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1 **A synthesis of methane dynamics in thermokarst lake environments**

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17 **Abstract**

18 Greenhouse gas emissions from physical permafrost thaw disturbance and subsidence, including
19 the formation and expansion of thermokarst (thaw) lakes, may double the magnitude of the
20 permafrost carbon feedback this century. These processes are not accounted for in current global
21 climate models. Thermokarst lakes, in particular, have been shown to be hotspots for emissions
22 of methane (CH₄), a potent greenhouse gas with 32 times more global warming potential than
23 carbon dioxide (CO₂) over a 100-year timescale. Here, we synthesize several studies examining
24 CH₄ dynamics in a representative first-generation thermokarst lake (Vault Lake, informal name)
25 to show that CH₄ production and oxidation potentials vary with depth in thawed sediments
26 beneath the lake. This variation leads to depth-dependent differences in both *in situ* dissolved
27 CO₂:CH₄ ratios and net CH₄ production responses to additional warming. Comparing CH₄
28 production, oxidation, and flux values from studies at Vault Lake suggests up to 99% of
29 produced CH₄ is oxidized and/or periodically entrapped before entering the atmosphere. We
30 summarize these findings in the context of CH₄ literature from thermokarst lakes and identify
31 future research directions for incorporating thermokarst lake CH₄ dynamics into estimates of the
32 permafrost carbon feedback.

33 **Keywords**

34 methane, methane oxidation, permafrost carbon feedback, rapid thaw

35 **1. Introduction**

36 The Arctic is currently experiencing a geologically abrupt climate warming event where major
37 portions have warmed by more than 1 °C per decade during the past 40 years (Jansen et al.,
38 2020). Temperatures in the Arctic are anticipated to increase a further 4-7 °C during the 21st
39 century (Post et al., 2019). Permafrost regions, which represent around a quarter of the Arctic,
40 are estimated to contain almost 1700 Pg of soil organic carbon (SOC) (Schuur et al., 2015),
41 making them a globally important carbon (C) stock. Permafrost warming and thaw, triggered by
42 climate warming, is expected to mobilize ~288 Pg of permafrost SOC by 2300 (Turetsky et al.,
43 2020). A portion of this mobilized permafrost C will be microbially mineralized into the
44 greenhouse gasses (GHG) methane (CH₄), carbon dioxide (CO₂), and nitrous oxide (N₂O), which
45 will contribute a positive feedback to climate warming called the permafrost carbon feedback
46 (PCF) and potentially turn the Arctic to a net C source to the atmosphere (Schuur et al., 2015;
47 Voigt et al., 2020). Estimates of the PCF suggest that C released from thawing permafrost can
48 increase global climate warming by 0.13-0.27 °C by 2100 (Schuur et al., 2015). However, these
49 models are based on gradual, top-down permafrost thaw (e.g. active layer deepening). The
50 impact of comparatively rapid physical disturbance thaw processes (e.g. thermokarst and coastal
51 erosion) are not accounted for in current estimates of the PCF (Tanski et al., 2019; Turetsky et
52 al., 2019, 2020).

53 In the Arctic, lakes have been identified as hotspots where terrestrial C, including C from
54 thawed permafrost, is mineralized and emitted as CH₄ and CO₂ (Kuhn et al., 2018; Walter et al.,
55 2006; Wik et al., 2016). Lakes are a common feature in northern latitudes; by surface area,
56 almost half of the world's lakes are located in Arctic and sub-Arctic regions (Smith et al., 2007;
57 Grosse et al., 2013). Thermokarst (thaw) lake landscapes are widespread in permafrost-underlain

58 regions and cover 1.3 million km², representing 20-40% of the permafrost region and 15-75% of
59 arctic lowland surface area (Olefeldt et al 2016). These lakes usually form when ice-rich
60 permafrost thaws or massive ice melts (thermokarst) in areas with low relief, thick
61 unconsolidated sediments, and high ground ice contents (Jones et al., 2011); the formation of this
62 type of lake is expected to become more widespread with climate change (Grosse et al., 2013;
63 Jones et al., 2011). Once formed, thermokarst lake expansion can rapidly introduce new C into
64 the active C cycle. Talik (thaw bulb) development beneath the lake can thaw and mobilize older,
65 deeper C that otherwise would not thaw during this century, even under the most severe active
66 layer deepening scenarios (Langer et al., 2016; Walter Anthony et al., 2018).

67 Thermokarst lakes currently contribute almost a quarter ($4.1 \pm 2.2 \text{ Tg CH}_4 \text{ yr}^{-1}$; Wik et al.,
68 2016) of the total CH₄ emitted annually from northern lakes. CH₄ and CO₂ emissions from
69 thermokarst lakes are expected to increase five-fold during the coming century, and models
70 suggest that accounting for GHG produced in and released from thermokarst lakes will more
71 than double the PCF this century (Schneider von Deimling et al., 2015; Walter Anthony et al.,
72 2018). Much of the C used to produce GHG originates from the surrounding terrestrial
73 environment, including reworked eroded and deposited sediments and permafrost thawing
74 beneath the lake. Carbon mass balance measurements of paleo thermokarst lake taliks by Walter
75 Anthony et al. (2014) suggest $28 \pm 12\%$ of thawed C beneath thermokarst lakes is converted to
76 GHG over the lakes' millennial-scale lifespans. Additional, modern C can be contributed by
77 primary productivity and vegetation within and surrounding the lakes (Dean et al., 2020; Elder et
78 al., 2018).

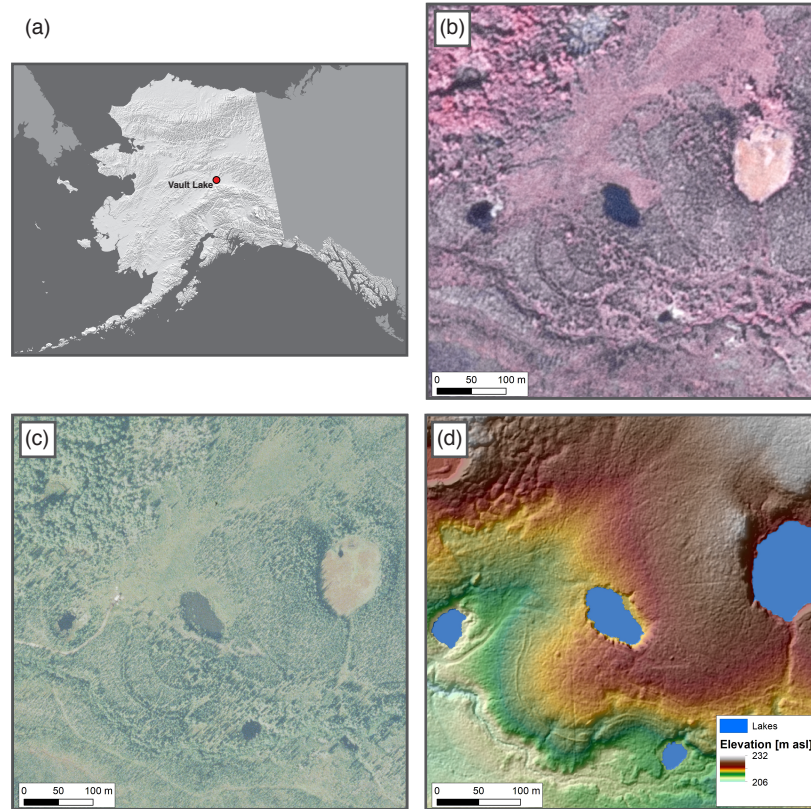
79 Understanding CH₄ emissions from thermokarst lakes is particularly important given that
80 CH₄ has 96x more global warming potential (GWP) than CO₂ over a 20-year timescale and 32x

81 more GWP than CO₂ over a 100-year timescale (Alvarez et al., 2018; Etminan et al., 2016). One
82 approach towards constraining estimates of CH₄ emissions in these rapidly changing landscapes
83 is examining CH₄ emission, production, and oxidation dynamics in thermokarst lake
84 environments. Here, we synthesize research conducted on CH₄ dynamics at a thermokarst lake in
85 central Alaska to provide insights into whole-lake CH₄ cycling in a thermokarst lake system and
86 the role CH₄ emissions from thermokarst lakes may play in the PCF. To our best knowledge, this
87 is the most detailed biogeochemical and microbiological study of a sediment core capturing the
88 full depth profile of a thermokarst lake talik. We used sediment from a lake core collected from
89 the center of Vault Lake (informal name) and extending through the entire talik to examine: (1)
90 CH₄ production potentials (Heslop et al., 2015); (2) aerobic (Martinez-Cruz et al., 2015) and
91 anaerobic (Martinez-Cruz et al., 2017; Winkel et al., 2019) CH₄ oxidization; (3) temperature
92 sensitivities of CH₄ production (Heslop et al., 2019a); and (4) CH₄-producing and oxidizing
93 microbial communities (Heslop et al., 2019a; Martinez-Cruz et al., 2018; Winkel et al., 2019).
94 We also measured *in situ* sediment CH₄ concentrations (Winkel et al., 2019) and the whole-lake
95 CH₄ emission budget (Sepulveda-Jauregui et al., 2015) at the same lake for comparison with
96 results from our laboratory studies. Our examination provides unique and valuable insights on C
97 cycling dynamics in thawed sediments beneath thermokarst lakes. Further, we draw upon
98 research from lakes in other northern and global environments to fill knowledge gaps in CH₄
99 dynamics not studied at Vault Lake, to place our findings in a global context, and to identify
100 future research directions for incorporating thermokarst lake CH₄ dynamics into estimates of the
101 PCF.

102 **2. Study site: Vault Lake**

103 *Lake location and characteristics*

104 Vault Lake (65.0293 °N, 147.6987 °W, elevation 216 m) is a 3,200 m² thermokarst lake located
105 in a lowland boreal region of Interior Alaska approximately 40 km north of Fairbanks. The
106 region has a continental climate (mean annual air temperature -2.39 °C and 274.6 mm mean
107 annual precipitation; 1981-2010 data from the National Climate Data Center) and is
108 characterized by discontinuous permafrost reaching up to 120-m thick (Jorgenson et al., 2008).
109 At the Vault site, permafrost is comprised of 36-m thick ice- (21-208 wt% gravimetric ice
110 content with ice wedges up to 3 m width) and organic-rich (1-3 wt% SOC) aggraded syngenetic
111 loess-like silt (yedoma) underlain by schistose bedrock. These yedoma sediments were largely
112 deposited in the unglaciated region during the early and middle Wisconsin periods, and observed
113 ice wedge and sediment deformation suggests displacement (e.g. from landslides) following
114 permafrost formation (Schirrmeister et al., 2016). Similar ice- and organic-rich yedoma deposits
115 are found in unglaciated lowland regions across Alaska, Northeast Siberia, and Northwest
116 Canada (Olefeldt et al., 2016; Walter Anthony et al. 2018).



