 | 21.80 | 20.85 | 21.51 | 21.63 |
| MgO | 10.28 | 10.81 | 7.10 | 8.26 | 11.88 | 11.50 | 7.83 | 9.65 | 0.42 | 0.48 | 0.40 | 0.23 | 0.37 | 0.31 |
| MnO | 0.19 | 0.15 | 0.26 | 0.31 | 0.19 | 0.21 | 0.20 | 0.26 | 0.83 | 0.88 | 0.95 | 1.05 | 1.08 | 1.05 |
| CaO | 11.03 | 11.26 | 11.16 | 11.22 | 11.21 | 10.90 | 11.53 | 11.62 | 10.46 | 10.63 | 9.72 | 10.27 | 10.49 | 10.40 |
| Na ₂ O | 2.43 | 2.18 | 2.34 | 2.23 | 2.05 | 2.68 | 1.33 | 1.32 | 1.81 | 1.94 | 2.09 | 2.42 | 2.21 | 2.20 |
| K ₂ O | 1.07 | 0.97 | 1.31 | 1.43 | 1.00 | 0.66 | 1.03 | 0.74 | 1.75 | 1.76 | 1.91 | 1.96 | 2.21 | 2.22 |
| Total | 96.59 | 95.99 | 96.91 | 97.12 | 97.23 | 96.47 | 96.30 | 97.09 | 96.21 | 97.37 | 96.56 | 98.16 | 98.08 | 98.20 |
| Formula (O=23) | | | | | | | | | | | | | | |
| Si | 6.20 | 6.29 | 6.07 | 6.28 | 6.17 | 5.97 | 6.28 | 6.55 | 6.07 | 6.00 | 6.16 | 5.97 | 6.00 | 5.99 |
| Ti | 0.37 | 0.33 | 0.40 | 0.28 | 0.08 | 0.21 | 0.09 | 0.02 | 0.23 | 0.24 | 0.10 | 0.09 | 0.09 | 0.09 |
| Al | 2.13 | 2.12 | 2.35 | 2.03 | 2.49 | 2.62 | 2.22 | 1.82 | 2.01 | 1.97 | 1.91 | 1.97 | 2.08 | 2.06 |
| Cr | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.01 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.01 |
| Fe ³⁺ | 0.14 | 0.10 | 0.03 | 0.12 | 0.45 | 0.39 | 0.53 | 0.63 | 1.15 | 1.29 | 1.42 | 1.68 | 1.44 | 1.46 |
| Fe ²⁺ | 1.85 | 1.72 | 2.49 | 2.38 | 1.25 | 1.32 | 2.09 | 1.82 | 3.13 | 3.04 | 3.00 | 2.84 | 2.92 | 2.94 |
| Mg | 2.33 | 2.45 | 1.64 | 1.90 | 2.62 | 2.57 | 1.80 | 2.18 | 0.10 | 0.12 | 0.10 | 0.06 | 0.09 | 0.08 |
| Mn | 0.02 | 0.02 | 0.03 | 0.04 | 0.02 | 0.03 | 0.03 | 0.03 | 0.12 | 0.12 | 0.13 | 0.14 | 0.15 | 0.14 |
| Ca | 1.80 | 1.83 | 1.86 | 1.86 | 1.78 | 1.75 | 1.91 | 1.89 | 1.85 | 1.86 | 1.71 | 1.79 | 1.83 | 1.81 |
| Na | 0.72 | 0.64 | 0.70 | 0.67 | 0.59 | 0.78 | 0.40 | 0.39 | 0.58 | 0.61 | 0.66 | 0.76 | 0.70 | 0.69 |
| K | 0.21 | 0.19 | 0.26 | 0.28 | 0.19 | 0.13 | 0.20 | 0.14 | 0.37 | 0.37 | 0.40 | 0.41 | 0.46 | 0.46 |
| Total | 15.76 | 15.69 | 15.83 | 15.84 | 15.66 | 15.76 | 15.55 | 15.47 | 15.60 | 15.62 | 15.60 | 15.70 | 15.74 | 15.73 |
| X _{Mg} | 0.56 | 0.59 | 0.40 | 0.44 | 0.68 | 0.66 | 0.46 | 0.55 | 0.03 | 0.04 | 0.03 | 0.02 | 0.03 | 0.02 |

Table 4. Selected major and trace element data of the anorthosite and the felsic rock suites.

