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2 COMPARISON OF GAS, KLINKENBERG AND LIQUID PERMEABILITY OF SANDSTONE

3

– FLOW REGIME AND PORE SIZE.

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ABSTRACT

13
14 Liquid permeability of sedimentary rocks is relevant in several contexts, but gas permeability is
15 easier to measure, so liquid permeability is typically estimated from gas permeability via empirical or
16 semi-empirical correction procedures. A frequently used and trusted procedure is the well-known
17 Klinkenberg correction, which is based on the pressure dependence of gas permeability. However,
18 from gaseous and liquid flow-through experiments on a series of Fontainebleau, Castlegate,
19 Bentheim, and Obernkirchen sandstones, this study indicates that the equivalent liquid permeability
20 derived from gas permeability via the Klinkenberg correction, only compares with liquid
21 permeability, when the gaseous flow adheres to Darcy's law. The lower and upper limits to Darcy
22 flow are defined by the Knudsen and Reynolds numbers, respectively. Both numbers can be
23 estimated from porosity and pore throat distribution, so from these properties, it is possible to assess
24 the flow and pressure limits for the applicability of the Klinkenberg correction. For the studied
25 sandstones, non-Darcy flow is indicated for the largest pores with diameters above around 10 μm ,
26 causing an erroneous Klinkenberg' correction. Knudsen diffusion takes place in pores smaller than
27 around 0.1 μm , but the contribution to the overall gas permeability of these small pores is however
28 insignificant in these sandstones. Liquid permeability modelled from contributions from each pore
29 size by using Kozeny's equation and NMR T_2 surface relaxation data, shows that the largest pores
30 have no positive effect on permeability due to the existence of pore throats, rather they may have a
31 negative effect on permeability due to turbulence.

INTRODUCTION

32

33 Permeability is a key reservoir parameter because it describes the transport of fluid through the
34 connected pore space of a sediment or sedimentary rock. According to the classical Darcy definition,
35 permeability, k , is an intrinsic hydraulic material property, which describes the degree to which a
36 fluid can flow through the material in one direction. The precondition is a linear relation between
37 fluid discharge and pressure drop. Permeability has the unit of area and thus in principle is
38 independent of fluid properties. However, it is well known from experimental studies, that whereas
39 transport of liquids in the general case is independent of the pressure level, gas permeability, k_g , is
40 pressure dependent. Because it is much easier to measure gas permeability, Klinkenberg (1941)
41 proposed an empirical method for correcting gas permeability to the equivalent liquid permeability.
42 Although validated initially by using Jena glass filters and undefined core samples with unknown
43 petrophysical properties (Klinkenberg 1941), the method has been widely applied and trusted, and
44 has also formed the basis for theoretical and empirical modifications (e.g., Sampath and Keighin,
45 1982; Tanikawa and Shimanoto, 2009; Civan, 2010; Moghadam and Chalaturnyk, 2014; Al-Jabri et
46 al., 2015; Li and Sultan, 2017). In order to evaluate the applicability of Klinkenberg's original
47 method to sandstones, the limited number of experiments with a direct comparison of Klinkenberg
48 and liquid permeability (e.g. Sampath and Keighin, 1982; Tanikawa and Shimanoto, 2009; Carles et
49 al., 2007; Chen et al., 2016) were complemented with new data. Thirteen sandstone samples were
50 selected for flow through experiments using liquid (freshwater or brine) as well as gas (argon), so
51 that both liquid and a series of gas permeabilities were obtained, and the equivalent liquid
52 permeability was derived by Klinkenberg's procedure. The results were evaluated by taking into
53 account the constraints on Darcy's equation concerning flow regime inherent to the Klinkenberg
54 procedure. Similar studies have been conducted on shales and tight rocks in general (e.g. Civan,

55 2010; Freeman et al., 2011; Fathi et al., 2012; Ziarani and Aguilera, 2012; Heller et al., 2014;
56 Moghadam and Chalaturnyk, 2014; Dadmohammadi et al., 2017a).

57 In order to account for effects related to the flow regime in flow-through experiments,
58 parameters such as the Knudsen (Kn) and Reynolds (Re) numbers are needed. The Knudsen number
59 is the ratio of the molecular mean free path to a characteristic length scale, whereas the Reynolds
60 number is the ratio of inertial to viscous forces. Both numbers thus depend on a characteristic length,
61 which for fluid flow in porous rocks typically is estimated as the pore- or pore throat diameter
62 assuming a circular cross-section of the pores.

63 FIGURE 1

64 For fluid flow in small pores with a diameter comparable to or smaller than the molecular
65 mean free path of the fluid in flow, a statistical formulation governs the fluid dynamics, and the
66 Knudsen number defines the end of a transition from free molecular flow (diffusion) to slip flow
67 (where Darcy's equation applies). At the opposite end of the flow scale, continuum mechanics
68 governs in large pores, and the Reynolds number separates the linear flow regime and the transition
69 to turbulent flow. By choosing pore throat size as the characteristic length scale, the end of a
70 transition to Darcy (slip) flow (d_{Kn}) as well as the limit of a transition to non-linear Darcy flow (d_{Re})
71 can be determined (Figure 1). This is done by combining the commonly used Knudsen and Reynolds
72 numbers for regime separation (Kn_{trans} and Re_{trans}) with data of mean pressure and discharge recorded
73 from gaseous flow-through experiments. This procedure is especially important in high permeable
74 material (>100 mD, $9.87E^{-14}$ m²), where a high specific fluid discharge increases the risk of flow
75 conditions in the non-linear transition from Darcy- to turbulent flow or even in the fully turbulent
76 regime. However, is it not common practice to quantify and evaluate the flow regime in flow-through

77 experiments on sandstones when the permeability reaches orders of mD, as it was done by
78 Bloomfield and Williams (1995) and Huang and Ayoub (2008).

79 Reliable permeability prediction from non-hydraulic petrophysical characteristics of a random
80 sedimentary rock involves two points of interest: 1) inaccuracy in assumptions for an estimation of
81 input characteristics and 2) physically relating the non-hydraulic material characteristics to a
82 hydraulic flow property, i.e. permeability. In general, inaccuracies in assumptions are practically
83 unavoidable. In order to relate material characteristics to a flow property, several authors consider the
84 relationship between permeability, porosity and specific surface by using various modifications of the
85 well-known Kozeny's equation (Kozeny, 1927) (e.g., Costa, 2006; Xu and Yu, 2008; Henderson et
86 al., 2010; Latief and Fauzi, 2012; Rabbani and Jamshidi, 2014; Rosenbrand et al., 2014). Kozeny's
87 equation accounts for pore geometry through Kozeny's factor, c , which in many studies is a pure
88 empirical constant, but for homogeneous rocks it may be estimated from porosity (Mortensen et al.
89 1998). In order to estimate the homogeneity of the studied brine-saturated sandstones, the electrical
90 resistivity was measured, and the corresponding porosity exponent, m , was derived from Archie's
91 equation. A low m was assumed to result from an inhomogeneous or internally cracked sample.

92 To assess to which extent permeability of the studied sandstones is governed by the a given
93 pore size and the size of the corresponding pore throat (Rosenbrand et al., 2015b), the concept of
94 Hossain et al. (2011) was followed. It relates an individual pore size as derived from low field
95 Nuclear Magnetic Resonance (NMR) to the corresponding individual permeability contribution by
96 using Kozeny's equation. The contributions to permeability from each pore size were then cumulated
97 from small pores up to match the measured liquid permeability. An upper cut-off pore size was
98 defined as the largest pore size to take into account, thus identifying pores smaller than this cut-off
99 size as effective for the overall permeability.

THEORY

100

101 **Gas permeability, flow regime, and Klinkenberg effect**

102 For linear Darcy flow, permeability, as measured using a liquid (e.g. a brine), is typically lower than
 103 permeability as measured using gas (e.g., Heid et al., 1950; Jones and Owens, 1980). According to
 104 Poiseuille's equation and Kozeny's concept, the velocity of a liquid flowing in a porous medium is
 105 zero at the pore walls due to friction. Across a pore, the resulting velocity profile becomes a parabola
 106 due to viscosity. Inter-molecular collisions dominate in liquid flow when the molecular mean free
 107 path is small as compared to the pore size. The molecular mean free path of gases is in general orders
 108 of magnitudes larger than that of liquids and Klinkenberg (1941) argued in his studies that flowing
 109 gas molecules experience a so-called slipping associated with a non-zero velocity at the pore walls.
 110 The resulting higher permeability is denoted the Klinkenberg effect, and although the theoretical
 111 basis is not rigid, Klinkenberg (1941) proposed deriving the equivalent liquid permeability (i.e. the
 112 Klinkenberg permeability, k_K) by theoretically extrapolating the mean free path of the applied gas to
 113 that of a liquid. Because the mean free path is pressure dependent at isothermal conditions, such an
 114 extrapolation corresponds to letting the pressure approach infinity. Thus, after deriving k_g from
 115 Darcy's equation for a series of pressures, Klinkenberg made a linear correlation of k_g and the inverse
 116 mean pressure, $1/P_m$, and determined the equivalent liquid permeability, k_K , by extrapolating k_g to
 117 zero $1/P_m$. Inherent to the proposed correlation is inevitably that the experimental flow conditions
 118 obey Darcy's law. In Darcy flow, the average flux (specific discharge), q , has an upper limit defined
 119 by the Reynolds number:

$$120 \quad Re = \frac{\rho_f q d_{Re}}{\mu \phi}, \quad (1)$$

121 where ρ_f is the fluid density, μ is the dynamic viscosity of the fluid, ϕ is porosity and d_{Re} is the
 122 characteristic length scale. The specific discharge, q , is determined from measured discharge, Q and

123 cross-sectional area, A ($q = Q/A$). The lower limit for Darcy flow is defined by the Knudsen number.
 124 For very low mean pressure or very small pores with diameter comparable to or smaller than the
 125 molecular mean free path, collisions between gas molecules and the pore wall dominate the flow as
 126 discovered experimentally and described theoretically by Knudsen (1909). This Knudsen diffusion or
 127 flux is limited by the Knudsen number:

$$128 \quad Kn = \frac{\lambda}{d_{Kn}}, \quad \lambda = k_b T (\sqrt{2} \pi d_p^2 P)^{-1}, \quad (2)$$

129 where λ is the molecular mean free path, T is absolute temperature, P is pressure assumed equal to the
 130 mean pressure, P_m , d_p is the diameter of the flowing molecule, k_b is the Boltzmann constant and d_{Kn} is
 131 the characteristic length scale. In terms of the Knudsen number, the limits noted by Karniadakis et al.
 132 (2005) are generally agreed upon as separators for gaseous flow, where in the regime of $Kn > 10$ the
 133 flow is dominated by diffusion, while $10 > Kn > 0.1$ characterize the transitional regime to linear
 134 Darcy flow. Linear Darcy flow with slippage is characterized by $Kn < 0.1$, and in terms of the
 135 Reynolds number is defined by $1 < Re < 10$, whereas the transition to non-linear Darcy flow is
 136 represented by $Re > 10$ (Scheidegger, 1960; Bear, 1972; Hassanizadeh and Gray, 1987). However, in
 137 tight low-permeable formations, Dadmohammadi et al. (2017b) found that there can be
 138 circumstances where regime limits are in disagreement with the ones generally accepted.

139 A sandstone typically contains pores of differing size, where each, for a given flow or mean
 140 pressure can define different flow regimes (equation 1 and 2) and determine if flow and pressure
 141 conditions are in or outside the framework of Darcy's law (Figure 1 and 2). Similarly, by knowledge
 142 of pore throat distribution before a planned series of gaseous flow-through experiments, pressures as
 143 well as specific discharges defining thresholds of transition zones can be determined (equation 1 and
 144 2), thus enabling an experimental choice of pressure levels so that Darcy's law is valid and
 145 Klinkenberg's procedure is applicable.

146 FIGURE 2

147 **Modelling permeability from Kozeny's equation and NMR**

148 For a homogenous sedimentary rock with a surface to volume ratio (specific surface) of the pores, S_p ,

149 Kozeny (1927) relates permeability, k_z , to porosity as:

$$150 \quad k_z = c \frac{\phi}{S_p^2}. \quad (3)$$

151 Kozeny defined c as an empirical factor and found it to be approximately 0.25 for sandstones. In

152 practice, c can account for the geometry of the pore space, including flow obstruction and

153 heterogeneity.

154 By envisaging the pore space as 3D orthogonally arranged and interpenetrating tubes,

155 Mortensen et al. (1998) applied Poiseuille's law to derive the fraction of the pore space that controls

156 the flow in one direction. By assuming a homogenous distribution of the specific surface, hence only

157 accounting for shielding effects, they derived an expression for Kozeny's factor, assuming a circular

158 pore cross-section as a function of porosity defined as:

$$159 \quad c_M = \left\{ 4 \cos \left[\frac{1}{3} \arccos \left(\phi \frac{64}{\pi^3} - 1 \right) + \frac{4\pi}{3} \right] + 4 \right\}^{-1}. \quad (4)$$

160 The expression by Mortensen et al. (1998) is denoted as the shielding factor, c_M . For porosities of

161 interest to most sedimentary rocks (2-40%), the expression in equation 4 approximates a linear

162 relation (Figure 3) and the expression can be simplified to:

$$163 \quad c_M = 0.155\phi + 0.175, \quad 0.02 < \phi < 0.4. \quad (5)$$

164 FIGURE 3

165 The distribution of specific surface in the pore-space can be derived from low field Nuclear

166 Magnetic Resonance (NMR) relaxometry. The transverse relaxation time, T_2 , derived from the

167 relaxation exponent has contributions from the solid-fluid interface ($T_{2,\text{Surface}}$), bulk relaxation in the
 168 fluid ($T_{2,\text{Bulk}}$) and molecular diffusion in the field gradient ($T_{2,\text{Diffusion}}$):

$$169 \quad \frac{1}{T_2} = \frac{1}{T_{2,\text{Surface}}} + \frac{1}{T_{2,\text{Bulk}}} + \frac{1}{T_{2,\text{Diffusion}}} . \quad (6)$$

170 In the laboratory, instrumental settings can allow surface relaxation to dominate, and equation 6
 171 reduces to:

$$172 \quad \frac{1}{T_2} \approx \frac{1}{T_{2,\text{Surface}}} \approx \rho_2 S_p \quad (7)$$

173 where ρ_2 is the surface relaxivity related to the mineralogy. Paramagnetic atoms in minerals or on
 174 mineral surfaces significantly affect ρ_2 , resulting in shorter relaxation times.

175 By combining equations 3 and 7 in accordance with Hossain et al. (2011), an expression for
 176 permeability, k_{NMR} , based on NMR and Kozeny's equation can be formulated as:

$$177 \quad k_{\text{NMR}} = c_M \phi (T_2 \rho_2)^2 , \quad (8)$$

178 where c from equation 3 is approximated by c_M . If expanded to represent incremental contributions to
 179 permeability, k_{NMR} equals:

$$180 \quad k_{\text{NMR},i} = c_M \phi f_{\text{NMR},i} (T_{2,i} \rho_2)^2 , \quad (9)$$

181 where $f_{\text{NMR},i}$ is the fraction of the porosity corresponding to $T_{2,i}$. The incremental porosity, $\phi'_{\text{inc,NMR}}$, is
 182 defined as $\phi f_{\text{NMR},i}$ and the cumulated permeability $k_{\text{NMR,cum}}$ as the summation of individual $k_{\text{NMR},i}$
 183 from the smallest pore size. Note that c_M is a bulk term accounting for shielding effects in a
 184 connected pore space and must hence be estimated from the total porosity.

185 **Pore and pore throat distributions**

186 By assuming a circular cross-section of all pore bodies, the pore diameter, d_{NMR} , can be estimated
 187 from NMR T_2 relaxation time through an expansion of equation 7:

$$188 \quad \frac{1}{T_2} = \rho_2 S_P = \rho_2 \left(\frac{2\pi r l}{\pi r^2 l} \right) = \rho_2 \frac{4}{d_{\text{NMR}}} \Leftrightarrow d_{\text{NMR}} = 4\rho_2 T_2, \quad (10)$$

189 where r is the pore radius and l is the pore length.

190 A pore throat distribution can be derived from Mercury Injection Capillary Pressure (MICP)
191 data through Washburn's equation:

$$192 \quad P_c d_{\text{Hg}} = -4\gamma \cos(\theta) \Leftrightarrow d_{\text{Hg}} = \frac{-4\gamma \cos(\theta)}{P_c}, \quad (11)$$

193 where d_{Hg} is pore throat diameter assuming a circular cross-section, P_c is the capillary pressure, γ is
194 the surface tension, and θ is the contact angle of the applied injection fluid. Mercury (Hg) is
195 commonly applied because of its non-wetting character and high density. The incremental porosity,
196 $\phi'_{\text{inc,Hg}}$, is the volume increment of injected fluid (Hg) corresponding to a capillary pressure interval
197 divided by the bulk volume.

198 Archie's equation

199 From experiments on partly saturated sedimentary rocks, Archie (1942) found the following
200 relationship between water saturation, S_w , porosity, the electrical resistivity of the pore water, R_w , and
201 the electrical resistivity of the rock, R_t :

$$202 \quad S_w = \left(\frac{1}{\phi^m} \frac{R_w}{R_t} \right)^{\frac{1}{n}}. \quad (12)$$

203 Archie (1942) defined n as the saturation exponent, whereas he denoted m as the cementation
204 exponent (Archie, 1942). The term "cementation exponent" is, however, misleading as it implies a
205 non-existing connection to cementation. Hence, the denotation of m as porosity exponent is adopted
206 here in accordance with recent studies (Revil et al., 2014; Corbett et al., 2017; Niu and Zhang, 2018).
207 By defining the formation factor, F , as the ratio between the electrical resistivity of the fully water
208 saturated rock, R_0 , and R_w , the commonly applied equation 13 is obtained:

209
$$F = \frac{R_0}{R_w} = \frac{1}{\phi^m}, \quad (13)$$

210 which only applies when the electric current is carried primarily by an electrolytic pore fluid.
211 Equation 13 does not account for electrical surface conductivity associated with clay minerals which
212 become significant in weak electrolytes. From experiments on various sandstones Archie found m to
213 be in the relatively narrow range between 1.8 and 2.0. By assuming that the majority of the material
214 in Archie's study were unfractured or only fractured on a microscale level with insignificant
215 influence on permeability, a porosity exponent in the range from 1.8 to 2.0 hence quantitatively
216 represents an unfractured core. For core plugs with a significant yet invisible fracture, R_0 should be
217 low, which from equation 13 would result in a porosity exponent below 1.8.