117

118 **Figure 1.** Location map of Vault Lake (a) and historical and recent georeferenced aerial images
 119 and LiDAR digital elevation model of the Vault Lake study area: (b) false color near-infrared
 120 aerial photo from 22 August 1986 from the Alaska High-Altitude Aerial Photography (AHAP)
 121 program; (c) true color aerial photo from 18 August 2007 from the USGS Digital Orthophoto
 122 Quadrangle (DOQQ) program for Alaska; and (d) LiDAR-based digital elevation model of the
 123 same study area from May 2011 produced by the Alaska Division of Geological and Geophysical
 124 Surveys (DGGS).

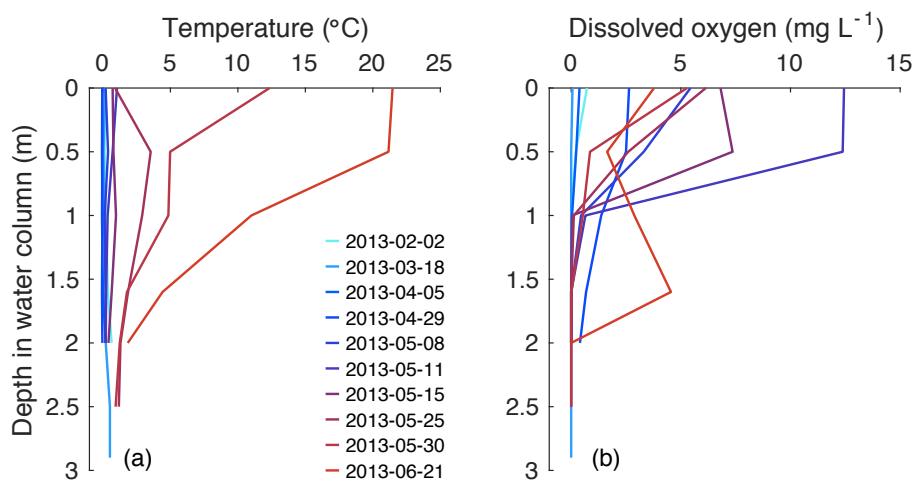
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126 Vault Lake is representative of a young thermokarst lake in a yedoma permafrost setting.

127

128 We note that not all thermokarst lakes are located in yedoma-type permafrost and, compared to
 129 thermokarst lakes formed in non-yedoma environments, thermokarst lakes in yedoma permafrost
 130 are typically deeper due to higher ground subsidence from melting massive ice (Grosse et al.,
 131 2013). Vault Lake has an average water depth of 3.7 m (maximum depth 4.6 m; Heslop et al.,
 132 2015) and reached a maximum lake ice thickness of 60-80 cm in late winter during 2012-2014
 133 (Bondurant, *unpublished data*). Temperature and dissolved oxygen profiles measured in the
 water column during February-June 2013 indicate Vault Lake stratifies at 1.0-1.5 m depth (Fig.

134 2), leading to an anoxic hypolimnion and talik. Talik depths, measured in a 2013 borehole
 135 transect, range from 5.5 to 8.8 m (Heslop et al., 2015). For comparison, other Fairbanks-region
 136 thermokarst lake taliks measured via boreholes (Walter Anthony et al., 2018; Walter Anthony et
 137 al., *submitted*; Winkel, *unpublished data*) and nuclear magnetic resonance (Parsekian et al.,
 138 2013) range from 5 to 22 m depth. No evidence of prior drained thermokarst lakes at the site
 139 indicate Vault Lake is thawing into the yedoma for the first time and, therefore, is a first-
 140 generation lake. Steep eroding bluffs along the lake margin are a sign of active lateral lake
 141 expansion and numerous active ebullition seeps emitting CH₄ with a ¹⁴C-depleted signature
 142 characteristic of Pleistocene-aged substrates (¹⁴C-CH₄ age up to 28,500 ± 140 yrs BP; δD -383 to
 143 -413‰; Winkel et al., 2019) indicate the talik is still deepening and thawing the underlying
 144 yedoma permafrost. Silt mobilized from soils in the lake's watershed is carried into the lake
 145 during rain events via overland flow, with mud flows moving over the top of the organic horizon,
 146 between tussocks and other types of vegetation. In addition to shore erosion, these mudflows
 147 contribute to the accumulation of organic-rich (3.83 ± 1.66 wt% SOC; Heslop et al., 2015)
 148 mineral sediments in the lake at an estimated rate of 0.2 to 0.8 cm yr⁻¹ (Bondurant, *unpublished*
 149 *data*).



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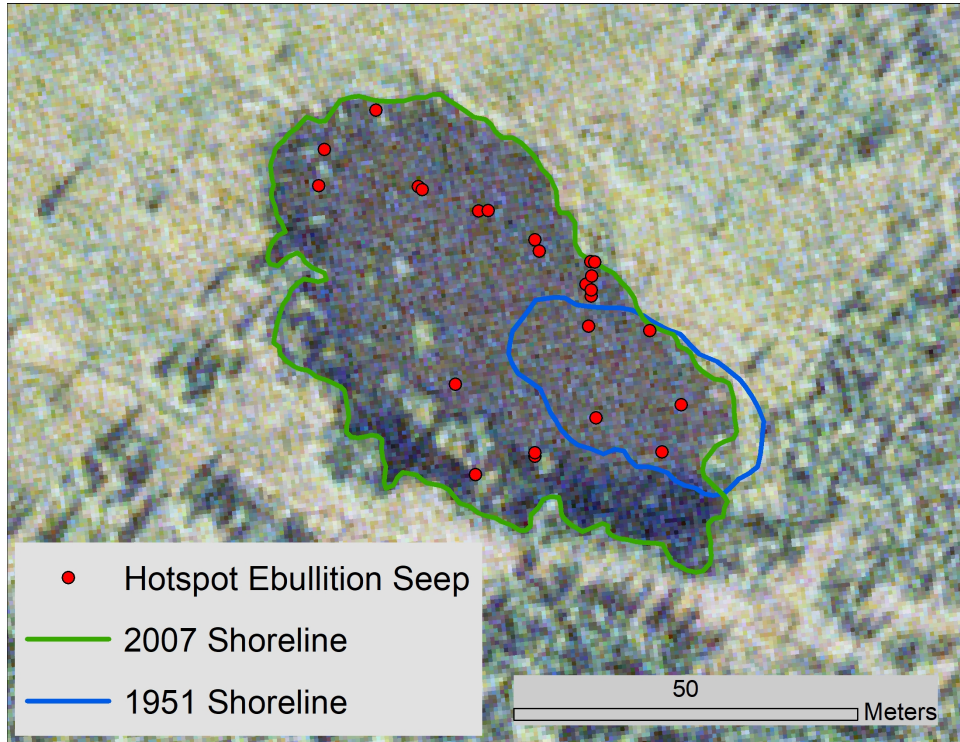
151 **Figure 2.** Water column temperature (a) and dissolved oxygen (b) profiles measured at Vault
152 Lake during February-June 2013. The data suggest Vault Lake stratifies at 1.0-1.5 m depth and
153 show the hypolimnion is anoxic.
154

155 *Sediment core*

156 In March 2013, we collected a 590-cm long sediment core from the center of Vault Lake. Core
157 drilling, processing, and storage methods are described in detail in Heslop et al. (2015). The
158 Vault Lake core captured the near-complete deposited lake sediment and thawed talik sediment
159 sequence beneath the lake, in addition to the top 40 cm of permafrost beneath the talik. Sediment
160 characteristics (Heslop et al., 2015) and organic matter composition (Heslop et al., 2017) were
161 used to classify the core into five facies: (1) Organic-rich mud (0-152 cm); (2) Lacustrine silt
162 (152-330 cm); (3) Taberite (330-508 cm); (4) Recently thawed taberite (508-550 cm); and (5)
163 Transitional permafrost (550-590 cm). The Organic-rich mud (mean 3.83 wt% SOC; 1.86 wt%
164 IC) and Lacustrine silt (1.04 wt% SOC; 0.56 wt% IC) facies represent terrestrial sediments
165 deposited following the lake formation that were exposed to the lake water column during
166 erosion and re-deposition (Heslop et al., 2015). Taberite (0.84 wt% SOC; 0.38 wt% IC)
167 represents yedoma sediment which thawed *in situ* and remained underneath the lake (Walter
168 Anthony et al., 2014; Schirrmeister et al., 2011); the bottom 43 cm of the taberite, representing
169 the most recently thawed sediments, was designated as Recently-thawed taberite (1.36 wt%
170 SOC; 0.70 wt% IC). The Transitional permafrost (1.52 wt% SOC; 0.88 wt% IC) facies consisted
171 of permafrost beneath the talik thaw front that was frozen at the time of core collection; we refer
172 to this as “transitional permafrost” because it is close to the thaw transition and therefore
173 contains a large amount of unfrozen water in the inter-pore space (Williams and Smith, 1989).

174 *Lake age*

175 We estimate Vault Lake formed 100-400 years before present (yr BP). Using AMS radiocarbon
176 ages of macrofossils picked from the Vault Lake sediment core, we had originally estimated the
177 lake age to be around 400 yr BP (Heslop et al., 2015). However, estimating the date of lake
178 formation using expansion rates between 1951 and 2007 shorelines (Fig. 3) yields a lake age
179 estimate of 100 yr BP. Using the depth of the deposited Organic-rich mud (0-152 cm) facies and
180 assuming a sediment deposition rate of 0.2 to 0.8 cm yr⁻¹ yields lake age estimates of 190-760 yr
181 BP. The variance in lake age estimates obtained using different methods highlights the
182 challenges of accurately dating thermokarst lake initiation. Using AMS radiocarbon ages of
183 macrofossils has the potential to overestimate lake age if pre-aged materials enter the sediment
184 column. Similarly, lake expansion and sediment disposition rates are unlikely to be constant over
185 the lakes' history, which can lead to over- or underestimating lake age. These challenges to
186 accurately estimating thermokarst lake age point towards the difficulties of modeling and
187 projecting CH₄ emissions from thermokarst lakes in the coming centuries, given that younger
188 thermokarst lakes typically have higher CH₄ emission rates than older thermokarst lakes
189 (Turetsky et al., 2020; Walter Anthony et al., 2018).



190

191 **Figure 3.** Locations of hotspot ebullition seeps (red) mapped at Vault Lake during October 2014
 192 in relation to the 2007 shoreline (green) and 1951 shoreline (blue).