| Rock type Sample | A,w 97-33b | A,w 97-44 | A,w 98-45 | A,px 97-13 | A,px 97-31 | A,ol 97-28 | A,ol 98-23 | GN 97-04 | GN 97-05 | GN 97-31b | GN 98-68 | T 97-92 | T 98-41 | T 98-230 | T _{grt} 97-104 | G 98-180a | S 98-40 | S 99-20 | S Sy-25 | SD,a 98-201 | SD,a 98-200 | SD,a 98-204 |
|--------------------------------|---------------|--------------|--------------|---------------|---------------|---------------|---------------|-------------|-------------|--------------|-------------|------------|------------|-------------|----------------------------|--------------|------------|------------|------------|----------------|----------------|----------------|
| <i>wt. %</i> | | | | | | | | | | | | | | | | | | | | | | |
| SiO ₂ | 47.51 | 51.03 | 49.22 | 50.41 | 46.23 | 49.87 | 49.77 | 43.91 | 47.95 | 37.67 | 51.09 | 49.82 | 49.47 | 49.41 | 48.93 | 75.56 | 65.80 | 61.64 | 59.29 | 56.51 | 62.00 | 51.97 |
| TiO ₂ | 0.11 | 0.38 | 0.13 | 1.29 | 2.39 | 0.12 | 0.19 | 4.24 | 2.95 | 7.25 | 0.27 | 0.11 | 0.13 | 0.08 | 0.15 | 0.25 | 0.25 | 0.61 | 0.61 | 1.16 | 0.05 | 0.04 |
| Al ₂ O ₃ | 27.27 | 25.78 | 28.85 | 23.53 | 22.79 | 26.27 | 27.77 | 15.85 | 18.91 | 17.26 | 22.41 | 22.89 | 25.70 | 26.38 | 24.52 | 11.27 | 16.44 | 17.52 | 13.80 | 14.18 | 20.11 | 23.43 |
| Fe ₂ O ₃ | 0.77 | 1.53 | 0.85 | 1.62 | 2.25 | 1.20 | 0.90 | 4.00 | 2.60 | 3.92 | 1.35 | 1.54 | 0.86 | 0.93 | 1.31 | 1.53 | 1.87 | 2.08 | 6.38 | 14.16 | 2.95 | 4.02 |
| FeO | 0.11 | 0.22 | 0.25 | 3.50 | 4.87 | 2.59 | 2.05 | 9.24 | 7.54 | 11.38 | 3.95 | 4.81 | 3.80 | 2.91 | 3.71 | 1.38 | 1.02 | 2.52 | n.a. | n.a. | n.a. | n.a. |
| MnO | 0.03 | 0.04 | 0.03 | 0.08 | 0.08 | 0.05 | 0.04 | 0.25 | 0.25 | 0.18 | 0.05 | 0.07 | 0.07 | 0.05 | 0.07 | 0.05 | 0.05 | 0.09 | 0.15 | 0.22 | 0.04 | 0.14 |
| MgO | 0.37 | 0.59 | 0.20 | 1.92 | 2.57 | 3.56 | 2.81 | 4.42 | 2.82 | 5.25 | 2.77 | 6.46 | 5.43 | 5.25 | 5.00 | 0.12 | 0.30 | 0.80 | 1.41 | 0.82 | 0.41 | 0.05 |
| CaO | 9.84 | 10.35 | 13.43 | 8.81 | 8.86 | 11.08 | 12.03 | 10.14 | 8.04 | 7.52 | 9.42 | 8.96 | 11.22 | 11.78 | 10.76 | 0.33 | 2.00 | 3.32 | 4.50 | 4.03 | 2.83 | 9.99 |
| Na ₂ O | 4.65 | 6.13 | 4.40 | 4.91 | 4.62 | 3.73 | 3.45 | 3.35 | 4.47 | 3.05 | 4.80 | 3.80 | 3.19 | 3.14 | 3.28 | 3.11 | 6.07 | 6.76 | 6.15 | 7.20 | 9.68 | 3.62 |
| K ₂ O | 2.80 | 0.71 | 0.52 | 1.25 | 0.60 | 0.27 | 0.23 | 1.27 | 1.78 | 0.44 | 0.77 | 0.51 | 0.27 | 0.17 | 0.36 | 5.28 | 4.33 | 3.34 | 4.03 | 0.19 | 0.07 | 4.06 |
| P ₂ O ₅ | 0.01 | 0.11 | 0.01 | 0.20 | 0.06 | 0.01 | 0.03 | 0.16 | 0.12 | 0.03 | 0.45 | 0.02 | 0.01 | 0.01 | 0.01 | 0.12 | 0.05 | 0.19 | 0.18 | 0.27 | 0.14 | 0.03 |
| S | b.d.l. | b.d.l. | 0.02 | 0.04 | 0.08 | b.d.l. | 0.02 | 0.16 | 0.06 | 0.19 | 0.02 | b.d.l. | 0.02 | 0.02 | b.d.l. | b.d.l. | b.d.l. | b.d.l. | <0.02 | 0.03 | <0.02 | <0.02 |
| GV | 5.83 | 3.09 | 2.30 | 1.45 | 3.61 | 0.85 | 1.26 | 1.61 | 1.68 | 4.73 | 2.15 | 0.42 | 0.27 | 0.31 | 1.21 | 0.37 | 0.95 | 0.62 | 3.17 | 0.40 | 0.79 | 1.66 |