218 MATERIALS AND METHODS

219 Sandstones

220 A series of plug samples from outcrop sandstones, originating from 1) Fontainebleau, France, 2)
221 Castlegate, USA, 3) Bentheim, Germany, and 4) Obernkirchen, Germany were studied. They were
222 selected such that a wide range of porosity and permeability is represented. The bulk mineral
223 composition from X-ray diffraction (XRD), specific surface by N_2 adsorption (BET-method;
224 Brunauer et al., 1938) as well as polished thin sections for backscatter electron micrography (BSEM)
225 were obtained from side trims. Quartz dominates all samples, and among clay minerals, kaolinite and
226 illite were detected in Castlegate sandstone whereas only kaolinite was detected in Obernkirchen
227 samples (Table 1, Figure 4d and e). In Fontainebleau sandstone, no clay minerals were detected
228 (Table 1, Figure 4a and b). Neither XRD nor BET data for Bentheimer samples indicate the presence
229 of clay minerals, but clusters of locally distributed kaolinite within the pore space are, however,

230 visible on backscatter electron micrographs (BSEM, Figure 4c). Consequently, a clay content of 2.7
231 mass % was listed in Table 1 in accordance with Peksa et al. (2017).

232 TABLE 1

233 FIGURE 4

234 **Experimental methods**

235 The sandstone plugs of 25 mm in diameter and 50 mm in length were oven-dried (60°C) and
236 equilibrated. Plugs were weighed and the dry density was derived before measurements of grain
237 density and gas-porosity by N₂ expansion.

238 Each sandstone plug was placed in a core holder designed for flow-through experiments, and
239 at a confining stress of 5 MPa, the discharge and pressure gradient from flow of argon gas was
240 measured. The gas permeability, k_g , was derived at steady state from Darcy's equation corrected to
241 account for gas compressibility by the ideal gas law (Dullien, 1979; Bloomfield and Williams, 1995;
242 Tanikawa and Shimanoto, 2009) as well as for the pressure dependency of fluid viscosity. The
243 discharge and pressure gradient was measured at a minimum of three upstream pore pressure levels
244 against atmospheric pressure downstream, and all measurements were done twice. Plugs of
245 Fontainebleau sandstone were then saturated with demineralized water, and the other sandstone plugs
246 with a 0.5M KCl brine solution before being weighed for determining saturated bulk density and the
247 degree of saturation, and afterwards were placed in the core holder. Each core plug was then exposed
248 to a fixed discharge of demineralized water or brine at identical confining stress to that used with the
249 argon experiments. The resulting pressure gradient was measured at steady state from which the
250 water permeability, k_w , was derived from Darcy's equation. Measurements were done twice, and
251 following the second measurement, Fontainebleau plugs were oven dried (60°C) and re-saturated
252 with the 0.5M KCl brine solution. The electrical resistivity was then measured on the brine saturated

253 core plugs at a uniaxial stress of 3 MPa. Measurements were conducted in a 1 kHz AC circuit and
254 with 5 V power supply.

255 NMR data were measured on brine saturated core plugs using a GeoSpec2 NMR Core
256 Analyzer at atmospheric pressure and a frequency of 2.25 MHz at a temperature of 35 °C.
257 Measurements were conducted with a recycle delay (repetition time) of 25 s, 16,000 echoes and a
258 CPMG inter echo spacing (τ) of 50 μ s. Using the Carr-Purcel-Meiboom-Gill (CPMG) pulse
259 sequences, T_2 relaxation spectra were derived using the WinDXP (Oxford Instruments UK) software.
260 From the calibrated signal amplitude of the saturating brine, signal amplitudes were converted to
261 derive the total NMR porosity, ϕ_{NMR} , and the incremental NMR porosity, $\phi'_{\text{inc,NMR}}$. The pore size
262 distribution was derived from equation 10 by adjusting the surface relaxation such that pore size
263 distributions obtained from NMR align with the pore throat distributions from MICP (Marschall et
264 al., 1995).

265 Saturated core plugs were methanol cleaned by the Soxhlet method before chopping off rock-
266 chips for measurements of Mercury Injection Capillary Pressure (MICP) using a Poremaster PM 33-
267 GT-12 instrument. The volume of intruded Hg was recorded at stepwise pressures up to 415 MPa.
268 Intruded Hg volumes were converted to incremental porosity, $\phi'_{\text{inc,Hg}}$, and the d_{Hg} distribution was
269 derived from the capillary pressure curves using Washburn's equation (equation 11).

270 By using the atomic diameter of argon gas (0.38 nm), combined with specific discharge,
271 porosity and mean pressure as input for equations 1 and 2 (using $Re = 10$ and $Kn = 0.1$), the specific
272 pore throat sizes d_{Re} and d_{Kn} (Figure 1) were established. From these diameters and the pore throat
273 distribution, it is possible to identify if non-linear flow takes place in parts of the pore space (Figure
274 1). For instance, if parts of the pore volume have pore throats corresponding to $Re > 10$, non-linear
275 Darcy flow is expected (Figure 2, third row).

RESULTS

276

277 **Porosity and pore size distribution of the studied sandstones**

278 In accordance with XRD analysis, a grain density close to 2.65 g/cm^3 (Table 2) corresponds to the
279 dominance of quartz (Table 1). A grain density of 2.71 g/cm^3 found in Castlegate samples indicates
280 the presence of heavier minerals. N_2 -porosity of the studied material ranges from 0.05 to 0.31 (Table
281 2). The saturation degree ranges between 0.84 and 0.98 with lowest values generally found in low
282 porosity samples.

283 TABLE 2

284 NMR porosity, ϕ_{NMR} , is in general found to be 0.02 lower than $\phi_{\text{N}} S_w$, even 0.05 lower for Bentheimer
285 and Castlegate samples (Tables 2 and 3). Presumably related to experimental errors, the difference
286 increases with porosity. Adjusting the surface relaxation (ρ_2) to match peak on peak of, respectively,
287 pore size and pore throat distributions derived from, respectively, NMR and MICP generally shows a
288 good agreement between curve shapes (Figure 5; Appendix A-1). The derived surface relaxation
289 ranges from 10 to $55 \mu\text{m/s}$ (Figure 5; Appendix A-1).

290 FIGURE 5

291 The derived porosity exponents, m , range from 1.86 to 2.30 with clay containing samples typically in
292 the higher range (Table 3). The values, however, indicate that the samples are unfractured. The
293 shielding factor, c_M , ranges from 0.18 to 0.22 (Table 3).

294 TABLE 3

295

296 **Pore throat distribution and flow-through experiments**

297 FIGURE 6 FIGURE 7

298 For all Castlegate samples and samples F31.21 and F81.2 from Fontainebleau, the limiting d_{Kn} and
299 d_{Re} values and pore throat distributions indicate that flow in most of the pore space is linear Darcy
300 (slip) flow for all choices of discharge, and accordingly the Klinkenberg correction gives a k_K
301 estimate of liquid permeability not too far from the measured (k_w) (Figures 6, 7; Appendix B-1). For
302 the two samples from Obernkirchen and samples F61.1 and F61.2 from Fontainebleau, linear Darcy
303 (slip) flow is predicted for the lowest chosen discharges, but for the highest discharges (q_{max}), the
304 limit defined by the Reynolds number is transgressed. This results in nonlinear flow and low gas
305 permeability at the highest discharge. Accordingly, only the low-discharge data can be used for the
306 Klinkenberg correction (Figure 6, 7; Appendix B-1). For the single sample from Bentheim as well as
307 samples F21.1, F22.11 and F7.1 from Fontainebleau, only the lowest discharge (q_{min}) corresponds to
308 d_{Re} and Klinkenberg correction is not possible because collecting data at even lower discharge was
309 unsuccessful (Figures 6, 7; Appendix B-1). Without an evaluation of the flow regime, widely wrong
310 Klinkenberg corrections would have resulted from unknowingly fitting data in the non-linear flow
311 regime, and neither would accepting any gas permeability as k_K give trustworthy results (Figure 7).

312 FIGURE 8 FIGURE 9

313 Liquid permeability generally increases with increasing porosity, but no single trend was found,
314 representing all the studied sandstones. The Fontainebleau samples follow the same liquid
315 permeability-porosity trend as found by Doyen (1988) and Chen et al. (2016) (Figure 8a). The
316 Bentheimer sample plots similar to data from Al-Yaseri et al. (2015) and close to a cluster of
317 Castlegate samples, whereas Obernkirchen samples have much lower permeability for a given

318 porosity (Figure 8a). For the sandstones in this study, the Klinkenberg corrected argon permeability
319 matches the liquid permeability well for experiments where Darcy conditions were met (Root Mean
320 Square Error (RMSE) is found to be 66) (Figure 8b). The largest offset is seen for Fontainebleau
321 samples with the lowest permeability. By contrast, results by Chen et al. (2016) show generally lower
322 liquid permeability than Klinkenberg permeability (Figure 8b). With respect to permeability
323 prediction, using an input of specific surface from BET measurements in Kozeny's equation gives a
324 good prediction of liquid permeability for Fontainebleau samples with $\phi_N > 0.06$, whereas for
325 Bentheimer, Obernkirchen and Castlegate samples, Kozeny permeability is significantly lower than
326 the measured water permeability, with the largest offset for Castlegate and Bentheimer sandstones
327 (Table 3 and Figure 8c). The cumulative NMR permeability as derived from equation 9 indicates that
328 pore sizes below the order of $0.1 \mu\text{m}$ do not contribute significantly to liquid permeability (Figure 9;
329 Appendix C-1). For Fontainebleau samples F2.11, F22.11 and F7.1 as well as Castlegate and
330 Bentheimer samples, the maximum required effective pore size to match the measured liquid
331 permeability ranges from 10 to $30 \mu\text{m}$, whereas for the other Fontainebleau and the Obernkirchen
332 samples the maximum required pore size ranges from 5 to $8 \mu\text{m}$ (Figure 9; Appendix C-1).

333 DISCUSSION

334 **Klinkenberg correction**

335 Among the 13 studied sandstone samples, measuring gas permeability and the performing
336 Klinkenberg correction was unproblematic for five samples. For four samples, it was experimentally
337 too challenging, whereas it was possible for four samples when care was taken not to include high
338 discharge data. High discharge data, where a significant part of the gas flow is in the non-linear
339 regime provides too steep lines on the k^{-1}/P_m cross plot, so that equivalent liquid permeability
340 becomes too low (Figure 7). Interestingly, results reported by Klinkenberg (1941) also have a

341 steepening trend for decreasing $1/P_m$ below one atm^{-1} (Figure 10). Unfortunately, data on pore throat
342 distributions, as well as flow-through discharge needed to evaluate the flow conditions in
343 Klinkenberg's experiments are not reported by Klinkenberg (1941). Apparently overlooking the
344 downward trend found in gaseous flow-through experiments, Klinkenberg found good agreement
345 with measurements using isooctane (Figure 10). Also, in the present study, in some cases good
346 agreement was found between water permeability k_K using the downward trend, thus disregarding
347 non-linear flow conditions. However, this only applies to samples with water permeability less than
348 the order of 20 mD ($19.7\text{E}^{-15} \text{ m}^2$) and may be a coincidence (Figure 7, dashed lines).

349 FIGURE 10

350 The significance of the present discussion of the applicability of the Klinkenberg correction is
351 illustrated by an excellent published data set. On a series of Fontainebleau samples with a wide
352 porosity range, Bourbié and Zinszner (1985) derived gas/air permeability (Figure 11) and by judging
353 the Klinkenberg correction as insignificant assumed $k_g = k_w$ without including a verification from
354 liquid flow-through experiments. Revil et al. (2014) reported identical experimental results, but stated
355 that the data were Klinkenberg corrected, and then defined a porosity-liquid permeability curve. This
356 is a concern because our data indicate that gas permeability significantly overestimates liquid
357 permeability (Figure 8). To assess the consequence for the Bourbié and Zinszner (1985) as well as
358 the Revil et al. (2014) data set, porosity and permeability data for Fontainebleau and Bentheimer
359 sandstones from Bourbié and Zinszner (1985), Doyen (1988), Revil et al. (2014), Al-Yaseri et al.
360 (2015), Chen et al. (2016), and the present study were cross plotted (Figure 11). For porosity values
361 below 0.1, model predictions by Revil et al. (2014) are in good agreement with both gas and liquid
362 permeability (Figure 11), but for increasing porosity, the model overestimates liquid permeability
363 more and more. The discrepancy can be related to one or more errors, where the experimentalists: 1)

364 omitted to confirm the presence of linear Darcy-conditions in their conducted flow-through
 365 experiments, 2) applied a too narrow range of mean pressures for classical Klinkenberg correction or
 366 3) applied an unsuitable empirical relation between gas- and liquid permeability.

367 Based on measured k_w , a new porosity-liquid permeability trend that appears less steep was
 368 constructed and shows a liquid permeability approaching 3 D for $\phi > 0.3$, thus an order of magnitude
 369 less than the model of Revil et al. (2014) (Figure 11).

370 FIGURE 11

371 **Permeability modelling**

372 Kozeny's equation gives a good estimate of liquid permeability for the homogeneous and clay
 373 free Fontainebleau samples with $\phi > 0.06$ when applying S_p derived from BET measurements and
 374 using c_M calculated from porosity (Figure 8c). This is probably because the clay free Fontainebleau
 375 samples have the highest homogeneity in the distribution of the specific surface and the preconditions
 376 for Kozeny's equation thus is met. The offset between k_z and k_w seen for the clay containing samples
 377 is significant, and as the clay is heterogeneously distributed (Figure 4), the precondition of
 378 homogeneity for Kozeny's equation is consequently not met. These results are in accordance with the
 379 modelled NMR-permeability indicating that pores smaller than $0.1 \mu\text{m}$ do not contribute significantly
 380 to the liquid permeability of these relatively permeable sandstones.

381 As recognized from petrography, the larger pores do not form a continuous path in the pore
 382 space. Consequently, the fluid flow is controlled by the smaller pores (Figure 4). Based on pore size
 383 distribution, permeability increments, $k_{\text{NMR},i}$, should hence be cumulated to $k_{\text{NMR,cum}}$ starting from the
 384 smallest pores, defining a cut-off at $k_{\text{NMR,cum}} = k_w$. This is opposite to the common practice of
 385 defining a cut-off assuming the smaller pores to be irrelevant for the overall permeability. In order to
 386 illustrate this point advocated by Rosenbrand et al. (2015b), the pore size distribution and $k_{\text{NMR,cum}}$ for

387 samples F7.1 and B11.11 were compared because these two samples have similar porosity and water
388 permeability, but only F7.1 follows Kozeny theory (Figure 8). The pore size distributions of the two
389 samples are similar for small pores to a pore size of 20-30 μm , however sample B11.11 has
390 additional larger pores, which, however, do not contribute to permeability (Figures 12b and c). The
391 pore throat distributions of the two samples are practically identical further illustrating that pore
392 throats and the corresponding small pore bodies below the upper cut-off size control the permeability
393 (Figure 12a).

394 FIGURE 12

395 CONCLUSIONS

396 On a selection of different sandstones, a sequence of single fluid, liquid and gas flow through
397 experiments were performed from which respectively water and gas permeability were derived.
398 Using the Reynolds number and knowledge of the pore throat distributions from capillary pressure
399 measurements, the validity of the experimental flow conditions were tested and thus it was identified
400 from which experiments Darcy's equation is valid for deriving gas permeability. Where the results
401 indicated conditions of linear Darcy flow, the Klinkenberg correction was applied and a good
402 agreement with liquid permeability was found. Besides the relevant of knowledge of the respective
403 pore throat distribution, this illustrates the applicability and importance of the Klinkenberg
404 correction. Experiments where NMR-derived pore size predicted a non-linear Darcy flow resulted in
405 too low gas permeabilities, and in extreme cases resulted in an erroneous negative Klinkenberg
406 permeability.

407 In combination with published data, the present results show that extreme care should be
408 taken when gas permeability of highly porous sandstones is applied to estimate the equivalent liquid
409 permeability by Klinkenberg's procedure. Measurements with a high discharge pore liability

410 concerns on estimates of gas permeability from Darcy's law because inherent constraints on flow
411 conditions may be violated if not evaluated further.

412 For clay free sandstone, liquid permeability was well modelled when using Kozeny's
413 equation with the specific surface area derived from BET measurements as well as a theoretically
414 derived Kozeny-factor. By combining specific surface area derived from NMR with Kozeny's
415 equation, permeability increments for each pore size were modelled. Cumulated from the smallest
416 pores this equals the measured liquid permeability at a cut-off pore size smaller or equal to the
417 largest pores. Pore sizes smaller than the cut-off size (order of 10 μm) are thus defined as the
418 controlling ones for the overall permeability, whereas larger pores can detract from gas permeability
419 due to a possible transition to turbulent flow.

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568 LIST OF FIGURES

569 Figure 1. Conceptual schematics of gaseous flow regimes from diffusion to turbulence based
 570 on the Knudsen (Kn) and Reynolds (Re) number. Limits of Kn are from Karniadakis et al. (2005).
 571 Limits of Re are from Scheidegger (1960) and Bear (1972). Kn_{trans} , d_{Kn} and Re_{trans} , d_{Re} mark the
 572 thresholds to transitions zones towards diffusive respectively turbulent flow, where Kn and Re refer
 573 to the Knudsen and Reynolds number, whereas d_{Kn} and d_{Re} refer to corresponding characteristic
 574 lengths.

575 Figure 2. Conceptual schematics illustrating the use of the pore throat distribution to identify
 576 flow regimes in experiments with gaseous flow using steps of increased discharge and pressure as
 577 suggested in Klinkenberg's procedure. q is specific discharge (Q/A) and related to the Reynolds
 578 number (equation 1). P_m is mean pressure and related to the Knudsen number (equation 2).

579 Figure 3. Modelled Kozeny' factor, c_M , versus porosity after Mortensen et al. (1998)
 580 (equation 4). The red line shows the approximated best linear fit in the porosity range from 0.02 to
 581 0.4.