193

194 *Talik temperatures*

195 Temperatures measured in two vertical profiles within the talik over a seven-year period (2013-

196 2020; Fig. 4) indicate sediments in the lake center (4.0 m water depth) range from -0.40 to 4.51

197 °C (annual mean \pm SD 1.11 ± 1.42 °C) and sediments near the lake margin (1.4 m water depth)

198 range from -0.40 to 14.45 °C (annual mean \pm SD 2.50 ± 3.37 °C). Temperatures at all depths in

199 both talik profiles were significantly correlated with atmospheric temperatures. Temperatures

200 from the shallower sediment depths (0.5 and 1.0 m) in the profile near the lake margin were

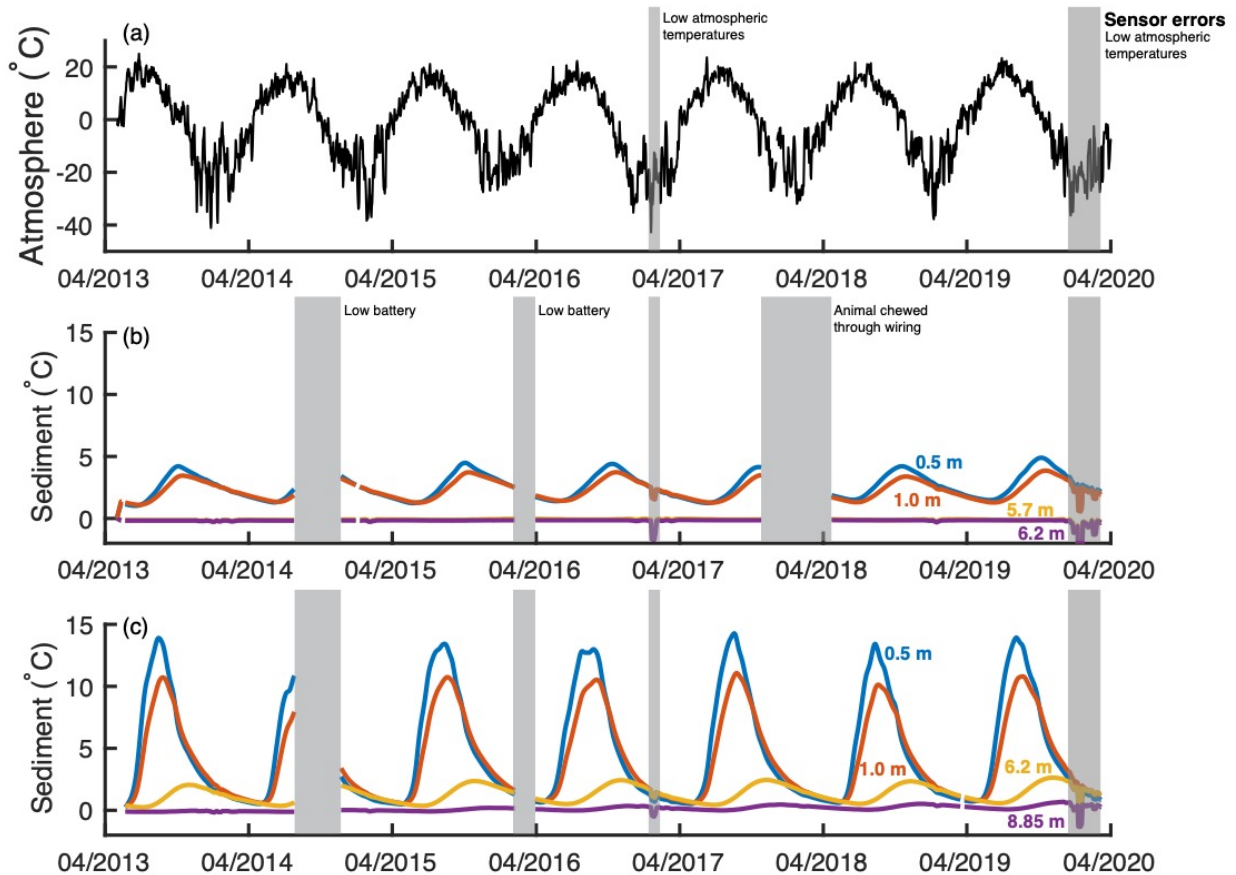
201 positively correlated with atmospheric temperatures (Pearson coefficient $r = 0.59$ and 0.44 ,

202 respectively; $p < 0.001$), while temperatures in the same sediment depths in the lake center were

203 negatively correlated with atmospheric temperatures (Pearson coefficient $r = -0.34$ and -0.52 ; p

204 < 0.001). This is most likely a result of: (1) the deeper water in the lake center causing less

205 efficient heat transfer from the atmosphere to the sediments, resulting in a lagged temperature
 206 response compared to the sediments in the lake margin profile, and (2) water beneath the
 207 hypolimnion in the lake center remaining colder during the thaw season than water in the
 208 shallower littoral zones.



209
 210 **Figure 4.** Mean daily atmospheric temperatures measured at Fairbanks, Alaska (a; data from
 211 ACIS) and hourly temperatures measured at the Vault Lake talik in the lake center (b; 4.0 m
 212 water depth) and near the lake margin (c; 1.4 m water depth). Temperature data were recorded
 213 continuously, except for gaps and sensor errors noted in gray on the figure.

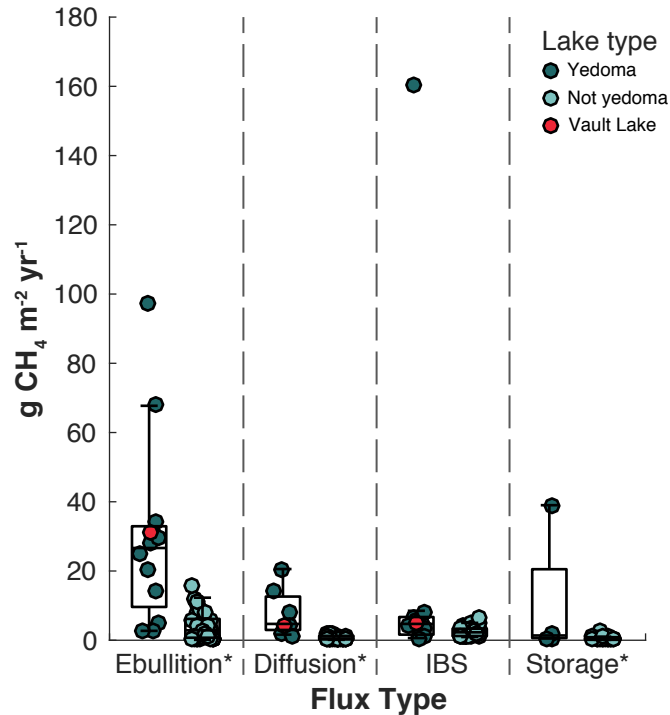
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 215 **3. Methane emission**

216 High latitude lakes emit CH₄ by four main seasonal pathways: diffusion, direct ebullition, ice
 217 bubble storage flux, and storage flux turnover events (Greene et al. 2014). Atmospheric diffusion
 218 occurs during ice-free conditions when the lake water becomes oversaturated in CH₄, leading to

219 positive diffusive flux to the atmosphere. Diffusive CH₄ flux is the lowest contribution of CH₄ to
220 the atmosphere, contributing approximately 5% of total annual emissions from interior Alaska
221 yedoma lakes (Sepulveda-Jauregui et al., 2015). Ebullition is the dominant pathway of CH₄
222 emission in most thermokarst lakes (Walter et al., 2006). Direct ebullition occurs when high
223 concentrations of CH₄, which is less soluble than CO₂, accumulate in sediment pore space. When
224 concentrations exceed the solubility limits, free gas bubbles form. Bubbles form and migrate
225 through secondary pore channels in sediments to release gas to the atmosphere (Liu et al., 2016;
226 Scandella et al., 2011), with the highest ebullition rates occurring when hydrostatic pressure
227 drops (Varadharajan, 2009). In winter surface lake ice impedes bubble release to the atmosphere,
228 entrapping many CH₄ ebullition bubbles in and under surface lake ice. Up to 80% of their CH₄
229 content may be lost to the water column when CH₄ diffuses out of bubbles before the ice
230 thickens around them. When the ice melts in spring, the remaining entrapped CH₄ is released to
231 the atmosphere through a process called ice bubble storage flux (Greene et al., 2014). Combined,
232 ebullition and ice bubble storage flux typically account for around 86% of annual CH₄ emissions
233 from thermokarst lakes formed in yedoma-type permafrost (Sepulveda-Jauregui et al., 2015).
234 Elevated concentrations of dissolved CH₄ in the water column beneath ice are also subject to
235 diffusive flux as the ice melts and during overflow events (Greene et al., 2014). These emissions
236 associated with springtime storage flux turnover events, which derives largely from dissolution
237 of ebullition bubbles beneath ice in winter (Greene et al., 2014), contributes 1.2% of total annual
238 CH₄ emissions (Sepulveda-Jauregui et al., 2015).

239 At Vault Lake, total annual CH₄ emissions were estimated to be 40.9 g CH₄ m⁻² yr⁻¹; 77%
240 of the annual CH₄ flux is from ebullition, 11% is from ice bubble storage flux, and 12% is from
241 diffusive flux (Sepulveda-Jauregui et al., 2015). Diffusive flux and ice bubble storage flux values

242 at Vault Lake are near the median for the yedoma lakes, while ebullitive flux is slightly higher
243 than the median but still within the interquartile range for yedoma lakes (Fig. 5). Compared to
244 other reported CH₄ emissions values from northern lakes, net Vault Lake CH₄ emission values
245 are near mean values for yedoma-type lakes (50.2 g CH₄ m⁻² yr⁻¹; n = 13 lakes) and more than
246 non-yedoma type lakes (mean 6.37 g CH₄ m⁻² yr⁻¹; n = 36 lakes). This is consistent with prior
247 research suggesting CH₄ emissions from thermokarst lakes in regions underlain by yedoma-type
248 permafrost emit approximately 6x more CH₄ than lakes underlain by other permafrost types
249 (Sepulveda-Jauregui et al., 2015). It is important to note that, as a relatively young thermokarst
250 lake (100 - 400 yr BP), Vault Lake functions as a net C source to the atmosphere. As thermokarst
251 lakes mature and eventually drain (typically over millennial time scales), their net C emissions
252 become lower and, in some cases, negative due to C being sequestered in increased vegetative
253 biomass (Walter Anthony et al., 2014). Emissions from newly formed thermokarst lakes (< 60
254 years old) are typically even higher than from older thermokarst lakes (Walter Anthony et al.,
255 2018; Turetsky et al., 2020), such as Vault Lake. This variance in C emissions amongst
256 thermokarst lakes highlights the importance of examining lakes in different environments to
257 constrain their potential circum-arctic GHG emissions.



*Significant ($p < 0.05$) difference between yedomas and non-yedomas lakes

258

259 **Figure 5.** Magnitudes of CH₄ emission pathways from northern yedomas ($n = 13$) and non-
 260 yedomas ($n = 36$) lakes. IBS refers to ice bubble storage flux. Significant (Two-sample t test; $p <$
 261 0.05) differences between yedomas and non-yedomas lakes are denoted. Emissions at Vault Lake,
 262 measured by Sepulveda-Jauregui et al. (2015), are highlighted in red. CH₄ flux values for
 263 diffusive flux and IBS at Vault Lake are near the median for the yedomas lakes, while ebullitive
 264 flux is slightly higher than the median, but still within the interquartile range. Spring storage flux
 265 following ice-out was not determined at Vault Lake. Data are reported by Huttunen et al. (2003),
 266 Repo et al. (2007), Sepulveda-Jauregui et al. (2015), Walter Anthony et al. (2010), and Zimov et
 267 al. (1997).

268

269

Within individual thermokarst lakes, CH₄ emissions vary spatially and temporally.