| CO ₂ | n.a. | n.a. | n.a. | n.a. | n.a. | n.a. | n.a. | n.a. | n.a. | n.a. | n.a. | n.a. | n.a. | n.a. | n.a. | n.a. | 0.65 | 0.42 | n.a. | n.a. | n.a. | n.a. |
| H ₂ O | n.a. | n.a. | n.a. | n.a. | n.a. | n.a. | n.a. | n.a. | n.a. | n.a. | n.a. | n.a. | n.a. | n.a. | n.a. | n.a. | 0.30 | 0.20 | n.a. | n.a. | n.a. | n.a. |
| Total | 99.31 | 99.97 | 100.22 | 99.01 | 99.01 | 99.60 | 100.54 | 98.60 | 99.17 | 98.88 | 99.50 | 99.42 | 100.44 | 100.44 | 99.32 | 99.37 | 99.12 | 99.49 | 99.67 | 99.17 | 99.07 | 99.01 |
| <i>ppm</i> | | | | | | | | | | | | | | | | | | | | | | |
| Sc | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | 15 | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | 17 | 13 | 22 | 2.4 | 0.92 |
| V | 17 | 21 | 22 | 72 | 148 | 16 | 23 | 296 | 182 | 317 | 39 | 13 | 25 | 12 | 25 | 10 | b.d.l. | 19 | 76 | 17 | 24 | 13 |
| Cr | b.d.l. | 13 | b.d.l. | 11 | 46 | 12 | 15 | 36 | 25 | 70 | 12 | 24 | 24 | 27 | 46 | b.d.l. | b.d.l. | b.d.l. | 20 | <10 | <10 | <10 |
| Co | 21 | 53 | 54 | 56 | 56 | 45 | 62 | 47 | 96 | 46 | 104 | 58 | 45 | 64 | 71 | 138 | 43 | <10 | 12 | <10 | <10 | <10 |
| Ni | b.d.l. | 26 | b.d.l. | 23 | 70 | 94 | 75 | 36 | 34 | 159 | 36 | 187 | 157 | 185 | 119 | b.d.l. | b.d.l. | 6 | <5 | <5 | <5 | <5 |
| Zn | 45 | 16 | b.d.l. | 59 | 92 | 33 | 18 | 118 | 83 | 138 | 44 | 52 | 33 | 22 | 39 | 99 | 62 | 54 | 82 | 532 | 32 | 37 |
| Ga | b.d.l. | b.d.l. | 19 | b.d.l. | b.d.l. | b.d.l. | 29 | b.d.l. | b.d.l. | b.d.l. | 17 | b.d.l. | 25 | 26 | b.d.l. | 34 | 31 | 42 | 9 | 26 | 31 | 43 |
| Rb | 69 | 9 | b.d.l. | 21 | 9 | 5 | b.d.l. | 28 | 33 | 10 | 12 | 5 | b.d.l. | b.d.l. | 6 | 121 | 116 | 82 | 88 | 7 | <5 | 109 |
| Sr | 805 | 805 | 627 | 685 | 642 | 475 | 566 | 447 | 510 | 423 | 979 | 535 | 520 | 543 | 470 | 26 | 201 | 266 | 674 | 199 | 1309 | 4514 |
| Y | 13 | 11 | b.d.l. | 10 | 8 | 8 | b.d.l. | 19 | 16 | 8 | 13 | 8 | b.d.l. | b.d.l. | 8 | 107 | 79 | 84 | 33 | 72 | 62 | 19 |
| Zr | 28 | 59 | b.d.l. | 63 | 48 | 19 | b.d.l. | 104 | 99 | 107 | 39 | 18 | b.d.l. | b.d.l. | 19 | 617 | 694 | 1854 | 284 | 791 | 977 | 284 |
| Nb | 10 | 7 | b.d.l. | 5 | 5 | 5 | b.d.l. | 15 | 17 | 13 | b.d.l. | 5 | b.d.l. | b.d.l. | 5 | 63 | 55 | 34 | 138 | 68 | 163 | 32 |
| Mo | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | 239 | b.d.l. | b.d.l. | 7 | 13 | b.d.l. | b.d.l. |
| Ba | 1510 | 555 | 236 | 706 | 409 | 121 | 134 | 567 | 662 | 355 | 773 | 245 | 124 | 99 | 164 | b.d.l. | 1448 | 900 | 543 | 792 | 194 | 1013 |
| Pb | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | 8 | b.d.l. | b.d.l. | b.d.l. | b.d.l. | 25 | 29 | 25 | 28 | 14 | 21 | 59 |
| Th | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | 13 | b.d.l. | b.d.l. | b.d.l. | 121 | 19 |
| U | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | 16 | 11 |
| Ce | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | 150 | 184 | b.d.l. | b.d.l. | b.d.l. | b.d.l. |
| Nd | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | 70 | 91 | b.d.l. | b.d.l. | b.d.l. | b.d.l. |