582 Figure 4. a) to e) BSEM images of polished thin sections from side trims of Fontainebleau,
 583 Bentheimer, Obernkirchen and Castlegate sandstones. Q = quartz, K = kaolinite.

584 Figure 5. Normalized incremental porosity versus pore size and pore throat diameter. MICP
 585 curves are derived with an Hg surface tension and an Hg contact angle of $480 \text{ mN}\cdot\text{m}^{-1}$ and 140° ,
 586 respectively. NMR curves are derived by adjusting the surface relaxivity to match peak to peak on
 587 MICP curves.

588 Figure 6. Normalized incremental porosity versus pore throat diameter. Vertical dashed lines,
 589 $P_{m,\min}$ and $P_{m,\max}$, are the throat diameters corresponding to $Kn = 0.1$ derived from equation 1 and the
 590 maximum and minimum mean pressures, respectively. Vertical lines, denoted q_{\min} and q_{\max} , are the

591 throat diameters corresponding to $Re = 10$ derived from equation 2 and the maximum and minimum
 592 specific discharges, respectively.

593 Figure 7. Gas permeability versus inverse mean pressure. Gray color markers are data points
 594 considered valid for use in Klinkenbergs procedure based on flow conditions evaluated from Figure
 595 6. Black lines are best linear fits of gray markers from Klinkenbergs procedure. Dashed lines are best
 596 linear fits of gray and open markers. Black markers are water permeability from Table 2 and plotted
 597 at $1/P_m = 0$. mD = millidarcy, $1\text{mD} = 9.869\text{E}^{-16} \text{ m}^2$.

598 Figure 8. a) Cross plot of water permeability and porosity. For this work $\phi = \phi_N$. b) Cross plot
 599 of water permeability and Klinkenberg permeability. c) Cross plot of water permeability and
 600 permeability predicted from Kozeny's equation. All data are from Table 3. mD = millidarcy, $1\text{mD} =$
 601 $9.869\text{E}^{-16} \text{ m}^2$.

602 Figure 9. Incremental and cumulated NMR permeability versus NMR pore diameter (equation
 603 9). Measured water permeability, k_w , is plotted as a horizontal line. The hatched area represents pore
 604 sizes cumulated from the smallest ones until $k_{\text{NMR,cum}} = k_w$. mD = millidarcy, $1\text{mD} = 9.869\text{E}^{-16} \text{ m}^2$.

605 Figure 10. Gas permeability versus inverse mean pressure adopted from Klinkenberg (1941).
 606 Curves show a downward curve trend for inverse mean pressures below the order of 1 atm^{-1} . Black
 607 markers are isooctane permeability (Klinkenberg, 1941). mD = millidarcy, $1\text{mD} = 9.869\text{E}^{-16} \text{ m}^2$.

608 Figure 11. Cross plot of permeability and porosity. Comparison of studies on Fontainebleau
 609 and Bentheimer sandstones. The solid line shows the model derived by Revil et al. (2014). The
 610 dashed line is a reasonable fit based on measured water permeability. mD = millidarcy, $1\text{mD} =$
 611 $9.869\text{E}^{-16} \text{ m}^2$.

612 Figure 12. Extract from Figures 5 and 9 for samples F7.1 and B11.11. a) shows normalized
 613 incremental porosity versus MICP pore throat diameter (Figure 5), b) shows normalized incremental
 614 porosity versus NMR pore diameter (Figure 5) and c) shows incremental and cumulated NMR

615 permeability versus NMR pore diameter. Measured water permeability, k_w , is plotted as a horizontal
616 line. The hatched area represents pore sizes cumulated from the smallest ones until $k_{\text{NMR,cum}} = k_w$
617 (Figure 9). mD = millidarcy, $1\text{mD} = 9.869\text{E}^{-16} \text{ m}^2$.

618 Appendix A-1. Normalized incremental porosity versus pore size and pore throat diameter.
619 MICP curves are derived with an Hg surface tension and an Hg contact angle of $480 \text{ mN}\cdot\text{m}^{-1}$ and
620 140° , respectively. NMR curves are derived by adjusting the surface relaxivity to match peak to peak
621 on MICP curves.

622 Appendix B-1. Normalized incremental porosity versus pore throat diameter. Curves are
623 derived with an Hg surface tension and an Hg contact angle of $480 \text{ mN}\cdot\text{m}^{-1}$ and 140° , respectively.
624 Vertical dashed lines, denoted P_{min} and P_{max} , are the throat diameters corresponding to $Kn = 0.1$
625 derived from equation 1 and the maximum and minimum mean pressure, respectively. Vertical lines
626 denoted q_{min} and q_{max} , are the throat diameters corresponding to $Re = 10$ derived from equation 2 and
627 the maximum and minimum specific discharge, respectively.

628 Appendix C-1. Incremental and cumulated NMR permeability versus NMR pore diameter
629 (equation 9). Measured water permeability, k_w , is plotted as a horizontal line. The hatched area
630 represents pore sizes cumulated from the smallest ones until $k_{\text{NMR,cum}} = k_w$.

631

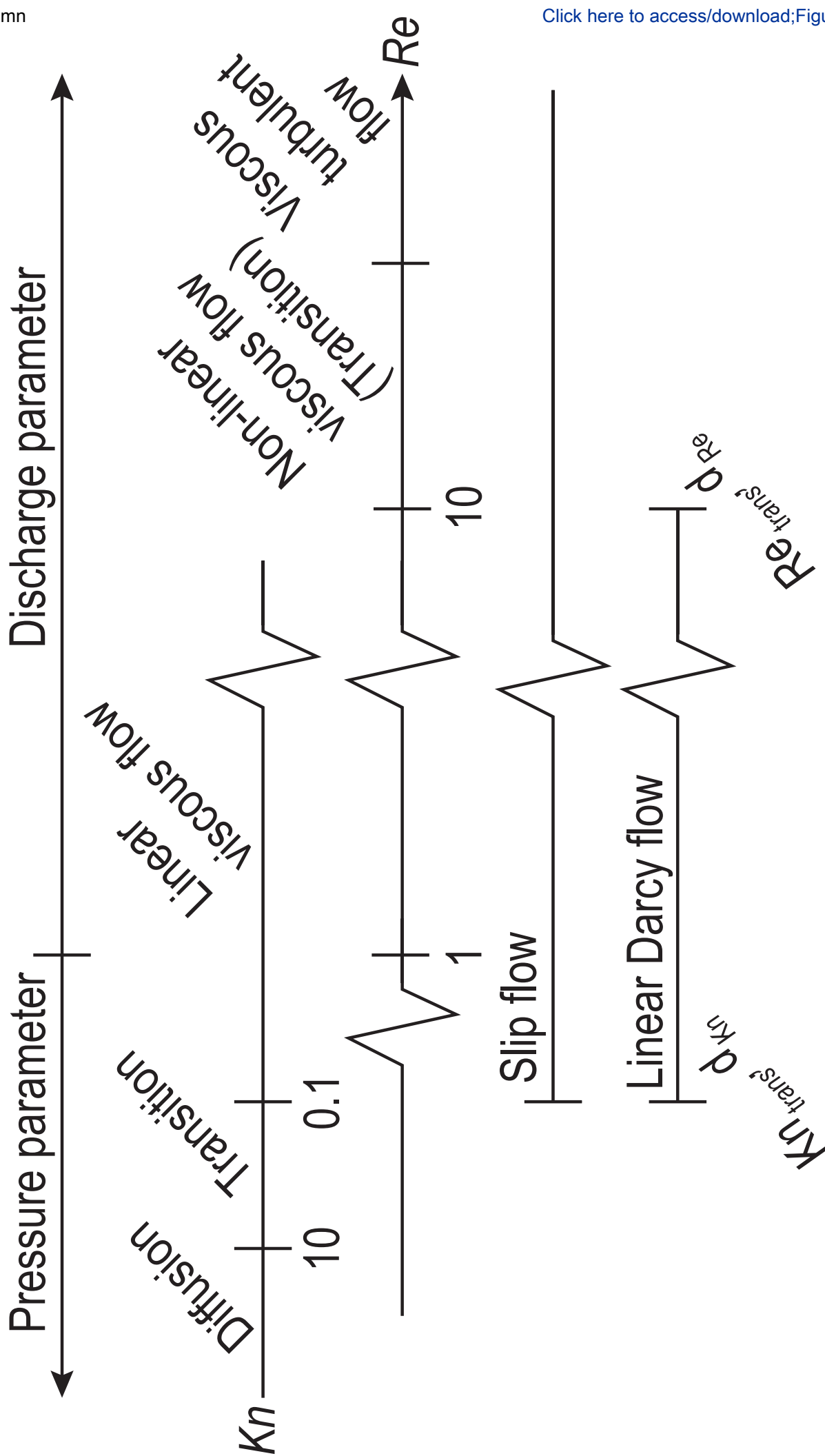
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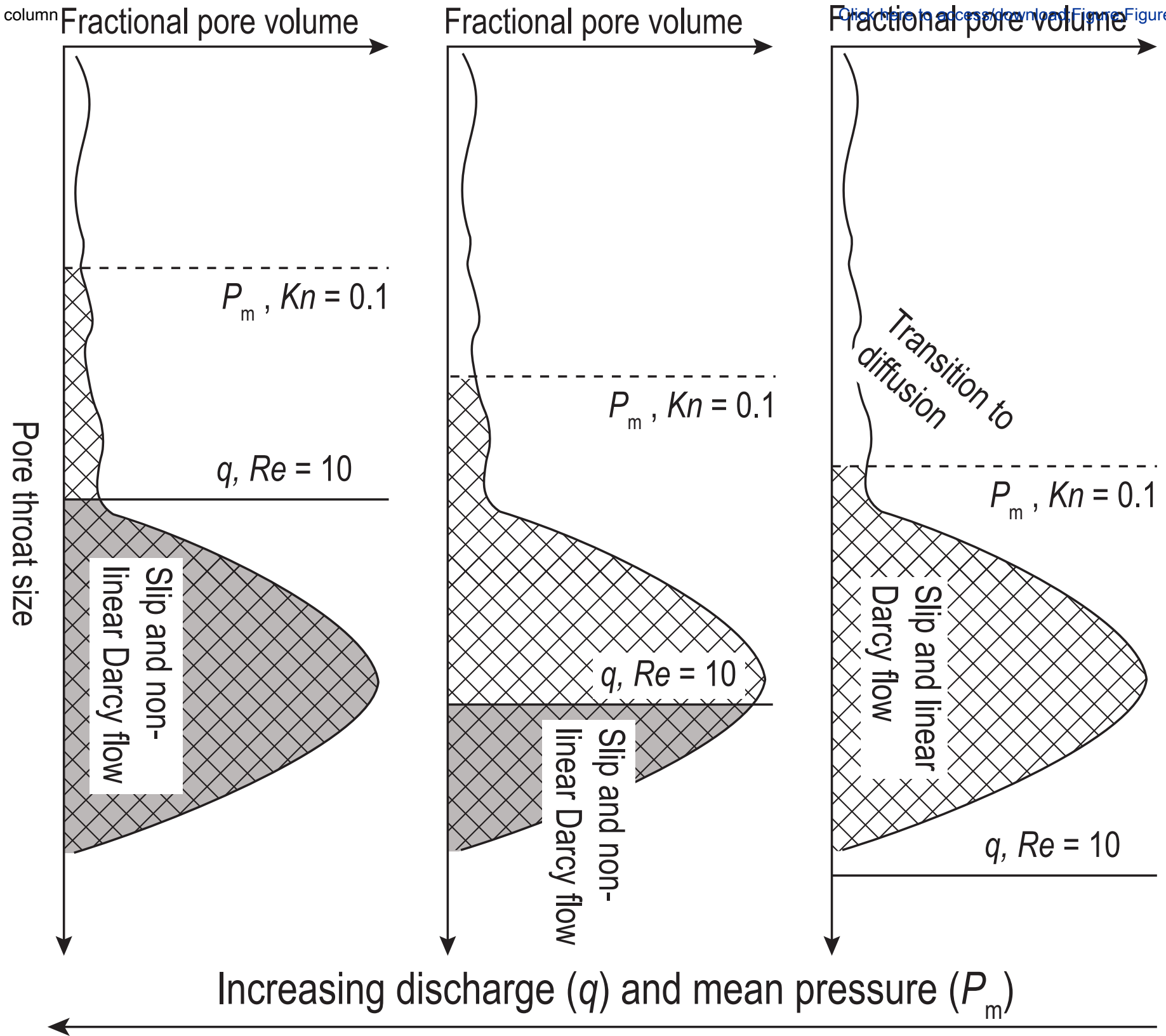
633 LIST OF TABLES

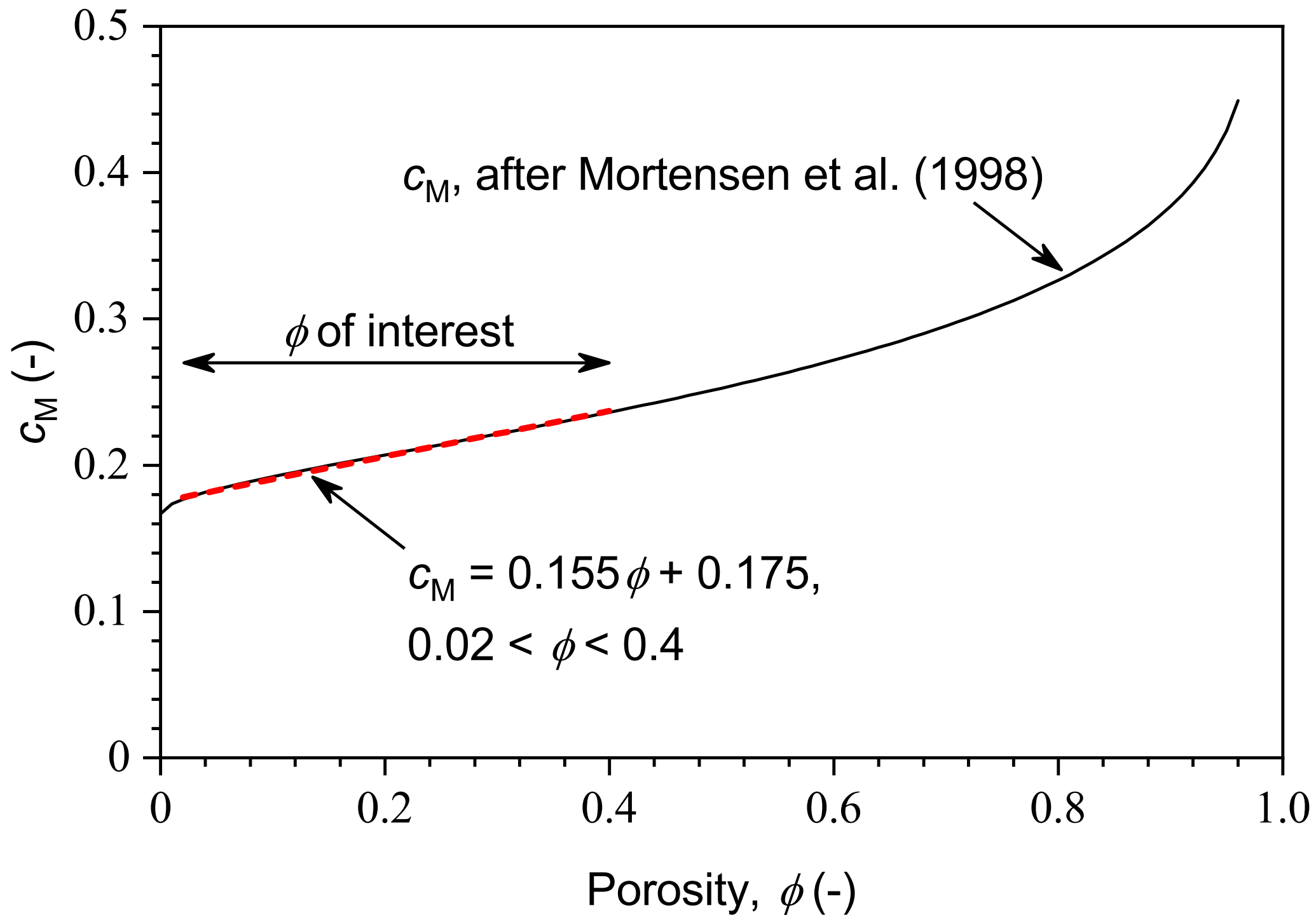
634 Table 1. Mineral content obtained from quantitative XRD analysis and specific surface area derived
635 from BET measurements on the sample material.

636 Table 2. Measured density and N₂-porosity of the studied rock material

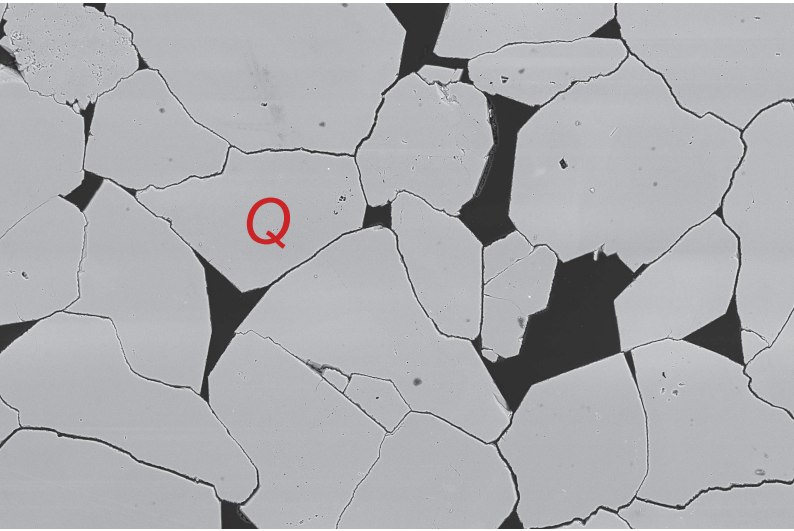
637 Table 3. Measured and derived properties of the studied rock material.