270

However, most studies of northern lakes only report diffusive CH₄ emissions based on sampling

271

< 1% of the lake surface area at 1-2 points in time (Engram et al., 2020). Further, the sporadic

272

nature and spatial variability of ebullitive flux make it difficult to accurately estimate its

273

magnitude. The bubble trap method, which is frequently used to measure ebullitive flux, is

274

strongly biased as: (i) bubble traps are intentionally fixed over ebullition seeps; (ii) it is almost

275

impossible to cover an stratified analysis for ebullition events due to variation in bottom

276

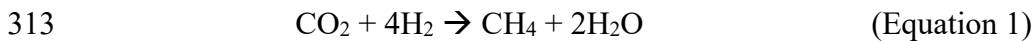
topography, water depth, and sediment properties; (iii) the different number of traps installed in

277 the ecosystems generates large uncertain in the literature reported; and (iv) the season and time
278 measuring bubbles by traps varies between studies and sites (Wik et al., 2016). This variation
279 makes it difficult to accurately quantify and compare CH₄ emissions from circum-arctic lakes. In
280 addition to the lack of information on spatial and temporal variability, measurements of CH₄
281 emissions from lakes are scarce compared to the number of lakes. A 2016 synthesis by Wik et al.
282 collected all known field measurements from northern lakes; the lakes studied (n = 733)
283 represented only 0.02% of the 3.8 million lakes north of 50° N (Engram et al., 2020). This
284 highlights the needs for additional data collection to constrain the spatial and temporal variability
285 in CH₄ emissions from both thermokarst and non-thermokarst northern lake systems. Remote
286 sensing methods have been suggested as one approach for constraining spatial and temporal
287 variability in circum-arctic lake CH₄ emissions (Engram et al., 2020; Matthews et al., 2020).

288 While few field studies measure year-round temporal variability in CH₄ emissions, results
289 from a process-based climate-sensitive lake biogeochemical model indicate that seasonal
290 variation in CH₄ emissions can be explained by energy input and substrate availability (Tan et
291 al., 2015). High temporal resolution CH₄ emissions have not been quantified at Vault Lake, but
292 Sepulveda-Jauregui et al. (2015) determined annual summer CH₄ flux (ebullition and diffusive
293 flux; 31.4 g CH₄ m⁻² yr⁻¹) was approximately 3 times greater than annual winter CH₄ flux
294 (ebullition and ice bubble storage; 9.4 g CH₄ m⁻² yr⁻¹). Spatially, both field observations (Engram
295 et al., 2020; Walter Anthony et al., 2016) and modeling results (Tan et al., 2015) show CH₄
296 emissions along the thermokarst margins are typically higher than emissions from non-
297 thermokarst margins and centers in the same lake. Mapping of ebullition hotspots at Vault Lake
298 during October 2014 suggest ebullition seep locations are generally more abundant in areas
299 where the lake has expanded since 1951 (Fig. 3).

300 4. Methane production

301 Methane emissions from northern lakes originate from different sources, including fossil
302 geological (microbial and thermogenic) sources and more recent microbial sources (Walter
303 Anthony et al., 2012). Recently, in temperate lakes, CH₄ production in the water column has also
304 been found to be associated with phytoplankton under oxic conditions (Günthel et al., 2020);
305 however, it is currently unknown if oxic CH₄ production is also present in northern lakes. δ¹³C-
306 CH₄ values collected from ebullition bubbles at Vault Lake (-67.7 ± 3.4‰; Winkel et al., 2019),
307 in combination with Vault Lake having a closed talik without geologic CH₄ seeps, indicate CH₄
308 production at Vault Lake is microbial. Microbial production of CH₄ during anaerobic respiration,
309 termed methanogenesis, occurs when *Archaea* called methanogens use simple substrate (CO₂,
310 H₂, acetate, methyl compounds) to produce CH₄. Methanogenesis using CO₂ and H₂ as substrate,
311 which is carried out by hydrogenotrophic methanogens (autotrophy), occurs when CO₂ is
312 reduced to CH₄ according to the reaction:



314 Methanogenesis using acetate as a substrate (heterotrophy), carried out by acetoclastic
315 methanogens, occurs when acetate is reduced to CH₄ according to the reaction:



317 The primary pathway for CH₄ production (hydrogenotrophic vs. acetoclastic) varies both
318 among yedoma thermokarst lakes and within the same lake depending on physiochemical
319 parameters and substrate availability (Walter et al. 2008). At Vault Lake, a combination of
320 archaeal sequencing measured from sediments and δ¹³C-CH₄ values measured from ebullition
321 bubble gas indicates both hydrogenotrophic and acetoclastic methanogenesis by
322 *Methanosarcina*, *Methanosaeta* and *Methanoregula* (Winkel et al., 2019). A less common

323 pathway of methanogenesis that has been shown to be activated in thawing permafrost uses
324 methyl compounds such as methylamines as a substrate (Canfield et al. 2005; Coolen and Orsi,
325 2015; Tveit et al., 2015). While we did not observe evidence of this CH₄ producing pathway at
326 Vault Lake, incubations of thermokarst lake sediments from Utqiagvik, Alaska (de Jong et al.,
327 2018) and Interior Alaska (Liebner, *unpublished data*) observed stimulation of methylotrophic
328 methanogens following substrate addition.

329 Through an anaerobic incubation study of the Vault Lake sediment core, we found that
330 two-thirds of total CH₄ measured during a 175-day incubation at 3 °C originated in the surface
331 organic-rich mud facies (0-152 cm depth), despite this facies only representing a quarter of the
332 core's total 590 cm thickness (Heslop et al., 2015). The organic-rich mud facies comprise of
333 reworked allochthonous and autochthonous materials deposited since lake formation, as
334 described by Farquharson et al. (2016) for other yedoma lakes. Archaeal communities were
335 dominated (84% of detected sequences) by the acetoclastic methanogens *Methanosaetaceae*,
336 with only a few hydrogenotrophic *Methanoregulaceae* and *Methanosarcinaceae* (Heslop et al.,
337 2019; Winkel et al., 2019). Given that all species within *Methanosaetaceae* only use acetate as
338 substrate, their high abundance in the organic-rich mud facies suggests high availability of
339 acetate. *Methanosaetaceae* and *Methanosarcinaceae* are frequently detected methanogens in
340 thermokarst lakes (Kallistova et al., 2020).

341 In an examination of bulk SOM composition using pyrolysis GC-MS, Heslop et al.
342 (2017) found that the organic-rich mud facies had both greater substrate availability (higher C
343 and N concentrations) and greater proportions of compounds associated with allochthonous
344 terrestrial materials (alkanes, alkenes, lignin products, and phenols and phenolic precursors)
345 compared to the underlying sediments. The abundance of these compounds was positively

346 correlated with CH₄ production potentials measured in 3 °C incubations. Based on ¹⁴C ages of
347 picked macrofossils from the top 214 cm of the lake core (170 – 429 years BP; Heslop et al.,
348 2015), the organic carbon (OC) in these deposited facies is far younger than in the Pleistocene-
349 aged permafrost studied in the nearby Vault Creek Permafrost Tunnel (>20,000 years BP;
350 Schirrmeister et al., 2016), which also underlies Vault Lake. Both ¹⁴C ages of *in situ* dissolved C-
351 CO₂ and C-CH₄ (Elder et al., 2018) and laboratory culture experiments (Douglas et al., 2020)
352 suggest CH₄ is produced faster from younger C than older C in northern lakes. However, we note
353 Elder et al. (2018) focused on the ¹⁴C ages of dissolved C-CO₂ and C-CH₄, which escapes lakes
354 by diffusion. In most arctic lakes, diffusion is a more minor component of total lake CH₄
355 emissions (~5% of total annual emissions; Sepulveda-Jauregui et al., 2015).

356 The radiocarbon age of CH₄ in naturally-emitted ebullition bubbles, which are the
357 dominant form of CH₄ emissions in northern lakes (Bastviken et al., 2011; Sepulveda-Jauregui et
358 al., 2015; Walter et al., 2006; Wik et al., 2016), is always ¹⁴C-depleted and older than CH₄
359 dissolved in the water column (Walter Anthony et al., 2018). At Vault Lake, this was supported
360 by the highest dissolved pore water CH₄ concentrations being at the talik thaw boundary, in the
361 recently-thawed taberite facies (Winkel et al., 2019). The sediments directly above the thaw front
362 also had high CH₄ production potentials in incubation studies of the Vault Lake core (Heslop et
363 al., 2015), and above-median CH₄ ebullition rates observed at Vault Lake compared to other
364 yedoma lakes (Fig. 4) suggest high CH₄ production rates and subsequent gas build up at the thaw
365 front. The ¹⁴C-CH₄ age (up to 28,500 ±140 yrs BP) and δD (-383 to -413‰) of ebullition bubbles
366 emitted from Vault Lake reflect Pleistocene-yedoma carbon and hydrogen (from Pleistocene ice
367 wedges) as the atomic components of the CH₄ molecules, as opposed to more modern
368 (Holocene) sources which have younger/heavier CH₄ isotopic values (Brosius et al., 2012). This

369 agrees with ebullition bubble mapping (Lindgren et al. 2016, 2019; Walter Anthony et al. 2016;
370 Walter Anthony et al., *submitted*) and modeling studies (Kessler et al., 2012) of thermokarst
371 lakes that suggest recently-thawed former permafrost at and directly above the thaw boundary is
372 a source of high CH₄ production. The high CH₄ production at the thaw boundary may be due to
373 high bioavailable substrate potential. The idea that permafrost dissolved OC (DOC) is more
374 biodegradable than younger DOC is supported by a metaanalysis showing higher observed DOC
375 biodegradability with increasing permafrost landscape extent (Vonk et al., 2015). In terrestrial
376 settings, DOC from recently thawed former permafrost has been shown to have higher
377 degradation potentials than DOC from the overlying seasonally-thawed active layer (Heslop et
378 al., 2019b; Liu et al., 2019; Selvam et al. 2017). Yedoma permafrost sediments have additionally
379 been shown to have high acetate concentrations (Drake et al., 2015; Ewing et al., 2015), which
380 can be directly utilized as substrate for acetoclastic methanogenesis following thaw. While
381 acetate concentrations were not directly measured in the Vault Lake core, higher relative
382 abundance of the acetoclastic methanogens *Methanosaeta* near the thaw front suggest higher
383 acetate substrate potential. Like the organic-rich mud facies, archaeal communities in the
384 recently-thawed taberite facies were dominated (89% of detected sequences) by the acetoclastic
385 and hydrogenotrophic methanogens *Methanosaetaceae*, *Methanosarcinaceae*, and
386 *Methanoregulaceae* (Heslop et al., 2019a; Winkel et al., 2019).

387 Former permafrost sediments that have thawed *in situ* beneath Vault Lake over century
388 timescales (152-508 cm core depth) had the lowest CH₄ production potentials indicated by both
389 incubation studies (Heslop et al., 2015) and dissolved CH₄ concentrations (Winkel et al., 2019).
390 The lacustrine silt and taberite facies, which represented 60% of the total Vault Lake core length,
391 only accounted for 21% of the whole-column CH₄ production potential in the 3 °C incubation

392 study (Heslop et al., 2015). Methanogens were only detected in the rare biosphere, which we
393 defined as taxa below 0.1% relative sequence abundance of all detected archaea in our 16S RNA
394 analyses. Detected methanogens included *Methanocellales*, *Methanoregulaceae*,
395 *Methanosarcinaceae*, and *Methanosaetaceae* (Heslop et al., 2019a; Winkel et al., 2019). Archaeal
396 communities in this region were dominated by CH₄-consuming methanotrophs (see Section 5).