Abbreviations: *A,w* white anorthosite; *A,ol* olivine anorthosite; *A,px* pyroxene anorthosite; *b.d.l.* below detection limit; *G* granite; *GN* leucogabbronorite; *n.a.* not analysed; *S* syenite; *SD* syenodiorite.

Table 5. Rare earth element data of the anorthosite and the felsic rock suites.

| Rock type | A,w | A,w | A,w | A,px | A,px | A,ol | A,ol | T | T | T | T |
|------------------------|--------|-------|--------|-------|--------|--------|--------|--------|--------|--------|--------|
| Sample | 97-33b | 97-44 | 98-45 | 97-13 | 97-31 | 97-28 | 98-23 | 97-92 | 97-104 | 98-41 | 98-230 |
| ppm | | | | | | | | | | | |
| Y | 11.00 | 8.30 | 1.10 | 6.10 | 3.10 | 1.30 | 1.20 | 0.40 | 1.20 | 1.20 | 0.43 |
| La | 8.10 | 14.00 | 3.20 | 7.40 | 4.40 | 2.90 | 1.90 | 2.00 | 2.20 | 1.80 | 1.10 |
| Ce | 17.00 | 27.00 | 5.50 | 16.00 | 8.70 | 5.70 | 3.60 | 3.20 | 4.20 | 3.90 | 2.00 |
| Pr | 2.10 | 3.20 | 0.62 | 2.00 | 1.10 | 0.64 | 0.50 | 0.37 | 0.43 | 0.43 | 0.25 |
| Nd | 8.30 | 12.00 | 2.30 | 8.20 | 4.30 | 2.50 | 2.00 | 1.40 | 1.80 | 1.70 | 1.10 |
| Sm | 2.00 | 2.50 | 0.41 | 1.70 | 0.89 | 0.49 | 0.40 | 0.27 | 0.38 | 0.34 | 0.16 |
| Eu | 1.00 | 1.70 | 0.91 | 1.60 | 1.08 | 0.63 | 0.63 | 0.71 | 0.69 | 0.54 | 0.44 |
| Gd | 1.90 | 2.10 | 0.37 | 1.60 | 0.76 | 0.39 | 0.35 | 0.14 | 0.34 | 0.31 | 0.13 |
| Tb | 0.28 | 0.30 | b.d.l. | 0.23 | 0.12 | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. |
| Dy | 1.90 | 1.70 | 0.21 | 1.30 | 0.63 | 0.28 | 0.27 | 0.12 | 0.27 | 0.25 | 0.10 |
| Ho | 0.36 | 0.31 | b.d.l. | 0.21 | b.d.l. | 0.06 | b.d.l. | b.d.l. | b.d.l. | b.d.l. | b.d.l. |
| Er | 1.10 | 0.89 | 0.08 | 0.62 | 0.34 | 0.14 | b.d.l. | 0.07 | b.d.l. | b.d.l. | b.d.l. |
| Tm | 0.16 | 0.12 | b.d.l. | 0.09 | 0.05 | b.d.l. | b.d.l. | b.d.l. | b.d.l. | 0.02 | b.d.l. |
| Yb | 0.93 | 0.75 | 0.08 | 0.56 | 0.30 | 0.12 | 0.10 | 0.05 | 0.12 | 0.12 | 0.05 |
| Lu | 0.13 | 0.12 | 0.02 | 0.08 | 0.05 | b.d.l. | 0.02 | 0.01 | 0.02 | 0.02 | b.d.l. |
| Sc | 2.10 | 5.20 | 0.35 | 6.00 | 5.60 | 1.60 | 1.70 | 1.60 | 3.30 | 3.00 | 1.80 |
| Total | 58.36 | 80.19 | 15.15 | 53.69 | 31.42 | 16.75 | 12.67 | 10.34 | 14.95 | 13.63 | 7.56 |
| Σ REE (ppm) | 45.26 | 66.69 | 13.70 | 41.59 | 22.72 | 13.85 | 9.77 | 8.34 | 10.45 | 9.43 | 5.33 |
| (Eu/Eu*) _{CN} | 1.57 | 2.27 | 7.14 | 2.96 | 4.01 | 4.40 | 5.15 | 11.16 | 5.87 | 5.08 | 9.32 |
| (La/Yb) _{CN} | 5.87 | 12.58 | 26.97 | 8.91 | 9.89 | 16.29 | 12.81 | 26.97 | 12.36 | 10.11 | 14.83 |
| (La/Nd) _{CN} | 1.89 | 2.26 | 2.69 | 1.75 | 1.98 | 2.24 | 1.84 | 2.76 | 2.37 | 2.05 | 1.94 |