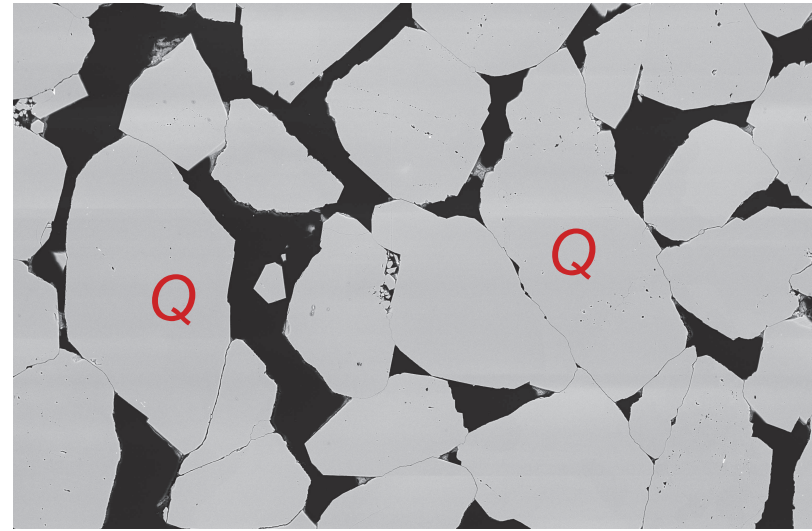


a) Fontainebleau, $\phi \sim 6-9\%$



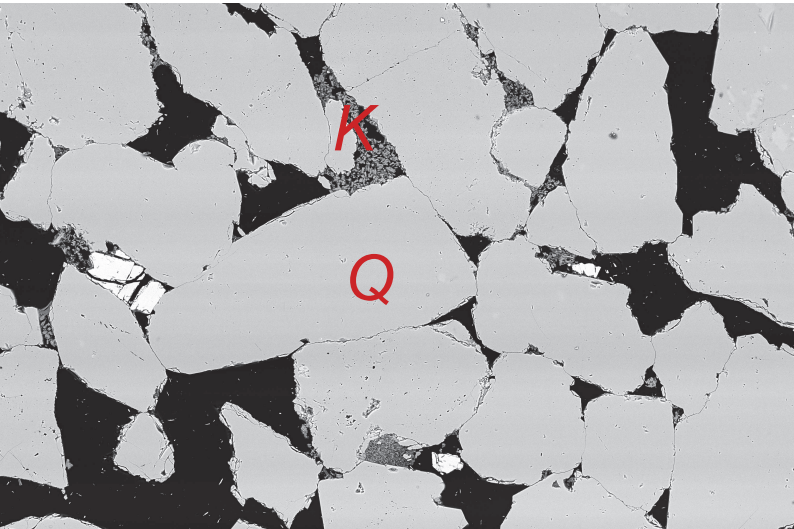
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b) Fontainebleau, $\phi \sim 23\%$



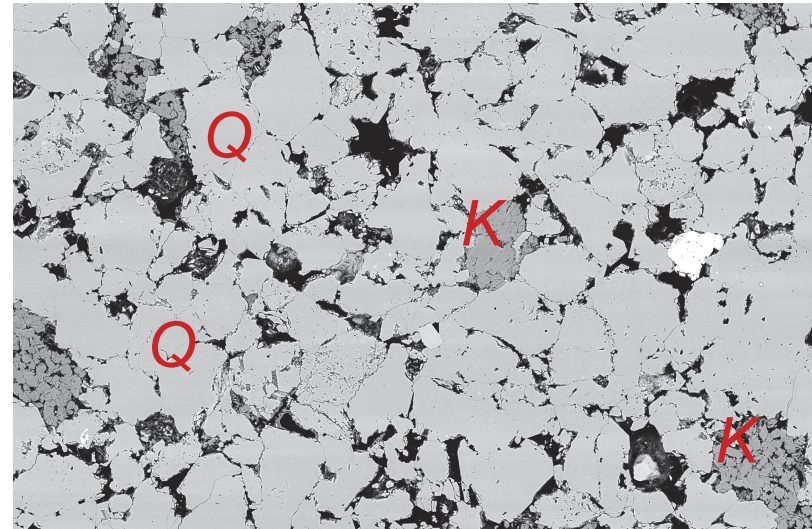
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c) Bentheimer, $\phi \sim 20\%$



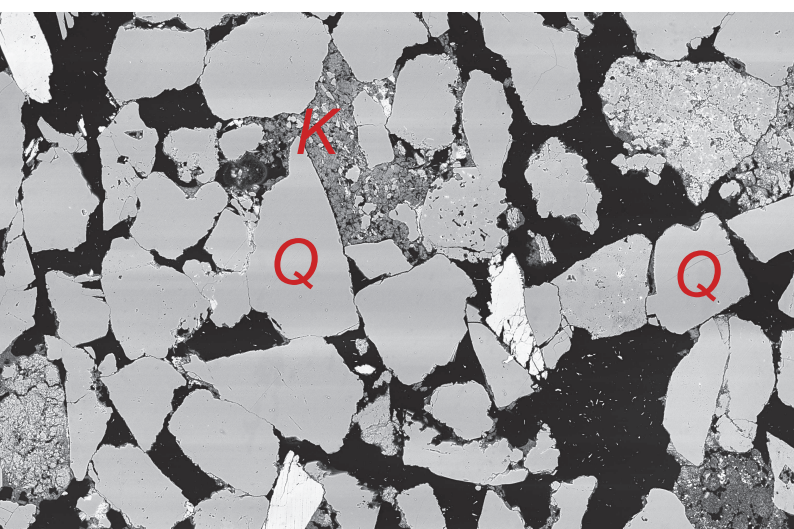
500 μm

d) Obernkirchen, $\phi \sim 20\%$

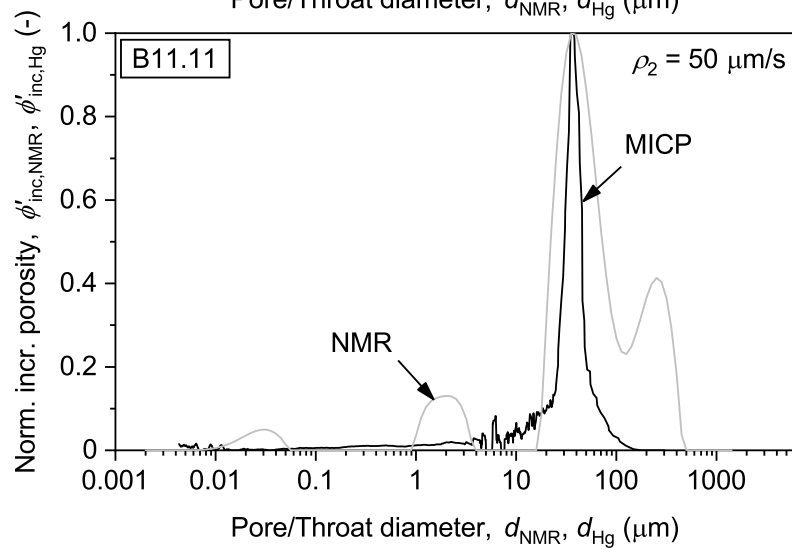
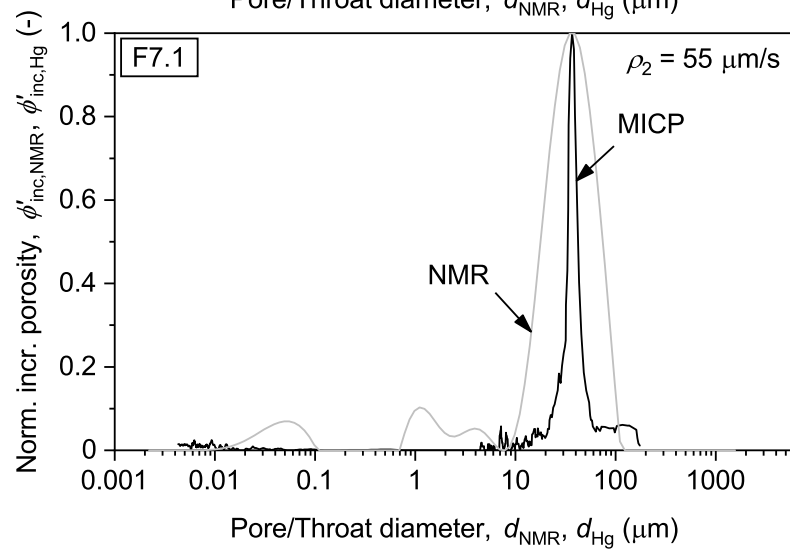
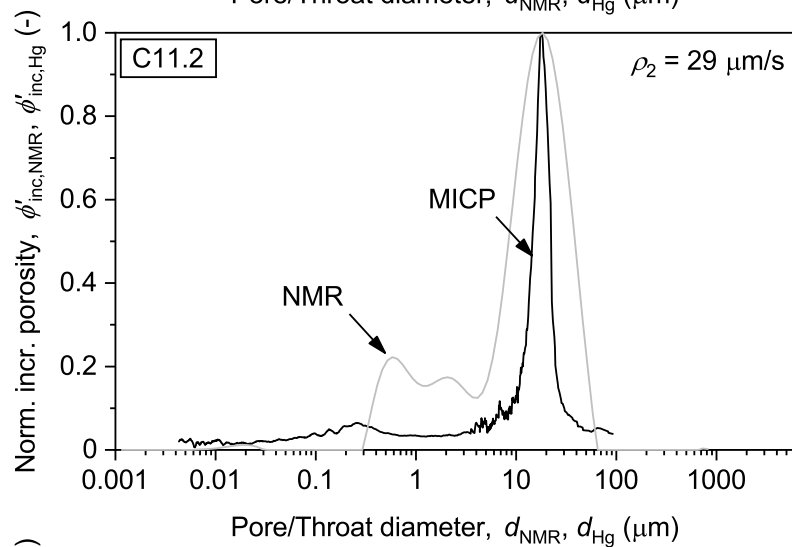
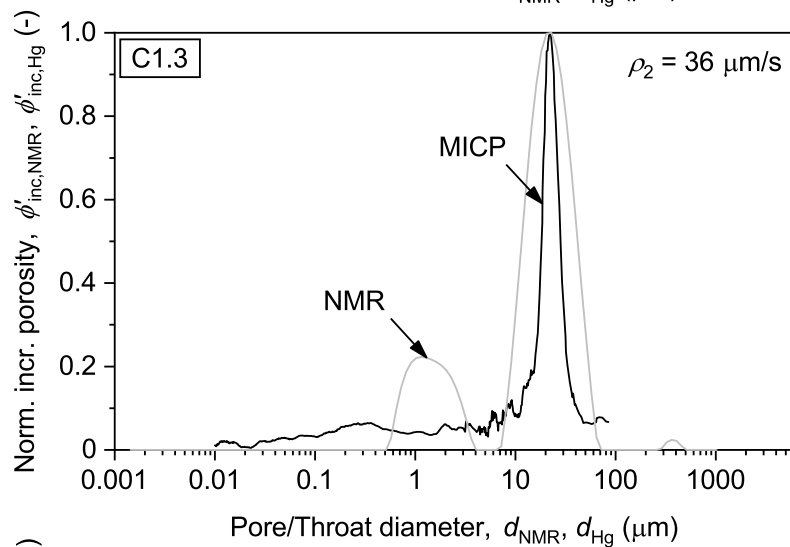
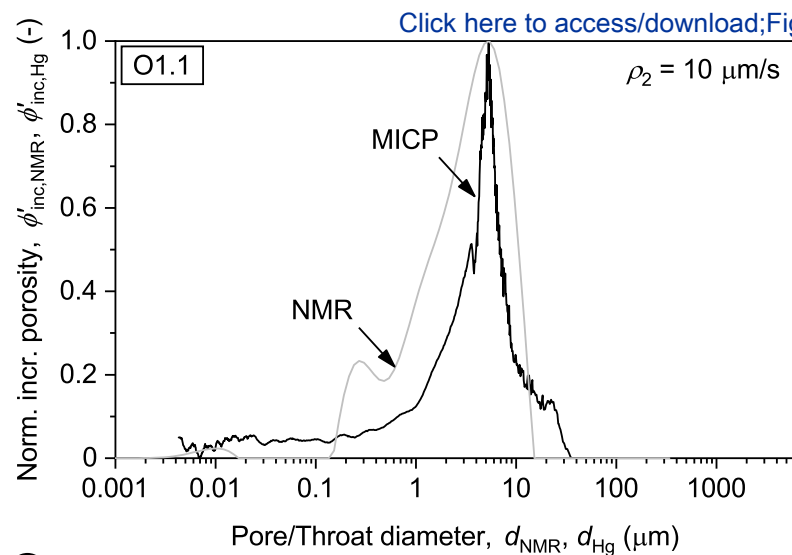
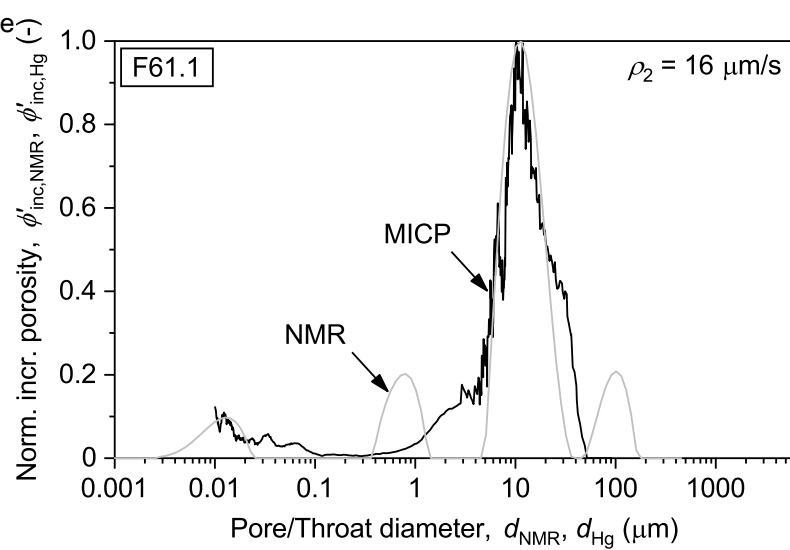