397 Using temperature sensitivity incubations, Heslop et al. (2019a) showed that these
398 deeper, thawed *in situ* sediments had higher temperature sensitivities for CH₄ production
399 compared to the overlying reworked and deposited material in the organic-rich mud facies. CH₄
400 production potentials did not increase with warming at incubation temperatures consistent with
401 temperatures measured in the talik (0 °C and 3 °C) but increased with higher incubation
402 temperatures (10 °C to 25 °C; Heslop et al., 2019a). This suggests CH₄ production in lacustrine
403 silt and taberite facies is substrate-limited, and the input of additional energy with higher
404 temperatures may increase CH₄ production potentials. In the lacustrine silt and taberite facies, we
405 hypothesize the most labile “fast carbon pool” (days to years turnover time; < 5% of permafrost
406 OC; Schädel et al., 2014) was previously exhausted. Per kinetic theory, the remaining, more
407 complex OC (turnover times of decades to centuries) require higher activation energies for
408 microbial decomposition due to the higher number of enzymatic steps needed for biological
409 degradation (Davidson and Janssens, 2006, Conant et al., 2011). Hence, higher temperatures are
410 needed to decompose the more recalcitrant sediment organic C. In contrast, the most deeply
411 thawed sediments located at the talik boundary (i.e. the permafrost thaw front) showed the
412 highest temperature sensitivity at a much lower temperature regime. The large increase in CH₄
413 production potentials when temperatures increased from 0 to 3 °C points to the higher

414 bioavailability of the recently thawed, more labile permafrost organic C at the thaw front. We
415 further discuss this in the context of changing climate in Section 7.

416 **5. Methane oxidation**

417 Methane production estimated from anaerobic laboratory incubations of the Vault Lake sediment
418 core (2,819 g CH₄ m⁻² at 3 °C ; Heslop et al., 2015) greatly exceeds the magnitude of CH₄
419 emitted annually by field flux measurements at the same lake (40.9 g CH₄ m⁻²; Sepulveda-
420 Jauregui et al., 2015). Using these values to calculate a first-order estimate suggests up to 99% of
421 produced CH₄ at Vault Lake either accumulates and is held in sediments (presumably released
422 during rare events not captured in field measurements) or is oxidized before being emitted to the
423 atmosphere. A third possibility is that incubation conditions in closed vessels led to inflated CH₄
424 production potentials (i.e. the bottle effect; Ionescu et al., 2015) compared to actual CH₄
425 production, which is unknown; however, the 3 °C temperature of our incubation was slightly
426 lower than the mean annual temperature of surface lake sediments (3.47 ± 3.07 °C at the 0.5 and
427 1.0 m depths), where most of the CH₄ was produced. This suggests that the actual rates in these
428 surface lake sediments could be higher than our estimate. We acknowledge that our CH₄
429 production estimate does not take into account spatial and temporal heterogeneity affecting *in*
430 *situ* CH₄ production and emissions at Vault Lake. Modelled CH₄ production at other Alaskan
431 thermokarst lakes also indicate significantly more CH₄ is produced in sediments than is emitted
432 into the atmosphere, with one study finding CH₄ production in a modeled thermokarst lake talik
433 was up to 10 times higher than observed emissions in the field (Kessler et al., 2012).

434 Both globally and in the Arctic, CH₄ oxidation plays an important mitigating role in
435 preventing a significant portion of CH₄ produced in lakes and wetlands from entering the
436 atmosphere (Oh et al., 2020; Osudar et al., 2016; Reeburgh 2006). Calculations along the length

437 of the Vault Lake core using stable isotopes suggest 41-83% of CH₄ produced in the core is
438 anaerobically oxidized *in situ* (Winkel et al., 2019). Under anaerobic conditions, CH₄ oxidation
439 often occurs as the result of syntrophic relationships between methanotrophic *Archaea* (ANME)
440 and bacteria that reduce inorganic electron acceptors (Knittel and Boetius, 2009). However,
441 recent research also showed that ANME can perform the process without a bacterial partner (Cai
442 et al., 2018), the existence of anaerobic bacteria that internally produce oxygen to oxidize CH₄
443 (Ettwig et al., 2010), and that aerobic methanotrophs that can perform CH₄ oxidation under
444 anaerobic conditions (Oswald et al., 2016). In the Vault Lake hypolimnion and sediments, which
445 are under anoxic conditions year-round (Fig. 2b), anaerobic oxidation of methane (AOM) can
446 occur using sulfate, nitrogen oxidizes, organic matter, chlorite, or metals (e.g. iron and
447 manganese) as terminal electron acceptors. AOM values reported in the Vault Lake core by
448 Winkel et al. (2019) ranged from 0 to 2.87 nmol CH₄ cm⁻³ d⁻¹. For comparison, reported AOM
449 rates in global freshwater lake systems (n = 66) were 0.01 to 100 nmol CH₄ cm⁻³ d⁻¹ (Martinez-
450 Cruz et al., 2018). The lower values reported by Winkel et al. (2019) compared to global
451 freshwater AOM rates are likely due to a combination of: (1) lower organic matter quality in the
452 taberite facies, which had been thawed beneath Vault Lake for centuries, compared to in near-
453 surface deposited sediments, which are where sediment AOM is typically quantified; and (2) the
454 incubations conducted by Winkel et al. 2019 being done at 4 °C, which is a lower temperature
455 than most of the incubations reported in literature.

456 To our knowledge, the study by Winkel et al. (2019) is the first to identify and quantify
457 AOM in cold thawing and former permafrost sediments beneath a thermokarst lake. Tracer
458 incubation studies of AOM potentials found low AOM potential rates in the transitional
459 permafrost (~0.9 pmol cm⁻³ d⁻¹), but high (~660 µM) concentrations of nitrite (Winkel et al.,

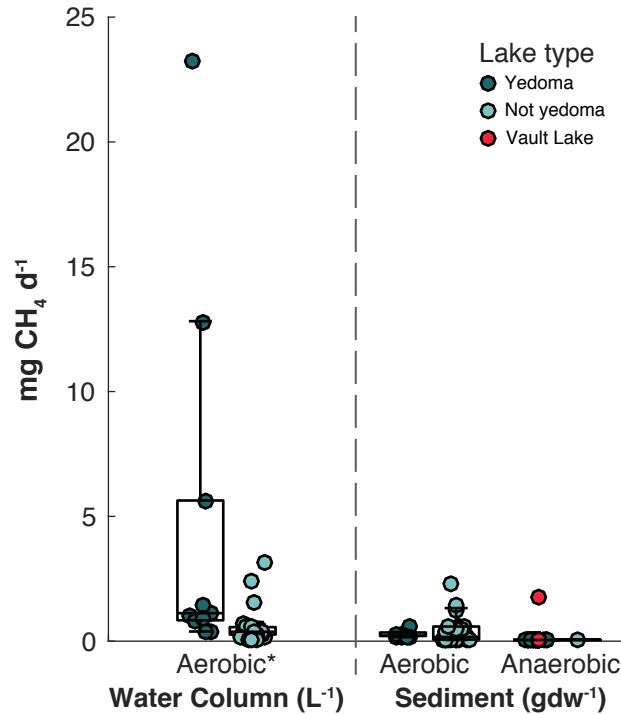
460 2019). Studies from another lake core collected in Interior Alaska suggest yedoma sediments
461 also have high concentrations of Fe, which may serve as an alternate electron acceptor for AOM
462 following thaw (Winkel, *unpublished data*). Evidence from the Vault Lake core suggests former
463 permafrost sediments that have been thawed *in situ* for centuries (311 to 532 cm depth; taberite
464 and recently-thawed taberite facies) are a region of high AOM in thermokarst lake environments.
465 Compared to the transitional permafrost, AOM rates doubled in the overlying recently-thawed
466 taberite facies. This region of the Vault Lake core had the highest AOM potential rates (up to
467 $2.88 \text{ pmol cm}^{-3} \text{ d}^{-1}$; mean $1.7 \pm 0.7 \text{ pmol cm}^{-3} \text{ d}^{-1}$), high nitrite concentrations, and the most
468 fractionated $\delta^{13}\text{C-CH}_4$ in the pore waters (Winkel et al., 2019). Archaea in this region were
469 almost exclusively the nitrate-driven AOM (Haroon et al 2013) or iron-driven AOM (Ettwig et al
470 2015, Cai et al 2018) archaea *Methanoperedenaceae* (Winkel et al., 2019). Sulfate
471 concentrations in the taberite (up to $35 \pm 2 \text{ mM}$) were also found to be sufficient for potentially
472 supporting AOM (Winkel et al., 2019).

473 A second region of high AOM was observed in the deposited near-surface sediments of
474 Vault Lake. Winkel et al. (2019) found the top 100 cm of the Vault Lake core to have high AOM
475 potential rates (up to $2.3 \text{ pmol cm}^{-3} \text{ d}^{-1}$ at $3 \text{ }^\circ\text{C}$). Mean AOM potential rates separately measured
476 by Martinez-Cruz et al. (2017) in sediments collected from the top 25 cm were 12.27 ± 3.96
477 $\text{nmol cm}^{-3} \text{ d}^{-1}$ (mean \pm SD) at $4 \text{ }^\circ\text{C}$. Compared to net CH_4 production potential rates measured
478 from the same 25 cm core at the same incubation temperature, this represents approximately one-
479 third of CH_4 produced in surface sediments at Vault Lake being anaerobically oxidized
480 (Martinez-Cruz et al., 2017). AOM in this region used different electron acceptors and had
481 different methanotrophs compared to the high AOM in the underlying taberite. This AOM was
482 mainly attributed to type I aerobic methanotrophs belonging to the genus *Methylobacter*; only

483 iron and manganese were found to be present in sufficient quantities to support the observed
484 AOM (Martinez-Cruz et al., 2017).

485 Synthesizing published rates of CH₄ oxidation from Alaskan yedoma (n = 38
486 observations) and non-yedoma (n = 53 observations) lakes revealed that lakes located in yedoma
487 regions did not have significantly different oxidation rates (aerobic or anaerobic) in their
488 sediment than non-yedoma northern lakes (Fig. 6). While we did not examine CH₄ oxidation in
489 the water column at Vault Lake, research from nearby Alaskan thermokarst lakes and other arctic
490 lakes suggests further CH₄ oxidation occurs in the water column and under ice under both
491 aerobic and anaerobic conditions. A study by Martinez-Cruz et al. (2015) in yedoma-type lakes
492 near Vault Lake found mean potential aerobic CH₄ oxidation rates of 0.71 mg L⁻¹ d⁻¹ in the
493 summer and 0.18 mg L⁻¹ d⁻¹ in the winter. Our literature synthesis suggests median potential CH₄
494 oxidation rates in the water columns of yedoma lakes are higher than in non-yedoma lakes (Fig.
495 6). A recent study by Thalasso et al. (2020) found that, in a stratified lake in north-central
496 Siberia, all CH₄ produced in the sediments and deep hypolimnion was removed below the
497 oxycline, suggestive of high AOM. This case study suggests, in northern lakes that stratify, it is
498 possible for most or all CH₄ produced in sediments to be anaerobically oxidized in the
499 hypolimnion, resulting in little to no CH₄ exchange with the epilimnion and/or atmosphere. In
500 northern Alaska, it has been suggested that the threshold depth for lakes to fully stratify is
501 approximately 4 m (Arp et al., 2015). At Goldstream Lake, a thermokarst lake located in interior
502 Alaska, approximately 50% of CH₄ in ebullition bubbles trapped under winter ice cover oxidize
503 prior to ice out (Elder et al., 2019). Greene et al. (2014) estimated the under-ice CH₄ oxidation at
504 the same lake to be 56%. ¹³C-CH₄ analyses of ice cores from thermokarst lake and lagoon
505 environments underlain by continuous permafrost in Siberia revealed that CH₄ at the ice-water

506 interface was oxidized to concentrations near atmospheric equilibrium, while the overlying ice
 507 was supersaturated in CH₄ compared to atmospheric concentrations (Spangenberg et al., *in*
 508 *review*).