| Rock type | GN | GN | GN | GN | S | S | S | SD,a | SD,a | SD,a | SD,a |
|------------------------|--------|--------|--------|--------|--------|--------|--------|--------|---------|--------|---------|
| Sample | 97-4 | 97-5 | 97-31b | 98-68 | 98-40 | 99-20 | Sy-25 | 98-201 | 98-103s | 98-204 | 98-200 |
| ppm | | | | | | | | | | | |
| Y | 19.00 | 13.00 | 1.40 | 15.00 | 72.00 | 46.00 | 33.00 | 72.00 | 46.00 | 19.00 | 62.00 |
| La | 9.30 | 11.00 | 3.00 | 13.00 | 80.00 | 73.00 | 60.00 | 93.00 | 227.00 | 107.00 | 437.00 |
| Ce | 23.00 | 24.00 | 5.90 | 30.00 | 145.00 | 123.00 | 139.00 | 201.00 | 366.00 | 182.00 | 530.00 |
| Pr | 3.20 | 3.30 | 0.60 | 4.10 | 18.00 | 16.00 | 18.00 | 29.00 | 37.00 | 19.00 | 74.00 |
| Nd | 15.00 | 14.00 | 2.40 | 18.00 | 69.00 | 63.00 | 84.00 | 126.00 | 129.00 | 63.00 | 229.00 |
| Sm | 3.90 | 3.30 | 0.42 | 4.00 | 13.00 | 12.00 | 16.00 | 26.00 | 18.00 | 9.00 | 33.00 |
| Eu | 1.70 | 1.90 | 0.80 | 2.00 | 4.30 | 3.80 | 4.60 | 6.80 | 3.60 | 2.00 | 6.50 |
| Gd | 4.10 | 3.20 | 0.39 | 3.30 | 13.00 | 10.00 | 12.00 | 21.00 | 13.00 | 6.30 | 21.00 |
| Tb | 0.66 | 0.50 | - | 0.54 | 2.20 | 1.60 | 1.50 | 3.40 | 2.00 | 0.79 | 2.70 |
| Dy | 3.80 | 2.70 | 0.30 | 3.20 | 13.00 | 9.00 | 7.90 | 17.00 | 9.60 | 4.40 | 12.00 |
| Ho | 0.75 | 0.52 | - | 0.59 | 2.90 | 1.90 | 1.30 | 3.20 | 1.60 | 0.81 | 2.30 |
| Er | 2.10 | 1.40 | 0.21 | 1.60 | 8.20 | 4.90 | 3.20 | 8.20 | 5.00 | 2.20 | 5.30 |
| Tm | 0.32 | 0.20 | 0.04 | 0.24 | 1.30 | 0.68 | 0.41 | 1.10 | b.d.l. | 0.33 | 0.83 |
| Yb | 1.90 | 1.30 | 0.26 | 1.40 | 9.20 | 5.20 | 2.40 | 7.30 | 5.00 | 2.00 | 4.90 |
| Lu | 0.29 | 0.20 | 0.05 | 0.21 | 1.40 | 0.80 | 0.32 | 1.20 | 0.77 | 0.29 | 0.68 |
| Sc | 44.00 | 22.00 | 13.00 | 17.00 | 7.20 | 5.90 | 13.00 | 22.00 | 4.80 | 0.92 | 2.40 |
| Total | 133.02 | 102.52 | 28.77 | 114.18 | 459.70 | 376.78 | 396.63 | 638.20 | 868.37 | 419.04 | 1423.61 |
| Σ REE (ppm) | 70.02 | 67.52 | 24.37 | 82.18 | 380.50 | 324.88 | 350.63 | 544.20 | 817.57 | 399.12 | 1359.21 |
| (Eu/Eu*) _{CN} | 1.30 | 1.79 | 6.04 | 1.68 | 1.01 | 1.06 | 1.02 | 0.89 | 0.72 | 0.81 | 0.75 |
| (La/Yb) _{CN} | 3.30 | 5.70 | 7.78 | 6.26 | 5.86 | 9.46 | 16.85 | 8.59 | 30.61 | 36.07 | 60.13 |
| (La/Nd) _{CN} | 1.20 | 1.52 | 2.42 | 1.40 | 2.24 | 2.24 | 1.38 | 1.43 | 3.40 | 3.40 | 3.40 |