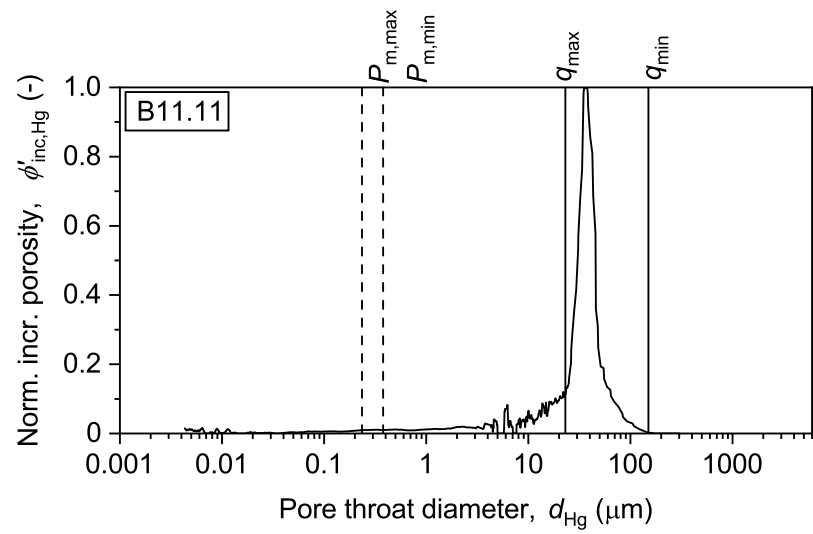
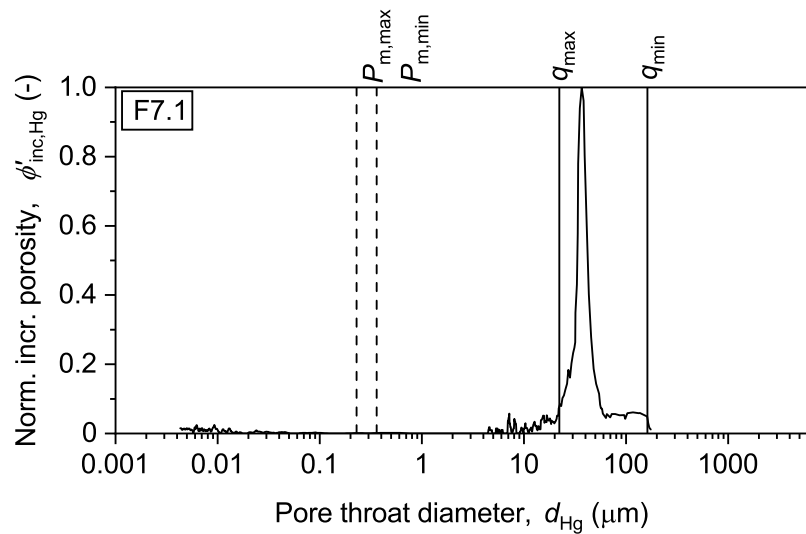
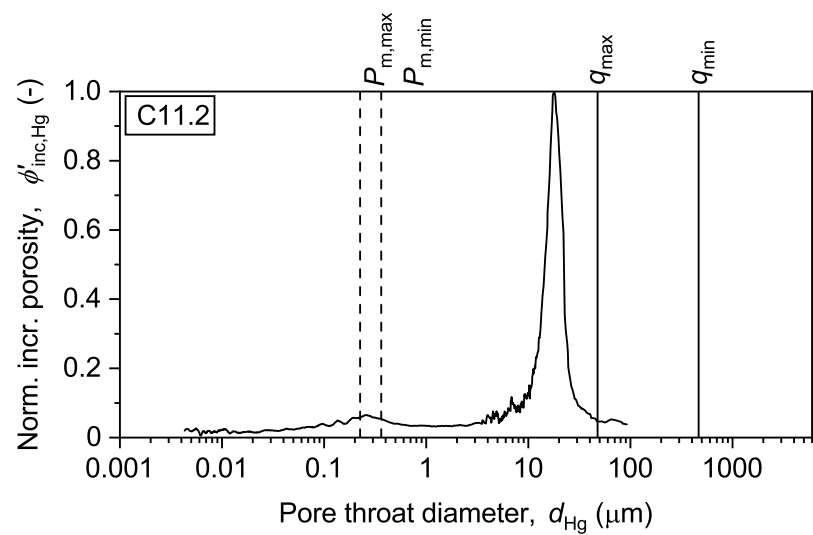
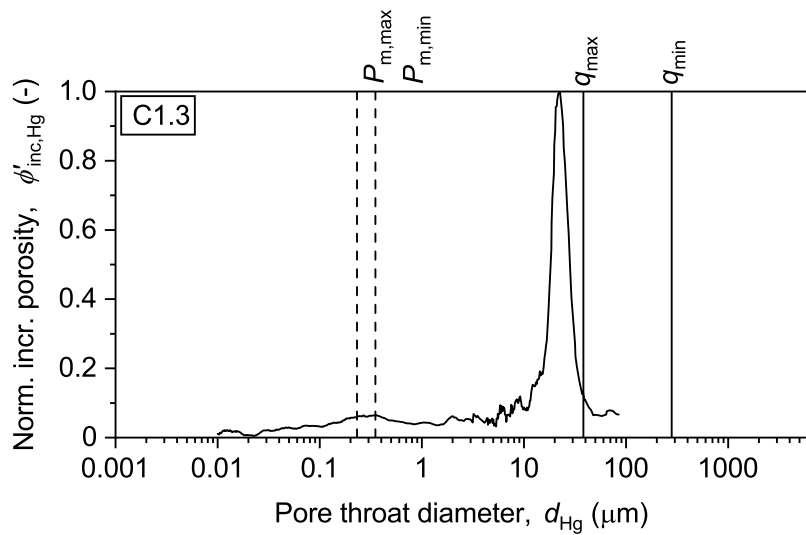
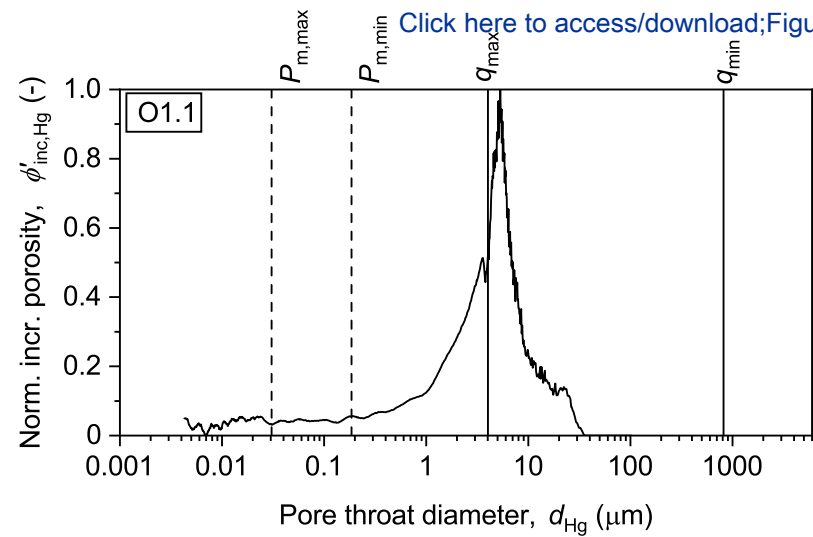
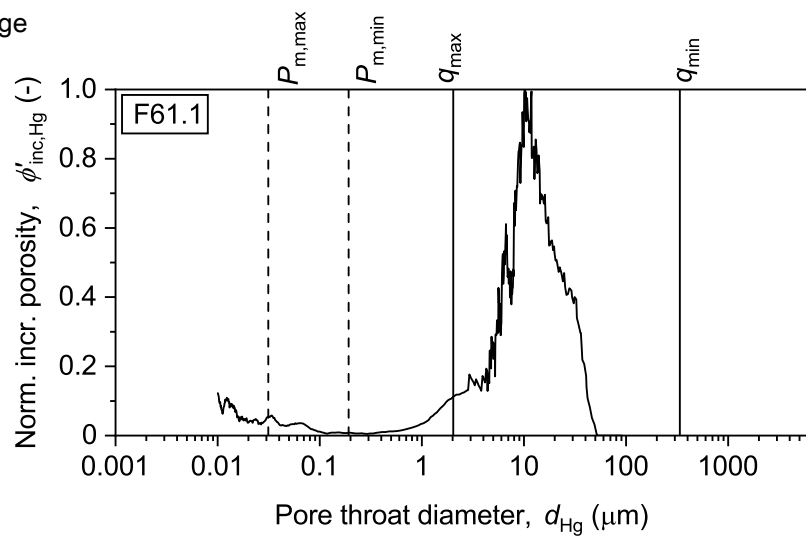
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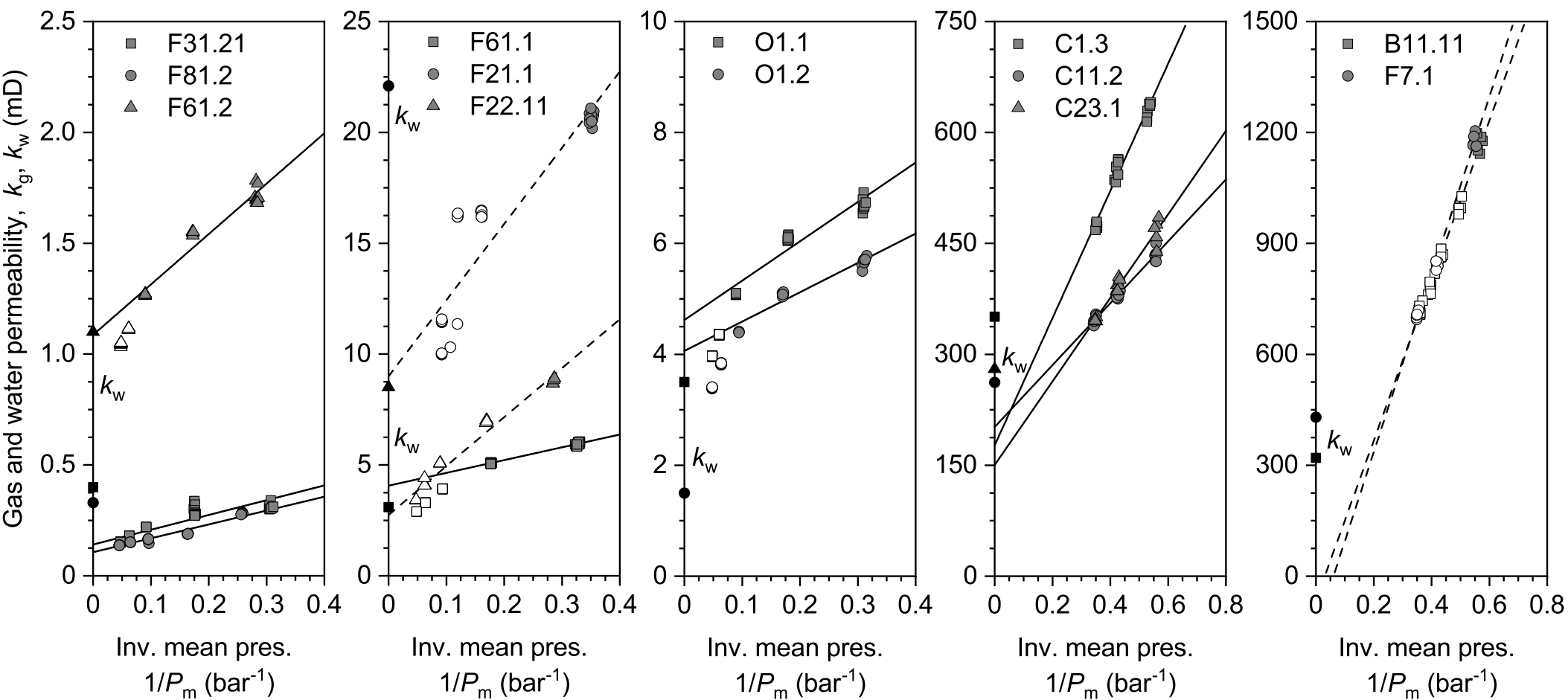
e) Castlegate, $\phi \sim 30\%$