*Significant ($p < 0.05$) difference between yedoma and non-yedoma lakes

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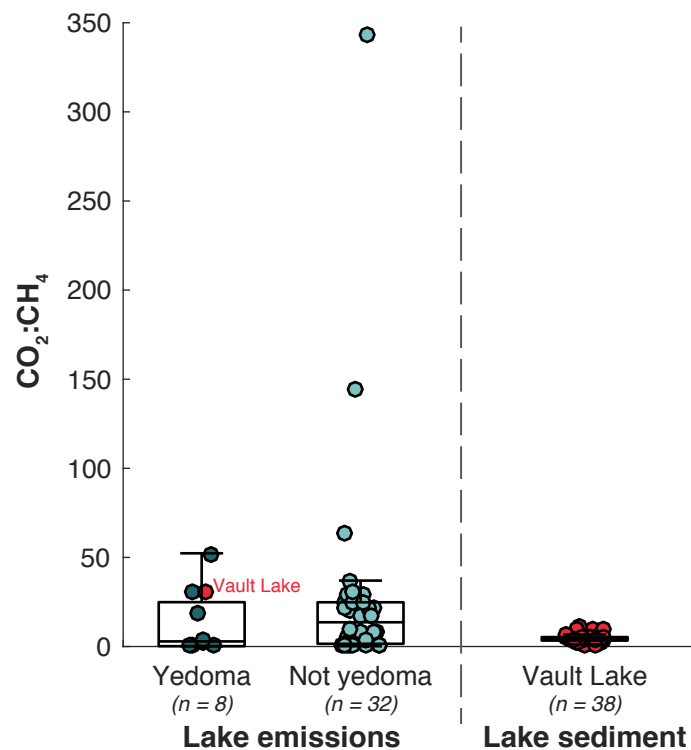
510 **Figure 6.** Rates of CH₄ oxidation from Alaskan yedoma (n = 38 observations) and non-yedoma
 511 (n = 53 observations) lakes. Significant (two-sample t-test; $p < 0.05$) differences between
 512 yedoma and non-yedoma lakes are denoted. Sediment AOM at Vault Lake, measured by
 513 Martinez-Cruz et al. (2017) and Winkel et al. (2019), are highlighted in red. Data are reported by
 514 de Jong et al. (2018), He et al. (2012), Martinez-Cruz et al. (2015, 2017), Miller et al. (2019),
 515 and Winkel et al. (2019).

516

517 6. Carbon dioxide to methane production ratios

518 One issue of great importance in determining the potential impact of C emissions from
 519 thermokarst lakes on the global C cycle is determining proportions of C released as CO₂ versus
 520 CH₄. Northern lakes emit more CO₂ than CH₄; on average 26x more CO₂ was emitted than CH₄
 521 in a transect of 40 lakes in Alaska (Fig. 7; Sepulveda-Jauregui et al., 2015). Several processes in
 522 both the lake sediments and overlying water column affect the magnitudes of CO₂ versus CH₄

523 emissions. Analysis of data from 5,118 boreal lakes found that CO₂ emissions were
524 predominately (60% of studied lakes) sustained by loading from the catchment (e.g. inorganic C
525 loading from surface and ground waters), as opposed to internal CO₂ production (Weyhenmeyer
526 et al., 2015). In the aerobic portion of the water and sediment column, internal processes that
527 produce CO₂ include: biological respiration in the water column (3.4–91% of internal CO₂
528 production in boreal lakes; Weyhenmeyer et al., 2015) and sediments (4.0-60% internal CO₂
529 production); inorganic C fixation; and, in shallow non-yedoma lakes (Tan et al., 2017), light
530 degradation (3.0-48% internal CO₂ production). In anaerobic portions of the water column and in
531 lake sediments, CO₂ is produced both as a by-product of methanogenesis and fermentation, and
532 as an end-product of CH₄ oxidation; CH₄ is produced as an end-product of methanogenesis. In
533 both the sediments and water column, under both aerobic and anaerobic conditions, a significant
534 amount of CH₄ is removed and CO₂ is produced during CH₄ oxidation.



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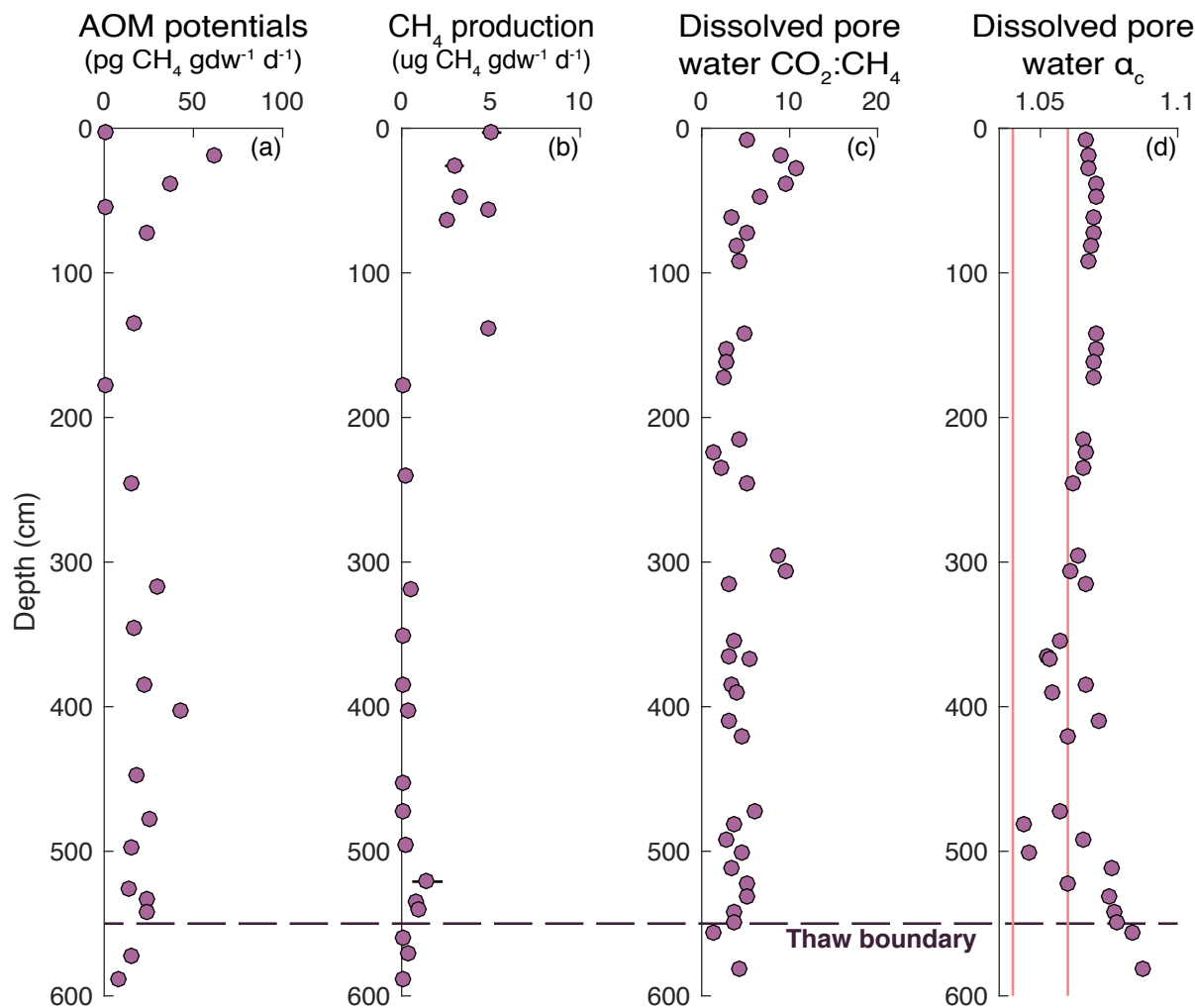
536 **Figure 7.** *In situ* ratios of CO₂:CH₄ emissions measured in Alaskan lakes vs. CO₂:CH₄ porewater
537 concentrations in Vault Lake sediments. Annual CO₂ and CH₄ emissions were reported by
538 Sepulveda-Jauregui et al. (2015). Emission CO₂:CH₄ ratios were not significantly different in
539 yedoma versus non-yedoma lakes (two-sample t-test; p = 0.51). Pore water dissolved gas
540 concentrations in anaerobic lake sediment, measured in the Vault Lake core, were reported by
541 Winkel et al. (2019).

542
543 It had been previously thought that, under stable anaerobic CH₄-producing conditions, the
544 lowest possible ratio of CO₂:CH₄ production is 1:1 (Conrad, 1999). However, some recent
545 studies from Arctic environments have reported CO₂:CH₄ ratios of less than 1, indicating more
546 CH₄ production than CO₂ production (Heslop et al., 2019b; Knoblauch et al., 2018; Lokshina et
547 al., 2019; Walter Anthony et al., 2014). At a eutrophic fen site in Siberia, the higher CH₄
548 production than CO₂ production was attributed to proportions of CH₄ produced through
549 acetoclastic (Equation 2) versus hydrogenotrophic (Equation 1) methanogenesis; adding an
550 inhibitor to change the major pathway from acetoclastic to hydrogenotrophic methanogenesis
551 increased CO₂:CH₄ ratios from 0.3-0.4 to 0.7-0.9 (Lokshina et al., 2019).