Abbreviations: *A,w* white anorthosite; *A,ol* olivine anorthosite; *A,px* pyroxene anorthosite; *GN* leucogabbroite; *S* syenite; *SD,a* altered syenodiorite; *T* leucotroctolite.

Table 6. U-Pb analytical results for zircon from syenodiorite Ku-98-201, Swartbooisdrif, Namibia.

| Sample ^a | Weight (mg) | Concentration (ppm) | | ²⁰⁶ Pb | Radiogenic Pb | | | Atomic ratios ^c | | | Apparent ages (Ma) ^d | | |
|---------------------|-------------|---------------------|-------------------|------------------------------|-------------------|-------------------|-------------------|----------------------------|-------------------|-------------------|---------------------------------|-------------------|-------------------|
| | | U | Pb _{tot} | | ²⁰⁴ Pb | ²⁰⁶ Pb | ²⁰⁷ Pb | ²⁰⁸ Pb | ²⁰⁶ Pb | ²⁰⁷ Pb | ²⁰⁷ Pb | ²⁰⁶ Pb | ²⁰⁷ Pb |
| | | | | Measured ratios ^b | | | | ²³⁸ U | ²³⁵ U | ²⁰⁶ Pb | ²³⁸ U | ²³⁵ U | ²⁰⁶ Pb |
| 1 | 0.123 | 172 | 47.6 | 3560 | 73.3 | 6.43 | 20.27 | 0.2333 | 2.8198 | 0.0877 | 1352 | 1361 | 1375 |
| 2 | 0.194 | 135 | 41.3 | 370.8 | 74.37 | 6.52 | 19.11 | 0.2305 | 2.7861 | 0.0877 | 1337 | 1352 | 1375 |
| 3 | 0.145 | 167 | 46.5 | 658 | 73.92 | 6.49 | 19.59 | 0.2215 | 2.6792 | 0.0877 | 1290 | 1323 | 1377 |
| 4 | 0.106 | 161 | 45.1 | 1060 | 73.64 | 6.46 | 19.9 | 0.2292 | 2.7740 | 0.0878 | 1331 | 1349 | 1377 |

^a Zircon concentrates were obtained using standard mineral-separation procedures and purified through separation by hand under the binocular microscope. All analyzed zircon crystals had a prismatic habit, were pink, and smaller than 100 μm. ²⁰⁵Pb-²³⁵U mixed tracer was added before sample dissolution. All samples were dissolved with 52% HF in autoclaves at 220°C for four days, dried, and transferred overnight into chloride-form using 6N HCl. Ion-exchange chromatography follows the procedure described by Krogh

^b Lead isotope ratios corrected for fractionation with 0.1% / a.m.u.