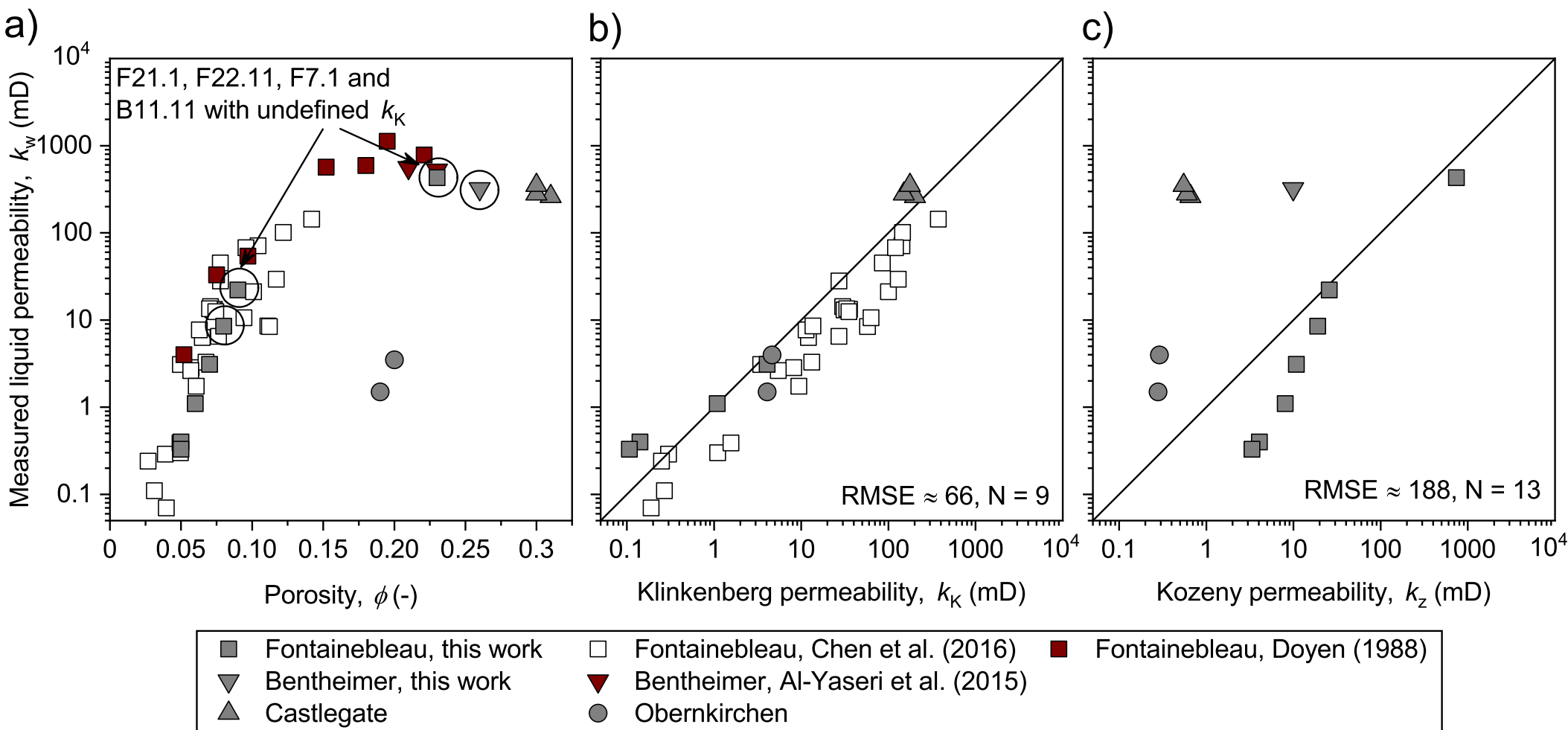


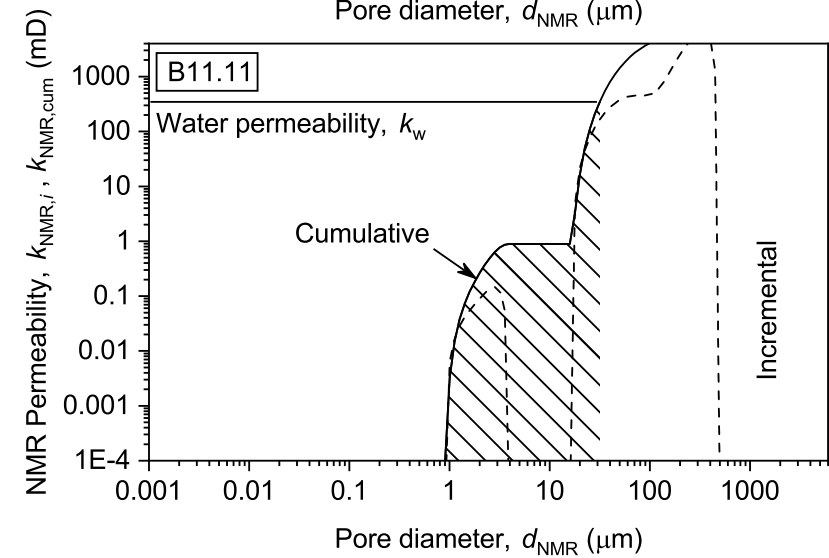
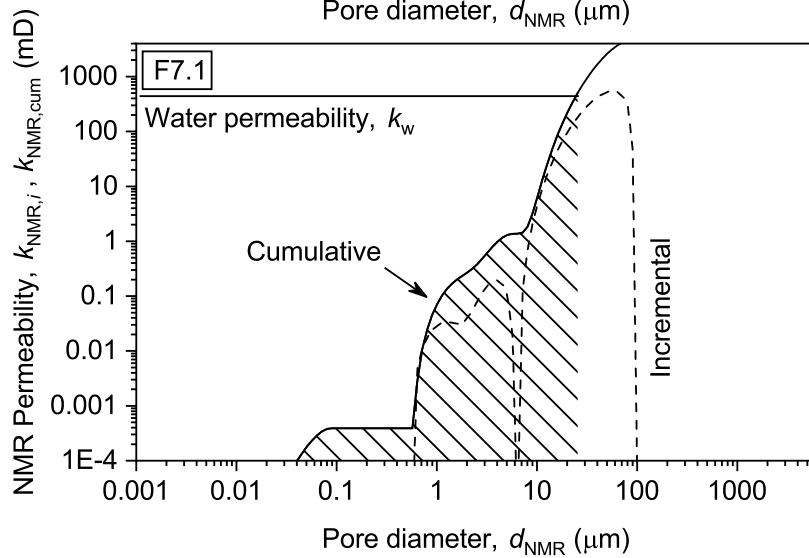
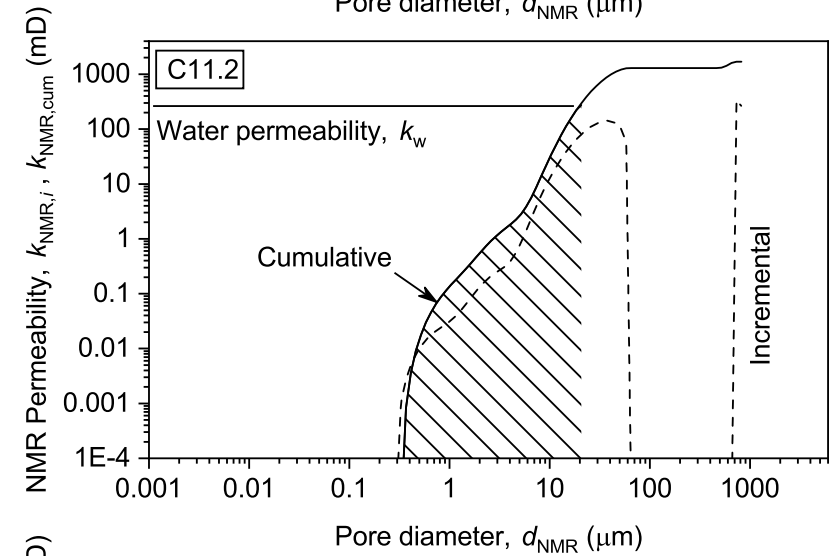
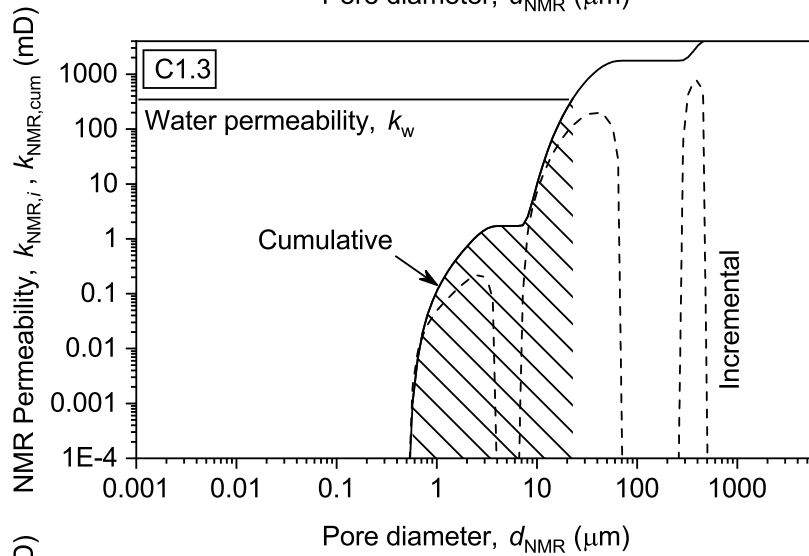
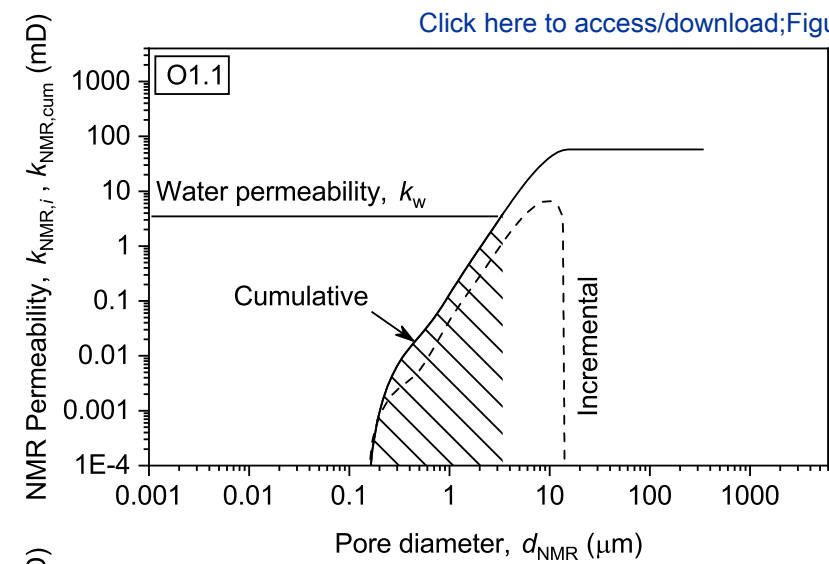
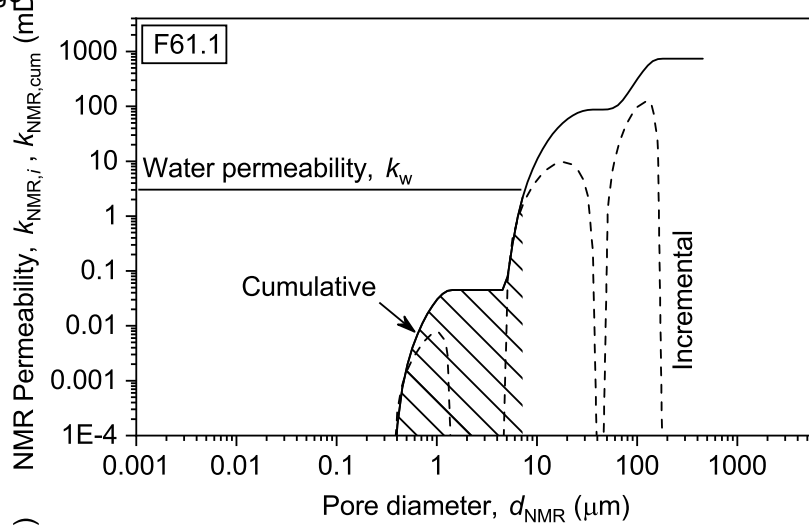
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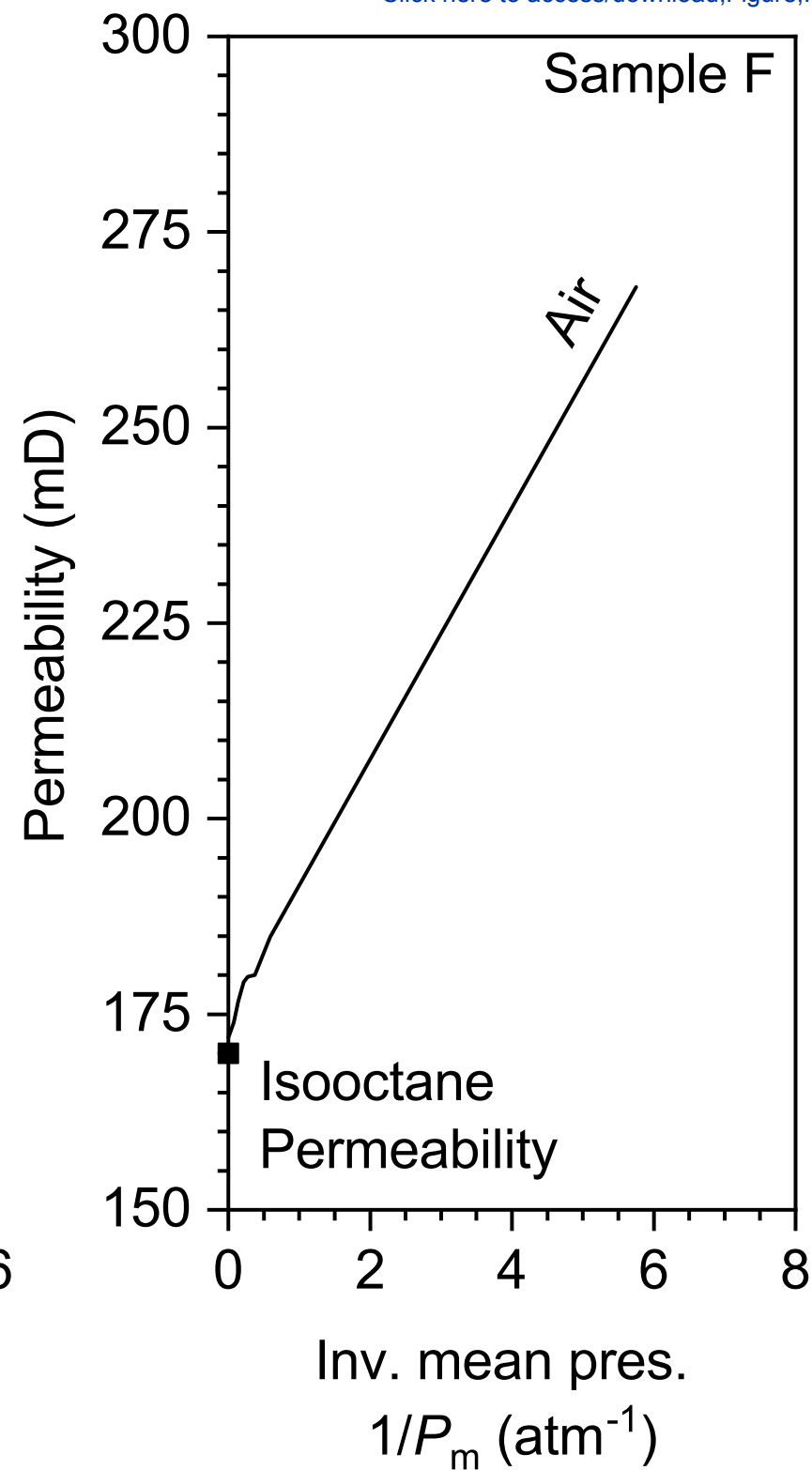
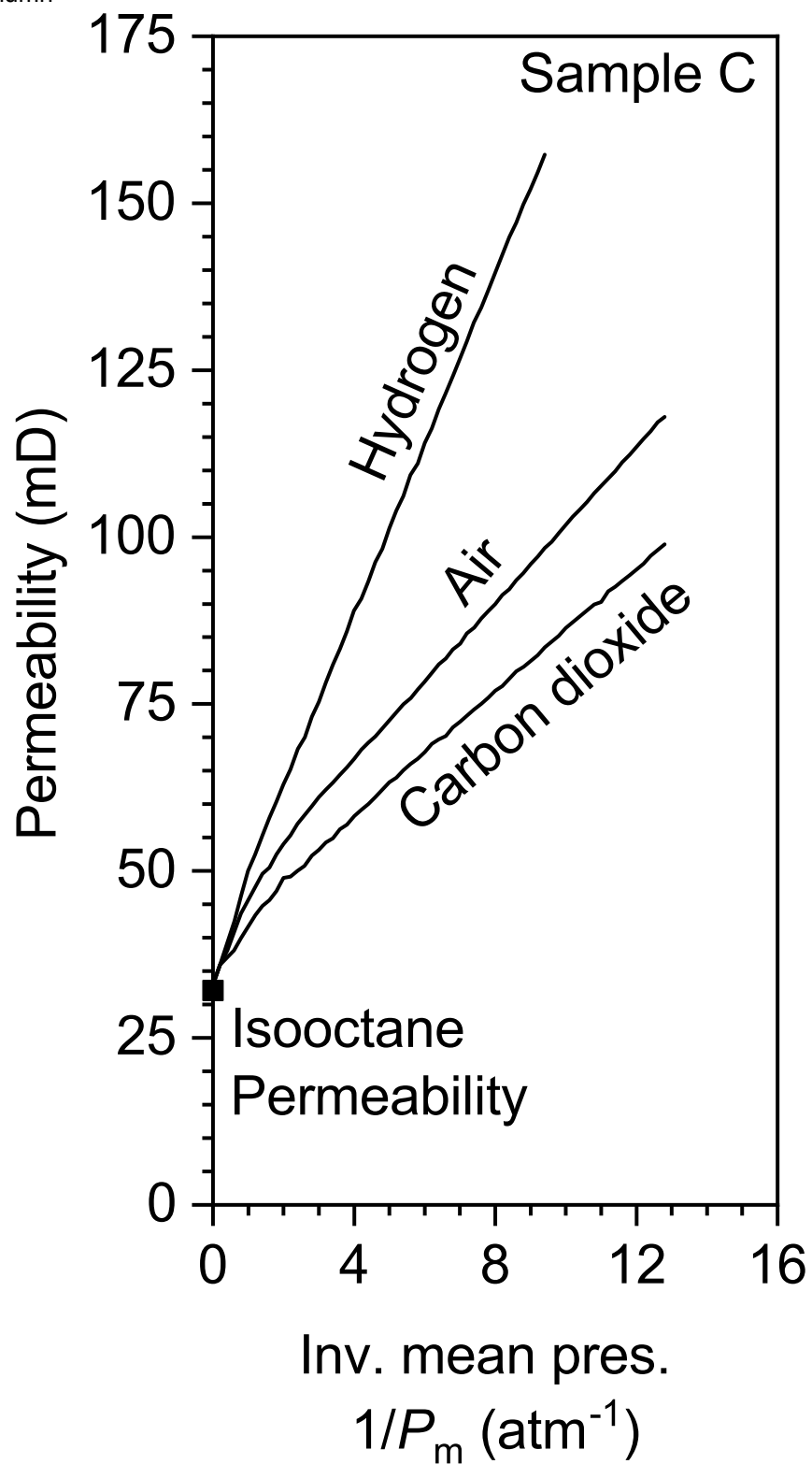


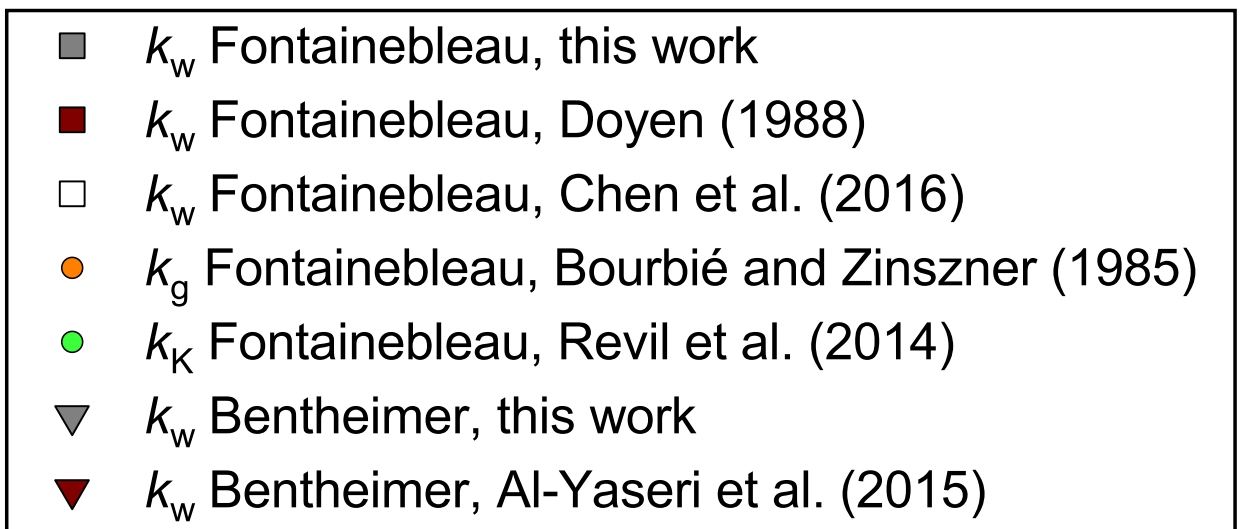
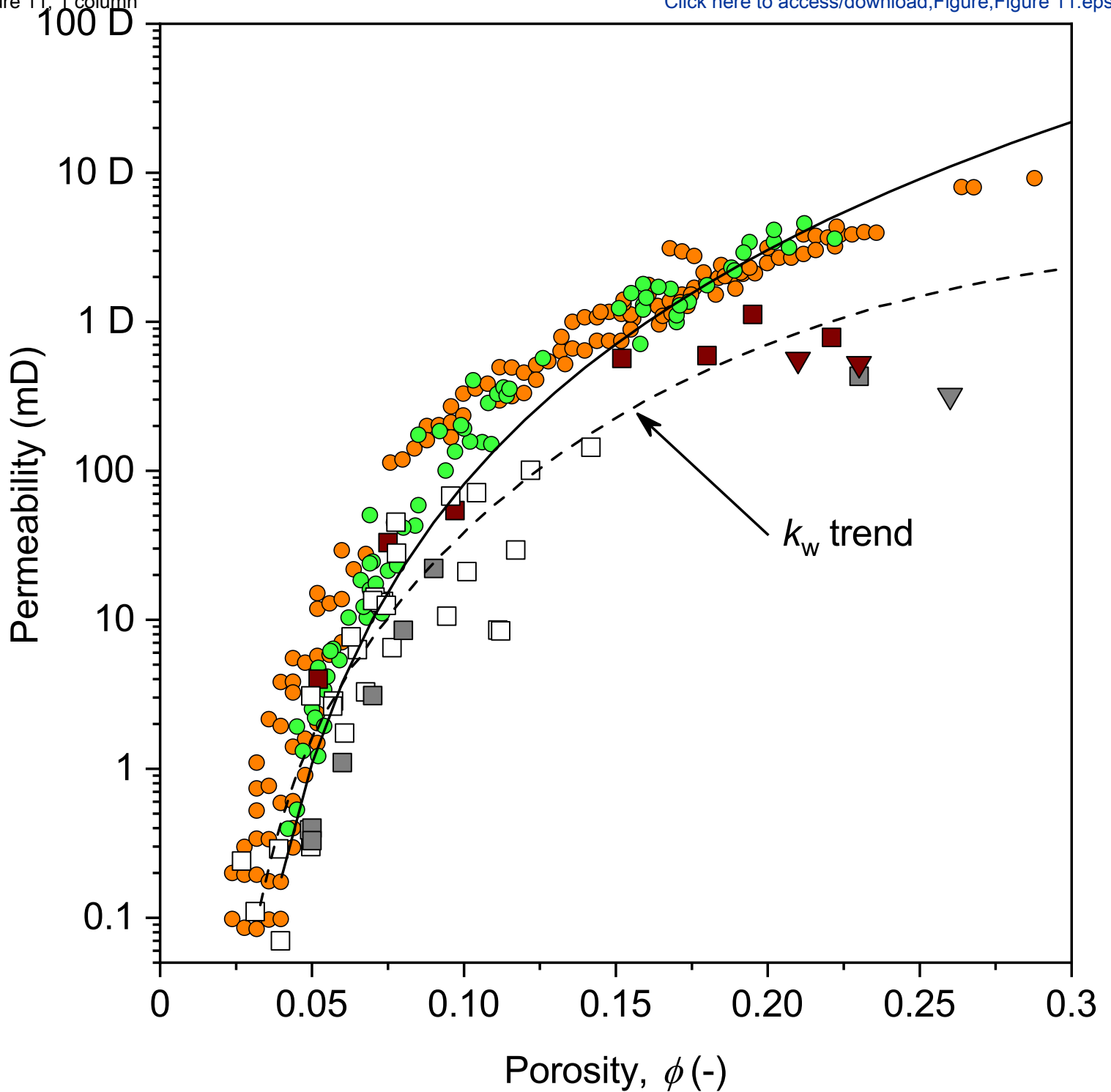