552 The notion that the CO₂:CH₄ production ratio can be less than 1:1 during anaerobic C
553 mineralization is also supported by field measurements and modeling. Due to the diffusion of
554 gases through meters of dense, silt-dominated sediments in yedoma lake taliks being
555 prohibitively slow, yedoma talik sediments favor ebullition through secondary pore channels as a
556 gas escape mechanism (Tan et al., 2015). ¹⁴C-CH₄ measured in ebullition bubbles is older than
557 ¹⁴C-CO₂ from the same bubbles (Walter Anthony et al. 2018), suggesting that the CH₄ in
558 ebullition bubbles comes from older, deeper sediment and the CO₂ comes from younger
559 sediments closer to the sediment-water interface. The concept that CH₄ production dominates in
560 deep taliks and that the dissolved gases giving rise to diffusive fluxes originate from surface lake
561 sediments is further supported by ¹⁴C-CH₄ ages of dissolved CH₄ being much younger than ¹⁴C-
562 CH₄ in ebullition bubbles (Elder et al., 2019). Ebullition bubbles released from deep with taliks

563 are typically > 85% CH₄ and < 2% CO₂ (Walter et al. 2006; Walter Anthony et al., *submitted*),
564 which may indicate that the CO₂:CH₄ production rates are skewed towards greater CH₄ than CO₂
565 production. In order for CH₄ to be produced in excess of CO₂, an inorganic source of H₂ is
566 required. Telling et al. (2015) showed that pulverization of silicate rocks beneath glaciers
567 produces abiogenic H₂; future research should explore the possibility that pulverization of rocks
568 generating wind-blown silt could have also led to abiogenic H₂ in syngenetically aggraded
569 yedoma permafrost sediments, enabling higher CH₄ production in deep yedoma talik sediments.

570 CO₂:CH₄ ratios of pore water dissolved gas measured in the Vault Lake core ranged from
571 1.2 to 10.7 (mean 4.6; Fig. 8; Winkel et al., 2019). The lowest CO₂:CH₄ ratio (1.2; 557 cm depth)
572 was found at the thaw boundary, a region which also had higher relative abundance of the
573 acetoclastic methanogens *Methanosaeta* (Winkel et al., 2019). These results would be consistent
574 with the suggestion that acetoclastic methanogenesis leads to lower CO₂:CH₄ production ratios.
575 However, the α_C values (an indicator of CH₄ production pathway; $\alpha_C > 1.06$ indicates CO₂
576 reduction; $\alpha_C < 1.04$ indicates acetate fermentation; Walter et al., 2008) in both the dissolved gas
577 and ebullition bubbles indicate that CH₄ in the deepest part of the talik is produced primarily by
578 CO₂ reduction (Fig. 8d). The α_C of dissolved gas in the pore water ranged from 1.044 to 1.087
579 (mean \pm SD 1.066 \pm 0.009; n = 38) and in bubbles released from the base of the talik in
580 boreholes was 1.066 \pm 0.005 (mean \pm SE; n = 5); α_C in hotspot ebullition seeps, which are
581 thought to be the deepest-seated naturally occurring ebullition seeps in taliks was 1.054 \pm 0.006
582 (n = 5); and, α_C in ebullition seeps originating from shallower sediments was 1.048 \pm 0.005 (n =
583 8).



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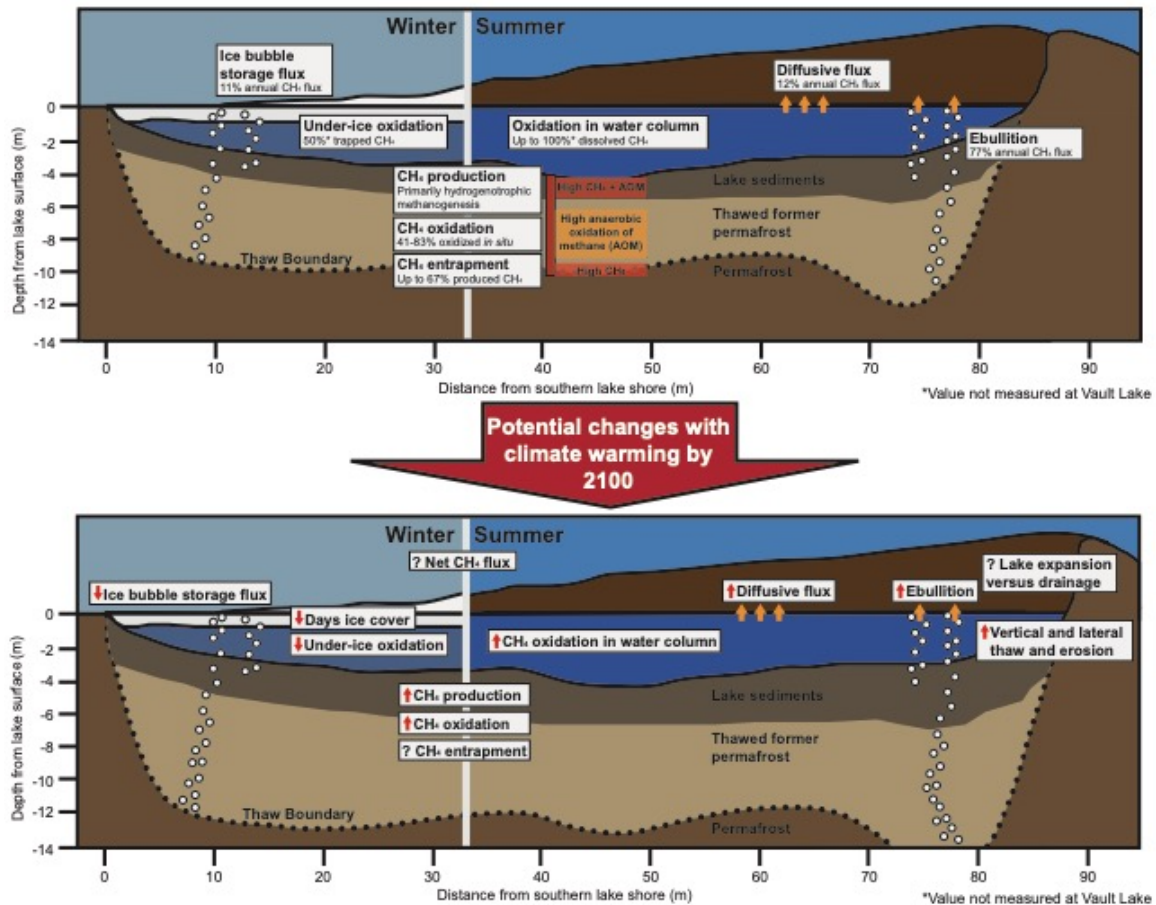
585 **Figure 8.** Magnitudes of AOM potentials (a; Winkel et al., 2019) and CH₄ production potentials
 586 (b; Heslop et al., 2015) measured in incubations of sediments from the Vault Lake core.
 587 Measurements of CO₂ and CH₄ concentrations in sediment pore waters from the Vault Lake core
 588 (Winkel et al., 2019) were used to calculate *in situ* CO₂:CH₄ ratios (c) and α_c (d).
 589

590 . The highest CO₂:CH₄ ratios in porewater dissolved gases (8.6-10.7) were found in the
 591 surface organic rich mud (18-38 cm depth) and in the taberite (296-306 cm depth) facies. This is
 592 consistent with regions in the core found to have higher rates of AOM, which would decrease
 593 CH₄ concentrations and increase CO₂ concentrations. δ¹³C-CO₂ becomes increasingly depleted
 594 as CH₄ is oxidized to CO₂. Hence, enriched values of δ¹³C-CO₂ indicate very little CH₄
 595 oxidation, while depleted values indicate more CH₄ oxidation. With the exception of one

596 borehole bubble gas sample, in which $\delta^{13}\text{C-CO}_2$ was -20.7 ‰ and the corresponding $\delta^{13}\text{C-CH}_4$
597 was -66.2 ‰, borehole bubbles at Vault Lake had enriched ‰ values ($\delta^{13}\text{C-CO}_2$ -5.1 to -15.4 ‰;
598 $\delta^{13}\text{C-CH}_4$ -74.2 to -75.5 ‰; Winkel et al., 2019). This indicates CH_4 produced at the base of the
599 talik is subject to less AOM, which is supported by the low $\text{CO}_2:\text{CH}_4$ porewater ratios at this
600 depth. In contrast, bubbles occurring from natural ebullition seeps draining free-phase gas from
601 shallower sediments had more depleted $\delta^{13}\text{C-CO}_2$ (-14.0 to -26.2 ‰) and more enriched $\delta^{13}\text{C-}$
602 CH_4 (-63.0 to -75.5 ‰), suggesting these bubbles were formed from pore space gas that was
603 more influenced by AOM.

604 **7. Potential changes with climate warming**

605 Thermokarst lake formation alters local thermal dynamics, leading to lateral and vertical thaw
606 and erosion, the formation of taliks (Brewer, 1958), and mobilization C previously sequestered in
607 permafrost (Fig. 9). A process-based climate-sensitive lake biogeochemical model by Tan et al.
608 (2015) indicated that seasonal variation in CH_4 emissions can be explained by energy input. In
609 mesocosm warming experiments conducted in a shallow pond in the Netherlands, warming by 4
610 °C increased ebullitive CH_4 flux by 51%, but did not significantly affect diffusive CH_4 flux
611 (Aben et al., 2017). The combination of increased energy input due to warmer temperatures and
612 increased substrate potential due to additional permafrost thaw suggest additional climate
613 warming will increase CH_4 fluxes and production in thermokarst lakes such as Vault Lake.



614

615 **Figure 9.** Current and future CH₄ dynamics in a thermokarst lake environment. Climate warming
 616 is expected to lead to increased lateral and vertical thaw and erosion, which may lead to lake
 617 expansion or lake drainage (see section 8 below). In addition, warmer temperatures are expected
 618 to lead to increased CH₄ production and oxidation and increased ebullitive and diffusive flux.
 619 Fewer days under ice cover are anticipated to reduce ice storage bubble flux and under-ice CH₄
 620 oxidation. The net effects on annual CH₄ flux remain uncertain, although literature suggests
 621 these combined factors will lead to increases in CH₄ flux from thermokarst lakes with additional
 622 warming. See Section 8 for additional discussion of uncertainties.

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Temperature sensitivity analyses conducted on sediments from the Vault Lake core suggest CH₄ production in recently-thawed former permafrost is highly sensitive to warming and thaw (0 °C to 3 °C; Heslop et al., 2019a). Therefore, we postulate permafrost warming and thaw at the base of the talik increases the available activation energy to initiate the chemical reactions involved in CH₄ production (Contant et al., 2011), and subsequently increases the number of

629 methanogens and CH₄ production rates (Knoblauch et al., 2018, Wei et al., 2018). Acetoclastic
630 methanogens, which were in higher relative abundance near the thaw front, have lower
631 temperature optima than hydrogenotrophic methanogens (Allen et al., 2014) and therefore may
632 be more suited to metabolic processes at these colder temperatures. While incubated sediments
633 from near the thaw boundary had high temperature sensitivities at lower incubation temperatures,
634 the sediments did not produce significantly more CH₄ at higher incubation temperatures (10 °C
635 and 25 °C; Heslop et al., 2019a). This suggests current CH₄-producing microbial communities
636 near the thaw boundary are not substrate-limited and optimized for colder temperatures.