^c Lead corrected for fractionation, blank, tracer contribution, and initial lead. During the measurement period

^d Apparent ages were calculated using the constants of Jaffey et al. (1971) recommended by IUGS (Steiger

Table 7. Oxygen isotopic data of feldspar of the anorthosite and the felsic rock suites.

| Rock type | Sample | Mineral | Textural relationship | $\delta^{18}\text{O}_{\text{SMOW}}$ mineral |
|----------------------|-----------|-------------|-----------------------|--|
| White anorthosite | Ku-97-33 | plagioclase | cumulus mineral | 5.35 |
| | Ku-97-33b | plagioclase | cumulus mineral | 6.14 |
| | Ku-97-33a | plagioclase | cumulus mineral | 7.30 |
| | Ku-98-45 | plagioclase | cumulus mineral | 2.36 |
| | Ku-98-60 | plagioclase | cumulus mineral | 5.54 |
| Pyroxene anorthosite | Ku-97-31 | plagioclase | cumulus mineral | 6.10 |
| | Ku-97-92a | plagioclase | cumulus mineral | 5.83 |
| Leucogabbronorite | Ku-98-128 | plagioclase | cumulus mineral | 5.61 |
| | Ku-98-222 | plagioclase | cumulus mineral | 5.99 |
| Leucotroctolite | Ku-97-95 | plagioclase | cumulus mineral | 5.90 |
| | Ku-97-95 | olivine | interstitial | 3.07 |
| | Ku-97-105 | plagioclase | cumulus mineral | 3.19 |
| | Ku-98-125 | plagioclase | cumulus mineral | 6.13 |
| | Ku-98-220 | plagioclase | cumulus mineral | 5.63 |
| Felsic rock suite | Ku-98-40 | K-feldspar | phenocryst | 7.22 |
| | Ku-98-40 | plagioclase | phenocryst | 7.20 |
| | Ku-99-13 | K-feldspar | phenocryst | 7.50 |
| | Ku-99-20 | K-feldspar | phenocryst | 7.92 |
| | Ku-99-21 | K-feldspar | phenocryst | 7.36 |
| | Ku-99-21 | plagioclase | phenocryst | 7.52 |

Table 8. Sr and Nd isotope data for anorthosites and felsic rocks from NW Namibia.