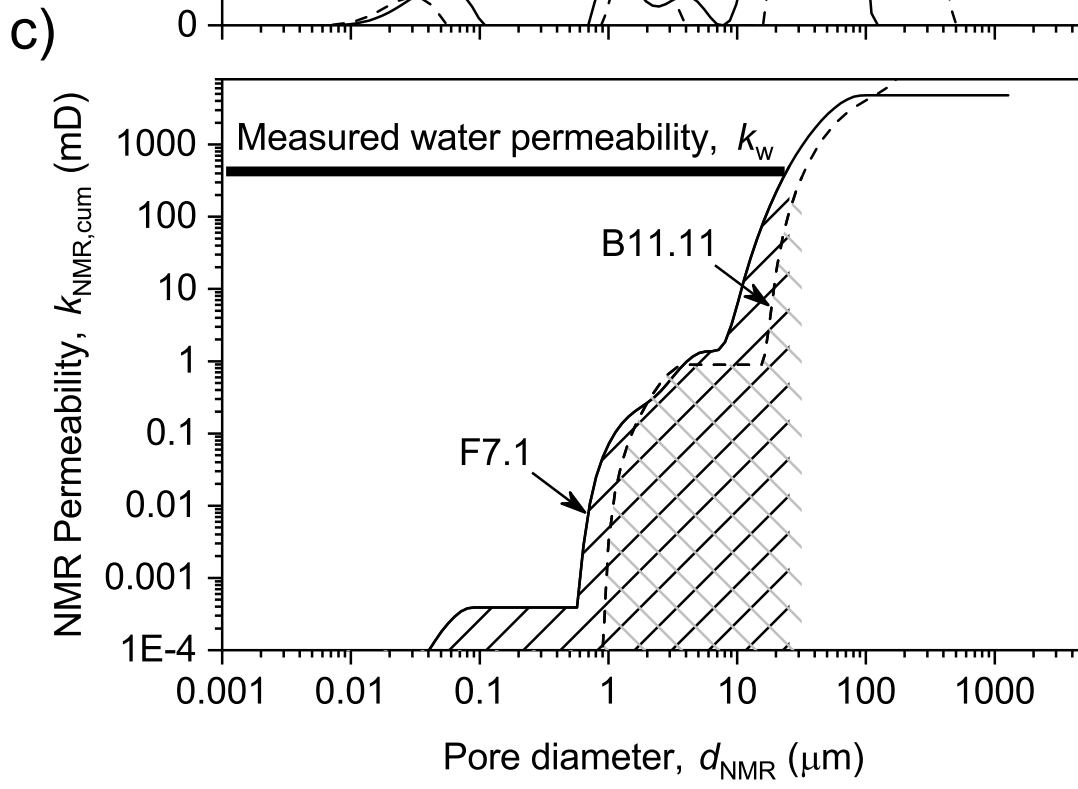
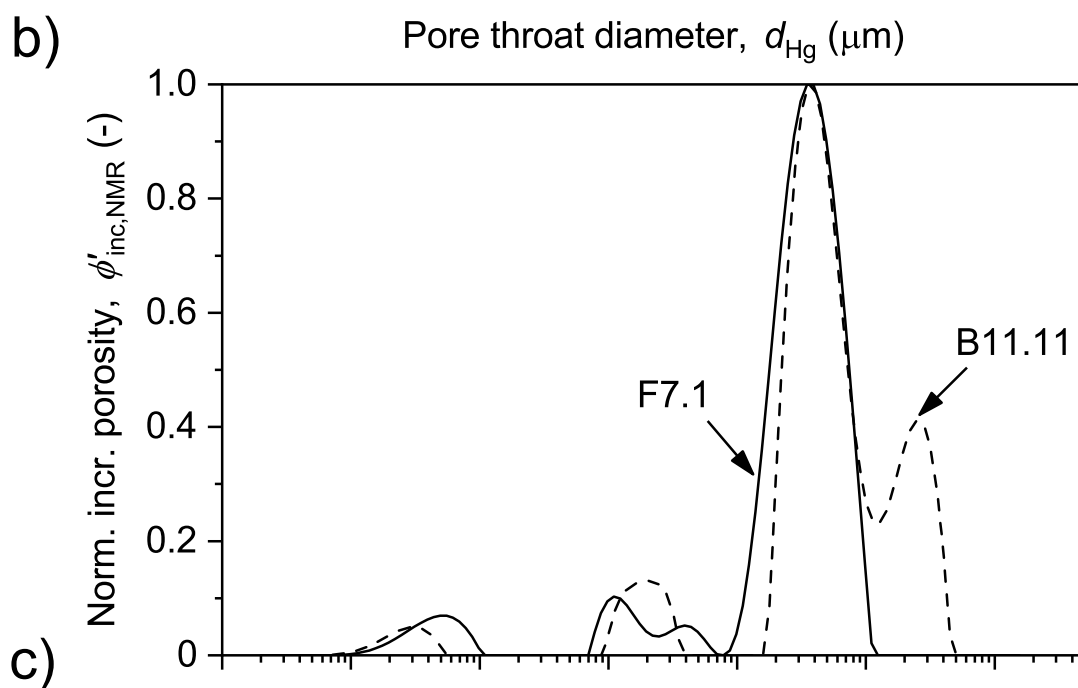
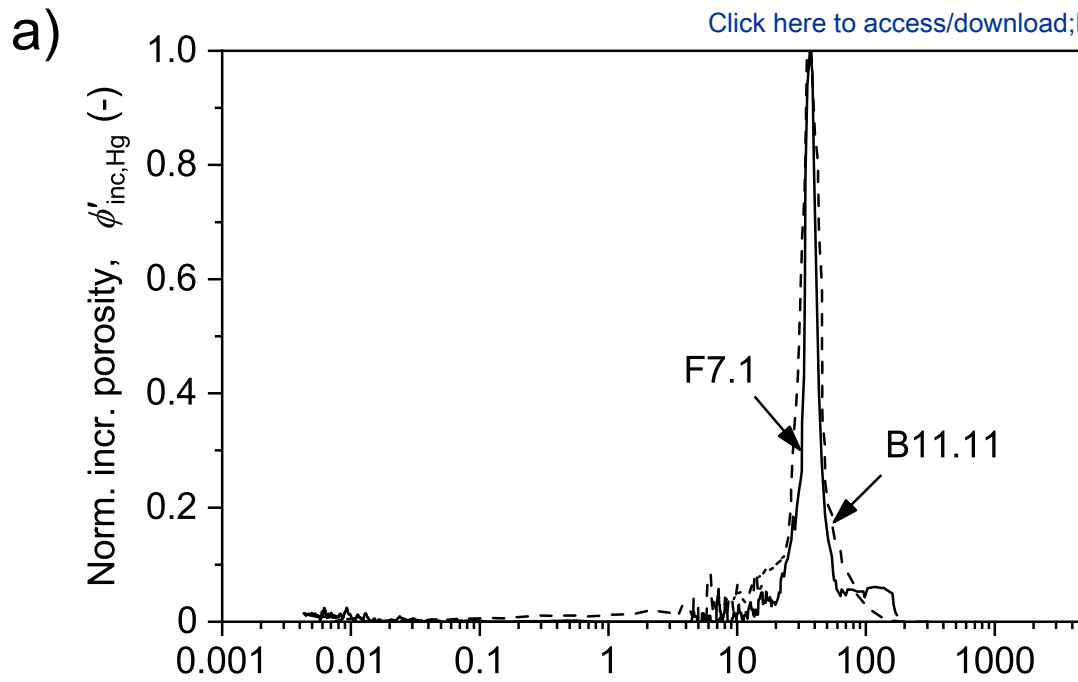


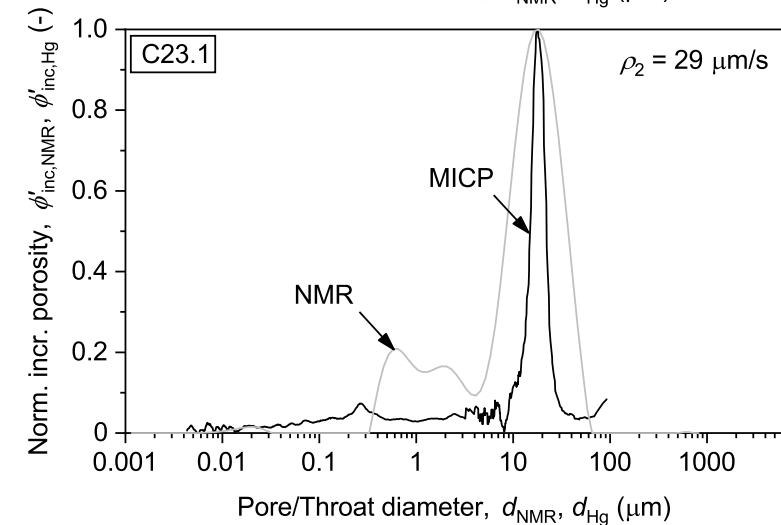
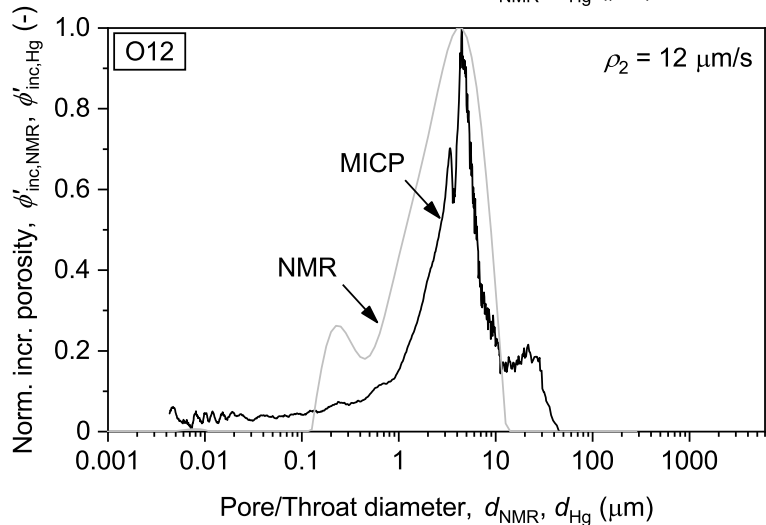
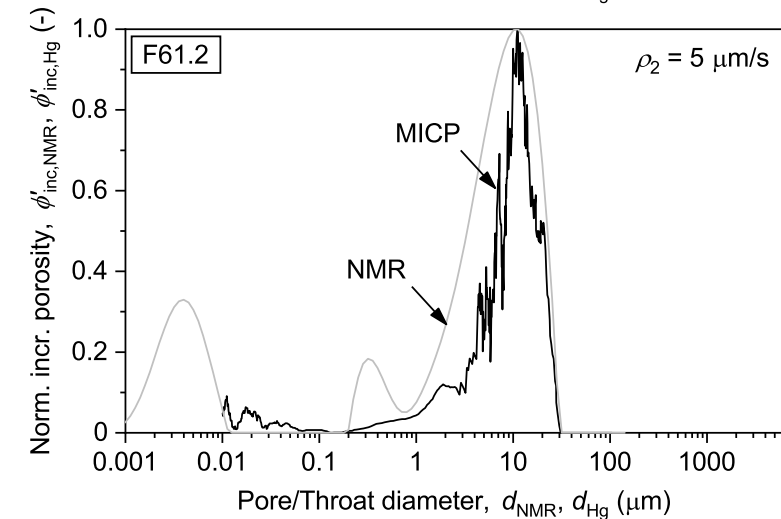
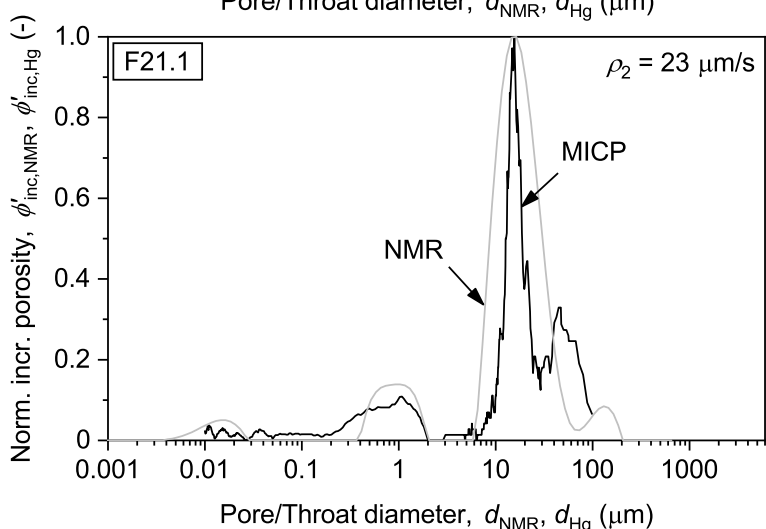
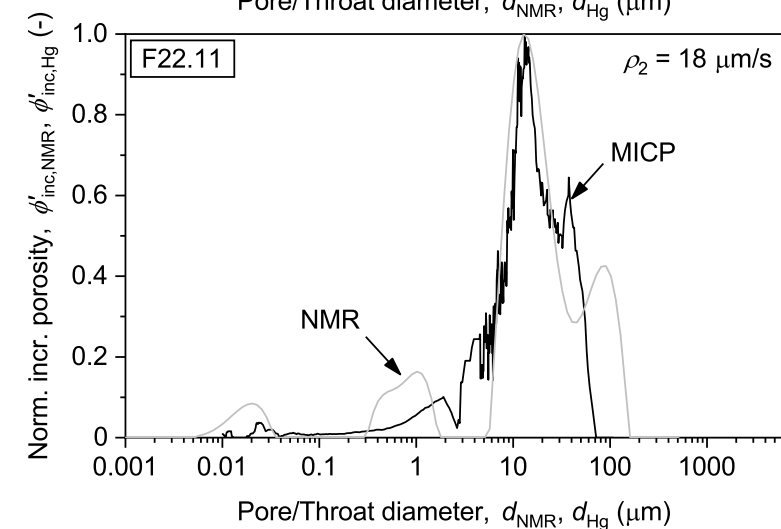
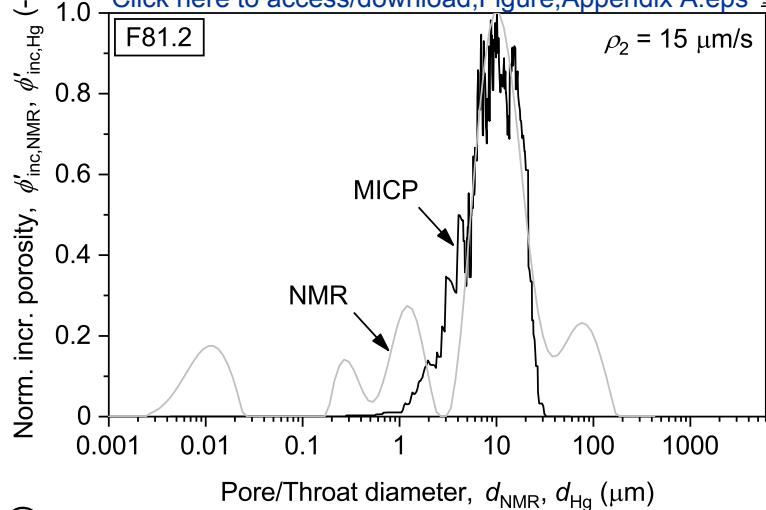
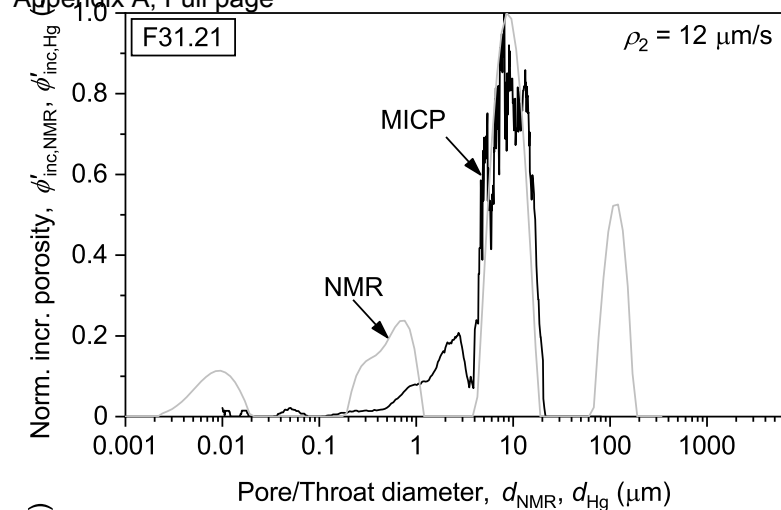


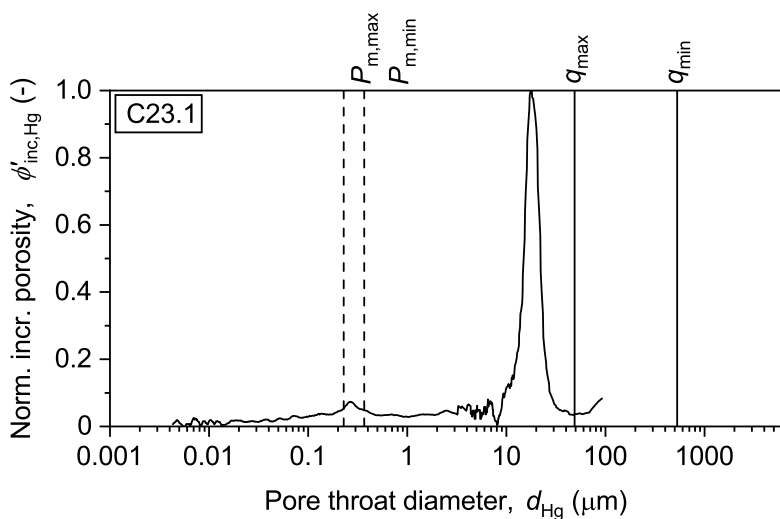
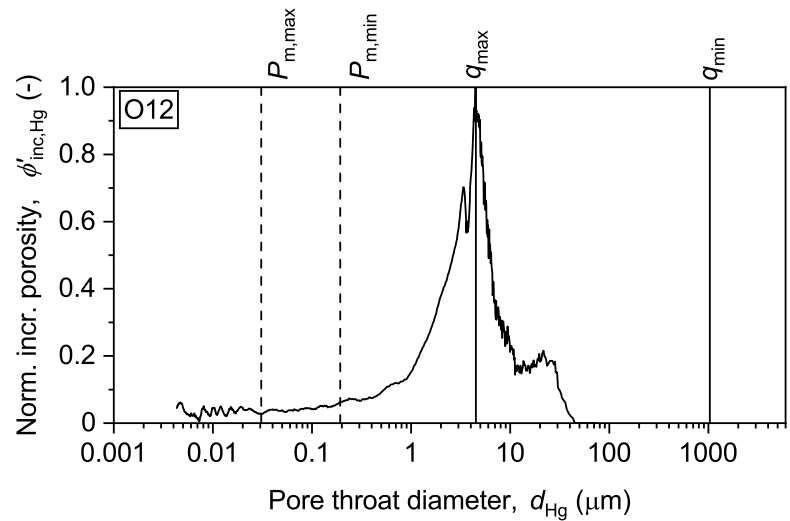
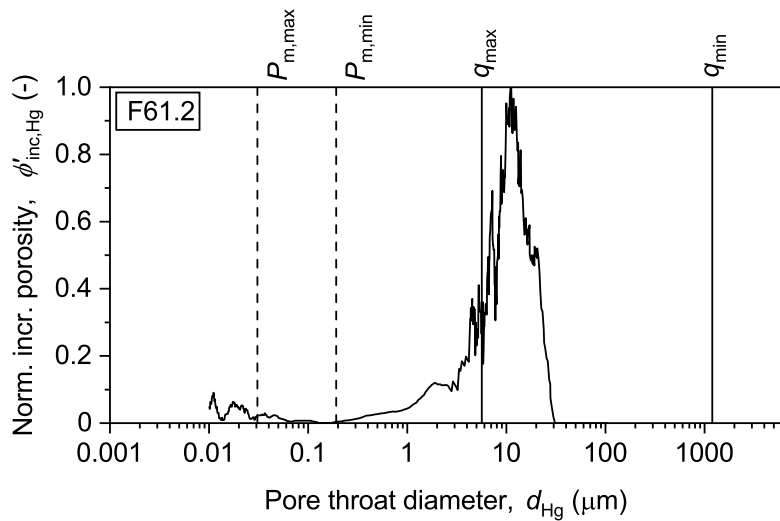
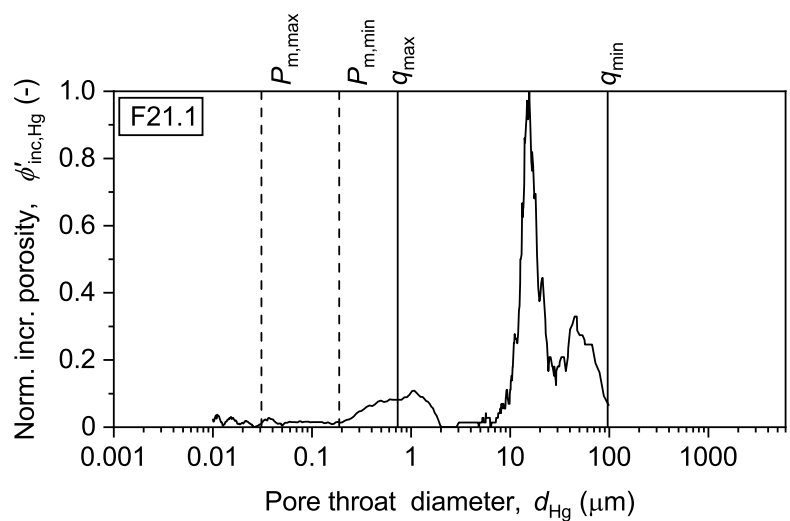
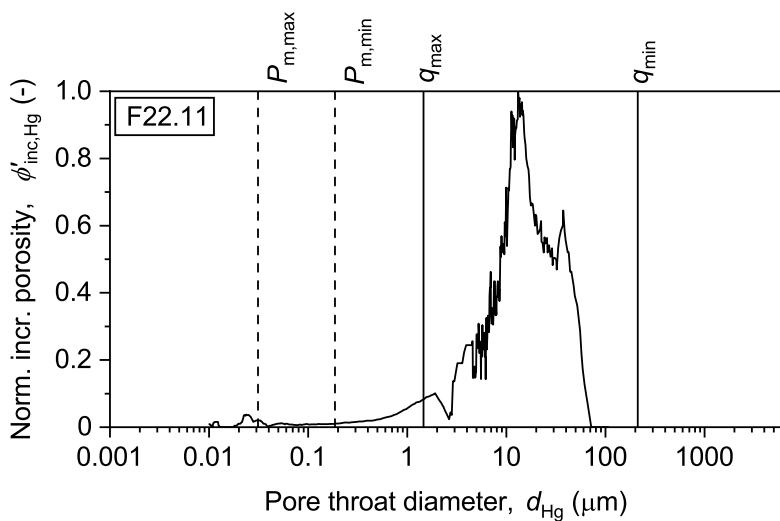
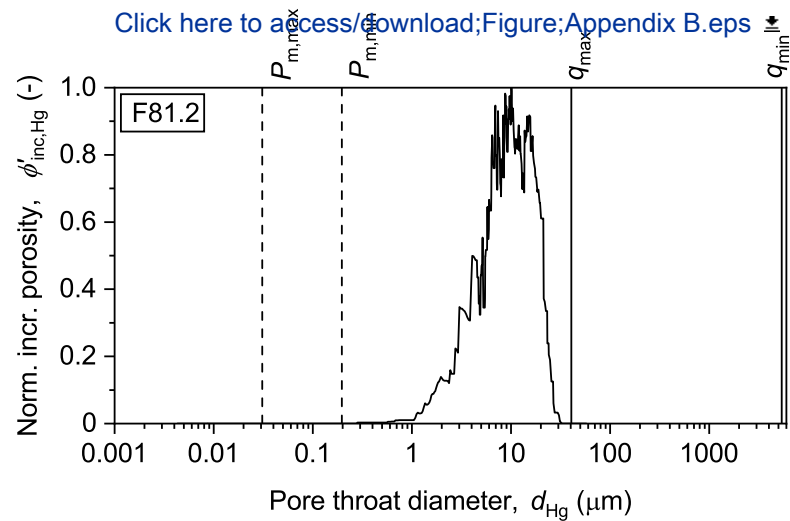
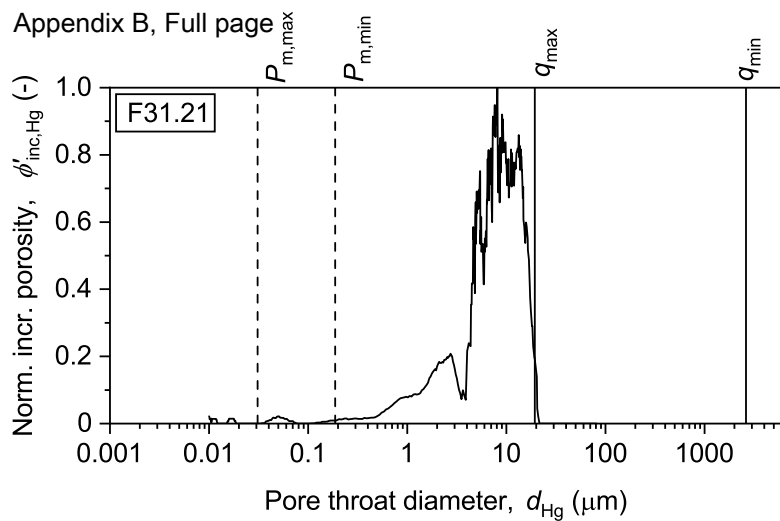