637 While increased energy input from permafrost thaw increases CH₄ production at lower
638 temperatures, at temperatures consistent with thermokarst lake taliks (4 °C) CH₄ production is
639 only at 17% of maximum activity (Metje and Frenzel, 2007). Laboratory incubation studies
640 suggest the optimal temperatures for methanogenesis are between 26-28 °C (Metje and Frenzel,
641 2007). This implies additional warming will cause a second increase in CH₄ production. In
642 sediments of the Vault Lake core estimated to have been thawed centuries (100 to 400 years),
643 CH₄ production did not significantly increase with warming at lower incubation temperatures
644 (0 °C to 3 °C) but increased when sediments were warmed to temperatures above those observed
645 *in situ* (10 °C and 25 °C; Heslop et al., 2019a). This implies that the former permafrost thawed
646 for centuries, which currently has low CH₄ production potentials (Heslop et al., 2015; Kessler et
647 al., 2012) due to low substrate bioavailability (Heslop et al., 2017) and/or the establishment of
648 AOM microbial communities (Winkel et al., 2019), has the potential to support additional CH₄
649 production following additional warming. This increased CH₄ production potential may be due
650 to increased substrate bioavailability due to higher ambient energy in warmer temperatures, a
651 shift in microbial communities from predominately methanotrophs to predominately

652 methanogens, or a combination thereof. In a landscape context, such additional warming of these
653 sediments may occur with warmer atmospheric temperatures and/or further thermokarst or
654 thermo-erosion exposing and mobilizing these sediments in the landscape.

655 While warming will increase CH₄ production rates in lake sediments, we expect the
656 increased production may be partially offset by increased CH₄ consumption rates (Oh et al.,
657 2020). However, in a nearby yedoma thermokarst lake in Interior Alaska, higher Q₁₀ values for
658 methanogenesis (Q₁₀ = 8.5 ± 0.1) versus aerobic CH₄ oxidation (Q₁₀ = 2.7 ± 0.3) in the water
659 column suggest increases in CH₄ production will outpace increases in CH₄ consumption
660 (Sepulveda-Jauregui et al., 2018). Further, it has been suggested that the shortening of ice cover
661 periods in thermokarst lakes due to warming and increasing snow thickness (Arp et al. 2018)
662 leads to more CH₄ being emitted into the atmosphere via direct ebullition instead of being
663 oxidized while trapped under ice (Elder et al., 2019; Greene et al., 2014). In Finnish boreal lakes,
664 more ice-free days have also been linked to greater diffusive CH₄ flux (Guo et al., 2020).
665 Additional research is necessary to determine how these different controls will affect net CH₄
666 emissions with warming in thermokarst and northern lakes.

667 **8. Future research directions**

668 *Thermokarst lake expansion versus drainage.*

669 Ongoing efforts to constrain the diverse impacts of climate change and permafrost thaw on
670 thermokarst lake expansion or drainage are critical for determining future CH₄ release potentials,
671 as the landscape wetting (lake expansion) versus drying (lake drainage) will influence whether
672 permafrost C is thawed under aerobic or anaerobic conditions (Lawrence et al., 2015). While in
673 the Interior Alaska region near Vault Lake, net thermokarst lake area has increased by ~43%
674 since 1949 (Walter Anthony et al., *submitted*), an examination of continental-scale Landsat

675 imagery transects in North America and Eurasia found northern net lake area decreased 1.45%
676 between 1999 and 2014 (Nitze et al., 2018). Despite net lake area decreases, thermokarst lake
677 expansion by lateral shore erosion in Arctic-Boreal lowlands with ice-rich permafrost continues
678 to cause significant gross lake area growth (transformation of land area to lake area; (Nitze et al.,
679 2018). This in turn leads to growth of taliks and thawed sediment volumes under lakes (Walter
680 Anthony et al., 2018). Mobilized permafrost C in taliks would be processed in anaerobic
681 environments, meaning it is released as both CH₄ and CO₂. On the other hand, lake drainage
682 changes local thermal dynamics such that: (1) taliks formed beneath a lake can refreeze and (2)
683 near-surface C is more likely to be mineralized under aerobic conditions as CO₂.

684 *Physical entrapment of CH₄.*

685 Our first-order calculations suggest up to 99% of CH₄ produced at Vault Lake is not emitted into
686 the atmosphere. Coupled with our estimates that 32-83% of produced CH₄ is oxidized *in situ*
687 (Martinez-Cruz et al. 2017; Winkel et al., 2019), this suggests that 16-67% of produced CH₄
688 becomes entrapped within the Vault Lake system. While we acknowledge these values are rough
689 first order estimates, they suggest periodic physical entrapment plays a substantial role in
690 thermokarst lake CH₄ budgets. The location and fate of this entrapped CH₄ is not well
691 constrained; however, rapid and sustained ebullition (for at least one day) observed in boreholes
692 distributed across Vault Lake drilled in 2013 to map the talik dimensions suggests a large
693 volume of gas was entrapped in sediments. Voids observed in freshly split sediment cores
694 collected from thermokarst lakes, including Vault Lake (Heslop, *unpublished data* Walter
695 Anthony, *unpublished data*), are suggestive of gas pockets storing CH₄ within the sediment.
696 While to our knowledge these gas bubbles have not been quantified in thermokarst lake
697 sediments, laboratory incubations examining gas bubble formation in wetland sediments suggest

698 gas-filled voids, which are a precursor to the formation of ebullition bubble conduits, store more
699 gas in clay and silt sediments (maximum 18.8% volumetric gas content) than sandy and silty
700 sediments (maximum 13.2% and 12.0% volumetric gas content, respectively; Liu et al., 2016). In
701 these laboratory incubations, hydrostatic head drop was linearly correlated with the magnitude of
702 ebullition episodes releasing stored gas (Liu et al., 2016).

703 Depending on the phase and location of the entrapped gas, it may or may not be subject
704 to CH₄ oxidation. Only the dissolved phase of CH₄ is subject to oxidation, so if gases are stored
705 as bubbles in sediments, then the bubble phase protects CH₄ from *in situ* oxidation. For instance,
706 the $\delta^{13}\text{C}\text{-CH}_4$ values of bubbles released from the borehole drilling at Vault Lake (mean -73‰, n
707 = 5 samples) suggests that this trapped gas was not enriched by oxidation. Additional, high-
708 resolution analyses in multiple thermokarst lakes to determine if and where entrapped CH₄ is
709 subject to additional oxidation and what physical factors determine if and how it can be released
710 will help better constrain the role of CH₄ emissions from thermokarst lakes in the PCF.

711 *Incorporating thermokarst C dynamics into models*

712 First-order models indicate up to 20% of the permafrost region is anticipated to
713 experience physical disturbance thaw processes by 2300 (Turetsky et al., 2020), with GHG
714 emissions from thermokarst lakes having the potential to more than double the PCF this century
715 (Schneider von Deimling et al. 2015, Walter Anthony et al. 2018). Yet, large-scale global models
716 predicting the magnitude of the PCF currently only consider gradual thaw processes (e.g. active
717 layer deepening) and do not incorporate potential C emissions from rapid thaw processes,
718 including thermokarst lakes (Turetsky et al., 2020). Improved understanding CH₄ dynamics
719 within thermokarst lake environments will aid in incorporating their GHG emissions into future
720 estimates of the PCF. While there have been studies modelling the formation of taliks beneath

721 water bodies and subsequent potential C emissions under different climate scenarios (Kessler et
722 al., 2012; Langer et al., 2016; Schneider von Deimling et al., 2015) and a process-based climate-
723 sensitive lake biogeochemical model predicting CH₄ emissions (Tan et al., 2015), additional
724 research is needed to incorporate these models into larger scale permafrost thaw and C dynamic
725 models. Our work at Vault Lake suggests that the sediment layers of the talik experience
726 different rates of methanogenesis and AOM due to varying substrate potentials and microbial
727 communities, supporting the suggestion by Tan et al. (2015) that variability of CH₄ emissions
728 can be primarily explained by energy input and substrate availability. In turn, this led to the
729 facies experiencing different responses to additional warming. Integrating this depth-dependent
730 variability in thermokarst lake GHG production dynamics into thermokarst models, including
731 spatially dynamic estimates of CO₂:CH₄ production ratios, CH₄ oxidation, and gas release versus
732 entrapment, is a first step in refining our estimates of how rapid thaw processes will contribute to
733 the PCF. Given the variation we see between sediment layers within the same core at one
734 thermokarst lake, it is necessary to examine additional thermokarst lake cores from diverse
735 Arctic environments to determine how permafrost thaw history, sediment background, and OM
736 and microbial characteristics influence CH₄ dynamics. Examining spatial variability within the
737 same lake, including variation in facies thicknesses, talik depth, microbial communities, and
738 substrate potential, will further refine estimates of potential GHG emissions from thermokarst
739 lakes. More drilling studies capturing entire talik profiles are necessary to determine if patterns
740 observed at Vault Lake hold true for other thermokarst lakes, including thermokarst lakes in non-
741 yedoma regions.

742 In addition to additional field sampling to constrain variability, we suggest one
743 approach to accounting for this spatial and temporal heterogeneity in models is to adopt a

744 microscale approach, where variabilities in substrate availability, the microbiome, and energy are
745 incorporated into geochemical rate models (e.g. Neumann et al., 2016) to predict CH₄ production
746 and consumption rates in each facies. The predicted GHG production and consumption rates can
747 then be integrated into whole-lake models that take into account thaw beneath lakes (e.g. Kessler
748 et al., 2012; Langer et al., 2016; Schneider von Deimling et al., 2015), and eventually into larger
749 models predicting the impact of thermokarst lakes on the PCF (e.g. Turetsky et al., 2020; Walter
750 Anthony et al., 2018).

751 **9. Summary and conclusions**

752 At Vault Lake in Interior Alaska, a typical recently-formed thermokarst lake in a lowland
753 permafrost setting with yedoma deposits, we observed depth-dependent variations in CH₄
754 production and oxidation. Variations in CH₄ production and oxidation potentials corresponded
755 with changes in CO₂:CH₄ ratios and net CH₄ production response to temperature increases. The
756 top 150 cm of the Vault Lake core had the highest CH₄ production potentials in 3 °C incubations,
757 but oxidation potentials suggest 41-83% of the produced CH₄ is anaerobically oxidized *in situ*.
758 This region also had higher CO₂:CH₄ ratios compared to the remainder of the core. The
759 sediments near the thaw boundary, which prior modelling and ebullition bubble mapping studies
760 of thermokarst lakes suggested are an important source of CH₄ production, had the highest *in situ*
761 CH₄ concentrations and high CH₄ production potentials in incubations. CH₄ production near the
762 thaw boundary was sensitive to warming at temperatures consistent with *in situ* temperatures in
763 the talik and did not respond to additional warming. Further, the lowest CO₂:CH₄ ratio was
764 observed near the thaw boundary, potentially due to higher proportions of CH₄ being produced
765 through acetoclastic versus hydrogenotrophic methanogenesis. Former permafrost sediments,
766 thawed centuries ago, had high CO₂:CH₄ ratios that were potentially influenced by high AOM

767 rates. These sediments also experienced significant increases in net CH₄ production when
768 warmed to temperatures above those observed *in situ*.

769 While not studied at Vault Lake, additional processes within the water column such as
770 CH₄ oxidation and heterotrophic C respiration affect CO₂:CH₄ ratios and how much CH₄ is
771 emitted into the atmosphere. The form of CH₄ in the thermokarst lake system (bubbles versus
772 dissolved) is a key control on whether CH₄ is subject to *in situ* oxidation. Therefore, more
773 research is necessary to better constrain how processes such as CH₄ storage within the system,
774 including storage in sediments and under ice cover, affect lake CH₄ dynamics and emissions.

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