| Sample ^a | Region | GPS-coord. | Age ^b (Ma) | Sm ^c (ppm) | Nd ^c (ppm) | ¹⁴³ Nd/ ¹⁴⁴ Nd ^d (meas) | ¹⁴³ Nd/ ¹⁴⁴ Nd ^e (T) | εNd ^e (T) | Rb ^c (ppm) | Sr ^c (ppm) | ⁸⁷ Sr/ ⁸⁶ Sr ^d (meas) | ⁸⁷ Sr/ ⁸⁶ Sr ^e (T) | εSr ^e (T) | Nd model age (Ma) |
|--|------------|--------------------|--------------------------|--------------------------|--------------------------|---|--|-------------------------|--------------------------|--------------------------|---|--|-------------------------|----------------------|
| <i>Anorthosite (NW Namibia)</i> | | | | | | | | | | | | | | |
| White anorthosite | | | | | | | | | | | | | | |
| Ku-97-30 | Mine | 19.505'' .48.110'' | 1385 | 0.38 | 2.30 | 0.511810±5 | 0.51090 | 1.1 | 15.0 | 581 | 0.705317±7 | 0.70385 | 14 | 1713 |
| Ku-97-33a | Mine | 19.505'' .48.110'' | 1385 | 0.26 | 1.40 | 0.511842±5 | 0.51095 | 2.0 | 10.0 | 480 | 0.705375±7 | 0.70413 | 18 | 1865 |
| Dark anorthosite | | | | | | | | | | | | | | |
| Ku-97-105 | Zebra Mts. | 25.100'' .38.500'' | 1385 | 0.33 | 1.70 | 0.512021±5 | 0.51097 | 2.3 | 1.9 | 389 | 0.703423±7 | 0.70314 | 4 | 1692 |
| Ku-98-219 | Zebra Mts. | 25.799'' .40.032'' | 1385 | 0.54 | 2.50 | 0.512132±5 | 0.51095 | 1.9 | 4.9 | 450 | 0.704266±7 | 0.70364 | 11 | 1754 |
| Ku-98-227 | Zebra Mts. | 23.899'' .32.546'' | 1385 | 0.12 | 0.69 | 0.511909±5 | 0.51094 | 1.7 | 0.7 | 500 | 0.703113±7 | 0.70303 | 2 | 1660 |
| Ku-98-236 | Zebra Mts. | 09.705'' .33.700'' | 1385 | 0.17 | 1.00 | 0.511919±5 | 0.51100 | 3.0 | 0.9 | 468 | 0.703052±7 | 0.70294 | 1 | 1613 |
| Ku-98-208 | Mine | 20.861'' .46.231'' | 1385 | 0.35 | 1.60 | 0.512004±5 | 0.51096 | 2.2 | 2.1 | 515 | 0.703042±7 | 0.70284 | -1 | 2010 |
| <i>Felsic rock suite (Swarthoisdrijf area, NW Namibia)</i> | | | | | | | | | | | | | | |
| Altered felsic rocks | | | | | | | | | | | | | | |
| Ku-98-201 | Mine | 21.367'' .47.170'' | 1377 | 26.0 | 126 | 0.512006±5 | 0.51088 | 0.4 | 1.5 | 197 | 0.707598±7 | 0.70718 | 61 | 1848 |
| Ku-98-204 | Mine | 20.954'' .47.155'' | 1377 | 8.30 | 61.0 | 0.511733±5 | 0.51100 | 2.5 | 97.0 | 5000 | 0.708678±7 | 0.70761 | 67 | 1578 |
| Ku-98-200 | Mine | 21.353'' .47.109'' | 1377 | 33.0 | 229 | 0.511776±5 | 0.51100 | 2.5 | 0.9 | 1301 | 0.706312±7 | 0.70627 | 48 | 1587 |
| Felsic rocks unaffected by an alteration | | | | | | | | | | | | | | |
| Ku-98-202 | Mine | 21.407'' .47.161'' | 1377 | 27.0 | 124 | 0.512029±5 | 0.51084 | -0.4 | 18.0 | 541 | 0.706138±7 | 0.70430 | 20 | 1953 |
| Ku-Sy-25 | Mine | 19.505'' .48.110'' | 1377 | 16.0 | 80.0 | 0.512067±5 | 0.51097 | 2.2 | 90.0 | 702 | 0.709539±7 | 0.70245 | -6 | 1682 |

^a Whole rock samples. Location as indicated in Fig. 2. Samples were dissolved with 52% HF for four days at 160°C on the hot plate. Digested samples were dried and taken up in 6N HCl. Sr and Nd were separated and purified using ion-exchange chromatography as described in Romer et al. (2005).

^b Age from Fig. 6 and Drüppel et al. (2000).

^c Concentrations of Nd and Sm were determined by ICP-MS using the procedures of Dulski (1994); Rb and Sr were determined by ICP-MS.

^d The Nd isotopic composition was determined on a Finnigan MAT262 multi-collector mass-spectrometer using dynamic multicollection. Individual ratios were normalized to ¹⁴⁶Nd/¹⁴⁴Nd = 0.7219. The Sr isotopic composition was determined on VG Sector 54-30 multi-collector mass-spectrometer using dynamic multicollection. Individual ratios were normalized to ⁸⁶Sr/⁸⁸Sr = 0.1194 and ¹⁴⁶Nd/¹⁴⁴Nd = 0.7219; analytical uncertainties are given at 2σ_m level. During the measurement period, reference material La Jolla and NBS 987 gave values of 0.511855 ± 8 (2σ, n = 15 samples) and 0.710246 ± 10 (2σ, n = 20 samples), respectively.

^e Initial compositions and ε values of Nd and Sr were calculated using the age of the complex and the following parameters and constants: ¹⁴⁷Smλ = 6.54E-12 y⁻¹; ¹⁴³Nd/¹⁴⁴Nd_{CHUR(0)} = 0.512638; ¹⁴⁷Sm/¹⁴⁴Nd_{CHUR(0)} = 0.1967; ⁸⁷Rbλ = 1.42E-11 y⁻¹; ⁸⁷Sr/⁸⁶Sr_{CHUR(0)} = 0.7045; ⁸⁷Rb/⁸⁶Sr_{CHUR(0)} = 0.0816. ε notation as in Jacobsen and Wasserburg (1979)