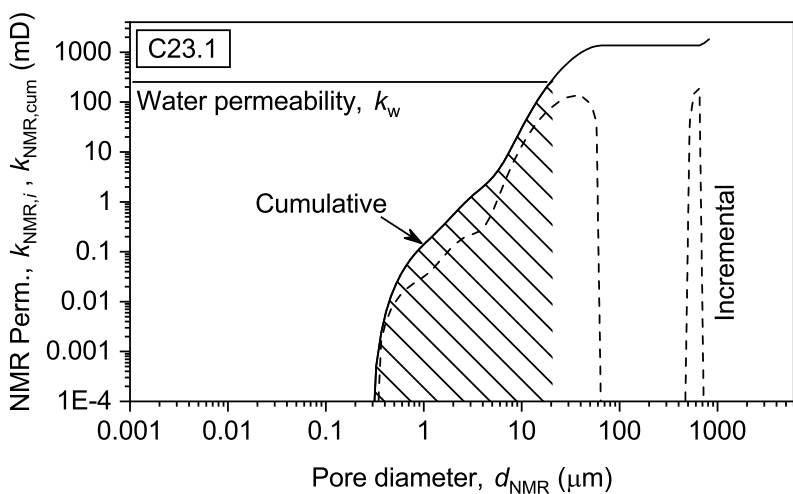
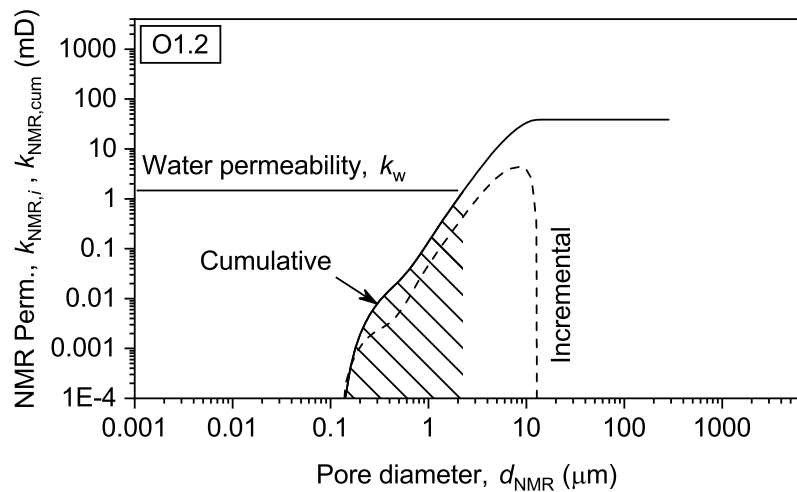
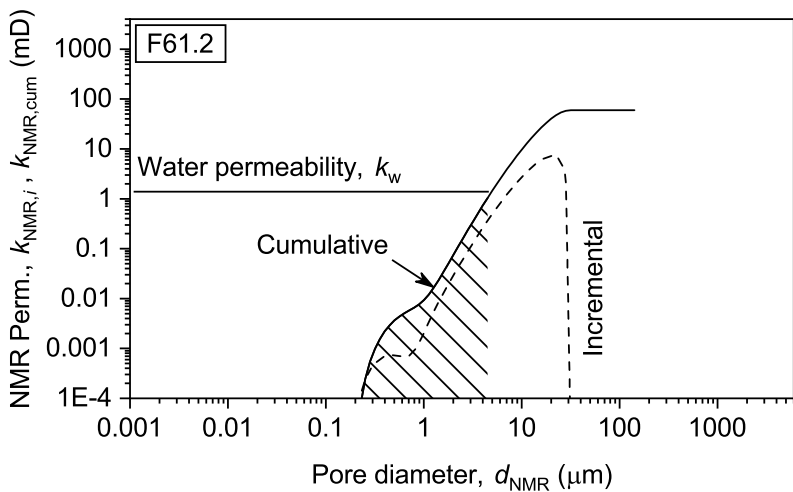
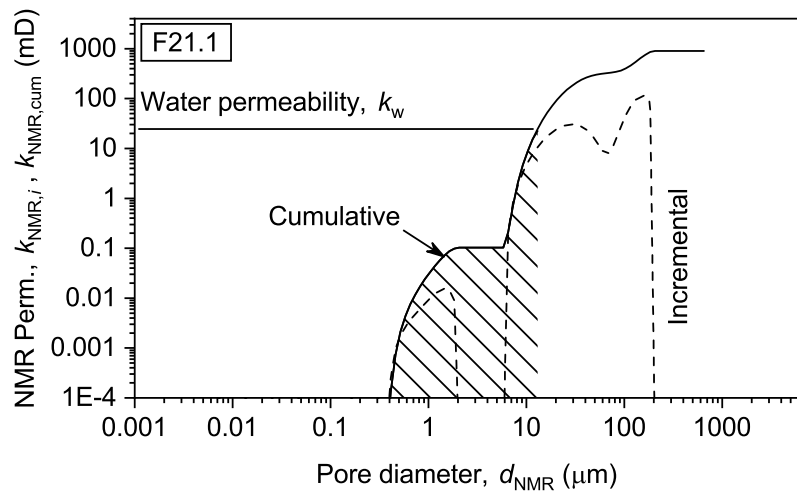
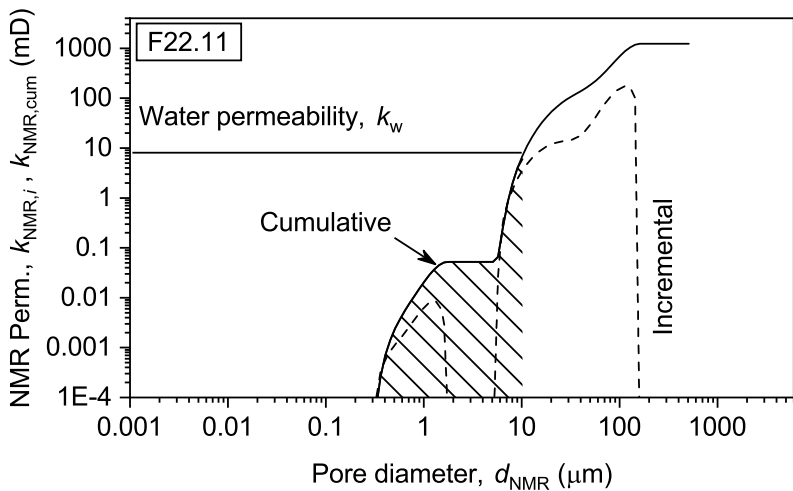
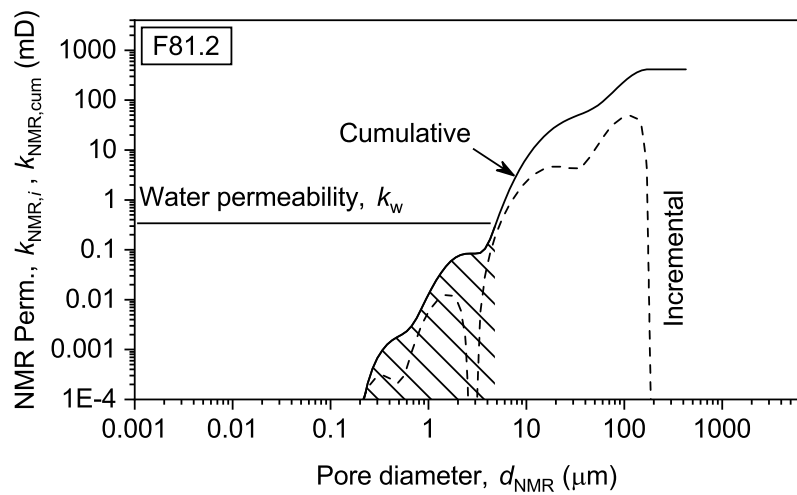
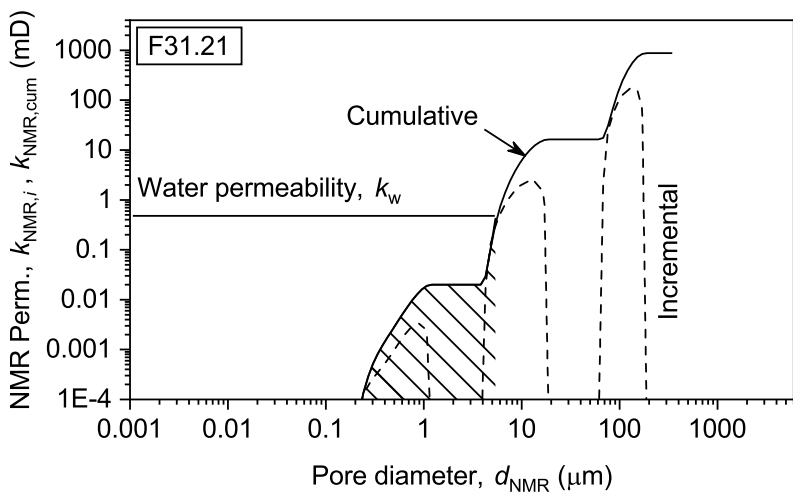












1 TABLE 1

Formation	Quartz	Feldspar	Clay ^a	BET, specific surface
	mass % of total solid			m ² /g
Fontainebleau	100			0.03
Castlegate	95.4	1.1	3.5	1.72
Bentheimer	95.3	4.7	(2.7) ^b	0.31
Obernkirchen	96.0		4.0	1.06

2 ^aClay includes illite and kaolinite. ^b(Peksa et al., 2017)

1 TABLE 2

Sample	Dry density	Grain density	N ₂ Porosity	KCL, Saturated density	Saturation degree
	ρ_{dry}	ρ_{min}	ϕ_{N}	ρ_{sat}	S_w
	g/cm ³	g/cm ³	-	g/cm ³	-
F31.21 ^a	2.52	2.65	0.05	2.56	0.84
F81.2 ^a	2.53	2.65	0.05	2.57	0.88
F61.1 ^a	2.47	2.66	0.07	2.53	0.93
F61.2 ^a	2.47	2.66	0.06	2.53	0.93
F21.1 ^a	2.43	2.66	0.09	2.49	0.93
F22.11 ^a	2.44	2.66	0.08	2.52	0.96
F7.1 ^a	2.04	2.65	0.23	2.27	0.98
B11.11 ^b	1.98	2.67	0.26	2.24	0.98
O1.1 ^c	2.15	2.68	0.20	2.34	0.95
O1.2 ^c	2.17	2.70	0.19	2.35	0.94
C1.3 ^d	1.86	2.71	0.31	2.16	0.94
C11.2 ^d	1.89	2.71	0.30	2.17	0.91
C23.1 ^d	1.88	2.70	0.30	2.16	0.93

2 ^aFontainebleau, ^bBentheimer, ^cObernkirchen, ^dCastlegate.

1 TABLE 3

Sample	$\phi_N S_w$	NMR	Electrical	Porosity	Shielding	Permeability		
		Porosity	resistivity ^e	exponent ^f	factor ^g	Kozeny ^h	Klink. ⁱ	water
		ϕ_{NMR}	R_0	m	c_M	k_z	k_K	k_w
	-	-	Ohm-m	-	-	mD (E ⁻¹⁵ m ²)	mD (E ⁻¹⁵ m ²)	mD (E ⁻¹⁵ m ²)
F31.21 ^a	0.04	0.03	95.8	2.12	0.18	4.06 (4.01)	0.14 (0.138)	0.40 (0.395)
F81.2 ^a	0.04	0.04	86.2	2.08	0.18	3.34 (3.30)	0.11 (0.108)	0.33 (0.325)
F61.1 ^a	0.06	0.05	25.9	1.96	0.19	10.8 (10.6)	4.07 (4.02)	3.10 (3.05)
F61.2 ^a	0.06	0.05	31.0	1.86	0.18	8.02 (7.92)	1.09 (1.07)	1.10 (1.09)
F21.1 ^a	0.08	0.07	15.7	1.89	0.19	25.7 (25.4)	- ^j	22.1 (21.8)
F22.11 ^a	0.08	0.05	24.0	1.97	0.19	18.9 (17.7)	- ^j	8.50 (8.38)
F7.1 ^a	0.23	0.21	3.2	2.02	0.21	739 (729)	- ^k	430 (424)
B11.11 ^b	0.25	0.21	2.8	2.09	0.21	9.92 (9.97)	- ^k	320 (316)
O1.1 ^c	0.19	0.18	6.1	2.23	0.21	0.29 (0.286)	4.62 (4.56)	3.50 (3.45)
O1.2 ^c	0.18	0.17	7.0	2.25	0.20	0.28 (0.276)	4.06 (4.01)	1.50 (1.48)
C1.3 ^d	0.29	0.23	2.4	2.23	0.22	0.66 (0.651)	177 (174)	351 (3.46)
C11.2 ^d	0.28	0.24	2.4	2.27	0.22	0.60 (0.651)	201 (198)	262 (259)
C23.1 ^d	0.28	0.24	2.7	2.30	0.22	0.55 (0.542)	150 (148)	280 (276)

2 ^aFontainebleau, ^bBentheimer, ^cObernkirchen, ^dCastlegate, ^eMeasured at unconfined stress conditions
3 and an axial stress of 3 MPa. Measured with a saturating 0.5M KCl solution with a density of 1.02
4 g/cm³ and an electrical resistivity, R_w , of 0.167 Ohm-m, ^fderived from equation 13 assuming $\phi = \phi_N$,
5 ^gderived using the simplified expression from equation 5 and $\phi = \phi_N$, ^hderived from equation 3 using
6 $\phi = \phi_N$, S_P determined from BET measurements (Table 1) and approximating c with c_M , ⁱKlinkenberg
7 permeability derived from Figure 7, ^jundefined Klinkenberg permeability because of insufficient data,
8 ^kundefined Klinkenberg permeability because of line crossing at negative part of y-axis. mD =
9 millidarcy, 1mD = 9.869E⁻¹⁶ m².